See back cover for an English translation of this cover



90939M



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Ahupūngao, Kaupae 1, 2013

90939M Te whakaatu māramatanga ki ētahi āhuatanga o te wera

2.00 i te ahiahi Rāhina 25 Whiringa-ā-rangi 2013 Whiwhinga: Whā

Paetae	Paetae Kaiaka	Paetae Kairangi
Te whakaatu māramatanga ki ētahi āhuatanga o te wera.	Te whakaatu māramatanga hōhonu ki ētahi āhuatanga o te wera.	Te whakaatu māramatanga matawhānui ki ētahi āhuatanga o te wera.

Tirohia mehemea e ōrite ana te Tau Ākonga ā-Motu (NSN) kei tō pepa whakauru ki te tau kei runga ake nei.

Me whakautu e koe ngā pātai KATOA kei roto i te pukapuka nei.

Tirohia mēnā kei a koe te Rau Rauemi L1-PHYSMR.

Ki roto i ō whakautu, whakamahia ngā whiriwhiringa tohutau mārama, ngā kupu, ngā hoahoa hoki/rānei ki hea hiahiatia ai.

Me hoatu te wae tika o te Pūnaha o te Ao (SI) ki ngā whakautu tohutau.

Ki te hiahia koe ki ētahi atu wāhi hei tuhituhi whakautu, whakamahia ngā whārangi kei muri i te pukapuka nei, ka āta tohu ai i ngā tau pātai.

Tirohia mehemea kei roto nei ngā whārangi 2–15 e raupapa tika ana, ā, kāore hoki he whārangi wātea.

HOATU TE PUKAPUKA NEI KI TE KAIWHAKAHAERE HEI TE MUTUNGA O TE WHAKAMĀTAUTAU.



Ka taea e koe te whakamahi ngā raraunga e whai ake mō ngā pātai.

Kahapuri wera¹ motuhake o te wai $= 4200 \text{ J kg}^{-1} \text{ K}^{-1}$

Wera torohū o te hononga tio = $3.3 \times 10^5 \text{ J kg}^{-1}$

Wera torohū o te tākohutanga o te wai $= 2.3 \times 10^6 \text{ J kg}^{-1}$

PĀTAI TUATAHI: WHAKAWHITI PŪNGAO WERA

(a) E toru ngā momo tikanga whakawhiti pūngao wera.

Whakaingoatia ia tikanga whakawhiti pūngao wera me te whakaingoa anō i te kawenga (totoka/wē/haurehu) mō ia tikanga e taea ai te whakawhiti pūngao wera mōrahi.

1.			

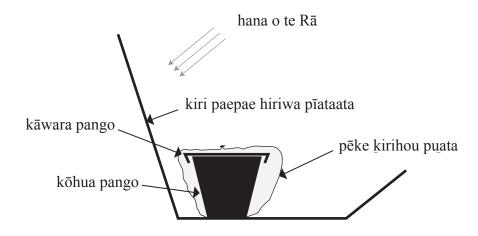
2.		

3.			

E whakaatu ana te pikitia i tētahi tō kōmaru e whakamahia ana ki te whakawera kai. He kiri paepae hanga-L tōna me tētahi pātū hiriwa o roto. He pūhara kei waenga mō tētahi kōhua. Ka whakatūhia tētahi kōhua pango me te kāwara² ki roto i tētahi pēke kirihou puata kati.



E whakaatu ana te hoahoa i raro i te tō mai i te taha.



¹ pōkākā

² taupoki, popoki

You are advised to spend 60 minutes answering the questions in this booklet.

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You may use the following data for any question:

Specific heat capacity of water $= 4200 \text{ J kg}^{-1} \text{ K}^{-1}$ Latent heat of fusion of ice $= 3.3 \times 10^5 \text{ J kg}^{-1}$ Latent heat of vaporisation of water $= 2.3 \times 10^6 \text{ J kg}^{-1}$

QUESTION ONE: HEAT TRANSFER

(a) There are three different heat transfer methods.

