



UNION PUBLIC SERVICE COMMISSION

EXAMINATION NOTICE NO. 5/2015-GEOL

DATED : 21.02.2015

(LAST DATE FOR SUBMISSION OF APPLICATIONS : 20.03.2015)

COMBINED GEO-SCIENTIST AND GEOLOGIST EXAMINATION, 2015

(COMMISSION'S WEBSITE : www.upsc.gov.in)

No. 4/1/2014-E.I(B) : A competitive examination for recruitment to the posts mentioned in para 2 below will be held by the Union Public Service Commission commencing from the **23rd May, 2015** in accordance with the Rules published by the Ministry of Mines in the Gazette of India dated the **21st February 2015** at the following centres.

| | |
|------------|--------------------|
| AHMEDABAD | JAIPUR |
| ALLAHABAD | JAMMU |
| BANGALORE | KOLKATA |
| BHOPAL | LUCKNOW |
| CHANDIGARH | MUMBAI |
| CHENNAI | PATNA |
| CUTTACK | SHILLONG |
| DELHI | SHIMLA |
| DISPUR | THIRUVANANTHAPURAM |
| HYDERABAD | |

The centres and the dates of holding the examination as mentioned above are liable to be changed at the discretion of the commission. Applicants should note that there will be a ceiling on the number of candidates allotted to each of the Centre, except Chennai, Delhi, Dispur, Kolkata and Ahmedabad. Allotment of Centres will be on the "first-apply-first-allot" basis, and once the capacity of a particular Centre is attained, the same will be frozen. Applicants, who cannot get a Centre of their choice due to ceiling, will be required to choose a Centre from the remaining ones. Applicants are, thus, advised that they may apply early so that they could get a Centre of their choice.

NB: Notwithstanding the aforesaid provision, Commission reserve the right to change the Centres at their discretion if the situation demands. Candidates admitted to the examination will be informed of the time table and place or places of examination. Candidates should note that no request for change of centre will be granted.

2(a). The categories of post to which recruitment is to be made on the results of this examination and the approximate number of vacancies in the various posts are given below :-

Category I : (Posts in the Geological Survey of India, Ministry of Mines).

- (i) Geologist, Group A : 150
- (ii) Geophysicist, Group A : 40
- (iii) Chemist, Group A : 50

Category II: (Posts in the Central Ground Water Board, Ministry of Water Resources)

- (i) Junior Hydrogeologists (Scientist B), Group A : 29

NOTE : 18 vacancies of Geologists are reserved for Physically Challenged candidates (16 for Hearing Impairment (partially Deaf)- and 2 for Locomotor disability or Cerebral Palsy) (OA, OL, PD).

2 vacancies of Geophysicist (1 each for hearing Impairment and Locomotor disability (OA, OL, PB, PD) or Cerebral palsy).

3 Vacancies of Chemists (2 for Hearing Impairment (partially Deaf) and 1 for Locomotor disability or Cerebral Palsy).

2 vacancies of Jr. Hydro-geologist (1 each for Hearing Impairment and Locomotor disability or Cerebral Palsy (OL)).

The above number of vacancies are liable to alteration.

Appointments will be made on a temporary basis in the first instance.

Reservations will be made for candidates belonging to Scheduled Castes, Scheduled Tribes, Other Backward Classes and Physically disabled

IMPORTANT

1. CANDIDATES TO ENSURE THEIR ELIGIBILITY FOR THE EXAMINATION:

Candidates applying for the examination should ensure that they fulfil all eligibility conditions for admission to the Examination. Their admission at all the stages of the examination will be purely provisional subject to satisfying the prescribed eligibility conditions.

Mere issue of Admission Certificate to the candidate will not imply that his/her candidature has been finally cleared by the Commission.

Verification of eligibility conditions with reference to original documents is taken up only after the candidate has qualified for Interview/Personality Test.

2. HOW TO APPLY :

Candidates are required to apply **online** only by using the website www.upsconline.nic.in. Brief instructions for filling up the online Application Form have been given in Appendix-II. Detailed instructions are available on the above mentioned website.

3. LAST DATE OF SUBMISSION OF APPLICATIONS:

The Online Applications can be filled upto **20th March, 2015 till 11.59 PM**, after which the link will be disabled.

4. The eligible candidates shall be issued an e-Admission Certificate two weeks before the commencement of the examination. The e-Admission Certificate will be made available in the UPSC website [www.upsc.gov.in] for downloading by candidates. No Admission Certificate will be sent by post. All the applicants are required to provide valid active E-Mail I.D. while filling up Online Application Form as the Commission may use electronic mode for contacting them at different stages of examination process.

5. SPECIAL INSTRUCTIONS :

Candidates are advised to read carefully "Special Instructions to the Candidates for Conventional Type Tests" (Appendix III).

6. FACILITATION COUNTER FOR GUIDANCE OF CANDIDATES :

In case of any guidance/information/clarification regarding their applications, candidature etc. candidates can contact UPSC's Facilitation Counter near 'C' Gate of its campus in person or over **Telephone No. 011-23385271/011-23381125/011-23098543** on working days between 10.00 hrs and 17.00 hrs.

7. Mobile Phones Banned:

(a) Mobile phones, pagers, bluetooth or any other communication devices are not allowed inside the premises where the examination is being conducted. Any infringement of these instructions shall entail disciplinary action including ban from future examinations.

(b) Candidates are advised in their own interest not to bring any of the banned item including mobile phones/pagers or any valuable/costly items to the venue of the examination, as arrangement for safe-keeping cannot be assured. Commission will not be responsible for any loss in this regard.

CANDIDATES ARE REQUIRD TO APPLY ONLINE AT www.upsconline.nic.in

ONLY. NO OTHER MODE IS ALLOWED FOR SUBMISSION OF APPLICATION

categories in respect of vacancies as may be fixed by the Government.

A candidate will be eligible to get the benefit of community reservation only in case the particular caste to which the candidate belongs is included in the list of reserved communities issued by the Central Government. If a candidate indicates in his/her Application Form for the Examination that he/she belongs to General Category but subsequently writes to the Commission to change his/her category, to a reserved one, such request shall not be entertained by the Commission.

While the above principle will be followed in general, there may be a few cases where there was a little gap (say 2-3 months)

between the issuance of a Government Notification enlisting a particular community in the list of any of the reserved communities and the date of submission of the application by the candidate. In such cases the request of change of community from general to reserved may be considered by the Commission on merit. Candidates seeking reservation/relaxation benefits available for SC/ST/OBC/PH/Ex-servicemen must ensure that they are entitled to such reservation/ relaxation as per eligibility prescribed in the Rules/Notice. They should also be in possession of all the requisite certificates in the prescribed format in support of their claim as stipulated in the Rules/ Notice for such bene-

fits, and these certificates should be dated earlier than the due date (closing date) of the application

2(b). A candidate may apply for admission to the examination in respect of any one or both the categories, mentioned in Para 2(a) above. Once an application has been made, no change will be allowed.

If a candidate wishes to be admitted for both the categories, he/she need fill in only one application. He/She will be required to pay the fee mentioned in para 4 below once only and will not be required to pay separate fee for each of the categories for which he/she applies.

A candidate may compete for any one or both the categories of Posts for which he/ she is eligible in terms of Rules. A candidate who qualifies for both the categories of Posts on the result of written part of examination will be required to indicate clearly in the Detailed Application Form the categories of Posts for which he/ she wishes to be considered in the order of preference so that having regard to his/ her rank in order of merit, due consideration can be given to his/her preference when making appointment.

N.B. (i): No request for addition/alteration in the preferences indicated by a candidate in his/her detailed application form will be entertained by the Commission.

N.B.(ii) The candidates competing for both the categories of the Posts will be allotted to Posts strictly in accordance with their merit position, preference exercised by them and number of vacancies.

3. ELIGIBILITY CONDITIONS :

(I) Nationality

A candidate must be either :

- (a) a Citizen of India, or
- (b) a subject of Nepal, or
- (c) a subject of Bhutan, or
- (d) a Tibetan refugee who came over to India before the 1st January, 1962 with the intention of permanently settling in India. or

(e) a person of Indian origin who has migrated from Pakistan, Burma, Sri Lanka or East African Countries of Kenya, Uganda, the United Republic of Tanzania, Zambia, Malawi, Zaire and Ethiopia or from Vietnam with the intention of permanently settling in India.

Provided that a candidate belonging to categories (b), (c), (d) and (e) above shall be a person in whose favour a certificate of eligibility has been issued by the Government of India.

A candidate in whose case a certificate of eligibility is necessary, may be admitted to the examination but the offer of appointment may be given only after the necessary eligibility certificate has been issued to him/her by the Government of India.

(II) Age Limits : A candidate for this examination must have attained the age criteria as prescribed below:-

(a) For Geologist and Geophysicist and Chemist (Group 'A') in the Geological Survey of India, an attached office of Ministry of Mines, Government of India: A candidate must have attained the age of 21 years and must not have attained the age of 32 years on the first day of the month of January of the year in which the Examination is to be held [i.e. as on 1st January, 2015] i.e. he/she must have been born not earlier than 2nd January, 1983 and not later than 1st January 1994.

Continued

(b) For Jr. Hydrogeologist (Scientist B) (Group A) in Central Ground Water Board, Ministry of Water Resources:- Must have attained the age of 21 years and must not have attained the age of 35 years on 1st January, 2015 i.e. he/she must have been born not earlier than 2nd January 1980 and not later than 1st January, 1994

NB: Candidates should ensure that they meet the age eligibility criteria for each post mentioned above for which they are applying.

(C) The upper age limit will be relaxable upto a maximum of 7 years in the case of Government servants, if they are employed in a Department mentioned in Column I below and apply for the corresponding post mentioned in column II.

| Column I | Column II |
|---|---|
| Geological Survey of India Central Ground Water Board | (i) Geologist Group 'A' (ii) Geophysicist Group 'A' (iii) Chemist Group 'A' |
| Central Ground Water Board | (i) Jr.Hydrogeologist, Gr. A |

d) The upper age limit prescribed above will be further relaxable as follows :

(i) Upto a maximum of five years if a candidate belongs to a Scheduled Caste or a Scheduled Tribe.

(ii) Upto a maximum of three years in the case of candidate belonging to Other Backward Classes who are eligible to avail of reservation applicable to such candidates.

(iii) Upto a maximum of five years if a candidate had ordinarily been domiciled in the State of Jammu & Kashmir during the period from the 1st January, 1980 to the 31st day of December, 1989.

(iv) Upto a maximum of three years in the case of Defence Services Personnel disabled in operations during hostilities with any foreign country or in a disturbed area and released as a consequence thereof:

(v) Upto a maximum of five years in the case of Ex-Servicemen including Commissioned Officers and ECOs/SSCOs who have rendered at least five years Military Service as on 1st January, 2015 and have been released (i) on completion of assignment (including those whose assignment is due to be completed within one year from 1st January, 2015) otherwise than by way of dismissal or discharge on account of misconduct or inefficiency, or (ii) on account of physical disability attributable to Military Service or (iii) on invalidment.

(vi) Upto a maximum of 5 years in the case of ECOs/SSCOs who have completed an initial period of assignment of 5 years Military Service as on 1st January, 2015 and whose assignment has been extended beyond 5 years and in whose case the Ministry of Defence issue a certificate that they can apply for civil employment and that they will be released on 3 month's notice on selection from the date of receipt of offer of appointment.

(vii) Upto a maximum of 10 years in the case of blind, deaf-mute and Orthopaedically handicapped persons.

NOTE I- Candidates belonging to the Scheduled Castes and the Scheduled Tribes and the Other Backward Classes who are also covered under any other clauses of Para 3(ii)(d) above, viz. those coming under the category of Ex-Servicemen, persons domiciled in the State of J & K, physically handicapped etc. will be eligible for grant of cumulative age-relaxation under both the categories.

NOTE II- The term Ex-Servicemen will apply to the persons who are defined as Ex-Servicemen in the Ex-servicemen (Re-employment in Civil Service and Posts) Rules, 1979, as amended from time to time.

NOTE III- The age concession under Para 3(ii)(d)(v) and (vi) will not be admissible to

Ex-servicemen and Commissioned Officers including ECOs/SSCOs, who are released on their own request.

