

Effect of Nanovoid on Fracture Process of Two-Phase $\alpha(\text{TiAl})+\gamma(\text{Ti}_3\text{Al})$ Alloy

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Abstract

The fracture processes of nanocrystalline metallic material is affected by dislocation, nanovoid and other defects. Existing studies of defect evolution in titanium-aluminium alloy cover the case that voids located in single crystals, inside grain in polycrystals and at the grain boundaries. Molecular dynamics simulation was performed to study the evolution of a spherical nanovoid in $\alpha+\gamma$ two-phase titanium-aluminium alloy under uniaxial tension.

Keywords: $\alpha+\gamma$ two phase TiAl alloy; void; molecular dynamics

1. Introduction

TiAl alloy has been used as structural material in aviation industry because its inherent advantages such as low density and self-diffusion rates, high elastic module and high strength [1]. However, single phase γ – TiAl generally brittle
5 at room temperatures and this limits their use in many other fields. Two-phase titanium aluminum alloys with proper phase distribution and grain size exhibit better mechanical performance compared with monolithic constituents $\gamma(\text{TiAl})$ and $\gamma(\text{Ti}_3\text{Al})$ alloy [2].

Two phase titanium aluminide alloys exhibit a much better mechanical performance than their monolithic constituents $\gamma(\text{TiAl})$ and $\alpha_2(\text{Ti}_3\text{Al})$, provided
10 that the phase distribution and grain size are suitably controlled. Two-phase titanium aluminide alloys with proper phase distribution and grain size exhibit better mechanical performance compared with monolithic constituents gamma and alpha. Brittle fracture in TiAl alloy strongly affects the safety of fracture

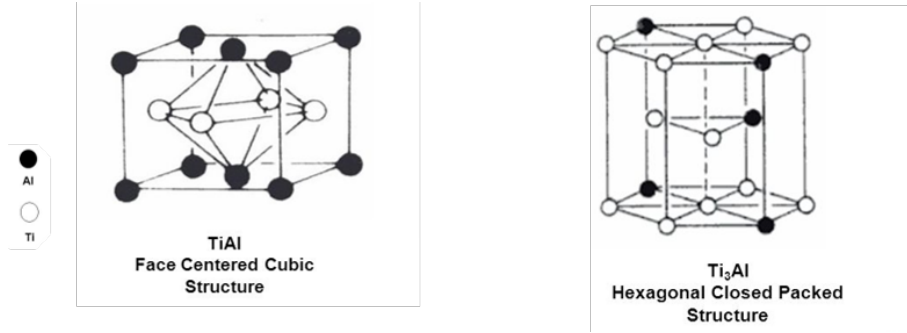


Fig. 1. unit cell

15 of structure like turbo of aircraft engine and combustion generator. [R]. Defects such as dislocation, void and segregation plays an significant role in the process of fracture [R]. In order to understanding the mechanism of brittle fracture, multiscale methods from micro to marco scale have been applied to investigate the behavior of fracture. It's necessary to carefully examine the revolution of de-
 20 fects and its influence on the fracture process at atomic scale. A previous study on void growth in gamma-TiAl single crystal has reveals that void with high volume fraction detracts incipient yield strength [R]. Molecular dynamics (MD) method has been use to investigate the evolution of void in materials in nano-scale [R]. The fracture mechanisms in the duplex micro-structure are plasticity-
 25 induced grain boundary de-cohesion and cleavage, while those in the lamellar microstructure are interface delamination and crack- ing across the lamellae [R].

2. Molecular Dynamics Simulation

2.1. Model Creation of Crystalline

30 γ TiAl has a fcc-centered tetragonal with an $L1_0$ structure [], and α TiAl is hcp structure, the structure of the two initial cell are shown in Fig. [], and the constructing parameters are givin by Table. []. The simulation cells of two phase poly crystal with an initially spherical void at different position are shown in figure []. Periodic boundary conditions are applied along all three directions, that makes poly crystal with periodic nanovoid structures. Each cuboidal model,

