

$$3P\dot{R} + R\dot{P} = -3R_u T k_c H P.$$

Var, $P = P_A + P_L g z$

$$\frac{dP}{dt} = BR$$

$$\dot{P} = P_L g \dot{z}$$

$$= P_L g \beta R$$

Substituting it into main equation:-

$$3P\dot{R} + R(P_L g \beta R) = -3R_u T k_c H P$$

$$\Rightarrow 3P\dot{R} + P_L g \beta R^2 = -3R_u T k_c H P.$$

Solve for $\dot{R} \Rightarrow$

$$3P\dot{R} = -3R_u T k_c H P - P_L g \beta R^2$$

$$\frac{dR}{dt} = -R_u T k_c H - \frac{P_L g \beta R^2}{3P}$$

Eliminating time:-

$$\frac{dR}{dP} = \frac{\dot{R}}{\dot{P}} = \frac{-R_u T k_c H - \frac{P_L g \beta R^2}{3P}}{P_L g \beta R}$$

Given Problem:-

For Physical Vapor Deposition:-

Assumptions:-

- 1D growth:- Crystal grows only in the x direction from $x=0$ to $x=L(t)$.
- Instantaneous crystallization:- every molecule arriving at the interface adheres with 100% efficiency.
- Isothermal interface:- Interface temperature $T_m(t)$ is prescribed.
- Constant properties
- Negligible vapor-side resistance: mass flux $J_m(t)$ is taken as known (or shaped into an empirical growth rate).

Governing Equations:-

- Heat conduction of the solid:-

$$\rho_s c_s \frac{\partial T}{\partial t} = k_s \frac{\partial^2 T}{\partial x^2}, \quad 0 < x < L(t)$$

- Stefan (energy) condition at $x = L(t)$

$$\rho_s \Delta H \frac{dL}{dt} = -k_s \left. \frac{\partial T}{\partial x} \right|_{x=L(t)}$$

Since, $u_0 = R_0^2$ constant

$$R^2 p^2 z = R_0^2 p_0^2 z - \frac{6A}{5} [p^{s_3} - p_0^{s_3}]$$

Disappearance condition $R \rightarrow 0$:-

Set $R=0$ at $p = p_{end}$,

$$0 = R_0^2 p_0^2 z - \frac{6A}{5} [p_{end}^{s_3} - p_0^{s_3}]$$

$$\Rightarrow p_{end}^{s_3} = p_0^{s_3} + \frac{5R_0^2 p_0^2 z}{6A}$$

$$\Rightarrow p_{end} = \left(p_0^{s_3} + \frac{5R_0^2 p_0^2 z}{6A} \right)^{\frac{1}{s_3}}$$

$$z_{end} = \frac{p_{end} - p_A}{PLg}$$