

given 10% SO<sub>2</sub>, 90% air, 1 atm total pressure., 30°C

$$\text{Average molecular weight} = \frac{(0.1)(64) + (0.9)(28.8)}{(SO_2) \quad (air)} = 32.3 \text{ g}$$

Resolving Ring 1/2 side or 50 mm.

Step 1: Assuming ideal gas

$$PV = nRT$$

$$PV = \frac{m}{(M_w)_{\text{avg}}} RT$$

$$\frac{P(M_w)_{\text{avg}}}{RT} = \frac{m}{V} = \rho_g \quad (\text{gas density})$$

$$\rho_g = \frac{(P_A) \times 101325 \times 32.3}{R(30+273)} = 1.298 \text{ kg/m}^3.$$

8.314 J/molK

Assumption Moi is pure (if not given)

take  $\rho_w = 1000 \text{ kg/m}^3$  always as the temp. doesn't fluctuate as much in questions. ( $\approx 20-30^\circ\text{C}$ )

to  $\frac{\text{Flow rate}}{\text{wg. wt}} = \text{Molar flow rate.}$

$$q_s = q_i (1 - y_1) = q_{ol} (1 - y_1)$$

$$k_s = L_o (1 - x_1) = L_i (1 - x_2)$$

$$q_s (y_1 - y_2) = L_s (x_1 - x_2)$$

$$L_s \times 18 \rightarrow \text{kg/hr (min)}$$

$x_1 = \frac{y_1}{1-y_1}$

$$= L_i + (SO_2) \times 64 \text{ kg/hr}$$

$$q_i = 1500$$

$$L_o = 37525$$

$$y_1 = 0.1$$

Step 2

fluid prop  
liquid ~  
vap dens

Step 3: flooding  
→ for ale  
→ for st

now its cel

L =

(100) →

=)

=)

→ for

→ for

→ Now,

how

$$G_i = 1500 \text{ kg/hr}$$

$$L_0 = 37525 \text{ length (cm)}$$

$$\gamma_f = 0.1$$

Fig 2 fluid properties  
 liquid viscosity  $\mu_L = 0.81 \text{ cp}$  (if not given we have to find out  
 air density  $\rho_L = 1000 \text{ kg/m}^3$  from the viscosity w/o table)

- Fig 3 flooding line  
 → for absorption, take bottom  $L_0 \propto G_i$   
 → for strapping take upper  $L' \propto G_i$  (less likely to occur).

Now its absorption (given in question)

$$L' = L_0 \quad c_1' = c_1$$

$$\Rightarrow \left( \frac{L'}{c_1'} \right) \left( \frac{\rho_A}{\rho_L} \right)^{0.5}$$

$$\Rightarrow \left( \frac{37525}{1500} \right) \left( \frac{1.298}{1000} \right)^{0.5} = 0.9 \rightarrow \text{this is the value of } n \text{ axis in fig 5.33 & 5.34.}$$

→ for Rasching Ring } 1st gen. use Fig 5.33  
 Berl ring } APDC (generalized pressure drop correlation)

→ for packing } 2nd gen use Fig 5.34

→ Now, draw perpendicular and find the pt where it intersects flooding line

here for  $n = 0.9 \quad y = 0.025$ .

$$x_1 = \frac{y_1}{1-y_1}$$

$$\log P_h$$

using no% safety/ stability criteria

$$y = f_p \times 70\% = 0.02C \times 0.7 \\ y = 0.017C$$

Now draw a horizontal line from  $y = 0.017C$  where it intersects  $x = 0.9$  line

→ for verifying

$$(p_f/l) = 0.115 f_p^{0.7}$$

↳ the only value of  $y$

$$f = 0.0175$$

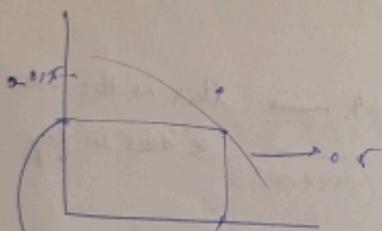
$$\alpha = 0.9$$

↳ packing factor

↳ for Packing → Table 5.9

and then find corresponding coordinate value of

$$y' = 0.016$$



$$y' = 0.016$$

for  $x = 0.9$  so curve

$$\left( \frac{1.1}{3} \right) \left( \frac{1.2}{3} \right)$$

which

is best

fit is  $(4/4)$

$\Rightarrow 0.5$

$$y' = (a') f_p (g_w/g_L) U_L^{0.2}$$

$$\underbrace{g_w g_L}_{16 \text{ ft}^3} \rightarrow$$

we'll find  $a'$ .