NOTE IV- Notwithstanding the provision of age-relaxation under para 3(ii)(d)(vii) above, a physically handicapped candidate will be considered to be eligible for appointment only if he/she (after such physical examination as the Government or appointing authority, as the case may be, may prescribe) is found to satisfy the requirements of physical and medical standards for the concerned Services/posts to be allocated to the physically handicapped candidates by the Government.

Save as provided above the age limits prescribed can in no case be relaxed.

The date of birth accepted by the Commission is that entered in the Matriculation or Secondary School Leaving Certificate or in a certificate recognised by an Indian University as equivalent to Matriculation or in an extract from a Register of Matriculates maintained by a University which must be certified by the proper authority of the University or in the Higher Secondary or an equivalent examination certificate. These certificates are required to be submitted only after the declaration of the result of the written part of the examination.

No other document relating to age like horoscopes, affidavits, birth extracts from Municipal Corporation, service records and the like will be accepted. The expression Matriculation/Higher Secondary Examination Certificate in this part of the instruction includes the alternative certificate mentioned above.

NOTE I:- Candidates should note that only the Date of Birth as recorded in the Matriculation/Higher Secondary Examination Certificate or an equivalent certificate as on the date of submission of applications will be accepted by the Commission and no subsequent request for its change will be considered or granted.

NOTE II:- Candidates should also note that once a Date of Birth has been claimed by them and entered in the records of the Commission for the purpose of admission to an **Examination**, no change will be allowed subsequently (or at any other Examination of the Commission) on any ground whatsoever.

NOTE III :- The candidates should exercise due care while entering their date of birth in online application form for the Examination. If on verification at any subsequent stage, any variation is found in their date of birth from the one entered in their matriculation or equivalent Examination certificate, disciplinary action will be taken against them by the Commission under the Rules.

N.B. - (i) The candidature of a person who is admitted to the examination under the age concession mentioned in para 3(ii)(c) above, shall be cancelled, if after submitting his/her application, he/she resigns from service or his/her services are terminated by his/her department/office, either before or after taking the examination. He/she will, however, continue to be eligible if he/she is retrenched from the service or post after submitting the application.

(ii) A candidate who, after submitting his/her application to his/her department is transferred to other department/office will be eligible to compete under departmental age concession for the post(s), for which he/she would have been eligible, but for his/her transfer, provided his/her application, duly recommended, has been forwarded by his/her parent Department.

(iii) Minimum Educational Qualification

(i) For Geologists Gr 'A' in Geological Survey of India

(a) Master's degree in Geological Science or Geology or Applied Geology or Geo-Exploration or Mineral Exploration or Engineering Geology or Marine Geology or Earth Science and Resource Management or Oceanography and

Coastal Areas Studies or Petroleum Geosciences or Petroleum Exploration or Geochemistry or Geological Technology or Geophysical Technology from a University incorporated by an Act of the Central or State Legislature in India or an educational institution established by an Act of Parliament or declared to be deemed university under section 3 of the University Grants Commission Act, 1956 (3 of 1956)

(b) For Geophysicists Gr 'A' in Geological Survey of India

M.Sc. in Physics or Applied Physics or M.Sc. (Geophysics) or Integrated M.Sc. (Exploration Geophysics) or M.Sc. (Applied Geophysics) or M.Sc. (Marine Geophysics) or M.Sc. (Tech.) (Applied Geophysics) from a University incorporated by an Act of Parliament or State Legislature in India or other educational institutes established by an Act of the Parliament or declared to be deemed universities under the University Grants Commission Act, 1956.

(c) For Chemists Gr 'A' in Geological Survey of India

M. Sc. in Chemistry or Applied Chemistry or Analytical Chemistry from a University incorporated by an Act of Parliament or State Legislature or other educational Institutes established by an Act of the Parliament or declared to be deemed Universities under section 3 of the University Grants Commission Act, 1956 i.e. recognized University.

Note: Master's degree in respective discipline means the post graduate degree or post graduate diploma of minimum two years duration after graduation from aforesaid University or Institute in the respective discipline, whether be it an integrated course or otherwise.

Note II: Candidates who have appeared at the final examinations of the minimum educational qualification specified in clause (a) to (c) of this paragraph, passing of which shall render them eligible to appear at the relevant year Examination, but have not been informed of the result, may also apply for admission to the Examination and in such case his or her admission to the relevant year Examination shall be provisional and subject to obtaining minimum educational qualification specified in clause (a) to (c) of this paragraph.

Note III : A candidate who is otherwise qualified but who possesses the Master's degree specified in clause (a) to (c) of this paragraph from a foreign University approved by the Government may also be admitted to the Examination by the Commission.

(a) For Junior Hydrogeologists (Scientist B), Group 'A' in Central Ground Water Board.

(a) Master's degree in Geology or Applied Geology or Marine Geology from a University incorporated by an Act of the Central or State Legislature in India or other educational Institutes established by an act of Parliament or declared to be deemed as Universities under Section 3 of the University Grants Commission Act, 1956; or

(b) Master's degree in Hydrogeology from a recognised University.

Note : Candidate possessing common qualifications in (i) (a) & (ii) (a) can apply for both the categories.

NOTE I- A candidate who has appeared at an examination, the passing of which would render him/her educationally qualified for this examination but has not been informed of the result, may apply for admission to the examination. A candidate who intends to appear at such a qualifying examination may also apply. Such candidates will be admitted to the examination, if otherwise eligible but their admission would be deemed to be provisional and subject to cancellation if they do not produce proof of having passed the requisite qualifying examination along with the detailed application form which will be required to be submitted by the candidates who qualify on the result of the written part of the examination.

NOTE II- In exceptional cases the Commission may treat a candidate who does not possess any of the qualifications prescribed in this rule, as educationally qualified provided that he/she has passed examinations conducted by other institutions, the standard of which in the opinion of the Commission, justifies his/her admission to the examination.

NOTE III- A candidate who is otherwise eligible but who has taken a degree from a Foreign University recognised by Government may also apply to the Commission and may be admitted to the examination.

(IV) PHYSICAL STANDARDS :

Candidates must be physically fit according to physical standards for admission to the Examination, 2015 as per Regulations given in Appendix-II of the rules for the Combined Geo-scientist and Geologist Examination, 2015 in Gazette of India dated 21st February, 2015.

4. FEE :

Candidates (excepting Female/SC/ST/ PH who are exempted from payment of fee) are required to pay a fee of **Rs. 200/- (Rupees Two Hundred Only)** either by depositing the money in any Branch of SBI by cash, or by using net banking facility of SBI, State Bank of Bikaner & Jaipur/ State Bank of Hyderabad/State Bank of Mysore/ State Bank of Patiala/State Bank of Travancore or by using Visa/Master Credit/ Debit Card.

NB 1. Applicants who opt for "Pay by cash" mode should print the system generated Pay-in-slip during Part-II registration and deposit the fee at the counter of SBI Branch on the next working day only. "Pay by cash" mode option will be deactivated at 23.59 hours of 19.03.2015 i.e one day before the closing date: however, applicants who have generated their Pay-in-slip before it is de-activated may pay at the counter of SBI Branch during banking hours on the closing date. Such applicants who are unable to pay by cash on the closing date i.e. during banking hours at SBI Branch, for reason whatever, even if holding a valid Pay-in-slip, will have no other offline option but to opt for online Debit/Credit Card or Internet Banking Payment mode on the closing date i.e. till 25.59 hours of 20.03.2015

Note: 2.. Candidates should note that payment of examination fee can be made only through the modes prescribed above. Payment of fee through any other mode is neither valid nor acceptable. Applications submitted without the prescribed fee/mode (unless remission of fee is claimed) shall be summarily rejected.

Note: 3. Fee once paid shall not be refunded under any circumstances nor can the fee be held in reserve for any other examination or selection.

Note:4.. For the applicants in whose case payments details have not been received from the bank, they will be treated as fictitious payment cases and their applications will be rejected in the first instance. A list of all such applicants shall be made available on the Commission website within two weeks after the last day of submission of online application. The applicants shall be required to submit the proof of their fee payment within 10 days from the date of such communication either by hand or by speed post to the Commission. On receipt of documentary proof, genuine fee payment cases will be considered and their applications will be revived, if they are otherwise eligible.

All Female Candidates and candidates belonging to Scheduled Castes/ Scheduled Tribes are not required to pay any fee. No fee exemption is, however, available to OBC candidates and they are required to pay the prescribed full fee.

Physically handicapped persons are exempted from the payment of fee provided they are otherwise eligible for appointment to the Posts to be filled on the results of this examination on the basis of the standards of medical fitness for these Posts (including any concessions specifically extended to the physically handicapped). A physically handicapped candidate claiming age relaxation/fee concession will be required by the Commission to submit along with their detailed application form, a certified copy of the certificate from a Government Hospital/Medical Board in support of his/her claim for being physically handicapped.

NOTE : Notwithstanding the aforesaid provision for age relaxation/fee exemption, a physically handicapped candidate will be considered to be eligible for appointment only if he/she (after such physical examination as the Government or the appointing authority, as the case may be, may prescribe) is found to satisfy the requirements of physical and medical standards for the concerned Services/ Posts to be allocated to Physically Handicapped candidates by the Government.

Note : Applications without the prescribed Fee (Unless remission of Fee is claimed) shall be summarily rejected.

5. How to Apply:

(a) Candidates are required to apply Online using the link www.upsconline.nic.in. Detailed instructions for filling up Online Applications are available on the above mentioned website.

(b) The applicants are advised to submit only single application. However, if due to any unavoidable situation, he/she submits another/multiple applications, then he/she must ensure that application with the higher RID is complete in all respects like applicants details, examination centre, photograph, signature, fee etc. The applicants who are submitting multiple applications should note that only the applications with higher RID (Registration ID) shall be entertained by the Commission and fee paid against one RID shall not be adjusted against any other RID.

(c) All candidates, whether already in Government Service, in Government owned industrial undertakings or other similar organisations or in private employment should submit their applications direct to the Commission.

Persons already in Government service, whether in permanent or temporary capacity or as workcharged employees other than casual or daily rated employees or those serving under the Public Enterprises are, however, required to inform in writing their Head of Office/Department that they have applied for the Examination. Candidates should note that in case a communication is received from their employer by the Commission withholding permission to the candidates applying for/attending at the examination, their application will be liable to be rejected/candidature will be liable to be cancelled.

NOTE 1 : While filling in his online application form, the candidates should carefully decide about his/her choice for the centre for the examination.

If any candidate appears at a centre other than the one indicated by the Commission in the e-Admission Certificate, the papers of such a candidate will not be valued and his/her candidature will be liable to cancellation.

NOTE 2 : Incomplete or defective applications shall be summarily rejected. No representation or correspondence regarding such rejection shall be entertained under any circumstances.

(d) Candidates are not required to submit hard copy of their application to the commission at this stage. **The candidates applying for the examination should ensure that they fulfil all the eligibility conditions for admission to the Examination. Their admission at all the stages of examination for which they are admitted by the Commission viz. Written Examination and Interview Test will be purely provisional, subject to their satisfying the prescribed eligibility conditions. If on verification at any time before or after the written Examination or Interview Test, it is found that they do not fulfil any of the eligibility conditions, their candidature for the examination will be cancelled by the Commission.**

Candidates are requested to keep ready the attested copies of the following documents for submission to the Commission soon after the declaration of the result of the written part of the examination which is likely to be declared in the month of July/August, 2015.

1. Certificate of Age.
2. Certificate of Educational Qualification.
3. Certificate in support of claim to belong to Scheduled Caste, Scheduled Tribe and Other Backward Classes where applicable.
4. Certificate in support of claim for age/fee concession, wherever applicable.
5. Certificate in support of being physically handicapped (wherever applicable).

Immediately after the declaration of the result of the written part of the examination successful candidates will be intimated by the Commission electronically and they shall be asked to submit Detailed Application Form (DAF) online. The successful candidates have to send the printout of this DAF with each page duly signed along with the attested copies of the above mentioned certificates to the

Commission at that time. Originals will have to be produced at the time of interview. The interview letter to the candidates may also be issued electronically.

