3

L3-CHEMMR





Te Mātauranga Matū, Kaupae 3, 2016

2.00 i te ahiahi Rāhina 21 Whiringa-ā-rangi 2016

PUKA RAUEMI mō 91390M, 91391M, 91392M

Tirohia tēnei puka hei whakatutuki i ngā tūmahi o ō Pukapuka Tūmahi, Tuhinga hoki.

Tirohia mēnā e tika ana te raupapatanga o ngā whārangi 2-3 kei roto i tēnei pukapuka, ka mutu, kāore tētahi o aua whārangi i te takoto kau.

KA TĀEA TĒNEI PUKA TE PUPURI HEI TE MUTUNGA O TE WHAKAMĀTAUTAU.

TE TAKA PŪMOTU

| <i>18</i> He 4.0 | 0 Ne 20.2 | 8 Ar 40.0 | .8.8 | % 31 | 5 Rn 222 | |
|-------------------------|-----------------------|-------------------------|-------------------------|-------------------------|------------------------|--------------------------|
| 1 2 H | | - | w | 54 Xe 13] | 8 | |
| 17 | 9 F 19.0 | 17 CI 35.5 | 35 Br 79.9 | 53 I 127 | 85 At 210 | |
| 16 | 8 O 16.0 | 16 S 32.1 | 34 Se 79.0 | 52 Te 128 | 84 Po 210 | 116 Lv 292 |
| 15 | 7 N 14.0 | 15 P 31.0 | 33 As 74.9 | 51 Sb 122 | 83 Bi 209 | |
| 14 | 6 C 12.0 | 14 Si 28.1 | 32 Ge 72.6 | 50 Sn 119 | 82 Pb 207 | 114 Fl 289 |
| 13 | 5 B 10.8 | 13 AI 27.0 | 31 Ga 69.7 | 49 In 115 | 81 Tl 204 | |
| • | | 12 | 30 Zn 65.4 | 48 Cd 112 | 80 Hg 201 | 112 Cn 277 |
| | | II | 29 Cu 63.5 | 47 Ag 108 | 79 Au 197 | 1111 Rg 272 |
| | | 01 | 28 Ni 58.7 | 46 Pd 106 | 78 Pt 195 | 110 Ds 271 |
| Papatipu Ngota | | 6 | 27 Co 58.9 | 45 Rh 103 | 77 Ir 192 | 109 Mt 268 |
| Papatip | | ∞ | 26 Fe 55.9 | 44 Ru 101 | 76 Os 190 | 108 Hs 265 |
| 1 H 1.0 | | K | 25 Mn 54.9 | 43 Tc 98.9 | 75 Re 186 | 107 Bh 264 |
| Tau Iraoho | | 9 | 24 Cr 52.0 | | | 106 Sg 263 |
| Tau | | 'n | 23 V 50.9 | 41 Nb 92.9 | 73 Ta 181 | 105 Db 262 |
| | | 4 | 22 Ti 47.9 | 40 Zr 91.2 | 72 Hf 179 | 104 Rf 261 |
| | | ω | 21 Sc 45.0 | 39 Y 88.9 | 71 Lu 175 | 103 Lr 262 |
| 2 | 4 Be 9.0 | 12 Mg 24.3 | 20 Ca 40.1 | 38 Sr 87.6 | 56 Ba 137 | 88 Ra 226 |
| I | 3 Li 6.9 | 11 Na 23.0 | | | 55 Cs 133 | 87 Fr 223 |

| | 57 | 58 | 59 | 09 | 61 | 62 | 63 | 64 | 65 | 99 | 29 | 89 | 69 | 70 |
|------------|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|-----|
| Raupapa | La | Ce | Pr | Nd | Pm | Sm | Eu | Сd | Tb | Dy | Ho | Er | Tm | ΛP |
| Lanthanide | 139 | 140 | 141 | 144 | 147 | 150 | 152 | 157 | 159 | 163 | 165 | 167 | 169 | 173 |
| | 68 | 06 | 91 | 92 | 93 | 94 | 95 | 96 | 26 | 86 | 66 | 100 | 101 | 102 |
| Raupapa | Ac | Th | Pa | Ω | Np | Pu | Am | Cm | Bk | Cf | Es | Fm | Md | No |
| Actinide | 227 | 232 | 231 | 238 | 237 | 239 | 241 | 244 | 249 | 251 | 252 | 257 | 258 | 259 |