Name each heat transfer method and name the medium (solid/liquid/gas) for each method that would allow for maximum heat transfer to occur.

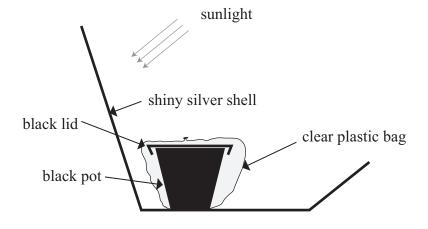
1.				

2.			

The picture shows a solar cooker being used to heat food. It consists of an L-shaped shell with a silver interior wall. There is a platform in the middle for a cooking pot. A black cooking pot with a lid will be placed inside a closed clear plastic bag.



The diagram below shows a side view of the cooker.



Ko ngā āhuatanga matua o te tō:

- te pātū pīataata o roto o te tō
- te mata pango o te kōhua
- te kiri paepae whakaata hanga-L

W/h	akamāramahia te take ka āwhina te kāwara ki te whakatere ake i te tunu i te kai.
VV 11	akamaramania te take ka awimia te kawara ki te whakatere ake i te tuhu i te kar.
17	1 (*) -1 1 1 1 2 2 4 1 2 4 1 2 4 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1 1 2 4 1
Ka	katia e te pēke kirihou puata kati tētahi apa ³ hau huri noa i te kōhua.
Wh	akamāramahia mai he pēhea te kōpani i te tō ki roto i te pēke kirihou puata e āwhina ai k
	vhakatere ake i te tunu kai.

MĀ TE KAIMĀKA ANAKE

³ papanga

The main features of the cooker are:

- the shiny interior wall of the cooker
- the black surface of the pot

the L-shaped reflector shell.
Explain how these features help heat the food efficiently.
Explain why the lid on the pot helps to cook the food more quickly.
The closed clear plastic bag traps a layer of air around the pot.
Explain how enclosing the cooker inside the clear plastic bag aids faster cooking.

PĀTAI TUARUA: TE WHAKAWERA WAI I ROTO I TĒTAHI TŌ KŌMARU

Ka putua ⁴ 100 kāwara mō te 2	g wai i te 23°C ki roto i te kōhua o te tō kōmaru ka waiho i te Rā me te kore 0 meneti.				
I te paunga o te	e rua meneti, ko te pāmahana o te wai he 42°C.				
Tātaihia te nui	o te pūngao wera i riro mai i te wai.				
	pūngao wera				
V 4- 20	Ka pau te 20 meneti e whakawerahia ana, ka korohū te wai i te 100°C. I te tangohanga mai o e kōhua i te tō, he 87 g anake te papatipu o te wai i roto i te kōhua.				
te kõhua i te tõ Whakamārama					
te kõhua i te tõ Whakamārama	, he 87 g anake te papatipu o te wai i roto i te kōhua. hia mai, e pā ana ki te nekehanga korakora, i ahatia te wai i roto i te 20 men				
te kōhua i te tō Whakamārama	, he 87 g anake te papatipu o te wai i roto i te kōhua. hia mai, e pā ana ki te nekehanga korakora, i ahatia te wai i roto i te 20 men				
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te kõhua i te tõ Whakamārama	, he 87 g anake te papatipu o te wai i roto i te kōhua. hia mai, e pā ana ki te nekehanga korakora, i ahatia te wai i roto i te 20 men				

MĀ TE KAIMĀKA ANAKE

⁴ tāringitia

QUESTION TWO: HEATING WATER IN A SOLAR COOKER

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State what is meant by the term temperature .
100 g of water at 23°C is poured into the solar cooker's pot and left in the sun without the l for 20 minutes.
Two minutes later, the temperature of the water is 42°C.
Calculate the amount of heat energy gained by the water.
heat energy
After 20 minutes of heating, the water boils at 100°C. When the pot was removed from the cooker, the mass of the water in the pot was only 87 g.
Explain, in terms of particle motion, what has happened to the water during the 20 minutes
Your answer should show an understanding of the terms heat energy and temperature .

