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Te Mātauranga Matū, Kaupae 1, 2016

2.00 i te ahiahi Rāhina 21 Whiringa-ā-rangi 2016

PUKAPUKA RAUEMI mō 90933M me 90934M

Tirohia tēnei pukapuka hei whakatutuki i ngā tūmahi o ō Pukapuka Tūmahi, Tuhinga hoki.

Tirohia mēnā e tika ana te raupapatanga o ngā whārangi 2–5 kei roto i tēnei pukapuka, ka mutu, kāore tētahi o aua whārangi i te takoto kau.

KA TĀEA TĒNEI PUKAPUKA TE PUPURI HEI TE MUTUNGA O TE WHAKAMĀTAUTAU.

Raupapatanga hohe

Ca Mg Al Zn Fe Pb (H) Cu Ag

Ngā ture mō te mehamehanga

ngā pākawa ota, NO ₃	Ka memeha te katoa
ngā pūhaumāota, Cl	Ka memeha te katoa, hāunga te AgCl, PbCl ₂
ngā kahautawa, I¯	Ka memeha te katoa, hāunga te AgI, PbI ₂
ngā pākawa pungatara, ${{ m SO_4}}^{2-}$	Ka memeha te katoa, hāunga te BaSO _{4,} PbSO _{4,} CaSO ₄
ngā waihā, OH¯	Ka memeha te katoa, hāunga te KOH, NaOH
ngā pākawa waro, ${\rm CO_3}^{2-}$	Ka memeha te katoa, hāunga te K ₂ CO ₃ , Na ₂ CO ₃

Taka katote

+1	+2	+3	-3	-2	-1
NH ₄ ⁺	Ca ²⁺	A1 ³⁺		O ²⁻	OH_
Na ⁺	Mg ²⁺	Fe ³⁺		S ²⁻	Cl ⁻
K ⁺	Cu ²⁺			CO ₃ ²⁻	Г
Ag ⁺	Pb ²⁺			SO ₄ ²⁻	NO ₃
H^{+}	Fe ²⁺				HCO ₃
Li ⁺	Ba ²⁺				F ⁻
	Zn ²⁺				

Activity series

Ca Mg Al Zn Fe Pb (H) Cu Ag

Solubility rules

nitrates, NO ₃ ⁻	All soluble
chlorides, Cl	All soluble except AgCl, PbCl ₂
iodides, I	All soluble except AgI, PbI ₂
sulfates, SO_4^{2-}	All soluble except BaSO ₄ , PbSO ₄ , CaSO ₄
hydroxides, OH	All insoluble except KOH, NaOH
carbonates, CO ₃ ²⁻	All insoluble except K ₂ CO ₃ , Na ₂ CO ₃

Table of ions

+1	+2	+3	-3	-2	-1
NH ₄ ⁺	Ca ²⁺	A1 ³⁺		O ²⁻	OH_
Na ⁺	Mg ²⁺	Fe ³⁺		S ²⁻	Cl ⁻
K ⁺	Cu ²⁺			CO ₃ ²⁻	Γ
Ag ⁺	Pb ²⁺			SO ₄ ²⁻	NO ₃
H^{+}	Fe ²⁺				HCO ₃
Li ⁺	Ba ²⁺				F-
	Zn ²⁺				

TE TAKA PŪMOTU

18	2 He	10 Ne	18 Ar	36 Kr	54 Xe	86 Rn	
	17	9 F	17 CI	35 Br	53 I	85 A t	
	91	© &	16 S	34 Se	52 Te	84 Po	116 Lv
	15	Z	15 P	33 As	51 Sb	83 Bi	
	14	O 9	14 Si	32 Ge	50 Sn	82 Pb	114 F1
	13	5 B	13 Al	31 Ga	49 In	18 TI	
			12	30 Zn	48 C d	\mathbf{H}	112 C n
			II	29 Cu	47 Ag	79 Au	1111 R g
			10	28 Z i	46 Pd	78 Pt	110 Ds
			6	27 Co	45 Rh	77 Ir	109 Mt
		1	∞	26 Fe	44 Ru	92 Os	108 Hs
	1 H		~	25 Mn	43 Tc	75 Re	107 Bh
	Tau Iraoho		9	24 Cr	42 Mo	74 W	106 S g
	Ta		'n	23 V	41 Nb	73 Ta	105 Db
			4	22 Ti	40 Zr	72 Hf	104 Rf
			'n	21 Sc	39 Y	71 Lu	103 Lr
	2	4 Be	12 Mg	20 Ca	38 Sr	56 Ba	88 Ra
	<i>F</i>	3 Li	11 Na	19 K	37 Rb	55 Cs	87 Fr

Ce Ce	59 Pr	PN	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	0H	68 Er	69 Tm	70 Yb
	91	92	93	94	95	96	97	98	99	100	101	102
	Pa	U	N p	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No

PERIODIC TABLE OF THE ELEMENTS

18 2 He	10 Ne	18 Ar	36 Kr	54 Xe	86 Rn	
17	9 F	17 CI	35 Br	53 I	85 At	
91	0 8	16 S	34 Se	52 Te	84 Po	116 Lv
15	Z	15 P	33 As	51 Sb	83 Bi	
14) 9	14 S i	32 Ge	50 Sn	82 Pb	114 F1
13	5 B	13 Al	31 Ga	49 In	81 TI	
		12	30 Zn	48 C d	80 Hg	112 Cn
		II	29 Cu	47 Ag	79 Au	Rg
		10	28 Ni	46 Pd	78 Pt	110 Ds
		6	27 Co	45 Rh	77 Ir	109 Mt
	1	8	26 Fe	44 Ru	92 Os	108 Hs
<u> </u>		~	25 Mn	43 Tc	75 Re	107 Bh
Atomic number		9	24 Cr	42 Mo	74 W	Sg
Atomic		3	23 V	41 Nb	73 Ta	105 Db
		4	22 Ti	40 Zr	72 Hf	104 Rf
		80	21 Sc	39 Y	71 Lu	103 Lr
2	4 Be	12 Mg	20 Ca	38 Sr	56 Ba	88 Ra
1	3 Li	11 Na	19 K	37 Rb	55 Cs	87 Fr

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$\frac{70}{\mathbf{Yb}}$	102 No
69	101
Tm	Md
68	100
Er	Fm
67	99
Ho	Es
66	98
Dy	Cf
65	97
Tb	Bk
64	96
Gd	C m
63	95
Eu	Am
62	94
Sm	Pu
61	93
Pm	Np
PN	92 U
59	91
Pr	Pa
58	90
Ce	Th
57	89
La	Ac

English translation of the wording on the front cover

Level 1 Chemistry, 2016

2.00 p.m. Monday 21 November 2016

RESOURCE BOOKLET for 90933 and 90934

Refer to this booklet to answer the questions in your Question and Answer Booklets.

Check that this booklet has pages 2–5 in the correct order and that none of these pages is blank.

YOU MAY KEEP THIS BOOKLET AT THE END OF THE EXAMINATION.