$$U_L = 0.81 C_p$$

$$f_p = 94 \text{ or } f_p = 1$$

$$g_w = 9.81 \text{ m/s}^2 = 4.18 \times 10^8 \text{ ft/hr}^2$$

we'll get

$a' =$

next step

Pack

$$C_p = (4/4) \text{ m}$$

$a'$

$y'$

$if$

$$(56/4) \cdot 0.2 = 28$$

$$(16/4) \cdot 1/4$$

$$(40/4) \cdot 1/4$$

villante

to g

no angle

3/4

we'll get  $G'$  from this

$$G' = 607.6 \text{ lb/ft}^2 \text{ hr} \quad (\text{from here})$$

$$G = G' = 1500 \text{ kg/hr} = 1500 / 0.4536$$

$$= 3306.87 \text{ lb/ft}^2 \text{ hr}$$

$$\text{area} = \frac{G}{G'} = \frac{3306.87}{607.6} = 5.44 \text{ ft}^2$$

$$\text{area} = \frac{\pi D^2}{4} = 5.44 \text{ ft}^2 \Rightarrow D = 2.63 \text{ ft}$$

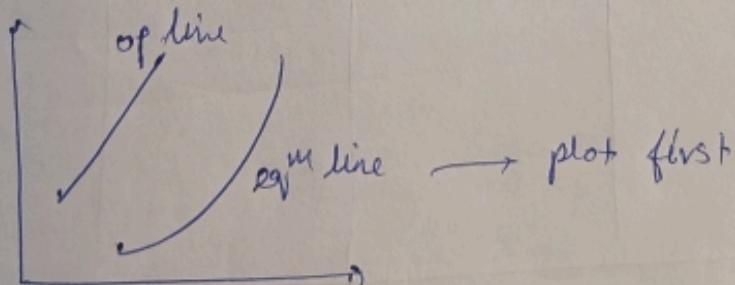
next step Packed bed height

$$h_p = (r_p - r_b) \int_{y_1}^{y_2} \frac{G' \text{ dy}}{k_{eq} (1-y)(y-y_1)}$$

$G'$  = gas flow rate per unit area ( $\text{lb/ft}^2 \text{ hr}$ )

$y_1$  = mol frac. of solute at interface ( $\text{eq}^M$  line)

if  $k_{eq}$  &  $k_{eq}$  not given find using 8h formula



to generate op. line

P/T

0.14

calculation of mol fraction of  $\text{SO}_2$  at exit of gas

$$\text{moles} = \frac{11 \times 100}{32.3} = 46.44 \text{ kmol/hr}$$

$$y_2 = \frac{0.14}{41.79}$$

$$\text{SO}_2 \text{ inlet } 10\% \text{ of total mass} = 4.64 \text{ kmol/hr}$$

$$90\% \text{ air} = 41.79 \quad (46.44 \times 0.9)$$

Outlet

97% removed

0.0027

$$\text{SO}_2 \text{ conc' (moles)} = 4.64(1 - 0.97) = 0.14 \text{ kmol/hr}$$

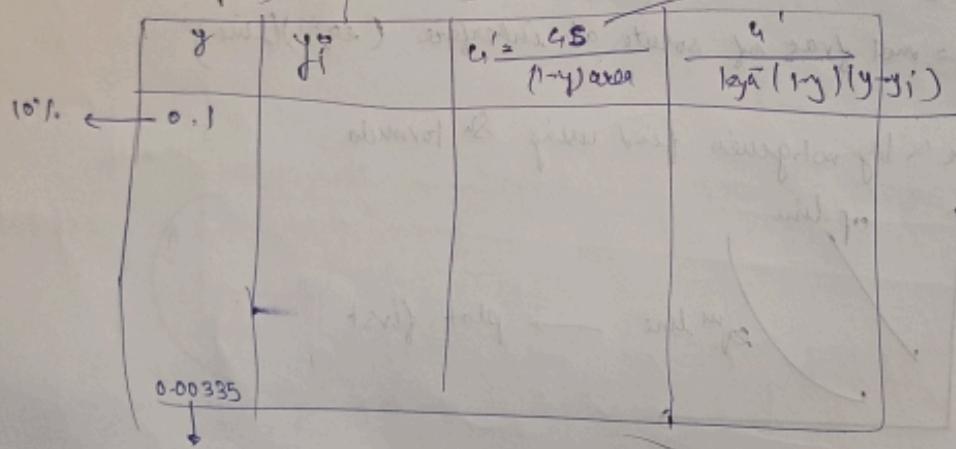
$$\text{SO}_2 \text{ removed & added to liquid} = 4.64(0.97) = 4.5$$

$$\text{air will remain const} \rightarrow q_s = 41.79 \text{ kmol/hr}$$

(air pure)

(Next) slope ka value

will be const



97% added

3% balance

$$q_s / (y_1 - y_2) = L_s (y_1 - x_2)$$

$$q_s = q_{10} (n y_2)$$

$$q_{10} (1/y_1)$$

~~$$q_{10} (n y_1)$$~~

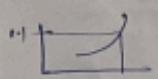
kmol/hr

or

depends on  
k<sub>y2</sub>

$$\rightarrow \text{as } y_2 = 0.1$$

$$q_s (0.1) = \frac{y_1}{1-y_1} = L_s (x_1 - x_2)$$



0.0033

0.0027

not

0.0027

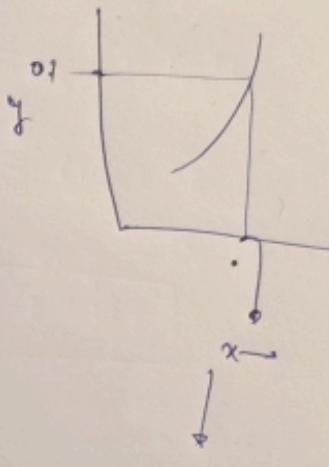
x<sub>2</sub> = 0 cuz water

put y<sub>2</sub>

$$q_s \left( \frac{y_1}{1-y_1} - 0.0033 \right) = L_s (x_1 - u_2)$$

↓  
inlet

initial



① get  $x$

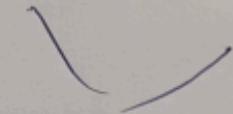
② put in eqn

③ get  $L_s$

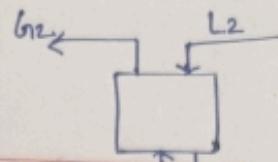
NOW  $L_s \times 1.25$

$$q_s \left( 0.111 - \frac{y}{1-y} \right) = L_s \times 1.25 \left( 0.00215 - \frac{u}{1-u} \right)$$

↓



$x \propto y$



Date / /

$$\underline{G_1} \Rightarrow G_1 = 515 \text{ kmol/h}$$

$$y_1 = 0.015$$

$$\text{Solute in the feed} = 515(0.015) = 7.8 \text{ kmol/h}$$

$$\text{carrier gas} = 515 - 7.8 = 507.2 \text{ kmol}$$

$$\text{moles solute absorbed} = \frac{7.8}{7.8 + 507.2} = 7.799 \text{ kmol/h}$$

$$x_1^* = y_1 / 1.07 = 0.01402 \quad (\text{For min. liq. rate})$$

$x_1^*$  lies on equi. line

$$y_2 \approx 0.0001 \approx 0 \quad x_2 = 0 \quad (\text{Pure Water})$$

$$G_1 y_1 - G_1 y_2 = L_1 x_1^* - L_2 x_2$$

$$L_1 = G_1 y_1 = 515 \times 0.015 = 550.99 \approx 551$$

$$L_1 = L_S = L_2 \quad L_2 = L_S = L_1 (1 - x_1^*)$$

$$L_2 = L_S = 543.3 \text{ kmol/h}$$

$$\text{Actual } L_S = (1.3) \times (L_S)_{\min} = 1.3 \times 543.3 = 706.3$$

$$\begin{aligned} \text{Total liquid rate at bottom} &= [L_S + \text{Moles Solute Absorbed}] \\ &= 706.3 + 7.799 \\ &= 714.099 \text{ kmol/h} \\ &= 12853.782 \text{ kg/h} \end{aligned}$$