PERIODIC TABLE OF THE ELEMENTS

| 18 He 4.0 | 0 Ne | Ar | 40.0 | 36 | Ţ. | 3.8 | | Xe | 131 | | Rn | 222 | | | |
|----------------------|----------------|----------------|------|----|----|------|----|------------------------|------|----|-------|-----|-------|------------|-----|
| 2 4 | 10 | 18 | | 36 | | | 54 | | | 98 | | 2 | | | |
| 17 | 9 F | 17 CI | 35.5 | 35 | Br | 79.9 | 53 | I | 127 | 85 | At | 210 | | | |
| 16 | 8 | 16 S | 32.1 | 34 | Se | 79.0 | 52 | Te | 128 | 84 | P_0 | 210 | 116 | Lv | 292 |
| 15 | 7 N | 15 P | 31.0 | 33 | As | 74.9 | 51 | $\mathbf{S}\mathbf{b}$ | 122 | 83 | Bi | 209 | | | |
| 14 | 6 C 12.0 | 14 Si | 28.1 | 32 | Ge | 72.6 | 50 | \mathbf{Sn} | 119 | 82 | Pb | 207 | 114 | F | 289 |
| 13 | 5 B | 113 Al | 27.0 | 31 | Са | 69.7 | | In | 115 | 81 | I | 204 | | | |
| l | | | 12 | 30 | Zn | 65.4 | 48 | Cd | 112 | | Hg | 201 | 112 | Cn | 277 |
| | | | II | 29 | Cu | 63.5 | 47 | Ag | 108 | 62 | Au | 197 | 1111 | Rg | 272 |
| nass | | | 0I | 28 | Z | 58.7 | 46 | Pd | 106 | 18 | Pt | 195 | 110 | Ds | 271 |
| Relative atomic mass | | | 6 | 27 | Co | 58.9 | | Rh | 103 | 77 | Ir | 192 | 109 | Mt | 268 |
| Relative | | | 8 | 26 | Fe | 55.9 | | Ru | 101 | | Os | 190 | 108 | Hs | 265 |
| 1 H 1.0 | | | 7 | | Mn | 54.9 | | Tc | 6.86 | | Re | 186 | 107 | Bh | 264 |
| mber | | | 9 | 24 | Cr | | | Mo | | | | 184 | | So | |
| Atomic number | | | 5 | 23 | > | 50.9 | | Sp | 92.9 | | La | 181 | | Dp | 262 |
| A | | | 4 | 22 | Ţ | 47.9 | | Zr | 91.2 | | Ht | 179 | 104 | Rf | 261 |
| | | | 3 | 21 | Sc | 45.0 | 39 | Y | 6.88 | | Lu | 175 | 103 | Γ r | 262 |
| ~ | 4 Be | 12 Mg | 24.3 | 20 | Ca | 40.1 | | Sr | 9.78 | | Ba | 137 | | Ra | 226 |
| I | 3 Li Li 69 | | 23.0 | | × | 39.1 | | Rb | 85.5 | | Cs | 133 | 8 8 | Fr | 223 |

| | 57 | 58 | 59 | 09 | 61 | 62 | 63 | 64 | 65 | 99 | 29 | 89 | 69 | 70 |
|-----------|-----|-----|-----|-----|-----|-----|-----|-----|------------------------|-----|-------|-----|-----|-----|
| anthanide | La | Ce | Pr | Nd | Pm | Sm | Eu | Сd | $\mathbf{T}\mathbf{b}$ | Dy | H_0 | Er | Tm | Vb |
| Series | 139 | 140 | 141 | 144 | 147 | 150 | 152 | 157 | 159 | 163 | 165 | 167 | 169 | 173 |
| | 68 | 06 | 91 | 92 | 93 | 94 | 95 | 96 | 26 | 86 | 66 | 100 | 101 | 102 |
| Actinide | Ac | Th | Pa | n | Np | Pu | Am | Cm | Bk | Cf | Es | Fm | Md | No |
| Series | 227 | 232 | 231 | 238 | 237 | 239 | 241 | 244 | 249 | 251 | 252 | 257 | 258 | 259 |

English translation of the wording on the front cover

Level 3 Chemistry, 2016

2.00 p.m. Monday 21 November 2016

RESOURCE BOOKLET for 91390, 91391, 91392

Refer to this sheet to answer the questions in your Question and Answer Booklets.

Check that this booklet has pages 2–3 in the correct order and that neither of these pages is blank.

YOU MAY KEEP THIS SHEET AT THE END OF THE EXAMINATION.