

$$\Delta G(N) = -Nk_B T \ln P^* + \gamma 4\pi R_0^2 N^{2/3}$$

$$P^* = P/P_e$$

**P:** Gas pressure

**P<sub>e</sub>:** Vapor pressure (increases if T increases)

**To promote P\* (supersaturation):**

**1. decrease P<sub>e</sub> (i.e., decrease T)**

Adiabatic expansion with cooling:

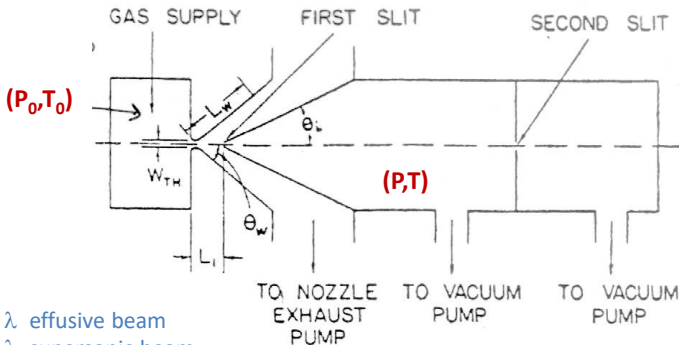
- Supersonic molecular beams

**2. increase P**

Production of condensable species:

- Chemical or photo-chemical decomposition in the gas phase
- sputtering
- thermal evaporation
- laser ablation
- ion implantation

## Free Jet Expansion

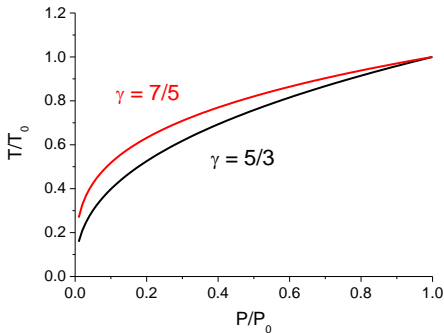


The vapor – mixed with an inert gas (*gas carrier*) – expands adiabatically through a nozzle with diameter  $d$  from the initial chamber at  $(P_0, T_0)$  in a vacuum chamber with pressure  $P < P_0$ :

## Free Jet Expansion

$$PV^\gamma = \text{const} \quad \gamma \equiv \frac{C_P}{C_V} > 1$$

$$\frac{P^{\frac{\gamma-1}{\gamma}}}{T} = \text{const} \quad T = T_0 \left( \frac{P}{P_0} \right)^{\frac{\gamma-1}{\gamma}}$$



During expansion the vapor cools down becoming supersaturated with the formation of *clusters*

The cluster density increases by:

- increasing  $P_0$  and  $d$
- decreasing  $T_0$

# Nucleation and growth

## Kinetics vs. Energetics

time

## 1. *Nuclei formation* (supersaturated solution)

- $\langle R \rangle =$  critical radius  $R^*$
- E.g., ion implantation: heterogeneous nucleation

$$R^* = \frac{2\gamma}{\Delta g_V}$$

## 2. *Diffusion limited aggregation (DLA)*

- non competitive growth: fixed n. of clusters
- the supersaturation decreases

$$R^2(t) = R_0^2 + K_1 Dt$$

(supersaturation)

## 3. *Coarsening (Ostwald ripening, OR)*

- competitive growth: the n. of cluster decreases
- the growth is controlled by the Gibbs-Thomson equation

$$R^3(t) = R_0^3 + K_2 Dt$$

(Gibbs-Thomson)

## 2. Diffusion limited aggregation (DLA)

Balance of the atomic flux at the surface of a cluster with radius  $R$  (boundary cluster-matrix):

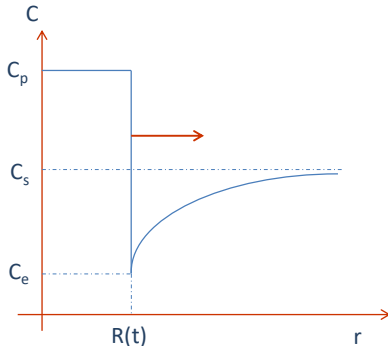
$$\Phi_{\Sigma}(\bar{J}) = Flux|_{r=R} = \left. \frac{dN}{dt} \right|_{out} - \left. \frac{dN}{dt} \right|_{in}$$

$$\left. \frac{dN}{dt} \right|_{in} = C_p \frac{d}{dt} \left( \frac{4\pi}{3} R^3 \right) = 4\pi R^2 C_p \frac{dR}{dt}$$

$$\left. \frac{dN}{dt} \right|_{out} = C_e \frac{d}{dt} \left( \frac{4\pi}{3} R^3 \right) = 4\pi R^2 C_e \frac{dR}{dt}$$

$$\bar{J} = -D \bar{\nabla} C \quad 1^a \text{ Fick's Law (Isotropic diffusion)}$$

$$\Phi_{\Sigma}(\bar{J}) = \int_{\Sigma} \bar{J} \cdot \hat{n} d\Sigma = -4\pi R^2 D \left. \frac{\partial C(r, t)}{\partial r} \right|_{r=R}$$