$$\Rightarrow F_L = \frac{L_1}{G_1} \left( \frac{P_A}{P_L} \right)^{0.5} = 0.0947$$

take  $\frac{\Delta P}{L} = 0.5$  If not given

$$\text{From Figure 5.34 } y_1 \text{-axis} = 0.92 = C_S F_P^{0.5} (V_L)$$

$$F_P = 27 \text{ ft}^{-1} \quad V_L = \frac{\mu}{P} = \frac{0.86 \times 100}{996} = 0.86$$

$C_S$  = we will find  $C_S$

$$C_S = \frac{4a}{\mu} \left[ \frac{P_A}{P_L - P_A} \right]^{0.5}$$

we will find  $4a$

Date / /

$$\text{tower Area} = \frac{G_1}{P_G \times H_G} \cdot \text{lb/s}$$

$$\frac{\pi}{4} d^2 = (\text{area}) \Rightarrow d = \underline{1} \text{ ft}$$

$$H_L = \frac{L_1}{P_L \times \text{area}} = \underline{\text{bed height}}$$

$$(P_L \times \text{area}) \approx 100 \text{ lb/in}^2 \cdot 1 \text{ ft}^2$$

$$Re_L = Re_L \cdot Sc_G \cdot Sc_L^{1/2} \cdot We_L \cdot Fr_L$$

$$(1000 \text{ s/m}^2) \approx 1000 \cdot 0.01 = 10$$

$$h_L =$$

$$K_L = (12)^{1/6} C_L \left( \frac{U_L \cdot D_L}{h_L \cdot L_1} \right)^{1/2} \cdot 10^{-1} \quad L_1 = 4 \text{ ft}$$

$$C_L = C_D \left( \frac{a_p}{L_1} \right)^{1/2} \frac{D_G}{(\epsilon - h_L)^{1/2}} (Re_L)^{3/4} (Sc_G)^{1/3}$$

$$\bar{a}_{eff} = 3(\epsilon)^{0.5} (Re_L)^{-0.2} (Fr_L)^{-0.45} (We_L)^{0.75}$$

$$a_p = 0.847 \times 84 = 71 \text{ ft}^2$$

$$k_x = k_L \times C$$

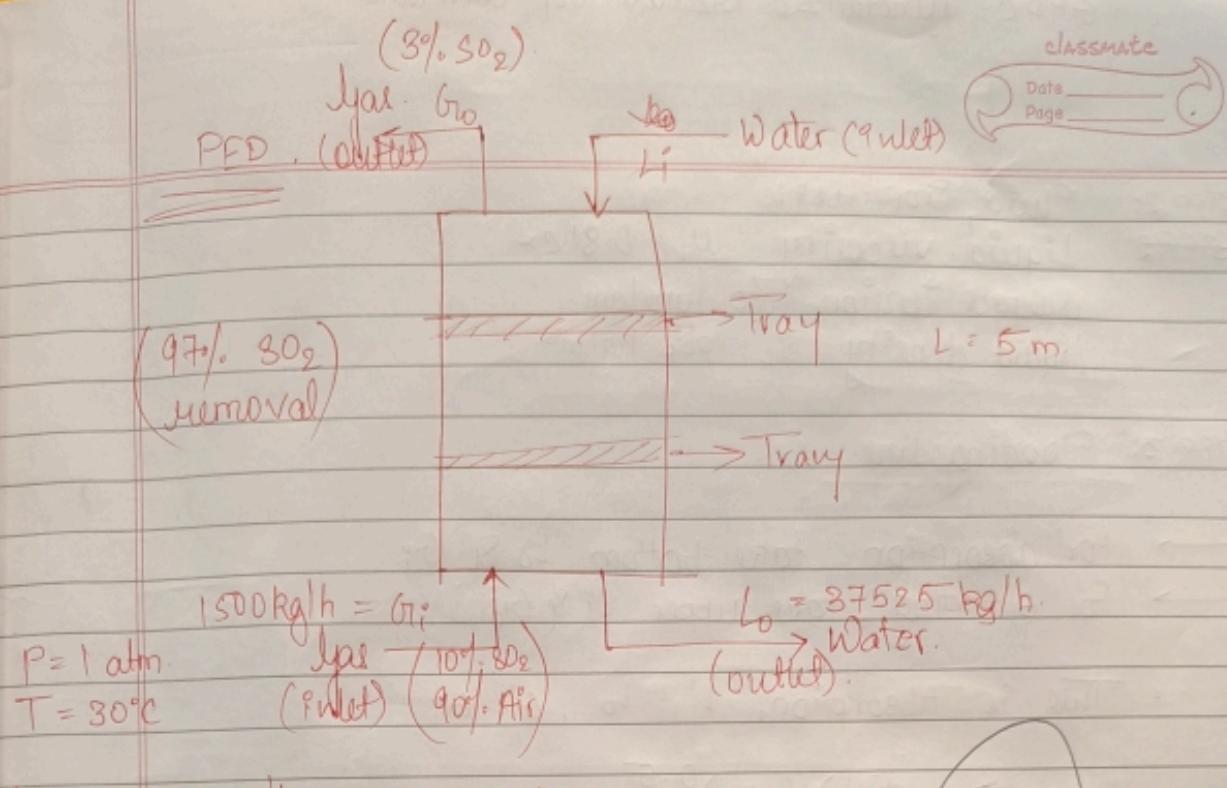
$$C = \underline{\text{Liquid at bottom area}} \times 4_L$$

