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L1-CHEMR





Level 1 Chemistry 2020

9.30 a.m. Friday 4 December 2020

RESOURCE BOOKLET

Refer to this booklet to answer the questions in your Question and Answer Booklets.

Check that this booklet has pages 2–4 in the correct order and that none of these pages is blank.

YOU MAY KEEP THIS BOOKLET AT THE END OF THE EXAMINATION.

Activity series

Ca Mg Al Zn Fe Pb (H) Cu Ag

Colours of selected ions and solids

| Colourless ions | chloride, iodide, sulfate, hydroxide, carbonate, calcium, magnesium, zinc, lead, barium, silver |
|-----------------|---|
| Blue ions | copper |
| Pale green ions | iron(II) |
| White solids | calcium sulfate, calcium hydroxide, calcium carbonate, magnesium hydroxide, magnesium carbonate, zinc carbonate, lead chloride, lead sulfate, lead carbonate, barium sulfate, barium hydroxide, barium carbonate, silver chloride |
| Green solid | iron(II) hydroxide, iron(II) carbonate |
| Blue solid | copper hydroxide |
| Yellow solid | lead iodide |
| Cream solid | silver iodide |

Solubility rules

| nitrates | All soluble |
|------------|---|
| chlorides | All soluble except silver chloride, lead chloride |
| iodides | All soluble except silver iodide, lead iodide |
| sulfates | All soluble except barium sulfate, lead sulfate, calcium sulfate |
| hydroxides | All insoluble except potassium hydroxide, sodium hydroxide |
| carbonates | All insoluble except potassium carbonate, sodium carbonate |

Table of ions

| +1 | +2 | +3 | -3 | -2 | -1 |
|------------------------------|------------------|------------------|----|-------------------------------|------------------|
| NH ₄ ⁺ | Ca ²⁺ | A1 ³⁺ | | O ²⁻ | OH_ |
| Na ⁺ | Mg ²⁺ | Fe ³⁺ | | S ²⁻ | Cl ⁻ |
| K ⁺ | Cu ²⁺ | | | CO ₃ ²⁻ | Г |
| Ag ⁺ | Pb ²⁺ | | | SO ₄ ²⁻ | NO ₃ |
| H^{+} | Fe ²⁺ | | | | HCO ₃ |
| Li ⁺ | Ba ²⁺ | | | | F- |
| | Zn ²⁺ | | | | |

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PERIODIC TABLE OF THE ELEMENTS

| 18 2 He | 10 Ne | 18 Ar | 36 Kr | 54 Xe | 86 Rn | 0g |
|----------------------|----------------|-----------------|------------------|-----------------|-----------------|-------------------|
| 17 | 9 F | 17 CI | 35 Br | 53 I | 85 At | 117 Ts |
| 91 | © 8 | 16 S | 34 Se | 52 Te | 84 Po | 116 Lv |
| 15 | Z | 15 P | 33 As | 51 Sb | 83 Bi | 115 Mc |
| 14 |) 9 | 14 Si | 32 Ge | 50 Sn | 82 Pb | 114 F1 |
| 13 | 5 B | 13 Al | 31 Ga | 49 In | 81 TI | 113 Nh |
| | | 12 | 30 Zn | 48 Cd | 80 Hg | 112 Cn |
| | | II | 29 Cu | 47 Ag | 97 Au | Rg |
| | | 10 | 28 Z i | 46 Pd | 78 Pt | 110 Ds |
| | | 6 | 27 Co | 45 Rh | 77 Ir | 109 Mt |
| | 1 | 8 | 26 Fe | 44 Ru | 92 Os | 108 Hs |
| H H | | 7 | 25 Mn | 43 Te | 75 Re | 107 Bh |
| Atomic number | | 9 | 24 C r | 42 Mo | 74 W | 106 S g |
| Atomic | | 8 | 23 V | 41 Nb | 73 Ta | 105 Db |
| | | 4 | 22 Ti | 40 Zr | 72 Hf | 104 Rf |
| | | S. | 21 Sc | 39 Y | 71 Lu | 103 Lr |
| 2 | 4 Be | 12 Mg | 20 Ca | 38 Sr | 56 Ba | 88 Ra |
| <i>F</i> | 3 Li | 11 Na | 19 K | 37 Rb | 55 Cs | 87 Fr |

| 57 | 58 | 59 | PN | 61 | 62 | 63 | 64 | 65 | 66 | 67 | 68 | 69 | 70 |
|-----------|----|-----------|-----------|-----------|-----------|----|-----------|-----------|-----------|-----------|-----|-----------|-----------|
| La | Ce | Pr | | Pm | Sm | Eu | Gd | Tb | Dy | Ho | Er | Tm | Yb |
| 89 | 90 | 91 | 92 | 93 | 94 | 95 | 96 | 97 | 98 | 99 | 100 | 101 | 102 |
| Ac | Th | Pa | U | Np | Pu | Am | Cm | Bk | Cf | Es | Fm | Md | No |