If any of their claims is found to be incorrect, they may render themselves liable to disciplinary action by the Commission in terms of Rule 14 of the combined Geo-Scientist and Geologist Examination, 2015 notified in the Gazette of India dated 21st February, 2015 and also reproduced below:

A candidate who is or has been declared by the Commission to be guilty of:

- (i) obtaining support for his/her candidature by any means, or
- (ii) impersonating, or
- (iii) procuring impersonation by any person, or
- (iv) submitting fabricated documents or documents which have been tampered with, or
- (v) making statements which are incorrect or false or suppressing material information, or
- (vi) resorting to any other irregular or improper means in connection with his/her candidature for the examination, or
- (vii) using unfair means during the examination, or
- (viii) writing irrelevant matter, including obscene languages or pornographic matter, in the script(s), or
- (ix) misbehaving in any other manner in the examination hall, or
- (x) harassing or doing bodily harm to the Staff employed by the Commission for the conduct of their examination, or

(xi) being in possession of or using mobile phone, pager or any electronic equipment or device or any other equipment capable of being used as a communication device during the examination; or

(xii) violating any of the instructions issued to candidates along with their e-Admission Certificates permitting them to take the examination, or

(xiii) attempting to commit or as the case may be abetting the commission of all or any of the acts specified in the foregoing clauses, may in addition to rendering himself/herself liable to criminal prosecution is liable.

(a) to be disqualified by the Commission from the examination for which he/she is a candidate and/or

(b) to be debarred either permanently or for a specified period

(i) by the Commission from any examination or selection held by them;

(ii) by the Central Government from any employment under them; and

(c) if he/she is already in service under Government to disciplinary action under the appropriate rules.

Provided that no penalty under these rules shall be imposed except after.

(i) giving the candidate an opportunity of making such representation, in writing as he/she may wish to make in that behalf; and

(ii) taking the representation, if any, submitted by the candidate within the period allowed to him/her into consideration.

6. LAST DATE FOR SUBMISSION OF APPLICATIONS :

The Online Applications can be filled upto **20th March, 2015 till 11.59 PM** after which the link will be disabled.

7. CORRESPONDENCE WITH COMMISSION:

The Commission will not enter into any correspondence with the candidates about their candidature except in the following cases:

(i) The eligible candidates shall be issued an e-Admission Certificate three weeks before the commencement of the examination. The e-Admission Certificate will be made available in the UPSC web-site [www.upsc.gov.in] for downloading by candidates. No Admission Certificate will be sent by post. For downloading the e-Admission Certificate/e-Admit Card the candidate must have his/her vital parameters like RID & Date of Birth or Roll No. (if

received) & date of birth or name, father's name & Date of Birth available with him/her. If a candidate does not receive his/her e-admission certificate or any other communication regarding his/her candidature for the examination two weeks before the commencement of the examination, he/she should at once contact the Commission. Information in this regard can also be obtained from the **Facilitation Counter** located in the Commission's Office either in person or over phone Nos. 011-23381125/23385271/23098543. In case no communication is received in the Commission's office from the candidate regarding non-receipt of his/her e-admission certificate at least 2 weeks before the examination, he/she himself/herself will be solely responsible for non-receipt of his/her e-admission certificate.

No candidate will ordinarily be allowed to take the examination unless he/she holds a e-admission certificate for the examination. On the receipt of e-Admission Certificate, check it carefully and bring discrepancies/errors, if any, to the notice of UPSC immediately.

The candidates should note that their admission to the examination will be purely provisional based on the information given by them in the application form. This will be subject to verification of all the eligibility conditions by the UPSC.

(ii) **The mere fact that an e-admission certificate to the Examination has been issued to a candidate, will not imply that the candidature has been finally cleared by the Commission or that entries made by the candidate in his/her application for the examination have been accepted by the Commission as true and correct. Candidates may note that the Commission takes up the verification of eligibility conditions of a candidate, with reference to original documents, only after the candidate has qualified for interview or Personality Test on the result of the Written Examination. Unless candidature is formally confirmed by the Commission, it continues to be provisional.** The decision of the Commission with regard to the acceptance of the application of a candidate and his/her eligibility or otherwise for admission to the Examination shall be final. Candidates should note that the name in the e-Admission Certificate in some cases, may be abbreviated due to technical reasons.

(iii) The candidates must ensure that their E-Mail IDs given in their Online Applications are valid and active as the Commission may use electronic mode of communication while contacting them at different stages of examination processes.

(iv) A candidate must see that communications sent to him/her at the address stated in his/her application are redirected, if necessary. Change in address should be communicated to the Commission at the earliest opportunity. Although the Commission make every effort to take account of such changes, they cannot accept any responsibility in the matter.

(v) In the event of a candidate downloading more than one admission certificate from the Commission's website he/she should use only one of these admission certificates for appearing in the examination and report about the other(s) to the Commission's Office.

(vi) If a candidate receives an e-admission certificate in respect of some other candidate from the Commission the same should be immediately brought to the notice of the Commission with a request to issue the correct e-Admission Certificate. Candidates may note that they will not be allowed to take the examination on the strength of an e-admission certificate issued in respect of another candidate.

Important : All communications to the

Commission should invariably contain the following particulars.

1. Name and year of examination.
2. Registration I.D. (RID) Number
3. Roll No. (if received)
4. Name of candidate in full and in Block Letters.
5. Complete Postal Address as given in the Application.
6. Valid and active E-mail I.D.

N.B. : (i) Communications not containing the above particulars may not be attended to.

N.B. (ii) If a letter/communication is received from a candidate after an examination has been held and it does not give his/her full name and Roll number, it will be ignored and no action will be taken thereon.

N.B. (iii) Candidates are strongly advised to keep a printout or soft copy of their online application for future references.

8. For being considered against the vacancies reserved for them, the physically disabled person should have disability of Forty per cent (40%) or more. However, such candidates shall be required to meet one or more of the following physical requirements/abilities which may be necessary for performing the duties in the concerned Services/Posts:-

| CODE | PHYSICAL REQUIREMENTS |
|------|--|
| F | 1. Work performed by manipulating (with Fingers) |
| PP | 2. Work performed by pulling and pushing |
| L | 3. Work performed by lifting |
| KC | 4. Work performed by kneeling and crouching |
| B | 5. Work performed by bending |
| S | 6. Work performed by sitting (on bench or chair) |
| ST | 7. Work performed by standing |
| W | 8. Work performed by walking |
| SE | 9. Work performed by seeing |
| H | 10. Work performed by hearing/speaking |
| RW | 11. Work performed by reading and writing |

The functional classification in their case shall be, one or more of the following, consistent with the requirements of the concerned Services/Posts:-

FUNCTIONAL CLASSIFICATION

| CODE | FUNCTIONS |
|------|---|
| BL | 1. both legs affected but not arms. |
| BA | 2. both arms affected <ul style="list-style-type: none"> a. impaired reach. 'b. weakness of grip. |
| BLA | 3. both legs and both arms affected |
| OL | 4. one leg affected (R or L) <ul style="list-style-type: none"> a. impaired reach. b. weakness of grip. c. ataxic. |
| OA | 5. one arm affected (R or L) <ul style="list-style-type: none"> a. impaired reach. b. weakness of grip. c. ataxic. |
| BH | 6. stiff back and hips (cannot sit or stoop) |
| MW | 7. muscular weakness and limited physical endurance |
| B | 8. the blind |
| PB | 9. partially blind |
| D | 10. the deaf |
| PD | 11. partially deaf |

9. WITHDRAWAL OF APPLICATIONS :

No request for withdrawal of candidature received from a candidate after he/she has submitted his/her application will be entertained under any circumstances.

10. Details about the scheme of examination, standard and syllabi of the subjects etc. may be seen in Appendix-I of this Notice.

**(G. SARAVANAN)
UNDER SECRETARY**

UNION PUBLIC SERVICE COMMISSION

APPENDIX-I

SCHEME OF EXAMINATION

1. The examination shall be conducted according to the following Plan :

Part I. - Written examination in the subjects as set out in para 2 below.

Part II. - Interview/ Personality Test of such candidates as may be called by the Commission, carrying a maximum of 200 marks.

2. The following will be the subjects for the written examination, for selection to the post of Geologist, Geophysicist and Chemist Group 'A' in GSI and Junior Hydro-geologist (Scientist 'B'), Group 'A' for Central Ground Water Board.

| Subject | Duration | Maximum Marks |
|--|----------|---------------|
| (1) | (2) | (3) |
| 1. General English (common for all category) | 3 hrs | 100 |
| Category - 1 For the post of Geologists | | |
| 2. Geology Paper I: | 3 hrs | 200 |
| 3. Geology Paper II: | 3 hrs | 200 |
| 4. Geology Paper III: | 3 hrs | 200 |
| Category - 2 For the post of Geophysicists | | |
| 2. Geophysics Paper I: | 3 hrs | 200 |
| 3. Geophysics Paper II: | 3 hrs | 200 |
| 4. Geophysics Paper III: | 3 hrs | 200 |
| Category - 3 For the post of Chemists | | |
| 2. Chemistry Paper I: | 3 hrs | 200 |
| 3. Chemistry Paper II: | 3 hrs | 200 |
| 4. Chemistry Paper III: | 3 hrs | 200 |
| Category - 4 For the Post of Jr. Hydrogeologist | | |
| 2. Geology Paper I | 3 hrs | 200 |
| 3. Geology Paper II | 3 hrs | 200 |
| 4. Hydrogeology | 3 hrs | 200 |

Note I: Candidates competing for selection to the posts of Geologist, Geophysicist, Chemist and Junior Hydrogeologist will be required to appear in all the subjects mentioned against respective category above.

Note II: Candidates competing for selection for both the posts of Geologist and Jr. Hydrogeologist will be required to appear in all the subjects mentioned against Categories 1 and 4 above.

Note III : If any candidate failed to appear in any one or more of above papers, meant for written examination for selection to the post of Geologist, Geophysicist, Chemist, their candidature shall stand rejected and part of written examination appeared by him/her shall not be evaluated and counted for any purpose.

3. THE EXAMINATION IN ALL THE SUBJECTS WILL BE OF CONVENTIONAL (ESSAY) TYPE.

4. All Question papers must be answered in English. The Question Papers will be set in English only.

5. The standard and syllabus of the examination will be as shown in the Schedule.

6. Candidates must write the papers in their own hand. In no circumstances will they be allowed the help of a scribe to write answers for them.

7. The Commission have discretion to fix qualifying marks in any or all the subjects of the examination.

8. If a candidate's handwriting is not easily legible, deduction will be made on this account from the total marks otherwise accruing to him/her.

9. Marks will not be allotted for mere superficial knowledge.

10. Credit will be given for orderly, effective and exact expression combined with due economy of words in all subjects of the examination.

11. In the question papers wherever necessary, questions involving the Metric System of Weights and Measures only will be set.

12. Candidates should use only International form of Indian numerals (e.g. 1, 2, 3, 4, 5, 6 etc) while answering question papers.

13. Candidates are permitted to bring and use battery operated pocket calculators for answering papers in this examination. Loaning or inter-changing of calculators in the Examination Hall is not permitted.

14. Interview / Personality Test : The candidate will be interviewed by a Board of competent and unbiased observers who will have before them a record of his/her career. The object of the interview is to assess his/her suitability for the posts for which he/she has competed. Special attention will be paid in the Personality Test to assessing the candidate's capacity for leadership, initiative and intellectual curiosity, tact and other social qualities, mental and physical energy, powers of practical application, integrity of character and aptitude for adapting themselves to the field life.

SCHEDULE STANDARD AND SYLLABUS

A paper in General English is compulsory and common for all the four categories and its standard will be such as may be expected of a science graduate. 3 compulsory papers each on Geology, Geophysics, Chemistry and Hydrogeology subjects will be approximately of the M.Sc. degree standard of an Indian University and questions will generally be set to test the candidate's grasp of the fundamentals in each subject.

There will be no practical examination in any of the subjects.

(1) GENERAL ENGLISH : 100 Marks

Candidate will be required to write a short Essay in English. Other questions will be designed to test their understanding of English and workmanlike use of words.

(2) GEOLOGY - PAPER I : 200 Marks

Section A : Geomorphology and Remote Sensing.

Basic principles. Weathering and soils, Mass wasting. Influence of climate on processes. Concept of erosion cycles. Geomorphology of fluvial tracts, arid zones, coastal regions, 'Karst' landscapes and glaciated ranges. Geomorphic mapping, slope analysis and drainage basin analysis. Applications of geomorphology in mineral prospecting, civil engineering, hydrology and environmental studies. Topographical maps. Geomorphology of India.

Concepts and principles of aerial photography and photogrammetry, satellite remote sensing — data products and their interpretation. Digital image processing. Remote-sensing in landform and land use mapping, structural mapping, hydrogeological studies and mineral exploration. Global and Indian Space Missions. Geographic Information System (GIS) — principles and applications.

