## ME 812 Conductive Heat Transfer

## Wilke Mixture Rule

The Chapman-Enskog model can be used for predicting of thermal conductivity of gas mixtures. Using Wilke's approach we have

$$k_{mix} = \sum_{i=1}^{n} \frac{x_i k_i}{\sum_{i=1}^{n} x_j \Phi_{ij}}$$

where

 $x_i$  is the mole fraction of the ith component

$$\Phi_{ij} = \frac{\left[1 + \left(\frac{\mu_i}{\mu_j}\right)^{0.5} \left(\frac{MW_j}{MW_i}\right)^{0.25}\right]^2}{\sqrt{8}[1 + (MW_i/MW_j)]^{0.5}}$$

Note that this equation reduces to

$$\Phi_{ii} = 1$$

The  $\boldsymbol{\mu}$  in the above equation is the dynamic viscosity for the gas, which can be calculated from the Chapman-Enskog model to be

$$\mu = (2.6709 \times 10^{-6}) \frac{\sqrt{\text{MW} \cdot \text{T}}}{\sigma^2 \Omega_{\mu}}$$

where  $\Omega_{\mu}$  is equal to  $\Omega_{k}$ .

## Example: Thermal Conductivity of Air as a Gas Mixture

We wish to determine the thermal conductivity of air at 500 K treating it as a gas mixture of  $N_2$  (72%) and  $O_2$  (28%).

## **Solution:**

We will use the Chapman-Enskog model with the Wilke mixture rule. We set up the following table:

Gas	MW	σ	ε/κ	Τκ/ε	$\Omega_{ m k}$	μ	k
$N_2$	28.013						
$O_2$	31.999						

Going to Table 7.1 to obtain  $\sigma$  and  $\varepsilon/\kappa$ .

Gas	MW	σ	ε/κ	Τκ/ε	$\Omega_{ m k}$	μ	k
$N_2$	28.013	3.798	71				
$O_2$	31.999	3.467	107				

Calculating  $T\kappa/\epsilon$  and reading  $\Omega_k$  from Table 7.2.

Gas	MW	σ	ε/κ	Τκ/ε	$\Omega_{ m k}$	μ	k
$N_2$	28.013	3.798	71	7.042	0.8719		
$O_2$	31.999	3.467	107	4.673	0.9393		

For the dynamic viscosity and thermal conductivity we will use

$$\mu = (2.6709 \times 10^{-6}) \frac{\sqrt{MW \cdot T}}{\sigma^2 \Omega_{\mu}}$$

$$k = k_{monatomic} + 1.32(c_P - 5/2 \frac{R_u}{MW}) (2.6709 \times 10^{-6}) \frac{\sqrt{MW \cdot T}}{\sigma^2 \Omega_k}$$

$$k_{monatomic} = (8.3127 \times 10^{-2}) \frac{\sqrt{T/MW}}{\sigma^2 \Omega_k}$$

The specific heat,  $c_P$ , will be calculated form the equations on Table A.11SI. We then obtain

Gas	MW	σ	ε/κ	Τκ/ε	$\Omega_{ m k}$	μ	k
$N_2$	28.013	3.798					
$O_2$	31.999	3.467	107	4.673	0.9393	2.9923 x 10 <sup>-05</sup>	0.0416

We can now apply our mixture rule,

$$k_{mix} = \sum_{i=1}^{n} \frac{x_i k_i}{\sum_{i=1}^{n} x_j \Phi_{ij}}$$

where

$$\Phi_{ij} = \frac{\left[1 + \left(\frac{\mu_i}{\mu_j}\right)^{0.5} \left(\frac{MW_j}{MW_i}\right)^{0.25}\right]^2}{\sqrt{8}[1 + (MW_i/MW_j)]^{0.5}}$$

Then

$$\Phi_{\text{N2N2}} = 1$$

$$\Phi_{N2N2}=1$$

$$\Phi_{N2O2} = 0.9791$$

$$\Phi_{O2N2}=1.0205$$

Finally, our mixture thermal conductivity is calculated:

$$k_{mix} = 0.0392 \text{ W/(m·K)}$$

From Icropera & DeWitt, we find that at 500 K they give an air thermal conductivity of 0.0407 W/(m·K) or a difference of 3.6%. Hence, this is a pretty good approximation.