

Radiometric Dating

Using the table and graph provided, please answer the following questions.

Number of half-lives elapsed	Fraction remaining	Percentage remaining	
0	1/1	100	
1	1/2	50	
2	1/4	25	
3	1/8	12	.5
4	1/16	6	.25
5	1/32	3	.125
6	1/64	1	.563
7	1/128	0	.781
...	
n	$1 / 2^n$	$100(1 / 2^n)$	

1. 1. Approximately what percentage of parent isotopes remains after 2 half-lives have passed?

25%

2. 2. If a rock initially contained 10 milligrams of a radioactive parent when it first crystallized, how much remains after 4 half-lives?

.625 milligrams

3. 3. What % parent remains after 10 half-lives?

.097%

4. 4. If a mineral contains 1.56% of its original parent isotopes, how many half-lives have passed?

6 half lives

5. 5. Approximately what percentage of parent isotopes remains after 0.5 half-lives have passed?

75%

6. 6. How many half-lives must pass before only 10 % of the original parent isotopes remain?

A little over 3

7. 7. Assume a parent isotope has a half-life of 100 million years, how old is a sample that contains 15 % of its original parent isotopes?

About 12.5 million years

8. 8. Assuming a parent isotope with a half-life = 4.5 billion years, what percentage of the original parents remains after 6.75 billion years have passed?

The half-life of U-235 decaying to Pb-207 is 713 million years. Note that this half-life can be obtained from the graph at the point where the decay and growth curves cross.

1. U-235 to Pb-207 = 713million years
2. RB-87 to Sr87 =
3. U-238 to Pb 206 =
4. C-14 to N-14 =

Hints

Question #5, % remaining = $100(1/2^n)$ where n = # of half lives or 0.5

Question #6, $10\% = 100(1/2^n)$, $0.1 = 1/2^n$, $\log 10 / \log 2 = n$

Question #7, $15\% = 100(1/2^n)$, $(\log(100/15)) / \log 2 = n$

Question #8, $n = 6.75 / 4.5$, $100(1/2^n)$