Section B : Structural Geology

Principles of geological mapping and map reading, projection diagrams. Stress-strain relationships of elastic, plastic and viscous materials. Measurement of strain in deformed rocks. Behaviour of minerals and rocks under deformation conditions. Structural analysis of folds, cleavages, lineations, joints and faults. Superposed deformation. Mechanism of folding and faulting. Time-relationship between crystallization and deformation. Unconformities and basement-cover relations. Structural behaviour of igneous rocks, diapirs and salt domes. Introduction to petrofabrics.

Section C : Geotectonics

Earth and the solar system, Meteorites and other extra-terrestrial materials, Planetary evolution of the earth and its internal structure. Heterogeneity of the earth's crust. Major tectonic features of the Oceanic and Continental crust. Continental drift — geological and geophysical evidence, mechanics, objections, present status. Gravity and magnetic anomalies at Mid-ocean ridges, deep sea trenches, continental shield areas and mountain chains. Palaeomagnetism. Seafloor spreading and Plate Tectonics. Island arcs, Oceanic islands and volcanic arcs. Isostasy, orogeny and epeirogeny. Seismic belts of the earth. Seismicity and plate movements. Geodynamics of the Indian plate.

Section D : Stratigraphy

Nomenclature and the modern stratigraphic code. Radioisotopes and measuring geological time. Geological time-scale. Stratigraphic procedures of correlation of unfossiliferous rocks. Precambrian stratigraphy of India. Stratigraphy of the Palaeozoic, Mesozoic and Cenozoic formations of India. Gondwana system and Gondwanaland. Rise of the Himalaya and evolution of Siwalik basin. Deccan Volcanics. Quaternary Stratigraphy. Rock record, palaeoclimates and palaeogeography.

Section E : Palaeontology

Fossil record and geological time-scale. Morphology and time-ranges of fossil groups. Evolutionary changes in molluscs and mammals in geological time. Principles of evolution. Use of species and genera of foraminifera and echinodermata in biostratigraphic correlation. Siwalik vertebrate fauna and Gondwana flora, evidence of life in Precambrian times, different microfossil groups and their distribution in India.

(3) GEOLOGY - PAPER II : 200 Marks

Section A : Mineralogy

Physical, chemical and crystallographic characteristics of common rock forming silicate mineral groups. Structural classification of silicates. Common minerals of igneous and metamorphic rocks. Minerals of the carbonate, phosphate, sulphide and halide groups.

Optical properties of common rock forming silicate minerals, uniaxial and biaxial minerals. Extinction angles, pleochroism, birefringence of minerals and their relation with mineral composition. Twinned crystals. Dispersion. Application of universal stage (4 axes/5 axes) in determining optical properties.

Section B : Igneous and Metamorphic Petrology

Forms, textures and structures of igneous rocks. Silicate melt equilibria, binary and ternary phase diagrams. Petrology and geotectonic evolution of granites, basalts, andesites and alkaline rocks. Petrology of gabbros, kimberlites, anorthosites and carbonatites. Origin of primary basic magmas.

Textures and structures of metamorphic rocks. Regional and contact metamorphism of pelitic and impure calcareous rocks. Mineral assemblages equilibrium/Reaction textures and geo-thermo barometry. Experimental and thermodynamic appraisal of metamorphic reactions. Characteristics of different grades and facies of metamorphism. Metasomatism and granitization, migmatites. Plate tectonics and metamorphic zones. Paired metamorphic belts.

Section C : Sedimentology

Provenance and diagenesis of sediments. Sedimentary textures. Framework matrix and cement of terrigenous sediments. Definition, measurement and interpretation of grain size. Elements of hydraulics. Primary structures, palaeocurrent analysis. Biogenic and chemical sedimentary structures. Sedimentary environment and facies. Facies modelling for marine, non-marine and mixed sediments. Tectonics and sedimentation. Classification and definition of sedimentary basins, Sedimentary basins of India. Cyclic sediments. Seismic and sequence stratigraphy. Purpose and scope of basin analysis. Structure contours and isopach maps.

Section D : Geochemistry

Earth in relation to the solar system and universe, cosmic abundance of elements. Composition of the planets and meteorites. Structure and composition of earth and distribution of elements. Trace elements. Elementary crystal chemistry and thermodynamics. Introduction to isotope geochemistry. Geochemistry of hydrosphere, biosphere and atmosphere. Geochemical cycle and principles of geochemical prospecting.

Section E : Environmental Geology

Concepts and principles. Natural hazards — preventive/precautionary measures — floods, landslides, earthquakes, river and coastal erosion. Impact assessment of anthropogenic activities such as urbanization, open cast mining and quarrying, river-valley projects, disposal of industrial and radio-active waste, excess withdrawal of ground water, use of fertilizers, dumping of ores, mine waste and fly-ash. Organic and inorganic contamination of ground water and their remedial measures. Soil degradation and remedial measures. Environment protection — legislative measures in India.

(4) GEOLOGY - PAPER III : 200 Marks

Section A : Indian mineral deposits and mineral economics

Occurrence and distribution in India of metalliferous deposits — base metals, iron, manganese, aluminium, chromium, nickel, gold, silver, molybdenum. Indian deposits of non-metals — mica, asbestos, barytes, gypsum, graphite, apatite and beryl. Gemstones, refractory minerals, abrasives and minerals used in glass, fertilizer, paint, ceramic and cement industries. Building stones. Phosphorite deposits. Placer deposits, rare earth minerals.

Strategic, critical and essential minerals. India's status in mineral production. Changing patterns of mineral consumption. National Mineral Policy. Mineral Concession Rules. Marine mineral resources and Law of Sea.

Section B : Ore genesis

Ore deposits and ore minerals. Magmatic processes of mineralisation. Porphyry, skarn and hydrothermal mineralisation. Fluid inclusion studies. Mineralisation associated with — (i) ultramafic, mafic and acidic rocks, (ii) greenstone belts, (iii) komatiites, anorthosites and kimberlites and (iv) submarine volcanism. Magmatic-related mineralisation through geological time. Stratiform and stratabound ores. Ores and metamorphism — cause and effect relations.

Section C : Mineral exploration

Methods of surface and subsurface exploration, prospecting for economic minerals — drilling, sampling and assaying. Geophysical techniques — gravity, electrical, magnetic, airborne and seismic. Geomorphological and remote sensing techniques. Geobotanical and geochemical methods. Borehole logging and surveys for deviation.

Section D : Geology of fuels

Definition, origin of coal. Stratigraphy of coal measures. Fundamentals of coal petrology, peat, lignite, bituminous and anthracite coal. Microscopic constituents of coal. Industrial application of coal petrology. Indian coal deposits. Diagenesis of organic materials.

Origin, migration and entrapment of natural hydrocarbons. Characters of source and reservoir rocks. Structural, stratigraphic and mixed traps. Techniques of exploration. Geographical and geological distributions of onshore and offshore petroliferous basins of India.

Mineralogy and geochemistry of radioactive minerals. Instrumental techniques of detection and measurement of radioactivity. Radioactive methods for prospecting and assaying of mineral deposits. Distribution of radioactive minerals in India. Radioactive methods in petroleum exploration — well logging techniques. Nuclear waste disposal — geological constraints.

Section E : Engineering Geology

Mechanical properties of rocks and soils. Geological investigations for river valley projects — Dams and reservoirs; tunnels — types, methods and problems. Bridges — types and foundation problems. Shoreline engineering. Landslides — classification, causes, prevention and rehabilitation. Concrete aggregates — sources, alkali-aggregate reaction. Aseismic designing — seismicity in India and earthquake-resistant structures. Problems of groundwater in engineering projects. Geotechnical case studies of major projects in India.

(2) GEOPHYSICS - PAPER I: 200 Marks

PART-A: 100 Marks

a. Solid Earth Geophysics:

Introduction to Geophysics its branches and relationship with other sciences. Solar system, its origin, characteristics of planetary members, Earth; its rotation and figure. Age of earth & various methods of determination. Tectonics and Geodynamics, Thermal history and its characteristics. Gravity field of earth and Isostasy. Geomagnetism, elements of earth's magnetism: Internal, External fields and their causes, Paleomagnetism, Polar wandering paths, Seafloor spreading, geophysical evidences. Elastic waves, internal structure of earth, variation of physical properties in the interior of earth.

b. Earthquake and Engineering Seismology:

Seismology, earthquakes, focal depth, epicenter, great Indian earthquakes, Intensity and Magnitude scales, Energy of earthquakes, foreshocks, aftershocks, Elastic rebound theory, Fault plane solutions, Seismicity and Seismotectonics of India, Frequency-Magnitude relation (b values), Velocity structure, Vp/Vs studies. Elastic waves, their propagation characteristics. Seismic ray theory for spherically and horizontally stratified earth, basic principles of Seismic Tomography and receiver function analysis, Seismic network and arrays, telemetry systems, Earthquake prediction; dilatancy theory, short-term, middle-term and long- term predictions, Seismic microzonation studies, application for engineering problems, Seismometry, Principle of electromagnetic seismograph, displacement meters, velocity meter, accelerometer, WWSSN stations, Strong motion seismograph, seismic arrays for detection of nuclear explosions, Broadband seismometry.

c. Mathematical methods in Geophysics:

Properties of scalars, vectors and tensors, Elements of vector analysis, Gradient, Divergence and Curl, Gauss's divergence theorem, Stokes theorem, Definition of fields, Gravitational field, Newton's Law of gravitation, Gravitation potential and fields due to bodies of different geometric shapes, Electrostatic field, Coulomb's law, Electrical permittivity and dielectric constant, Basic guiding equations, Magneto static field, Origin of Magnetic field, Ampere's law, Biot and Savart's law, Geomagnetic fields, Magnetic fields due to different type of structures, Solution of Laplace equation in Cartesian Coordinate, Cylindrical Polar Coordinate and Spherical Polar Coordinate, Complex Variables in Potential theory, Green's theorem in Potential Theory. Concept of Image in Potential Theory, Analytical continuation in Potential fields, Numerical Methods in Potential Theory. Electrical fields in geophysics, point source, continuous distribution and double layers, equipotential and line of force. Current and potential in the earth, basic concept and equations of electromagnetic, Maxwell's equations, boundary conditions, elliptic polarization, electromagnetic potential and waves, radiation from dipoles, retarded potential, near and far fields, radiation resistance, EM field of a loops of wire on half space, multi-layered media, impedance and its application.

d. Geophysical Inversion:

Fundamental concepts of inverse theory, Basic definition of inversions with application to Geophysics. Probability, Inverses with discrete and continuous models. Forward problems versus Inverse problems. Formulation of inverse problems and their relation to a matrix problem, linear inverse problems, classification of inverse problems, least square solutions and minimum norm solution, concept of norms, concept of 'a priori' information, constrained linear least square inversion, review of matrix theory Introduction to finite difference method, forward, backward and central difference method. Application of finite difference method for solving Helmholtz equation. Introduction to finite element method, various steps, simple examples showing application of finite element method. Models and data spaces, householder transformation, data resolution matrix, model resolution matrix, Eigen values and Eigen

vectors, singular value decomposition (SVD), generalized inverses, Non-linear inverse problems, Gauss Newton method, steepest descent (gradient) method, Marquardt-Levenberg method, Earthquake location problem, tomography problem. Probabilistic approach of inverse problems, maximum likelihood and stochastic inverse methods, Backus-Gilbert method, Global optimization techniques, genetic algorithm, simulated annealing methods, examples of inverting geophysical data.

