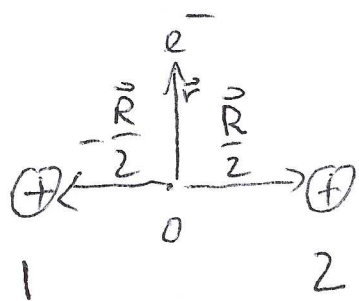


$e^- - e^-$ interaction prevents us from solving H_2 analytically.

→ find approximation methods

Linear combinations of Atomic Orbitals (LCAO)

H_2^+



$$\psi(\vec{r} + \frac{\vec{R}}{2}) = a(\vec{r})$$

$$\psi(\vec{r} - \frac{\vec{R}}{2}) = b(\vec{r})$$

$$\psi_{MO} = c_a a(\vec{r}) + c_b b(\vec{r})$$

should be normalized:

$$\begin{aligned} \int |\psi_{MO}|^2 d\vec{r} &= c_a^2 \underbrace{\int |a(\vec{r})|^2 d\vec{r}}_1 + c_b^2 \underbrace{\int |b(\vec{r})|^2 d\vec{r}}_1 \\ &\quad + 2 c_a c_b \underbrace{\int a(\vec{r}) b(\vec{r}) d\vec{r}}_{\text{overlap integral } S_{ab}} = 1 \end{aligned}$$

