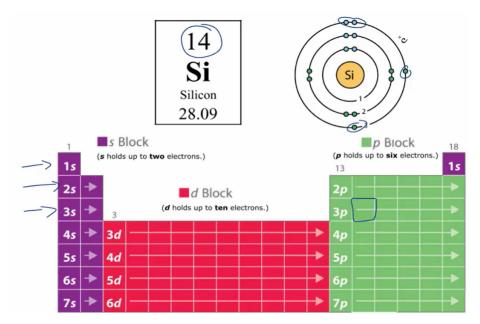
This document explains from Bohr model of atoms to how electrons and holes move within a semiconductor crystalline structure.

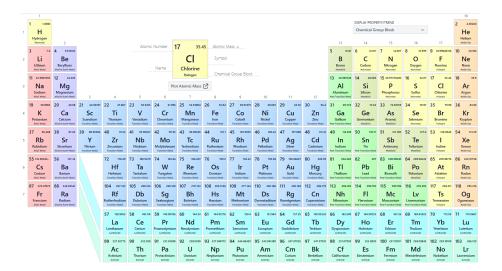
1 Bohr Model and periodic table

Useful Resource: Atomic Structure for Silicon Video, Periodic Table

Quick review that Silicon's atomic number is 14, which means a neutral atom of silicon has 14 electrons. The horizontal count by rows on the periodic table means the number of electrons in each shell.



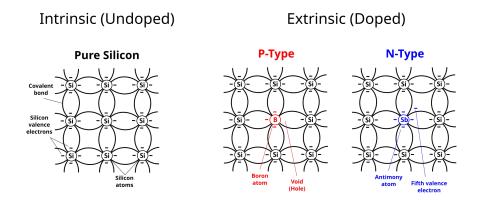
We can see that silicon has 4 electrons in the valence shell(outer most shell), since the count from left to right on the third row is 4.



2 Doping

Useful Resource: Intrinsic and Extrinsic semiconductors Si^+ means that a silicon atom lost an electron, so only 3 electrons in the valence shell, Vice versa for Si^- .

Doping is inserting different neutral atoms into a silicon crystal. A crystal structure is when atoms are nicely lined up, repeated arrangement of atoms, just like the picture below.



From the periodic table horizontal element count, we can see that silicon's valence shell is missing 4 electrons, and it just happens to have 4 free electrons in the valence shell as well. So in a pure silicon crystalline structure the Si atoms will share the 4 electrons. The specifics of covalent bond will not be omitted here. Some common element for doping includes

Acceptors, valence shell have 1 less(than 4) electron, aka trivalent impurities. This forms P-type extrinsic semiconductor:

- Boron(B, 5)
- Aluminum(Al, 13)
- Gallium(Ga, 31)

Donors, valence shell have 1 more(than 4) electron, aka pentavalent impurities. This forms N-type extrinsic semiconductor:

- Phosphorus(P, 15)
- Arsenic(As, 33)
- Antimony(Sb, 51)

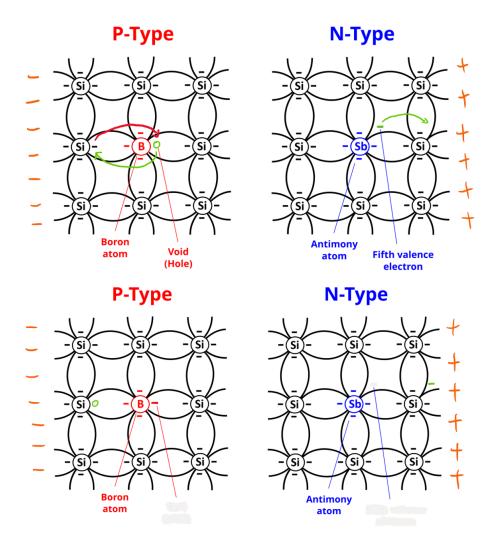
We can see the positions of these elements relative to silicon's position on the periodic table. Acceptors to the left, donors to the left. Note that due to reasons like thermal ionization and particle bombardment, the minority carriers are still there. For example, in N-type semiconductor material, there are still holes in the crystal. Vice versa for P-type semiconductor.

3 Electron & Hole Movement and Conventional Current

Useful Resource: Electron vs Hole Flow Video

Remember that conventional current is the opposite direction as the electron moving direction.

In P-type semiconductor, holes are the majority current carrier whereas in N-type semiconductor, electrons are the majority current carrier. In both types, the conventional current flow and electron flow directions are same. See the movement of electron in the following images:



The orange label are just indicators of electric field, showing that free electron flow is from left to right in both semiconductors. This shows that **conventional current in both semiconductors are from right to left**. Holes move in the same direction as conventional current.

This is why the source terminal in N and P type MOSFETs are called source because they are the source of charge carriers