PART-B: 100 Marks

a. Mathematical Methods of Physics:

Dimensional analysis, Vector algebra and vector calculus, Linear algebra, matrices, Cayley-Hamilton Theorem. Eigen values and eigenvectors. Linear ordinary differential equations of first & second order, Special functions (Hermite, Bessel, Laguerre and Legendre functions). Fourier series, Fourier and Laplace transforms. Elements of complex analysis, analytic functions; Taylor & Laurent series; poles, residues and evaluation of integrals. Elementary probability theory, random variables, binomial, Poisson and normal distributions. Central limit theorem. Green's function. Partial differential equations (Laplace, wave and heat equations in two and three dimensions). Elements of computational techniques: root of functions, interpolation, and extrapolation, integration by trapezoid and Simpson's rule, solution of first order differential equation using Runge-Kutta method. Finite difference methods. Tensors. Introductory group theory: SU (2), O (3). Complex Variables, Beta, Gamma functions and special functions, Laplace Transform & Fourier series, Fourier Transforms, Introductory methods of Numerical analysis. Operators and their properties.

b. Thermodynamics and Statistical Physics:

Laws of thermodynamics and their consequences; Thermodynamic potentials, Maxwell relations; Chemical potential, phase equilibria; Phase space, micro- and macrostates; Micro canonical, canonical and grand-canonical ensembles and partition functions; Free Energy and connection with thermodynamic quantities; First- and second-order phase transitions; Classical and quantum statistics, ideal Fermi and Bose gases; Principle of detailed balance; Blackbody radiation and Planck's distribution law; Bose-Einstein condensation; Random walk and Brownian motion; Introduction to non equilibrium processes; Diffusion equation.

c. Electrodynamics:

Gauss Theorem, Poisson's equation, Laplace's equation, solution to Laplace's equation in Cartesian coordinates, spherical, cylindrical coordinates, use of Laplace's equation in the solutions of electrostatic problems. Ampere's circuital law, magnetic vector potential, displacement current, Faraday's law of electromagnetic induction. Maxwell's equations, differential and integral forms, physical significance of Maxwell's equations. Wave equation, plane electromagnetic waves in free space, in non conducting isotropic medium, in conducting medium, electromagnetic vector and scalar potentials, uniqueness of electromagnetic potentials and concept of gauge, Lorentz gauge, Columb gauge, charged particles in electric and magnetic fields, charged particles in uniform electric field, charged particle in homogeneous magnetic fields, charged particles in simultaneous electric and magnetic fields, charged particles in non homogeneous magnetic fields. Lienard - Wiechert potentials, electromagnetic fields from Lienard - Wiechert potentials of a moving charge, electromagnetic fields of a uniformly moving charge, radiation due to non-relativistic charges, radiation damping, Abrahama-Lorentz formula, Cherenkov radiation, radiation due to oscillatory electric dipole, radiation due to small current element. Condition for plasma existence, occurrence of plasma, magneto hydrodynamics, plasma waves. Transformation of electromagnetic potentials, Lorentz condition in covariant form, invariance or covariance of Maxwell field equations in terms of 4 vectors, electromagnetic field tensor, Lorentz transformation of electric and magnetic fields.

d. Introductory Atmospheric and Space Physics:

The Neutral atmosphere, atmospheric nomenclature, the Hydrostatic equation, geopotential height, expansion and contraction, fundamental forces in the atmosphere, apparent forces, atmospheric composition, solar radiation interaction with the neutral atmosphere, climate change. Electromagnetic radiation and propagation of Waves: EM Radiation, fundamentals of EM waves, effects of environment, Antennas-basic considerations, types of antennas. Propagation of Waves: ground wave, sky wave, and space wave propagation, troposcatter communication and extra terrestrial communication. The Ionosphere, morphology of ionosphere, the D, E and F-regions, chemistry of the ionosphere, ionospheric parameters, E and F region anomalies and irregularities in the ionosphere. Global Positioning Systems (GPS)-basic concepts, overview of GPS system, augmentation services, GPS system segment, GPS signal characteristics, GPS errors, multi path effects, GPS performance, satellite navigation system and applications.

(3) GEOPHYSICS - PAPER II : 200 Marks

PART-A: 100 Marks

a. Geophysical Potential Fields (Gravity and Magnetic)

Geophysical potential fields, Inverse square law of field, Principles of Gravity and Magnetic methods, Geoid, Spheroid, Nature of gravity and its variation, Properties of Newtonian potential, Laplace's and Poisons equations, Green's theorem, Gauss law, Concept of Bouguer gravity anomaly, Rock densities, factors controlling rock densities, determination of density, theory of isostasy, Earth's main magnetic field, origin, temporal variations, Geomagnetic elements, Columb's law of magnetic force and fields, intensity of magnetization and induction, magnetic potential and its relation to field, units of measurement, origin of magnetic anomalies, interrelationship between different components of anomalies, Poisson's relation, Magnetic susceptibility, factors controlling susceptibility (Bulk chemistry, cooling history, metamorphism..), magnetic minerals, rock classification, Natural and remnant magnetism, Asiatic and Spinner magnetometers, demagnetization effects. Principles of Gravity and Magnetic instruments, Plan of conducting GM surveys, reduction of gravity and magnetic data, Airborne magnetic surveys and magnetic gradient surveys, Shipborne surveys, Gravity and Magnetic data reduction, IGSN Gravity bases, International Gravity formula, IGRF corrections for magnetic field. Separation of regional and residual anomalies, ambiguity in interpretation, Application of GM surveys for Geodynamic studies, Mineral Exploration, Environmental studies...Data processing and interpretation of anomalies, modeling of anomalies.

b. Electrical and Electromagnetic methods:

Electrical properties of rocks and their measurement, concepts and assumptions of horizontally stratified earth, anisotropy and its effects on electrical fields, the geo electric section and geological section, D.C Resistivity method, fundamental laws, concept on natural electric field, electrode configuration, choice of methods, Profiling, Vertical Electrical Sounding, SP Method, Origin of SP, application of SP surveys, Origin of Induced Polarization, Membrane and Electrode potential, time and frequency domains of measurement, IP, chargeability, percent frequency effect and metal factor, dipole theory of IP, Application of IP surveys for mineral exploration (disseminated sulphides). Electromagnetic methods/ Telluric/Magneto Telluric methods, Passive and Active source methods, Maxwell's equations, electromagnetic potential and wave equations, boundary conditions, long wave length approximation, depth of penetration, amplitude and phase relations, real and imaginary components, Principles of EM prospecting, various EM methods, Dip angle method, Turam method, moving source-receiver methods-horizontal loop (Slingram) method, AFMAG, and VLF methods, Airborne EM systems - rotary field method, INPUT method, EM Profiling and sounding, Interpretation of EM anomalies, Principles of Ground Penetrating Radar (GPR), Origin and characteristics of MT fields, Instrumentation, Field methods and interpretation of MT data and applications.

c. Seismic Prospecting:

Basic principles of seismic methods, Fermat's principle, Snell's law, Reflection, refraction and diffraction from multilayered medium, Reflection and transmission coefficients, propagation model for exploration seismology, Seismic resolution, Seismic absorption and anisotropy, Seismic data acquisition, sources of energy, Geophones, geometry of arrays, Instrumentation, digital recording Seismic Surveys: Principle for multilayer refraction Travel time curves, corrections, Interpretation of data, Reflection principles, CDP, data processing, corrections, NMO correction, Interpretation of data, Fundamental of VSP method, Seismic Tomography. Principles of High Resolution Seismic (HRS) for coal exploration

d. Borehole Geophysics (Principles of Well logging):

Objectives of well logging, fundamental concepts in borehole geophysics, borehole conditions, properties of reservoir rock formations, formation parameters and their relationships-formation factor, porosity, permeability, formation water resistivity, water saturation, irreducible water saturation, hydrocarbon saturation, residual hydrocarbon saturation; Archie's and Humble's equations; principles, instrumentations, operational procedures and interpretations of various geophysical logs, SP log, resistivity and micro resistivity logs, nuclear/radioactive logs, acoustic impedance and propagation logs, temperature log, caliper log and directional logs; production logging; clean sand and shaly sand interpretations; overlay and cross-plots of well-log data, determination of formation lithology, sub-surface correlation and mapping, delineation of fractures; application of well-logging in hydrocarbon, groundwater, coal, metallic and non-metallic mineral exploration.

PART-B: 100 Marks**a. Atomic and Molecular Physics and Properties and Characterization of materials:**

Quantum states of an electron in an atom; Electron spin; Stern-Gerlach experiment; Spectrum of Hydrogen, helium and alkali atoms; Relativistic corrections for energy levels of hydrogen; Hyperfine structure and isotopic shift; width of spectral lines; LS & JJ coupling; Zeeman, Paschen Back & Stark effect; X-ray spectroscopy; Electron spin resonance, Nuclear magnetic resonance, chemical shift; Rotational, vibrational, electronic, and Raman spectra of diatomic molecules; Frank - Condon principle and selection rules; Spontaneous and stimulated emission, Einstein A & B coefficients; Lasers, optical pumping, population inversion, rate equation; Modes of resonators and coherence length. Thermal properties, optical properties, fundamentals of transmission electron microscopy, study of crystal structure using TEM, study of microstructure using SEM. Resonance methods- Spin and an applied field- the nature of spinning particles, interaction between spin and a magnetic field, population on energy levels, the Larmor precession, relaxation times - spin-spin relation, spin-lattice relaxation, Electron spin resonance- Introduction, g factor, experimental methods, Nuclear Magnetic resonance- equations of motion, line width motional narrowing, hyperfine splitting, Nuclear Gamma Resonance: Principles of Mossbauer Spectroscopy, Line width, Resonance absorption, Mossbauer Spectrometer, Isomer Shift, Quadrupole splitting, magnetic field effects, applications.

b. Nuclear and Particle Physics:

Basic nuclear properties: size, shape, charge distribution, spin and parity; Binding energy, semi-empirical mass formula; Liquid drop model; Fission and fusion; Nature of the nuclear force, form of nucleon-nucleon potential; Charge-independence and charge-symmetry of nuclear forces; Isospin; Deuteron problem; Evidence of shell structure, single-particle shell model, its validity and limitations; Rotational spectra; Elementary ideas of alpha, beta and gamma decays and their selection rules; Nuclear reactions, reaction mechanisms, compound nuclei and direct reactions; Classification of fundamental forces; Elementary particles (quarks, baryons, mesons, leptons); Spin and parity assignments, isospin, strangeness; Gell-Mann-Nishijima formula; C, P, and T invariance and applications of symmetry arguments to particle reactions, parity non-conservation in weak interaction; Relativistic kinematics.

Crystalline and amorphous structure of matter; Different crystal systems, space groups; methods of determination of crystal structure; X-ray diffraction, scanning and transmission electron microscopes; Band theory of solids-conductors, insulators and semiconductors; Thermal properties of solids, specific heat, Debye theory; Magnetism: dia, para and ferromagnetism; elements of superconductivity; Meissner effect, Josephson junctions and applications; elementary ideas about high temperature superconductivity.

c. Electromagnetic Theory:

Electrostatics: Gauss' Law and its applications; Laplace and Poisson equations, boundary value problems; Magnetostatics: Biot-Savart law, Ampere's theorem, electromagnetic induction; Maxwell's equations in free space and linear isotropic media; boundary conditions on fields at interfaces; Scalar and vector potentials; Gauge invariance; Electromagnetic waves in free space, dielectrics, and conductors; Reflection and refraction, polarization, Fresnel's Law, interference, coherence, and diffraction; Dispersion relations in plasma; Lorentz invariance of Maxwell's equations; Transmission lines and wave guides; Dynamics of charged particles in static and uni-

form electromagnetic fields; Radiation from moving charges, dipoles and retarded potentials

d. Classical Mechanics:

Newton's laws; Phase space dynamics, stability analysis; Central-force motion; Two-body collisions, scattering in laboratory and centre-of-mass frames; Rigid body dynamics, moment of inertia tensor, non-inertial frames and pseudoforces; Variational principle, Lagrangian and Hamiltonian formalisms and equations of motion; Poisson brackets and canonical transformations; Symmetry, invariance and conservation laws, cyclic coordinates; Periodic motion, small oscillations and normal modes; Special theory of relativity, Lorentz transformations, relativistic kinematics and mass-energy equivalence. .