	Tātaihia te nui o te pūngao wera i whakawhitihia ki te wai i roto i te 20 meneti.	
	nui o te pūngao wera	
)	Tātaihia te hiko toharite o te tō kōmaru i roto i te 20 meneti.	
	hiko	

l)	(i)	During 20 minutes of heating, the temperature of the water rises from 23°C to 100°C, and the mass of the water decreases from 100 g to 87 g.		
		Calculate the amount of heat energy transferred to the water in 20 minutes.		
		amount of heat energy		
	(ii)	Calculate the average power of the solar cooker during the 20 minutes.		
		power		

PĀTAI TUATORU: NOHO PUNI

MĀ TE
KAIMĀKA
ANAKE

a)	Ina noho a Sonya ki tētahi tūru puni, ka pā mai te mōhio ki a ia he mātao ake te anga maitai ki te papanga o te tūru, ahakoa kei te pāmahana ōrite te maitai me te papanga.				
	Whakamāramahia mai tēnei rerekētanga.				
))	Ka tahuna e Sonya he ahi kōpae. E mōhio ana ia me eke te pāhamana o te wahia ki te 350°C kia tahu ai. Ka kite ia he māmā ake te tiki rārā iti hei tahu tēnā i ngā wahia nui.				
	Whakamahia ngā kupu kahapuri wera me te pūngao wera hei whakamārama he aha i māmā ake ai ki te whakawera i ngā rārā iti ki te 350°C hei tahu i te ahi tēnā i ngā wahia nui ake.				
)	E roa ana te noho a Sonya i te taha o te ahi, ka tīmata tana werawera.				
	Whakamāramahia mai he pēhea e āwhina ai te werawera kia noho kōangiangi te tinana.				

QUESTION THREE: CAMPING

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E	xplain this difference.
	onya lights a camp fire. She knows that wood must reach a temperature of 350°C to burn he finds that it is easier to get small twigs to burn than large logs.
	se the terms heat capacity and heat energy to explain why it is easier to heat the small rigs to 350°C to start the fire than the large logs.
Α.	fter being near the fire for a while, Sonya begins to sweat.
Εž	xplain how sweat helps to cool her body.

Tātaihia ta wā mā ta tia 100 g i ta 0°C kie rowa hai wai i ta 0°C				
Tātaihia te wā mō te tio 100 g i te 0°C kia rewa hei wai i te 0°C.				
	wā			

13			
The next morning, Sonya melts some ice using a gas cooker. The power available from the cooker is 0.15 kW.	ASSES		
Calculate the time for 100 g of ice at 0°C to melt into water at 0°C.			
time			

	He puka anō mēnā ka hiahiatia.		
TAU PĀTAI	Tuhia te (ngā) tau pātai mēnā e hāngai ana.		
PATAI			

	Extra paper it required.			
OUESTION	Write the question number(s) if applicable.			
QUESTION NUMBER	Time the question number (e) in approach			

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English translation of the wording on the front cover

Level 1 Physics, 2013

90939 Demonstrate understanding of aspects of heat

2.00 pm Monday 25 November 2013 Credits: Four

Achievement	Achievement with Merit	Achievement with Excellence
Demonstrate understanding of aspects of heat.	Demonstrate in-depth understanding of aspects of heat.	Demonstrate comprehensive understanding of aspects of heat.

Check that the National Student Number (NSN) on your admission slip is the same as the number at the top of this page.

You should attempt ALL the questions in this booklet.

Make sure that you have Resource Sheet L1-PHYSR.

In your answers use clear numerical working, words and/or diagrams as required.

Numerical answers should be given with an appropriate SI unit.

If you need more space for any answer, use the page(s) provided at the back of this booklet and clearly number the question.

Check that this booklet has pages 2–15 in the correct order and that none of these pages is blank.

YOU MUST HAND THIS BOOKLET TO THE SUPERVISOR AT THE END OF THE EXAMINATION.