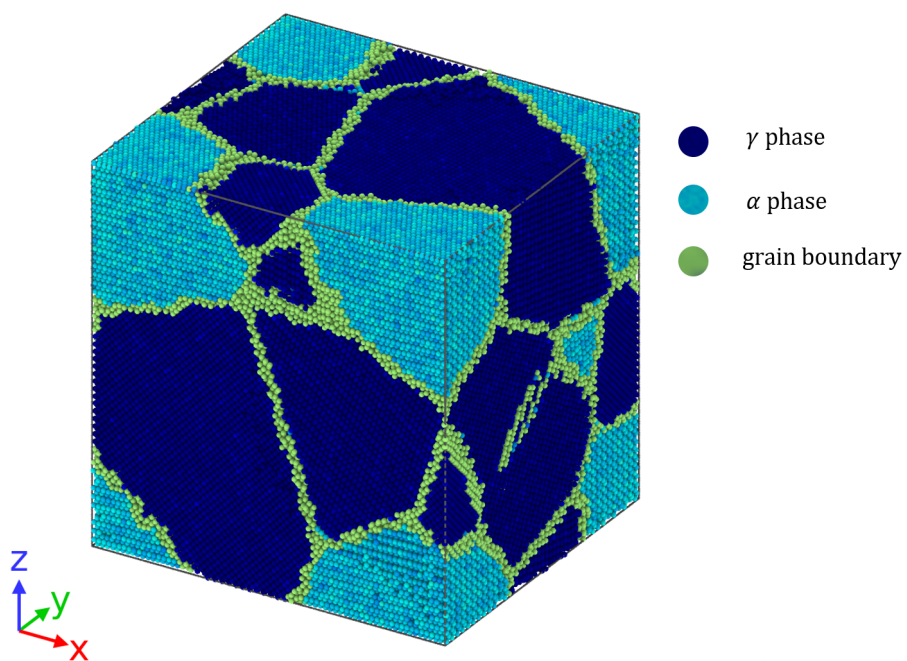


Fig. 2. Simulation box

Table 1

Parameters of nanocrystalline

Phase	Space group	Designation	Parameters
α_2	P6 ₃ /mmc	0 ₁₉	$a = 0.5765$ $c = 0.46833$
γ	tP4	L1 ₀	$a = 0.3997$ $c = 0.4062$

35 containing about 4.6 million atoms, The initial dimension of simulation cell is $L_x = nm$, $L_y = nm$, $L_z = nm$. The grain orientation and size were randomly created with Voronoi method with code ATOMSK [1], and resulting in the arbitrary shape and orientation of the grains. Only one spherical void defect was placed intragranularly or intergranularly within each simulation model void
40 within each simulation model. The intragranular spherical void was located in grain interior of the largest grain of the simulation model, as shown in Fig. [2]. The intergranular spherical void was at the center of the simulation cell, as shown in Fig. [3].

2.2. Atomic Potential

45 The interaction of particle in the material is determined by interatomic potential. Many reported examples of crack propagation in metal materials were performed with embedded atom method due to its better accuracy in metal lattice compared with F-S and L/J [4]. The embedded atom method (EAM) potential developed by Zope and Mishin [5] was used in the study. The simulation is submitted by MD simulations with the Large-scale Atomic/Molecular
50 Massively Parallel Simulator (LAMMPS) open-source code [6]. We performed constant-pressure and constant-temperature(NPT) molecular dynamics simulation.

$$E = \sum \quad (1)$$

2.3. Analysis method

55 *CSP.* In view of the fact that dislocations cannot be easily spotted in a three-dimensional crystal, a centrosymmetry parameter [] is used to identify the nucleation and propagation of dislocations. The said centrosymmetry parameter is defined as follow:

$$P = \sum_{i=1}^6 |\vec{R}_i + \vec{R}_{i+6}|^2 \quad (2)$$

where \vec{R}_i and \vec{R}_{i+6} are the vectors corresponding to the six pairs of opposite
60 nearest neighbors in the fcc lattice. The centrosymmetry parameter(CSP) is zero for atoms in a perfect lattice. In other words, if the lattice is distorted the value of P will not be zero. Instead, the parameter will have a value within the range corresponding to a particular defect. By removing all the perfect and surface atoms within the bulk, the existence of dislocation atoms become visible.