(4) GEOPHYSICS - PAPER III : 200 Marks**PART-A: 100 Marks****a. Radiometric Exploration / Airborne Geophysical surveys for Geological Mapping:**

Principles of radioactivity, radioactivity decay processes, units, radioactivity of rocks and minerals, Instruments, Ionisation chamber, G-M counter, Scintillation meter, Gamma ray spectrometer, Radiometric prospecting for mineral exploration (Direct/Indirect applications), Radiometric prospecting for beach placers, titanium, zirconium and rare-earths, portable gamma ray spectrometry and radon studies in seismology, environmental Applications, logging methods, radiometric dating techniques. Airborne geophysical surveys, planning of surveys, sensors, data corrections, flight path recovery methods, applications in geological mapping, interpretation of maps, identification of structural features, altered zones.

b. Marine Geophysics:

Oceans and Seas, origin of continents and oceans, salinity, temperature and density of sea water. Introduction to Sea-floor features: Physiography, divisions of sea floor, continental shelves, slopes, aprons and abyssal planes, growth and decline of ocean basins, turbidity currents, submarine sedimentation and stratigraphy, occurrence of mineral deposits and hydrocarbons in offshore. Geophysical surveys and instrumentation, Gravity and Magnetic surveys, Instrumentation used in ship borne surveys, towing cable and fish, data collection and survey procedures, corrections and interpretation of data. Oceanic magnetic anomalies, sea floor spreading, Vine-Matthews hypothesis, geomagnetic time scale and dating sea floor, linear magnetic anomalies, Oceanic heat flow, ocean ridges, basins, marginal basins, rift valleys. Seismic surveys, energy sources, Finger, Boomer, Sparker, Exploder, Air gun, Vapour cook, Hydrophones, processing, data reduction and interpretation. Bathymetry, echo sounding, bathymetric charts, sea bed mapping, seabed sampling, dredging and coring, Navigation methods and Position location methods.

c. Geophysical Signal Processing:

Various types of signals, sampling theorem, aliasing effect, Fourier series and periodic waveforms, Fourier transform and its properties, Discrete Fourier transform and FFT, Auto and cross correlations, Power spectrum, Delta function, unit step function. Time domain windows, Z transform and properties, Inverse Z transform.

Principles of digital filters, types of filters, moving average and recursive and non recursive filters Amplitude and phase response filters low pass, band pass and high pass filters, Processing of Random signals. Signal enhancement for gravity and magnetic maps; regional residual separation, continuations, evaluation of derivatives, pseudo gravity transformations, reduction to poles and equator, Improvement of signal to noise ratio, source and geophone arrays as spatial filters. Earth as low pass filter.

d. Remote Sensing and GIS applications:

Fundamental concepts of remote sensing, electromagnetic radiation spectrum, energy-frequency-wavelength relationship, Boltzman Law, Wien Law, electromagnetic energy and its interactions in the atmosphere and with terrain features; elements of photographic systems, reflectance and emittance, false color composites, remote sensing platforms, flight planning, geosynchronous and sun synchronous orbits, sensors, resolution, parallax and vertical exaggeration, relief displacement, mosaic, aerial photo interpretation and geological application. Fundamentals of photogrammetry, satellite remote sensing, multi-spectral scanners, thermal scanners, microwave remote sensing, fundamental of image processing and interpretation for geological applications. Introduction to Geographic Information Systems (GIS) spatial data structures, visualization and querying, spatial data analysis.

PART-B: 100 Marks**a. Solid State Physics:**

Crystalline and amorphous structure of matter; Different crystal systems, space groups; methods of determination of crystal structure; X-ray diffraction, scanning and transmission electron microscopes; Band theory of solids-conductors, insulators and semiconductors; Thermal properties of solids, specific heat, Debye theory; Magnetism: dia, para and ferromagnetism; elements of superconductivity; Meissner effect, Josephson junctions and applications; elementary ideas about high temperature superconductivity.

b. Laser systems: Light amplification and relation between Einstein A and B coefficients. Rate equations for three level and four level systems. Ruby laser, Nd-YAG laser, CO₂ laser, Dye laser, Excimer laser, Semiconductor laser.

c. Laser cavity modes: Line shape function and full width at half maximum (FWHM) for natural broadening, collision broadening, Doppler broadening, saturation behaviour of broadened transitions, longitudinal and transverse modes. ABCD matrices and cavity stability criteria for confocal resonators. Quality factor, Q-switching, mode locking in lasers. Expression for intensity for modes oscillating at random and modes lock in phase. Methods of Q-switching and mode locking. Optical fiber waveguides, Fiber characteristics.

d. Electronics and devices:

Semiconductor devices (diodes, junctions, transistors, field effect devices, homo and hetero junction devices) device structure, device characteristics, frequency dependence and applications. Opto-electronic devices (solar cells, photo detectors, LEDs) Operational amplifiers and their applications. Digital techniques and applications (registers, counters, comparators and similar circuits). A/D and D/A converters. Microprocessor and microcontroller basics. Data interpretation and analysis. Precision and accuracy. Error analysis, propagation of errors. Least square fitting. Intrinsic extrinsic semiconductors, pn-p and n-p-n transistors; Amplifiers and oscillators; Op-amps; FET, JFET and MOSFET; Digital electronics-Boolean identities, De

morgan's laws, logic gates and truth tables; simple logic circuits; thermistors, solar cells, fundamentals of microprocessors and digital computers.

e. Digital electronics, Radar systems, Satellite communications:

Digital circuits, Number systems and codes, Combination logic circuits, sequential logic circuits, microprocessor architecture, functional diagram, Pin description, Timing diagram of read cycle, timing diagram of write cycle. Data transfer techniques-Serial transfer, parallel transfer etc. Radar systems, signal and data processing, satellite communication-Fundamentals Designing a surveillance radar, tracking radar, signal and data processing, radar antenna parameters, satellite systems-communication satellite systems, communication satellites, orbiting satellites, satellite frequency bands, satellite orbit and inclinations. Multiple access techniques, earth station technology.

f. Quantum Mechanics:

Wave-particle duality; Wave functions in coordinate and momentum representations; Commutators and Heisenberg's uncertainty principle; Matrix representation; Dirac's bra and ket notation; Schroedinger equation (time-dependent and time-independent); Eigen value problems such as particle-in-a-box, harmonic oscillator, etc.; Tunneling through a barrier; Motion in a central potential; Orbital angular momentum, Angular momentum algebra, spin; Addition of angular momentum; Hydrogen atom, spin-orbit coupling, fine structure; Time-independent perturbation theory and applications; Variational method; WKB approximation; Time dependent perturbation theory and Fermi's Golden Rule; Selection rules; Semi-classical theory of radiation; Elementary theory of scattering, phase shifts, partial waves, Born approximation; Identical particles, Pauli's exclusion principle, spin-statistics connection; Relativistic quantum mechanics: Klein Gordon and Dirac equations.

(2) CHEMISTRY PAPER-I (Inorganic Chemistry): 200 Marks

Chemical periodicity:

Periodic table, group trends and periodic trends in physical properties. Classification of elements on the basis of electronic configuration. Modern IUPAC Periodic table. General characteristic of s, p, d and f block elements. Effective nuclear charges, screening effects, atomic radii, ionic radii, covalent radii. Ionization potential, electron affinity and electro-negativity. Group trends and periodic trends in these properties in respect of s-, p- and d-block elements. General trends of variation of electronic configuration, elemental forms, metallic nature, magnetic properties, catenation and catalytic properties, oxidation states, aqueous and redox chemistry in common oxidation states, properties and reactions of important compounds such as hydrides, halides, oxides, oxy-acids, complex chemistry in respect of s-block and p-block elements.

Chemical Bonding and structure:

Ionic bonding: Size effects, radius ratio rules and their limitations. Packing of ions in crystals, lattice energy, Born-Lande equation and its applications, Born-Haber cycle and its applications. Solvation energy, polarizing power and polarizability, ionic potential, Fajan's rules. Defects in solids. Covalent bonding: Lewis structures, formal charge. Valence Bond Theory, Molecular orbital Theory, hybridizations, VSEPR theory. Partial ionic Character of covalent bonds, bond moment, dipole moment and electro negativity differences. Concept of resonance, resonance energy, resonance structures. Schrodinger equation for the H-atom.

Coordinate bonding: Werner theory of coordination compounds, double salts and complex salts, Lewis acid-base. Ambidentate and polydentate ligands, chelate complexes. IUPAC nomenclature of coordination compounds. Coordination numbers, Geometrical isomerism. Stereoisomerism in square planar and octahedral complexes. Hydrogen bonding. Metallic bonding: qualitative idea of band theory, conducting, semi conducting and insulating properties.

Chemistry of coordination compounds:

Isomerism, reactivity and stability: Determination of configuration of cis- and trans-isomers by chemical methods. Labile and inert complexes, substitution reaction on square planar complexes, trans effect. Stability constants of coordination compounds and their importance in inorganic analysis. Structure and bonding: Elementary Crystal Field Theory: splitting of d_n configurations in octahedral, square planar and tetrahedral fields, crystal field stabilization energy; pairing energy. Jahn-Teller distortion. Metal-ligand bonding, sigma and pi bonding in octahedral complexes and their effects on the oxidation states of transitional metals. Orbital and spin magnetic moments, spin only moments of and their correlation with effective magnetic moments, d-d transitions; L-S coupling, spectroscopic ground states, selection rules for electronic spectral transitions; spectro-chemical series of ligands; charge transfer spectra.

Acid-Base reactions

Acid-Base concept: Arrhenius concept, theory of solvent system, Bronsted-Lowry's concept, relative strength of acids, Pauling rules. Lewis concept. Acidbase equilibria in aqueous solution and pH. Acid-base neutralisation curves; indicator, choice of indicators.

Precipitation and Redox Reactions:

Solubility product principle, common ion effect. Ion-electron method of balancing equation of redox reaction. Standard redox potentials, Nernst equation. Influence on complex formation, precipitation and change of pH on redox potentials; formal potential. Feasibility of a redox titration, redox potential at the equivalence point, redox indicators. Redox potential diagram of common elements and their applications. Disproportionation and comproportionation reactions.

Organic metallic compounds:

18-electron rule and its applications to carbonyls, nitrosyls, cyanides, and nature of bonding involved therein. Simple examples of metal-metal bonded compounds and metal clusters. Metal-olefin complexes: zeises salt, Ferrocene.

Nuclear chemistry:

Radioactive decay - General characteristics, decay kinetics, parent -daughter decay growth relationships, determination of half-lives, Nuclear models -shell model, liquid drop model, Fermi gas model, Collective model and optical model. Nuclear stability. Decay theories. Nuclear reactions- fission, fusion and spallation reactions. Definition of curie and related calculations, preparation of artificial radionuclides by bombardment, radiochemical separation techniques. Experimental techniques in the assay of radioisotopes, gas filled detectors-ionization chamber, proportional and Geiger-Muller counters -G.M. Plateau, dead time, coincidence loss, determination of dead time, scintillation counters, solid state detectors.

s-Block Elements :

Hydride, hydration energies, solvation and complexation tendencies of alkali and alkaline-earth metals, principle of metallurgical extraction, Chemistry of Li and Be, their anomalous behaviour and diagonal relationships, alkyls and aryls.

p-Block Elements :

Comparative study of group 13 & 14 elements with respect to periodic properties. Compounds such as hydrides, halides, oxides and oxyacids; diagonal relationship; preparation, properties, bonding and structure of diborane, borazine and alkali metal borohydrides. Preparation, properties and technical applications of carbides and fluorocarbons. Silicones and structural principles of silicates.

Chemistry of d- and f- block elements:

General comparison of 3d, 4d and 5d elements in term of electronic configuration, elemental forms, metallic nature, atomization energy, oxidation states, redox properties, coordination chemistry, spectral and magnetic properties. f-block elements: electronic configuration, ionization energies, oxidation states, variation in atomic and ionic (3+) radii, magnetic and spectral properties of lanthanides, comparison between lanthanide and actinides, separation of lanthanides (by ion-exchange method). Chemistry of some representative compounds: K₂Cr₂O₇, KMnO₄, K₄[Fe(CN)₆], K₂[Ni(CN)₄], H₂PtCl₆, Na₂[Fe(CN)₅NO].

(3) CHEMISTRY PAPER-II (Physical Chemistry) :200 Marks

Kinetic theory and the gaseous state:

Gaseous state: Gas laws, kinetic theory of gas, collision and gas pressure, derivation of gas laws from kinetic theory, average kinetic energy of translation, Boltzmann constant and absolute scale of temperature. Maxwell's distribution of speeds. Kinetic energy distribution, calculations of average, root mean square and most probable velocities. Principle of equipartition of energy and its application to calculate the classical limit of molar heat capacity of gases.

Collision of gas molecules, Real gases:

Collision diameter; collision number and mean free path; frequency of binary collisions; wall collision and rate of effusion. Real gases, Deviation of gases from ideal behaviour; compressibility factor; Andrew's and Amagat's plots; van der Waals equation and its characteristic features. Existence of critical state. Critical constants in terms of van der Waals constants. Law of corresponding state and significance of second virial coefficient. Boyle temperature. Intermolecular forces.

