

L1-MSCIE

RERESTANT SOUTH SOUTH SOUTH SOUTH SOUTH





Pūtaiao, Kaupae 1, 2022

TE PUKAPUKA RAUEMI mō 90940M me 90944M

Whakamahia tēnei pukapuka ki te whakaoti i ngā tūmahi i 90940M me 90944M.

Tirohia kia kitea ai e tika ana te raupapatanga o ngā whārangi 2–5 i tēnei pukapuka, ka mutu, kāore tētahi o aua whārangi i te takoto kau.

E ĀHEI ANA TŌ PUPURI KI TĒNEI PUKAPUKA HEI TE MUTUNGA O TE WHAKAMĀTAUTAU.

90940 Te whakaatu māramatanga ki ngā āhuatanga o te pūhanga manawa

$$v = \frac{\Delta d}{\Delta t} \qquad a = \frac{\Delta v}{\Delta t} \qquad F_{\text{net}} = ma \qquad P = \frac{F}{A} \qquad \Delta E_{\text{p}} = mg\Delta h$$

$$E_{\text{k}} = \frac{1}{2}mv^{2} \qquad W = Fd \qquad g = 10 \text{ N kg}^{-1} \qquad P = \frac{W}{t}$$

90944 Te whakaatu māramatanga ki ngā āhuatanga o te waikawa me te pāpāhua

Te Tūtohi o ngā Katote

+1	+2	+3	-3	-2	-1
NH ₄ ⁺	Ca ²⁺	Al ³⁺		O ²⁻	OH ⁻
Na ⁺	Mg^{2+}	Fe ³⁺		S ²⁻	Cl ⁻
K ⁺	Cu ²⁺			CO ₃ ²⁻	NO ₃
Ag^+	Pb ²⁺			SO ₄ ²⁻	HCO ₃
H ⁺	Fe ²⁺				F^{-}
	Ba ²⁺				Br ⁻
	Zn ²⁺				I-

90940 Demonstrate understanding of aspects of mechanics

$$v = \frac{\Delta d}{\Delta t}$$
 $a = \frac{\Delta v}{\Delta t}$ $F_{\text{net}} = ma$ $P = \frac{F}{A}$ $\Delta E_{\text{p}} = mg\Delta h$
$$E_{\text{k}} = \frac{1}{2}mv^{2}$$
 $W = Fd$ $g = 10 \text{ N kg}^{-1}$ $P = \frac{W}{t}$

90944 Demonstrate understanding of aspects of acids and bases

Table of Ions

+1	+2	+3	-3	-2	-1
NH ₄ ⁺	Ca ²⁺	Al ³⁺		O ²⁻	OH-
Na ⁺	Mg ²⁺	Fe ³⁺		S ²⁻	Cl ⁻
K ⁺	Cu ²⁺			CO ₃ ²⁻	NO ₃
Ag^+	Pb ²⁺			SO ₄ ²⁻	HCO ₃
H ⁺	Fe ²⁺				F ⁻
	Ba ²⁺				Br_
	Zn ²⁺				I-

TE TAKA PŪMOTU

18 He	Ne	Ar	Kr	Xe	Rn	8 O g
2	10	18	36	42	98 	0
17	9 F	17 CI	35 Br	53 I	85 At	117 Ts
91	© 8	16 S	34 Se	52 Te	84 Po	116 Lv
15	Z	15 P	33 As	51 Sb	83 Bi	115 Mc
14) 9	14 S i	32 Ge	50 Sn	82 Pb	114 F1
13	B	13 Al	31 Ga	49 In	81 TI	113 Nh
		12	30 Zn	48 Cd	80 Hg	112 Cn
		II	29 Cu	47 Ag	79 Au	111 Rg
		10	28 Ni	46 Pd	78 Pt	110 Ds
		0	27 Co	45 Rh	77 Ir	109 Mt
		80	26 Fe	44 Ru	76 Os	108 Hs
1 H		_	25 Mn	43 Tc	75 Re	107 Bh
Te tau iraoho		9	24 Cr	42 Mo	74 W	106 Sg
Te taı		8	23 V	41 Nb	73 Ta	105 Db
		4	22 Ti	40 Zr	72 Hf	104 Rf
		3	21 Sc	39 Y	71 Lu	103 Lr
$^{\mathcal{O}}$	4 Be	12 Mg	20 Ca	38 Sr	56 Ba	88 Ra
I	3 Li	11 Na	19 K	37 Rb	55 Cs	87 Fr

70	102
Yb	No
M L	101 Md
68	100
Er	Fm
67	99
Ho	Es
66 Dy	98 99 Es
65	97
Tb	Bk
64	96
Gd	Cm
63	95
Eu	Am
62	94
Sm	Pu
61	93
Pm	Np
pN	92 U
59	91
Pr	Pa
58	90
Ce	Th
57	89
La	Ac

PERIODIC TABLE OF THE ELEMENTS

18 2 He	10 Ne	18 Ar	36 Kr	54 Xe	86 Rn	0 g
17	9 F	17 CI	35 Br	53 I	85 At	117 Ts
16	8	16 S	34 Se	52 Te	84 Po	116 Lv
15	Z	15 P	33 As	51 Sb	83 Bi	115 Mc
14) 9	14 Si	32 Ge	50 Sn	82 Pb	114 F1
13	5 B	13 Al	31 Ga	49 In	81 TI	113 Nh
		12	30 Zn	48 Cd	80 Hg	112 Cn
		II	29 Cu	47 Ag	79 Au	111 Rg
		01	28 Ni	46 Pd	78 Pt	110 Ds
		6	27 C0	45 Rh	77 Ir	109 Mt
		&	26 Fe	44 Ru	92 Os	108 Hs
1 H			25 Mn	43 Tc	75 Re	107 Bh
Atomic number		9	24 Cr	42 Mo	74 W	106 Sg
Atomic		2	23 V	41 Nb	73 Ta	105 Db
		4	22 Ti	40 Zr	72 Hf	104 Rf
		~	21 Sc	39 Y	71 Lu	103 Lr
\sim	4 Be	12 Mg	20 Ca	38 Sr	56 Ba	88 Ra
I	3 Li	11 Na	19 K	37 Rb	55 Cs	87 Fr

Pr	p N	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67	68 Er	69 Tm	70 Yb
+	92	93	94	95	96	97	98	99	100	101	102
	U	N p	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	N o

English translation of the wording on the front cover

Level 1 Science 2022

RESOURCE BOOKLET for 90940M and 90944M

Refer to this booklet to answer the questions for 90940M and 90944M.

Check that this booklet has pages 2–5 in the correct order and that none of these pages is blank.

YOU MAY KEEP THIS BOOKLET AT THE END OF THE EXAMINATION.