Virial Stress. The atomic stress is calculated using the virial definition :

$$\sigma_t(i) = -$$

$$\sigma_t(i) =$$

65 3. Results and Discussion

3.1. Decrease of Strength

The existence of void detract the strength, and the void inside α phase grain have most significant impact on the strength, however the void on the grain boundary have little impact on incipient strength of the material. Detailed
70 observation of specimen with void inside the grain is shown in Figure []. 1. In many cases the orientation of slip slip is changed because the crystallographically available slip and directions are not continuous across the interface. This may significantly reduce the Schmid factor and thus impede slip transfer. At the γ/γ interfaces the orientation of the slip plan could change through a relevantly

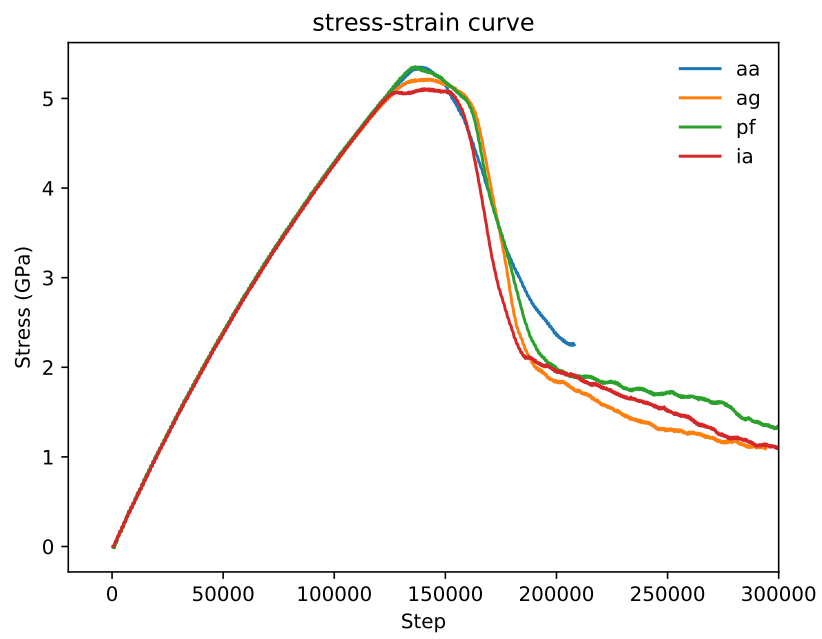


Fig. 3. Fracture process of specimen with void in α phase

75 large angle of about 90 degree. Reorientation of slip is always required at the α_2/γ interface; the smallest angle between the corresponding slip planes 111_γ and $10 - 10_{\alpha_2}$ is about 19 degree [].

The core of a dislocation intersecting an interface often needs to be transformed. For example, an ordinary $1/2[110]$ dislocation gliding in one γ grain has
80 to be converted in to a $[101]$ super dislocation with the double Burgers vector gliding in an adjacent γ grain. At the α/γ interface the dislocations existing in the $D0_{19}$ structure have to be transformed into dislocations consistent with the $L1_0$ structure. These core transformations are associated with a change of the dislocation line energy because the lengths of the Burgers vectors and the shear
85 module are different.

Dislocations crossing semi-coherent boundaries have to intersect the misfit dislocations, a process that involves elastic interaction, jog formation and the incorporation of gliding dislocations into the mismatch structure of the interface. When the slip is forced to cross α_2 lamellae, pyramidal slip of the α_2 phase
90 is required, which needs an extremely high shear stress.

3.2. Fracture Process

4. Conclusion

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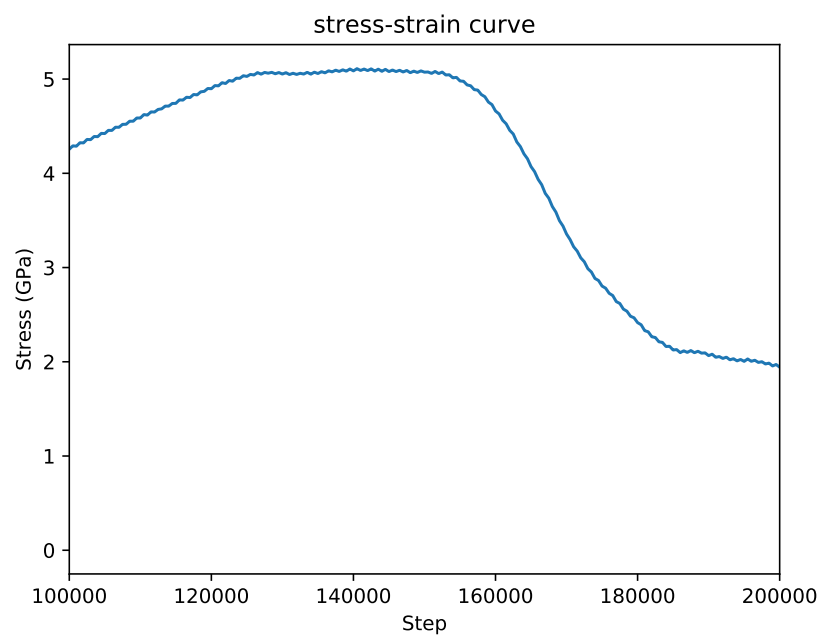


