L1-PHYSMR





# Ahupūngao, Kaupae 1, 2019

2.00 i te ahiahi Rātū 19 Whiringa-ā-rangi 2019

PUKAPUKA RAUEMI mō 91170M, 91171M, 91173M

Tirohia tēnei pukapuka hei whakatutuki i ngā tūmahi o ō Pukapuka Tūmahi, Tuhinga hoki.

Tirohia mēnā e tika ana te raupapatanga o ngā whārangi 2–3 kei roto i tēnei pukapuka, ka mutu, kāore tētahi o aua whārangi i te takoto kau.

KA TAEA TĒNEI PUKAPUKA TE PUPURI HEI TE MUTUNGA O TE WHAKAMĀTAUTAU.

Tērā pea ka whai hua ēnei tikanga tātai, raraunga hoki ki a koe.

90937M Te whakaatu māramatanga ki ētahi āhuatanga o te hiko me te autō

$$V = IR$$

$$P = IV$$

$$P = \frac{E}{t}$$

$$R_{T} = R_{1} + R_{2} + \dots$$

$$R = \frac{k I}{d}$$

$$k = 2.0 \times 10^{-7} \text{ T m A}^{-1}$$

90938M Te whakaatu māramatanga ki ētahi āhuatanga o te ngaru

$$v = \frac{d}{t}$$
  $v = f\lambda$   $f = \frac{1}{T}$ 

90939M Te whakaatu māramatanga ki ētahi āhuatanga o te wera<sup>1</sup>

$$Q = mc\Delta T$$
  $Q = mL$   $P = \frac{E}{t}$ 

### Raraunga whaihua

Kītanga wera motuhake o te tio =  $2100 \text{ J kg}^{-1} \,^{\circ}\text{C}^{-1}$ Kītanga wera motuhake o te wai =  $4200 \text{ J kg}^{-1} \,^{\circ}\text{C}^{-1}$ Wera moe o te honokarihi wai =  $330\,000 \text{ J kg}^{-1}$ Wera moe o te whakahaurehu wai =  $2300\,000 \text{ J kg}^{-1}$  You may find the following formulae and data useful.

### 90937 Demonstrate understanding of aspects of electricity and magnetism

$$V = IR$$

$$P = IV$$

$$P = \frac{E}{t}$$

$$R_{T} = R_{1} + R_{2} + \dots$$

$$R = \frac{k I}{d}$$

$$k = 2.0 \times 10^{-7} \text{ T m A}^{-1}$$

### 90938 Demonstrate understanding of aspects of wave behaviour

$$v = \frac{d}{t}$$
  $v = f\lambda$   $f = \frac{1}{T}$ 

## 90939 Demonstrate understanding of aspects of heat

$$Q = mc\Delta T$$
  $Q = mL$   $P = \frac{E}{t}$ 

#### Useful data

Specific heat capacity of ice =  $2100 \text{ J kg}^{-1} \,^{\circ}\text{C}^{-1}$ Specific heat capacity of water =  $4200 \text{ J kg}^{-1} \,^{\circ}\text{C}^{-1}$ Latent heat of fusion of water =  $330\,000 \text{ J kg}^{-1}$ Latent heat of vaporisation of water =  $2300\,000 \text{ J kg}^{-1}$ 

# English translation of the wording on the front cover

# Level 1 Physics, 2019

2.00 p.m. Tuesday 19 November 2019

RESOURCE BOOKLET for 91170, 91171, and 91173

Refer to this booklet to answer the questions in your Question and Answer Booklets.

Check that this booklet has pages 2–3 in the correct order and that none of these pages is blank.

YOU MAY KEEP THIS SHEET AT THE END OF THE EXAMINATION.