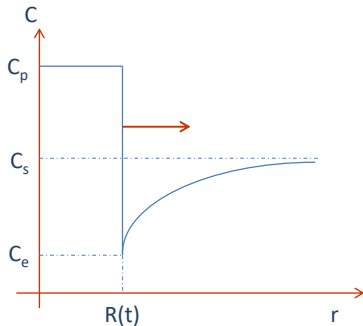
Net incoming flux (growth)

Substituting:

$$\left. \frac{dN}{dt} \right|_{out} = Flux|_{r=R} + \left. \frac{dN}{dt} \right|_{in}$$

$$C_e \frac{dR}{dt} = -D \left. \frac{\partial C(r,t)}{\partial r} \right|_{r=R} + C_p \frac{dR}{dt}$$

$$(C_p - C_e) \frac{dR}{dt} = D \left. \frac{\partial C(r,t)}{\partial r} \right|_{r=R}$$



$C = C(r, t)$  concentration field around the cluster

$C_s = C(r, 0)$  concentration in the matrix at  $t = 0$

$C_p$  equilibrium concentration in the cluster (density)

$C_e$  equilibrium concentration in the matrix (solubility limit)

## Linearizing the gradient:

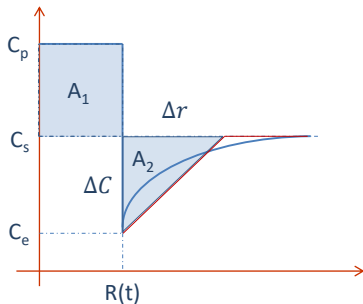
$$\left. \frac{\partial C(r, t)}{\partial r} \right|_{r=R} \approx \frac{\Delta C}{\Delta r} = \frac{(C_s - C_e)^2}{2(C_p - C_s)} \frac{1}{R}$$

$$(C_p - C_e) \frac{dR}{dt} \approx D \frac{(C_s - C_e)^2}{2(C_p - C_s) R}$$

$$R^2(t) = R_0^2 + \frac{(C_s - C_e)^2}{(C_p - C_e)(C_p - C_s)} Dt$$

$$R^2(t) = R_0^2 + K_1 Dt$$

$$K_1 = \frac{(C_s - C_e)^2}{(C_p - C_e)(C_p - C_s)} \approx \frac{C_s^2}{C_p^2}$$



$$A_1 = A_2$$

supersaturation



### 3. Ostwald Ripening (OR)

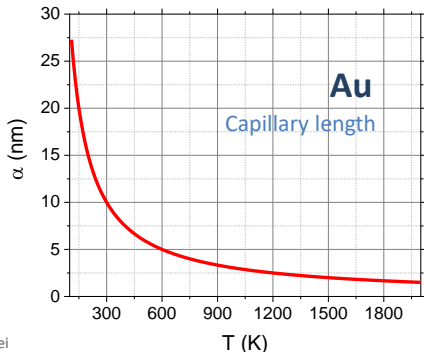
When the supersaturation is decreased: Gibbs-Thomson

$$k_B T \ln \frac{C_e(R)}{C_e(\infty)} = \frac{2\gamma}{\rho R} \quad C_p \equiv \rho$$

$$C_e(R) = C_e(\infty) \exp\left(\frac{2\gamma}{C_p k_B T R}\right) \sim C_e(\infty) \left(1 + \frac{\alpha}{R}\right)$$

$$\alpha \equiv \frac{2\gamma}{C_p k_B T} \quad \text{Capillary length}$$

$C_e(R)$  can be interpreted as the equilibrium solute concentration in the matrix at the cluster surface (it increases by decreasing  $R$ )



The atomic flux at the cluster surface is:

$$\frac{dN_{clu}}{dt} = C_p \frac{d}{dt} \left( \frac{4\pi}{3} R^3 \right) = 4\pi R^2 C_p \frac{dR}{dt}$$

$$\frac{dN_{clu}}{dt} = \Phi_{\Sigma}(\bar{J}) = -4\pi R^2 \left( D \frac{\partial C}{\partial r} \Big|_{r=R} \right) \approx -4\pi R^2 D \frac{dC_e(R)}{dR} = 4\pi R^2 D \frac{C_e(\infty)\alpha}{R^2}$$

$$\frac{dC_e(R)}{dR} = -C_e(\infty) \frac{\alpha}{R^2}$$

$$\frac{dN_{clu}}{dt} = 4\pi R^2 D \frac{C_e(\infty)\alpha}{R^2}$$

$$\frac{dR}{dt} = D \frac{C_e(\infty)\alpha}{C_p} \frac{1}{R^2}$$

$$R^3(t) = R_0^3 + 3 \frac{C_e(\infty)\alpha}{C_p} Dt = R_0^3 + 6 \frac{C_e(\infty)\gamma}{C_p^2 k_B T} Dt$$

$$R^3(t) = R_0^3 + K_2 Dt$$