$$H_{tG} = \frac{G_1'}{k_y \bar{a}_{eff}} = \frac{G_1 \cdot 1000}{\text{area} \times k_y \bar{a}_{eff}} = \underline{55 \text{ ft}} = \underline{55 \text{ ft}}$$

$$H_{tL} = L_1$$

$$S = \frac{m G_1}{L} \times 28.0 = N = 1.37 \text{ ft/s}$$

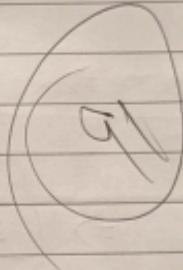
$$H_{tG} = H_{tG} + S H_{tL}$$



Given: For  $1\frac{1}{2}$  Raetig Rings.

To find:

- i) Diameter = ?
- ii) Height = 5 m.
- iii) Blower  $\leftarrow$  ?
- iv)  $\Delta P$  = ?



STEP 1: Assuming ideal gas law, [Gas density & molecular weight]

$$PV = nRT$$

$$PV = m RT \Rightarrow \frac{P \cdot (M_w)_{avg}}{RT} = \frac{m}{V} = \rho_g \text{ (density)}$$

NOTE  $(M_w)_{avg} = \sum_{i=1}^n y_i M_{w,i} \Rightarrow (M_w)_{avg}' = (0.9 \times 28.8) + (0.1 \times 64) = 32.3 \text{ g.}$

STEP 2:  $\rho_g = \frac{10132.5 \times 32.3}{R(30+273)} = 1.298 \text{ kg/m}^3$

$$\begin{aligned} \rho_g &= 1.298 \text{ kg/m}^3 \\ \rho_l &= 1000 \text{ kg/m}^3 \end{aligned}$$

# GPDC - Generalized Pressure drop correlation.

classmate

Date \_\_\_\_\_

Page \_\_\_\_\_

**STEP 2:** Fluid Properties:

Liquid viscosity  $\mu_L = 0.81 \text{ cP}$   
 Surface tension  $70 \text{ dynes/cm}$ .  
 Liquid density  $\rho_L = 1000 \text{ kg/m}^3$ .

**STEP 3:** Flooding Line:

- For absorption, take bottom  $L_0$  &  $G_0$ .
- For stripping, take upper  $L_i$  &  $G_i$ .

• This is Absorption,  $L' = L_0$ ,  $G' = G_i$ .

$$\Rightarrow \left( \frac{L'}{G'} \right) \left( \frac{\rho_G}{\rho_L} \right)^{0.5}$$

$$\Rightarrow \left( \frac{37625}{1500} \right) \left( \frac{1.298}{1000} \right)^{0.5} = 0.9 \quad \text{--- (i)}$$

• For Raschig Rings  $\Rightarrow$  1<sup>st</sup> generation, use Fig 5-33  
 Berl Saddles Pg. 245  
 (GPDC).