Liquid state:

physical properties of liquids and their measurements: surface tension and viscosity

Solids: Nature of solid state, law of constancy of angles, concept of unit cell, different crystal system, Bravais lattices, law of rational indices, Miller indices, symmetry elements in crystals. X-ray diffraction, Bragg's law, Laue's method, powder method, radius ratio and packing in crystals.

Thermodynamics:

Definition of thermodynamic terms. Thermodynamic functions and their differentials. Zeroth law, First law and Second law of thermodynamics. Cyclic, reversible and irreversible processes. Internal energy (U) and enthalpy (H); relation between Cp and Cv, calculation of w, q, ?U and ?H for expansion of ideal gas under isothermal and adiabatic conditions for reversible and irreversible processes including free expansion. Joule-Thomson Coefficient and inversion temperature. Application of First law of thermodynamics.

Application of Second law of thermodynamics.

Carnot cycle and its efficiency, Gibbs function (G) and Helmholtz function (A), Gibbs-Helmholtz equation, criteria for thermodynamic equilibrium and spontaneity of a process. Chemical equilibrium: chemical equilibria of homogeneous and heterogeneous systems, derivation of expression of equilibrium constants, Le Chatelier's principle of dynamic equilibrium.

Thermodynamics and Equilibrium:

Chemical potential in terms of Gibb's free energy and other thermodynamic state functions and its variation with temperature and pressure. Gibbs-Duhem equation; fugacity of gases and fugacity coefficient. Thermodynamic conditions for equilibrium, degree of advancement. Van't Hoff's reaction isotherm. Equilibrium constant and standard Gibbs free energy change. Definitions of KP, KC and Kx; van't Hoff's reaction isobar and isochore. Le Chatelier's principle. Activity and activity coefficients of electrolyte / ion in solution. Debye-Hückel limiting law.

Acids-bases and solvents:

Modern aspects of acids and bases: Arrhenius theory, theory of solvent system, Bronsted and Lowry's concept, Lewis concept with typical examples, applications and limitations. Strengths of acids and bases. Ionization of weak acids and bases in aqueous solutions, application of Ostwald's dilution law, ionization constants, ionic product of water, pH-scale, buffer solutions and their pH values, buffer actions & buffer capacity; hydrolysis of salts.

Solutions of non-electrolytes:

Colligative properties of solution, Raoult's Law, relative lowering of vapor pressure, osmosis and osmotic pressure; elevation of boiling point and depression of freezing point of solvents.

Chemical kinetics and catalysis:

Order and molecularity of reactions, rate laws and rate equations for first order and second order reactions; zero order reactions. Parallel and consecutive reactions. Determination of order of reactions. Temperature dependence of reaction rate, energy of activation. Enthalpy of activation, entropy of activation, effect of dielectric constant and ionic strength of reaction rate, kinetic isotope effect; collision theory & transition State Theory of reaction rate, Catalytic reactions.

Adsorption and Surface Chemistry:

Physisorption & Chemisorption, adsorption isotherms, Freundlich and Langmuir adsorption isotherm, BET equation, surface area determination, heterogeneous catalysis; colloids, electrical double layer and colloid stability, electro-kinetic phenomenon; elementary ideas about soaps & detergents, micelles, emulsions.

Electrochemistry:

Conductance: cell constant, specific conductance and molar conductance. Kohlrausch's law of independent migration of ions, ion conductance and ionic mobility. Equivalent and molar conductance at infinite dilution. Ostwald's dilution law. Debye-Hückel theory. Application of conductance measurement. Conductometric titrations. Determination of transport number by moving boundary method. Types of electrochemical cells, cell reactions, emf and change in free energy, ΔH and ΔS of cell reactions. Nernst equation. Standard cells. Half-cells /

electrodes, different types of electrodes. Standard electrode potential and principles of its determination. Types of concentration cells. Liquid junction potential. Glass electrode and determination of pH of a solution. Potentiometric titrations: acid-base and redox, electro chemical power sources; primary, secondary and fuel Cells, corrosion and inhibition of corrosion.

Photochemistry :

Frank-Condon principle and vibrational structure of electronic spectra. Bond dissociation and principle of determination of dissociation energy. Decay of excited states by radiative and non-radiative paths. Fluorescence and phosphorescence, Jablonsky diagram. Laws of photochemistry: Grothus-Draper law, Stark-Einstein law of photochemical equivalence and Lambert-Beer's law; quantum yield and its measurement for a photochemical process, actinometry. Photostationary state. Photosensitized reactions. Kinetics of H₂ decomposition, H₂-Br₂ reaction, dimerisation of anthracene.

Quantum Chemistry:

Wave-particle duality, Photoelectric and Compton effects, de Broglie hypothesis. Eigenfunctions and eigenvalues. Uncertainty relation, Expectation value. Hermitian operator. Schrodinger time-independent equation: nature of the equation, acceptability conditions imposed on the wave functions and probability interpretations of wave function. Schrodinger equation for one-dimensional box and its solution. Comparison with free particle eigenfunctions and eigenvalues.

Basic principles and application of spectroscopy:

Electromagnetic radiation, interaction with atoms and molecules and quantization of different forms of energies. Condition of resonance and energy of absorption for various types of spectra; origin of atomic spectra, spectra of hydrogen atoms, many electron atoms, spin and angular momentum. Rotational spectroscopy of diatomic molecules: rigid rotor model, selection rules, spectrum, characteristic features of spectral lines. Determination of bond length, effect of isotopic substitution. Vibrational spectroscopy of diatomic molecules: Simple Harmonic Oscillator model, selection rules, Raman Effect. Characteristic features and conditions of Raman activity with suitable illustrations. Rotational and vibrational Raman spectra.

UV Spectra: Electronic transition ($\sigma-\sigma^*$, $\pi-\pi^*$ and $n-\pi^*$), relative positions of λ_{max} considering conjugative effect, steric effect, solvent effect, red shift (bathochromic shift), blue shift (hypsochromic shift), hyperchromic effect, hypochromic effect (typical examples). IR Spectra: Modes of molecular vibrations, application of Hooke's law, characteristic stretching frequencies of O-H, N-H, C-H, C-D, C=C, C=N, C=O functions; factors effecting stretching frequencies

PMR Spectra: Nuclear spin, NMR active nuclei, principle of proton magnetic resonance, equivalent and non-equivalent protons, chemical shift (?), shielding / deshielding of protons, up-field and down-field shifts. NMR peak area, diamagnetic anisotropy, relative peak positions of different kinds of protons, substituted benzenes.

(4) CHEMISTRY PAPER-III:200 Marks

PART-A (Analytical Chemistry): 100 Marks

Theoretical basis of Quantitative inorganic analysis:

Law of mass action, chemical and ionic equilibrium, solubility, Solubility product and common ion effect, effect of temperature upon the solubility of precipitates, the ionic product of water, pH, effect of temperature on pH, Salt hydrolysis, hydrolysis constant, degree of hydrolysis, buffer solutions, different types of buffers and Henderson's equation.

Gravimetric Analysis:

General principles, stoichiometry, calculation of results from gravimetric data. Properties of precipitates. Nucleation and crystal growth, factors influencing completion of precipitation. Co-precipitation and post-precipitation, purification and washing of precipitates. Precipitation from homogeneous solution, a few common gravimetric determinations-chloride as silver chloride, sulphate as barium sulphate, aluminum as the oxinate and nickel as dimethyl glyoximate.

Sampling and treatment of samples for chemical analysis:

Techniques of collection of Solids, liquids and gaseous samples, dissolution of solid samples, attack with water, acids, and alkalis, fusion with Na₂CO₃, NaOH, Na₂O₂, K₂S₂O₇; Microwave assisted digestion techniques(Only elementary idea)

Volumetric Analysis:

Equivalent weights, different types of solutions, Normal solutions, Molar solutions, and molal solutions and their inter relations. Primary and secondary standard substances. principles of different type of titrations-i) acid-base titration, ii) redox titration, iii) complexometric titrations. Types of indicators - i) acid-base, ii) redox iii) metal-ion indicators. Principles in estimation of mixtures of NaHCO₃ and Na₂CO₃ (by acidimetry); Principles of estimation of iron, copper, manganese, chromium (by redox titration);

Acid base titrations: Principles of titrimetric analysis, titration curves for strong acid-strong base, weak acid-strong base and weak base-strong acid titrations, poly protic acids, poly equivalent bases, determining the equivalence point-theory of acid base indicators, colour change range of indicator, selection of proper indicator.

Redox Titrations: Principles behind the Iodometry, permanganometry, dichrometry, difference between iodometry and iodimetry.

Potentiometry: Fundamentals of potentiometry. Indicator and ion-selective electrodes. Membrane electrodes. Glass electrode for pH measurement, glass electrodes for cations other than protons. Liquid membrane electrodes, solid state ion selective detectors and biochemical electrodes. Applications of potentiometry. Direct potentiometric measurements-determination of pH and fluoride. Redox and potentiometer titrations- Balancing redox reactions, calculation of the equilibrium constant of the reaction, titration curves, visual end point detection. Redox indicators-theory, working and choice. Potentiometric end point detection. Applications of redox titrations.

Complexometric titrations: Complex formation reactions, stability of complexes, stepwise formation constants, chelating agents, EDTA-acidic properties, complexes with metal ions, equilibrium calculations involving EDTA, conditional formation constants, derivation of EDTA titration curves, effect of other complexing agents, factors affecting the shape of titration curves-completeness of reaction, indicators for EDTA titrations-theory of common indicators, titration methods employing EDTA-direct, back and displacement titrations, indirect determinations, titration of mixtures, selectivity, masking and de-masking agents, typical applications of EDTA titrations-hardness of water, magnesium and aluminium in antacids, magnesium, manganese and zinc in a mixture, titrations involving uni-dentate ligands-titration of chloride with Hg²⁺ and cyanide with Ag⁺.

Chromatographic methods of analysis:

Basic principles and classification of chromatography. Importance of column

chromatography and thin layer chromatography; Theory and principles of High Performance Liquid Chromatography (HPLC) and Gas Liquid Chromatography (GLC). Ion-exchange chromatography.

UV-Visible Spectroscopy:

Basic Principles of UV-Vis spectrophotometer. Lambert -Beer's Law and its limitations. Instrumentation consisting of source, monochromator, grating and detector. Spectrophotometric determination.

Flame photometry and Atomic absorption spectrometry:

Emission spectra Vs absorption spectra. Basic Principles and theory of flame photometry. Applications of Flame photometers. Basic Principles and theory of AAS. Three different modes of AAS - Flame-AAS, VGAAS, and GFAAS. Single beam and double beam AAS. Function of Halo Cathode Lamp (HCL) and Electrode Discharge Lamp (EDL). Different types of detectors used in AAS. Different types of interferences-Matrix interferences, chemical interferences, Spectral interferences and background correction in AAS. Use of organic solvents. Quantitative techniques-calibration curve procedure and the standard addition technique. Typical commercial instruments for FP and AAS. Applications. Qualitative and quantitative analysis. Relative detection abilities of atomic absorption and flame emission spectrometry.

X-ray methods of Analysis:

Introduction , theory of X-ray generation, , X-ray spectroscopy, , X-ray diffraction and X-ray fluorescence methods, Braggs law, instrumentation , dispersion by crystals, applications. Preparation of pallets, glass beads, quantitative and qualitative measurement.

Inductively coupled plasma spectroscopy:

Theory and Principles, plasma generation, utility of peristaltic pump, sampler - skimmer systems, ion lens, quadrupole mass analyzer, dynode /solid state Detector, different type of interferences- spectroscopic and non-spectroscopic interferences, isobaric and molecular interferences, applications.

Analysis of Minerals, Ores and Alloys:

Analysis of Minerals and Ores- estimation of (i) CaCO₃ , MgCO₃ in dolomite (ii) Fe₂O₃, Al₂O₃, and TiO₂ in Bauxite.(iii) MnO and MnO₂ in Pyrolusite.

Analysis of Metal and Alloys: (i) Cu and Zn in Brass (ii) Cu, Zn, Fe, Mn, Al and Ni in Bronze (iii) Cr, Mn, Ni, and P in Steel (iv) Pb, Sb, Sn in type metal.

Analysis of petroleum and petroleum products:

Introduction, constituents and petroleum fractionation. Analysis of petroleum products-specific gravity, viscosity, Doctor test, aniline point, colour determination, cloud point, pour point. Determination of water, neutralization value (acid and base numbers), ash content, Determination of lead in petroleum.

Analysis of coal and coke-Types, composition, preparation of sample, proximate and ultimate analysis calorific value by bomb Calorimetry.