$$\Rightarrow c_a^2 + c_b^2 + 2c_a c_b S_{ab} = 1$$

and for H_2^+ : $|c_a|^2 = |c_b|^2$

$$c^2 + c^2 + 2c^2 S_{ab} = 1$$

$$\approx c = \frac{1}{\sqrt{2 \pm 2S_{ab}}}$$

$$\psi_+ = \frac{1}{\sqrt{2+2S}} (\vec{a}(\vec{r}) + \vec{b}(\vec{r}))$$

symmetric

$g = \text{grade (even)}$

$$\psi_- = \frac{1}{\sqrt{2-2S}} (\vec{a}(\vec{r}) - \vec{b}(\vec{r}))$$

antisymmetric

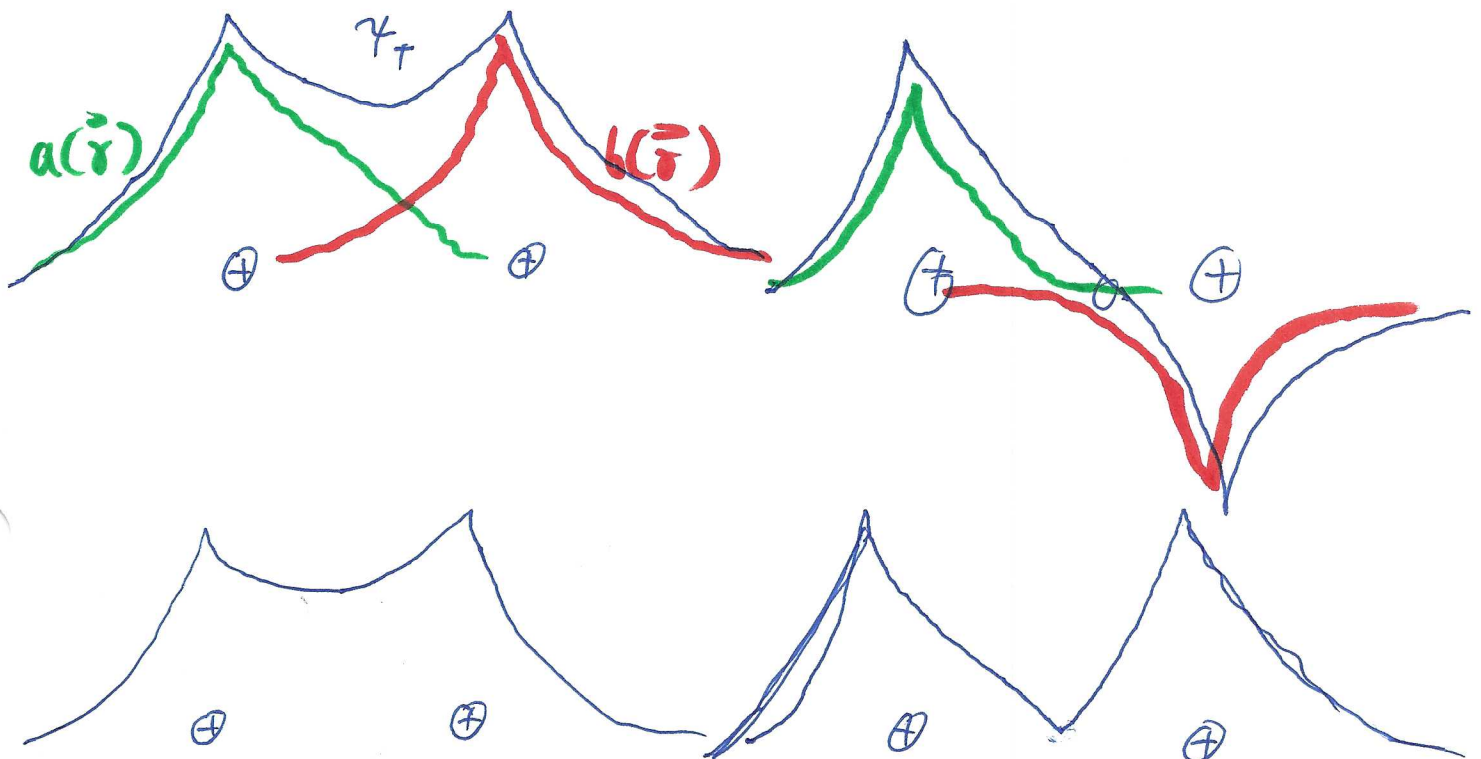
$u = \text{ungrade (odd)}$

$$\psi_+ = \frac{1}{\sqrt{2+5}} (a(\vec{r}) + b(\vec{r}))$$

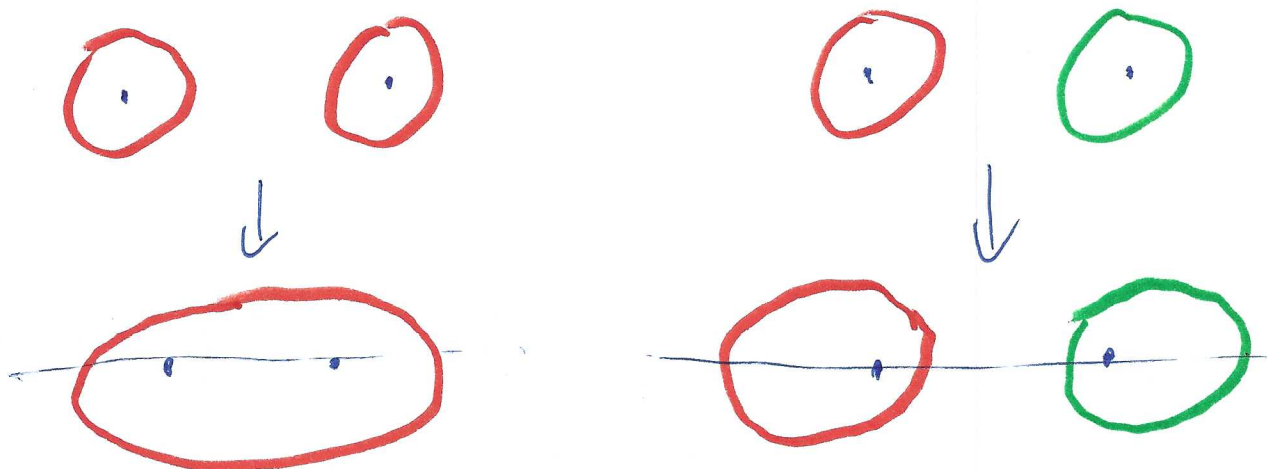
symmetric
g = gerade (even)

$$\psi_- = \frac{1}{\sqrt{2-5}} (a(\vec{r}) - b(\vec{r}))$$

antisymmetric
u = ungerade (odd)



alternative visualization



ψ_+ : increased probability in between nuclei

→ e^- attracted by both nuclei

→ lower in energy than atomic orbital

ψ_- : decreased probability in between nuclei

→ weaker binding

→ higher energy than atomic orbitals

molecular orbital diagrams:

