

Questions

1. Which of the following is the most paramagnetic in nature?

- a) N₂
- b) BC
- c) NO
- d) O₂**

2. Identify the increasing order of spectrochemical series

- a) I⁻ < Br⁻ < S²⁻ < en < NO₂⁻ < CN⁻ < CO
- b) I⁻ < Br⁻ < S²⁻ < en < NO₂⁻ < CO < CN⁻**
- c) CO > CN⁻ > NO₂⁻ > en > I⁻ < Br⁻ < S²⁻
- d) I⁻ < Br⁻ < S²⁻ = en = NO₂⁻ < CN⁻ < C

3. Calculate the number of fundamental vibrations for CO₂ and HCl molecules

- a) 1 and 3
- b) 4 and 1**
- c) 0 and 1
- d) 3 and 4

4. What happens to the vibrational frequency of molecule upon increasing bond strength

- a) Decreases
- b) Remains same
- c) Increases**
- d) No dependence

5. What happens to the absorbance of the sample upon increasing the path length (from 0.1 to 1 cm) of the sample tube?

- a) Decreases
- b) No change
- c) Increases**
- d) Sample's absorbance and path length are independent

6. Which among the following doesn't show rotational spectrum?

- a) HCl
- b) O₂**
- c) HBr
- d) H₂O

7. The ¹H NMR spectrum of ethanol consists of

- a) 0 signals
- b) 1 signal
- c) 3 signals**
- d) 4 signals

8. The order of increasing ionic radius of the following is

- a) $K^+ < Li^+ < Mg^{2+} < Al^{3+}$
- b) $K^+ < Mg^{2+} < Li^+ < Al^{3+}$
- c) $Li^+ < K^+ < Mg^{2+} < Al^{3+}$
- d) $Al^{3+} < Mg^{2+} < Li^+ < K^+$**

9. According to Fajan's rule, covalent bond is favoured by____

- a) Large cation and small anion
- b) Large cation and large anion
- c) Small cation and large anion**
- d) Small cation and small anion

10. The miller indices of crystal planes which cut through the crystal axes at (2a,3b,2c) are:

- (a) (326), **(b) (323)**, (c) (232), (d) (626)

11. In the Born-Oppenheimer approximation, which is the correct order?

Ans: **(a) $E_{el} > E_{vib} > E_{rot} > E_{tr}$** , (b) $E_{tr} > E_{rot} > E_{vib} > E_{el}$
(c) $E_{vib} > E_{rot} > E_{el} > E_{tr}$, (d) $E_{rot} > E_{vib} > E_{el} > E_{tr}$

12. A compound shows an NMR peak at 240 Hz downfield from the TMS peak in the spectrometer operating at 60 MHz.

The chemical shift (in ppm) is:

(a) 1, (b) 2, (c) 3, **(d) 4**

13. In X-ray diffraction, how the interplanar distance (d) and the angle between incident and transmitted beam (2θ) are related?

(a) d increases with increase in 2θ

(b) d decreases with increase in 2θ

(c) No relation between d and 2θ

14. The order for the electron affinity is:

(a) $\text{Ne} < \text{N} < \text{C} < \text{O} < \text{F}$, (b) $\text{C} < \text{N} < \text{O} < \text{F} < \text{Ne}$, (c) $\text{F} < \text{C} < \text{N} < \text{O} < \text{Ne}$, (d) $\text{Ne} < \text{F} < \text{O} < \text{N} < \text{C}$

15. How many signals are observed in the proton NMR of $(\text{CH}_3)_2\text{CHCH}_2\text{CH}_3$ and $\text{C}_6\text{H}_5\text{CH}_3$, respectively

(a) 5 and 2, (b) 5 and 3, **(c) 4 and 2**, (d) 3 and 2