• For Pall Ring  $\Rightarrow$  2<sup>nd</sup> generation, use Fig 5-34  
 (GPDC) Pg. 246.

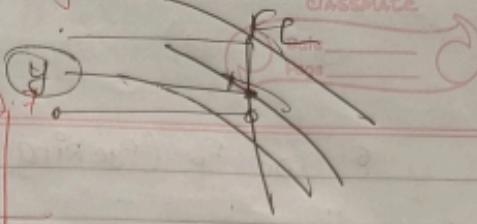
For 1½ Raschig Rings, use Fig. 5.33, & (i) on x-axis.

draw a vertical line till the point where it intersects  
 the flooding line, find max at y-axis.

hence, for  $x=0.9$

$$y=0.025$$

W 70% property to for stability maintaining classmate

$$y = y_F \times 70\% = 0.025 \times 0.7 = 0.0175$$


Now, draw a horizontal line from  $y = 0.0175$  which it intersects ' $x = 0.9$ ' line

\* To select the correct or most accurate curve:

Use  $\frac{(\Delta P/L)_{fl}}{F_p} = 0.115 F_p^{0.7}$  → for verification  
 only value of  $F_p$  Factor from table

$F_p$ : Packing Efficiency;  $(\Delta P/L)_{fl} \rightarrow$  value of different curves,  
 & choose the closest curve

OR

choose the curve closest to that point on the graph

For  $y = 0.0175$  &  $x = 0.9$

closest curve = 0.5

so, now, for  $x = 0.9$ , draw a straight line till curve 0.5 ( $= \Delta P/L$ )

and then find the corresponding y-coordinate value at

for  $x = 0.9$  & curve 0.5 ( $\Delta P/L = 0.5$ )

$$y' = 0.016$$

$$y' = (G')^2 F_p (S_w/S_l) \mu L^{0.2}$$

$\rightarrow 16 \text{ ft}^2 \text{ hr}$   $\rightarrow 0.81 \text{ cp}$   
 $S_w/S_l \text{ g.c.}$   $\rightarrow 4.18 \times 10^8$   
 $16 \text{ ft}^3$   $\rightarrow (\text{ft}^3/\text{hr}^2)$   
 or  $0.81 \text{ m}^3/\text{s}^2$

$S_w$  = density of water

$S_l$  = density of liquid

$S_g$  = density of gas

$F_p \rightarrow$  Table 5.9

Pg. 246

$$S_w/S_l = 1 \quad \mu_l = 0.81 \text{ cp} \quad F_p = 94.5 \text{ ft}^{-1}$$

$$g_c = 9.81 \text{ m/s}^2$$

\* For  $F_p$  (Packing Factor):

Rauchig Ring 1 1/2 inch

Table 5.9, Page 847, find  $F_p$

OR if not found use:

Treybal, Table 6.3, Pg. 210: (i) Find the desired Nominal size (packing)

(ii) Find  $C_p$  (Packing Factor)

$$F_p = C_p = 95 \text{ (from Treybal)}$$

STEP 4: DIAMETER ( $D$ )

$$y^* = \frac{(G')^2 F_p (S_w S_l) \mu_L^{0.2}}{g_s S_a S_l g_c}$$

$\frac{kg}{m^3} \cdot \frac{m}{m^2 \cdot m} \text{ out } G' = 607.6 \text{ lb/ft}^2 \cdot h = 0.4536 \text{ kg/m}^3 \cdot h$   
 $\text{in } G_i = 1500 \text{ kg/m}^3 = 1500 / 0.4536 = 3306.87 \text{ lb/ft}^3$

$$\text{Area} = \frac{G_i}{G'} \Rightarrow \frac{3306.87}{607.6} \Rightarrow 5.44 \text{ ft}^2$$

$$\text{Area} = \frac{\pi D^2}{4} = 5.44 \text{ ft}^2 \Rightarrow [D = 2.68 \text{ ft}]$$

STEP 5: Packed Bed Height:

$$h_T = \int_{y_1}^{y_2} \frac{G' dy}{k_y \bar{a} (1-y)(y-y_1)}$$