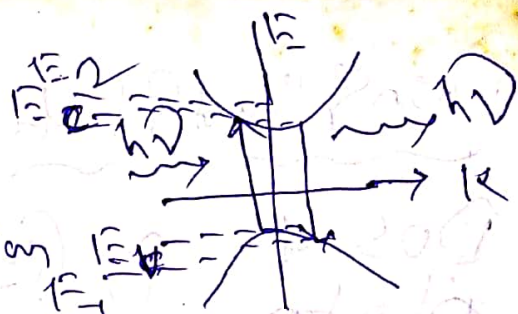


Optical Joint Density of State:



No. of emission (or) Absorption per unit volume is given by

$$n = \int n(E) dE = \int \rho(E) f(E) dE \rightarrow (1)$$

$$P = \int \rho(E) dE = \int \rho(E) [1 - f(E)] dE \rightarrow (2)$$

No. of Pairs of States in Conduction and Valence Band

$$E_2 = E_c + \frac{\hbar^2 k^2}{2m_c} \rightarrow (3)$$

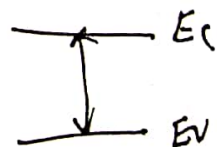
$$E_1 = E_v - \frac{\hbar^2 k^2}{2m_v} \rightarrow (4)$$

$$E_2 - E_1 = E_c - E_v + \frac{\hbar^2 k^2}{2} \left[\frac{1}{m_c} + \frac{1}{m_v} \right]$$

$$E_2 - E_1 = E_g + \frac{\hbar^2 k^2}{2m_r} \rightarrow (5)$$

$$\therefore h\nu = E_g + \frac{\hbar^2 k^2}{2m_r}$$

$$(or) (h\nu - E_g) = \frac{\hbar^2 k^2}{2m_r}$$



$$\therefore k^2 = \frac{2m_r}{\hbar^2} (h\nu - E_g) \rightarrow (6)$$

using eq (6) in eq (3) we get

$$E_2 = E_c + \frac{m_r}{m_c} (h\nu - E_g) \rightarrow (4)$$

$$\text{or } (E_2 - E_c) = \frac{m_r}{m_c} (h\nu - E_g) \rightarrow (5)$$

Optical Joint density of State

$$P(\nu) d\nu = P(E_2) dE_2 \rightarrow (6)$$

$$\begin{aligned} \therefore P(\nu) &= P(E_2) \frac{dE_2}{d\nu} \\ &= \frac{1}{2\pi^2} \left(\frac{2m_c}{\hbar^2} \right)^{3/2} (E_2 - E_c)^{1/2} h \frac{m_r}{m_c} \\ &= \frac{1}{2\pi^2} \left(\frac{2m_c}{\hbar^2} \right)^{3/2} \left(\frac{m_r}{m_c} \right)^{1/2} (h\nu - E_g)^{1/2} h \frac{m_r}{m_c} \\ P(\nu) &= \frac{1}{\pi \hbar^2} (2m_r)^{3/2} (h\nu - E_g)^{1/2} \rightarrow (7) \end{aligned}$$

Eq (7) is Optical Joint Dos