

RAYALASEEMA UNIVERSITY COLLEGE OF ENGINEERING, KURNOOL – 518007

B.Tech I Semester (RU23) I Sessional Examinations – Oct 2024

CHEMISTRY (23ASH9901T)

(Computer Science & Engineering)

Time: 90 min

Date: 15/10/2024 AN

Max. Marks: 30

Answer ONE FULL question from each unit

All the Questions Carry EQUAL marks

| Q. No | Question | Unit | BT Level | CO covered | Marks Allotted |
|-------|---|------|----------|------------|----------------|
| 1 | a) Derive Schrodinger wave equation for particle in one dimensional box. | I | L5 | CO1 | (8M) |
| | b) State Planks quantum theory | I | L1 | CO1 | (2M) |
| (OR) | | | | | |
| 2 | a) Draw the molecular orbital diagrams for O ₂ and CO molecules. | I | L5 | CO1 | (8M) |
| | b) Calculate the bond order for NO molecule. | I | L3 | CO1 | (2M) |
| 3 | a) Draw the molecular orbital diagram for π conjugated molecule | II | L5 | CO1 | (8M) |
| | b) Write the significance of ψ & ψ^2 ? | II | L1 | CO1 | (2M) |
| (OR) | | | | | |
| 4 | a) What is meant by super capacitor? Give it classification and applications. | II | L1 | CO1 | (8M) |
| | b) Write any four applications of semi conductors | II | L1 | CO1 | (2M) |
| 5 | a) Explain Band theory in detail with neat diagram. | III | L5 | CO2 | (8M) |
| | b) Define Super conductor. | III | L1 | CO2 | (2M) |
| (OR) | | | | | |
| 6 | a) Write short notes on CNT & Fullerenes. | III | L5 | CO2 | (8M) |
| | b) Define nano material. Give an example. | III | L1 | CO2 | (2M) |

| Q. No | Question | Unit | BT Level | CO covered | Marks Allotted |
|-------|---|------|----------|------------|----------------|
| 1 | a) Explain about electrochemical sensors. | III | L5 | CO3 | (8M) |
| | b) Define conductance. | III | L1 | CO3 | (2M) |
| (OR) | | | | | |
| 2 | a) Explain the working of Lithium ion battery. | III | L5 | CO3 | (8M) |
| | b) Primary batteries are not reused but secondary batteries are reused. Justify it. | III | L4 | CO3 | (2M) |
| 3 | a) Illustrate the preparation, properties and applications of following polymers i) Bakelite ii) Nylon 6, 6 | IV | L5 | CO4 | (8M) |
| | b) Thermoelastic plastics are recyclable but thermosetting plastics are not recyclable. Explain it? | IV | L5 | CO4 | (2M) |
| (OR) | | | | | |
| 4 | a) Explain mechanism of conduction and applications of polyacetylene as conducting polymer. | IV | L5 | CO4 | (8M) |
| | b) Write the preparation of polyethylene. | IV | L1 | CO4 | (2M) |
| 5 | a) Draw the schematic diagram for IR spectrophotometer. Explain its principle and applications IR spectroscopy. | V | L5 | CO5 | (8M) |
| | b) Draw the electromagnetic spectrum. | V | L5 | CO5 | (2M) |
| (OR) | | | | | |
| 6 | a) Demonstrate the principle, types and applications of HPLC | V | L1 | CO5 | (8M) |
| | b) Define Beer-Lamberts Law. | V | L1 | CO5 | (2M) |