PART-B (Organic Chemistry): 100 Marks

Basic organic chemistry:

Inductive effect, resonance and resonance energy. Homolytic and heterolytic bond breaking, electrophiles and nucleophiles; carbocations, carbanions and radicals (stability and reactivity). Alkanes, alkenes and alkynes: Synthesis and chemical reactivity of alkanes, mechanism of free-radical halogenation of alkanes. General methods of synthesis, electrophilic addition reactions and polymerization reaction (definition and examples only) of alkenes. General methods of synthesis, acidity, hydration and substitution reactions of alkynes.

Organometallic compounds:

Grignard reagents - preparations and reactions, application of Grignard reagents in organic synthesis. Organic compounds containing nitrogen: aromatic nitro compounds - reduction under different conditions. Methods of synthesis of aliphatic amines, Heinsberg's method of amine separation, Hofmann degradation, Gabriel's phthalimide synthesis, distinction of primary, secondary and tertiary amines; methods of synthesis of aromatic amines, basicity of aliphatic and aromatic amines. Sandmeyer reactions; synthetic applications of benzene diazonium salts.

Bonding and physical properties:

Valence bond theory: concept of hybridisation, resonance (including hyperconjugation), orbital pictures of bonding sp₃, sp₂, sp: C-C, C-N & C-O system). Inductive effect, bond polarization and bond polarizability, steric effect, steric inhibition of resonance. MO theory: sketch and energy levels of MOs of i) acyclic p orbital system ii) cyclic p orbital system, iii) neutral system. Frost diagram, Huckel's rules for aromaticity & antiaromaticity; homoaromaticity. Physical properties: bond distance, bond angles, mp/bp & dipole moment in terms of structure and bonding. Heat of hydrogenation and heat of combustion.

Aldol and related reactions:

Keto-enol tautomerism, mechanism and synthetic applications of aldol condensations, Claisen reaction, Schmidt reaction, Perkin reaction, Knovenogal, benzoin, Cannizaro reaction, Michael addition. Aromatic substitution reactions - electrophilic, nucleophilic and through benzenes - radical substitution of arenes - orientation of nucleophilic substitution at a saturated, carbon, SN1, SN2, SNI reactions -effect of structure, nucleophile, leaving group, solvent. Additions involving electrophiles, nucleophiles and free radicals.

Mechanism of some name reactions:

Aldol, Perkin, Benzoin, Cannizaro, Wittig, Grignard, Reformatsky, Hoffmann, Claisen and Favorsky rearrangements. Openauer oxidation, Clemmensen reduction, Meerwein - Pondorf and Verley and Birch reductions. Stork enamine reactions, Michael addition, Mannich Reaction, Diels - Alder reaction.

Electrocyclic Reactions:

Molecular orbital symmetry, frontier orbitals of ethylene, 1,3 Butadiene, 1,3,5-Hexatriene, allyl system, classification of pericyclic reactions FMO approach, Woodward-Hoffman correlation diagram method and perturbation of molecular (PMO) approach for the explanation of pericyclic reactions under thermal and photochemical conditions. Conrotatory and disrotatory motions (4n) and (4n+2).

Organic Reaction Mechanisms:

Addition Elimination Mechanisms: (a) Addition to carbon multiple bonds- hydrogenation of double and triple bonds, hydroboration, birch reduction, Michael reaction, addition of oxygen and N, (b) Addition to carbon-hetero atom multiple bonds: Mannich reaction, Reductions of Carbonyl compounds, acids, esters, nitrates, addition of Grignard reagents, Reformatsky reaction, Tollen's reaction, Wittig reaction: (c) Elimination reactions: Stereochemistry of eliminations in acyclic and cyclic systems,

orientation in eliminations - Saytzeff and Hoffmann elimination.

Organic Spectroscopy:

Infrared spectroscopy: Units of frequency wave length and wave number, molecular vibrations, factors influencing vibrational frequencies, the IR spectrometer, characteristic frequencies of organic molecules and interpretation of spectra.

Ultraviolet spectroscopy: Introduction, absorption laws, measurement of the spectrum, chromophores, definitions, applications of UV spectroscopy to Conjugated dines, trienes, unsaturated carbonyl compounds and aromatic compounds.

Nuclear Magnetic Resonance Spectroscopy: (Proton and Carbon -13 NMR) The measurement of spectra, the chemical shift: the intensity of NMR signals and integration factors affecting the chemical shifts: spin-spin coupling to 13C IH-IH first order coupling: some simple IH-IH splitting patterns: the magnitude of IH-IH coupling constants.

Mass spectroscopy: Basic Principles: instrumentation: the mass spectrometer, isotope abundances; the molecular ion, meta stable ions.

(4) HYDROGEOLOGY 200 Marks

Section A : Origin, occurrence and distribution of water.

Origin of water : meteroric, juvenile, magmatic and sea waters. Hydrologic cycle : precipitation, runoff, infiltration and evapotranspiration. Hydrographs, subsurface movement and vertical distribution of groundwater, springs, Classification of aquifers. concepts of drainage basin and groundwater basin. Hydrological properties of rocks-specific yield, specific retention, porosity, hydraulic conductivity, transmissivity, storage coefficient. Water table fluctuations-causative factors, concept of barometric and tidal efficiencies. Water table contour maps. Classification of rocks with respect to their water bearing characteristics. Hydrostratigraphic units. Groundwater provinces of India. Hydrogeology of arid zones of India, wet lands.

Section B : Well hydraulics and well design Theory of groundwater flow. Darcy's Law and its applications, determination of permeability in laboratory and in field.

Types of wells, drilling methods, construction, design, development and maintenance of wells, specific capacity and its determination. Unconfined, confined steady, unsteady and radial flow conditions. Pumps tests-methods, data analysis and interpretation for hydrogeologic boundaries. Evaluation of aquifer parameters using Teim, Theis, Jacob and Walton methods. Groundwater modelling-numerical and electrical models.

Section C : Groundwater chemistry

Groundwater quality-physical and chemical properties of water, quality criteria for different uses, graphical presentation of water quality data, groundwater quality in different provinces of India-problems of arsenic and fluoride. Saline water intrusion in coastal and other aquifers and its prevention. Radioisotopes in hydrogeological studies. Groundwater contamination.

Section D : Groundwater exploration

Geological-lithological and structural mapping, fracture trace analysis. Hydrological-lithological classification with respect of hydrologic properties. Hydraulic continuity in relation to geologic structures. Location of springs Remote sensing-hydrogeomorphic mapping of the terrain using different images of different satellite missions. Lineament mapping. Shallow groundwater potential zone mapping using satellite images. Surface geophysical methods-seismic, gravity, geoelectrical and magnetic. Subsurface geophysical methods-well logging for delineation of aquifers and estimation of water quality.

Section E : Groundwater problems and management

Groundwater problems related to foundation work, mining, canals and tunnels. Problems of over exploitation and groundwater mining. Groundwater development in urban areas and rain water harvesting. Artificial recharge methods. Groundwater problems in arid regions and remediation. Groundwater balance and methods of estimation. Groundwater legislation. Sustainability criteria and managing renewable and nonrenewable groundwater resources.

APPENDIX-II

INSTRUCTIONS TO THE CANDIDATES FOR FILLING ONLINE APPLICATIONS

Candidates are required to apply Online using the website www.upsconline.nic.in.

Salient features of the system of Online Application Form are given hereunder:

- Detailed instructions for filling up Online applications are available on the above mentioned website.
- Candidates will be required to complete the Online Application Form containing two stages viz. Part-I and Part-II as per the instructions available in the above mentioned site through drop down menus.
- The candidates are required to pay a fee of Rs.200/- (Rupees Two Hundred only) [excluding SC/ST/PH/Female candidates who are exempted from payment of fee] either by depositing the money in any branch of SBI by cash, or by using net banking facility of State Bank of India/ State Bank of Bikaner & Jaipur/State Bank of Hyderabad/State Bank of Mysore/ State Bank of Patiala/State Bank of Travancore or by using any Visa/Master Credit/Debit Card.
- Before start filling up of Online Application, a candidate must have his/her photograph and signature duly scanned in the .jpg format in such a manner that each file should not exceed 40 KB and must not be less than 3 KB in size for the photograph and 1 KB for the signature.

- The Online applications(Part I and II) can be filled from **21st February, 2015 to 20th March, 2015 till 11.59 p.m.**, after which link will be disabled.
- Applicants should avoid submitting multiple applications. However, if due to any unavoidable circumstances, any applicant submits multiple applications then he/she must ensure that the applications with higher RID is complete in all respects.
- In case of multiple applications, the applications with higher RID shall be entertained by the Commission and fee paid against one RID shall not be adjusted against any other RID.
- The applicants must ensure that while filling their Application Form, they are providing their valid and active E-Mail IDs as the Commission may use electronic mode of communication while contacting them at different stages of examination process.
- The applicants are advised to check their emails at regular intervals and ensure that the email address ending with @nic.in are directed to their inbox folder and not to the SPAM folder or any other folder.
- **Candidates are strongly advised to apply online well in time without waiting for the last date for submission of Online Applications.**

APPENDIX III

SPECIAL INSTRUCTIONS FOR CONVENTIONAL TYPE PAPERS

1. Articles permitted inside Examination Hall

Battery-operated pocket calculators of "non-programmable" type only, mathematical/engineering/drawing instruments, including a flat rule divided on the edges into inches and tens of an inch and into centimeters and millimeters, a slide rule, set squares, a protractor and a pair of compasses, pencils, coloured pencils, mapping pens, eraser, T-square and drawing board for use wherever necessary. Candidates are not allowed to bring with them any "Tables or Charts" for use in the Examination Hall.

Mobile phones, pagers, bluetooth or any other communication devices are not allowed inside the premises where the examination is being conducted. Any infringement of these instructions shall entail disciplinary action including ban from future examinations.

Candidates are advised in their own interest not to bring any of the banned items including mobile phones/pagers to the venue of the examination, as arrangements for safekeeping cannot be assured.

2. Tables to be supplied by UPSC

If it is considered necessary for answering the questions set in any paper, the Commission may supply any of the following for reference purpose only:-

- (i) Mathematical/Physical, Chemical and Engineering Tables (including Logarithmic Tables);
- (ii) Steam Table (including Mollier Diagrams for Temperature up to 800°C and Pressure up to 500 Kgf/Cm);
- (iii) National Building Code of India 1970 or 1983 Group 2 Part VI;
- (iv) Any other special articles as may be necessary for the candidates to answer the questions set in the question paper.

After conclusion of the examination, return the above items to the Invigilator.

3. Answers to be written in own hand

Write the answers in your own hand in ink. Pencil may be used for maps, mathematical drawings or rough work.

4. Check Answer Book

The candidate must write his/her roll number (and not his/her name) only in the

space provided for the purpose on every answer book used by him/her. Before writing in the answer book, please see that it is complete. In case there are any missing pages, it should be got replaced.

Do not tear out any pages from the Answer Book. If you use more than one Answer Book, indicate on the cover of first Answer Book the total number of Answer Books used. Do not leave any blank, unused spaces between answers. If such spaces are left, score them out.

5. Answers in excess of prescribed number will be ignored

The candidate must attempt questions strictly in accordance with the directions given on each question paper. If questions are attempted in excess of the prescribed number only the questions attempted first upto the prescribed number shall be valued and the remaining answers will be ignored.

6. Questions relating to graph/precis should be attempted only on graph/precis sheets to be supplied on demand by the Invigilators. All loose sheets such as precis sheet, drawing papers, graph sheets etc. whether used or not, should be placed inside the answer books and fastened along with the additional answer book(s), if any. Candidates who fail to observe this instruction will be penalized. Do not write your roll number on these sheets.

7. Unfair means strictly prohibited

Do not copy from the papers of any other candidate nor allow your papers to be copied nor give nor attempt to give nor obtain nor attempt to obtain irregular assistance of any description. It will be responsibility to every candidate to ensure that his/her answers are not copied by another candidate. Failure to do so will invite penalty, as may be awarded by the Commission for adoption of unfair means.

8. Conduct in Examination Hall

Do not misbehave in any manner or create disorderly scene in the examination hall or harass or bodily harm the staff deployed for the conduct of examination. You will be severely penalized if you attempt to do so.

9. Please read carefully and abide by the instructions printed on the Question Paper and on the Answer Book supplied in the Examination Hall.