Fig. 4. stress-strain curve

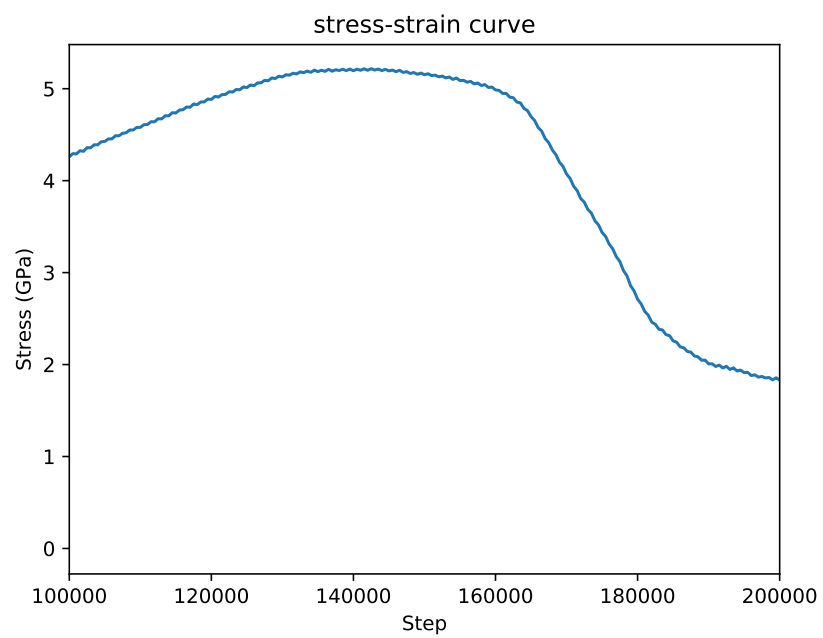


Fig. 5. stress-strain curve

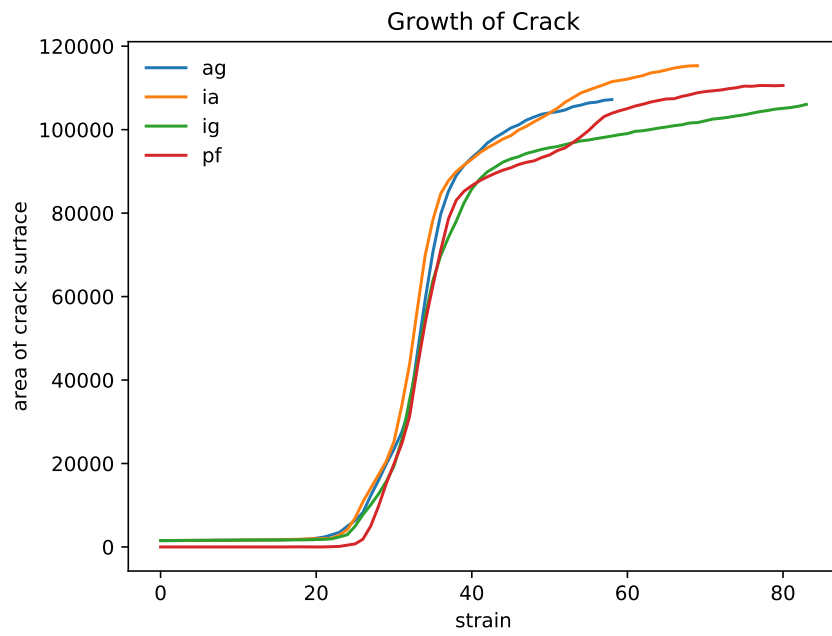


Fig. 6. stress-strain curve