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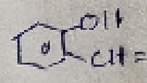
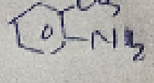
Boiling point data of some organic compounds:

Formic Acid: 100°C.
Acetic Acid: 118°C.
Methyl Alcohol: 65°C.
Ethyl Alcohol: 78°C
Benzene: 80°C
Toluene: 110°C
Cyclohexane: 80°C.
p/m/o-xylene: 137 °C / 139°C / 142°C.
Acetaldehyde: 21°C
Propanal: 49°C.
Acetone: 56°C
Benzaldehyde: 179°C.
Acetophenone: 202°C

: 196°C
Ethylamine: 18°C.
Diethylamine: 55°C.
Triethyl amine: 89°C.
Aniline: 183°C.

: 197°C.
Nitromethane: 101°C.
Nitroethane: 114°C.
Nitrobenzene: 210°C.

Phenol: 181°C.
o-Cresol: 190°C
p-Cresol: 201°C.
methyl Salicylate: 224°C.
n-butyric acid: 163°C.
Formaldehyde: 98°C.
Pentaerithritol: 253°C
Glycerol: 290°C.
Ethylene glycol: 197°C.
cyclohexanol: 160°C.
Formaldehyde: −19°C.
CH_3Br: 3.56°C.
CH_3I: 42°C.
CH_3F: −78.4°C.

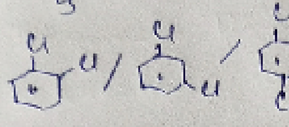
: 370°C.

Boiling point of different halides / organic compound (NCERT).

a) $CH_3-CH_2CH_2CH_2Br$: 375K.

b) $CH_3-CH_2-\underset{Br}{CH}-CH_3$: 364 K.

c) $H_3C-\overset{CH_3}{\underset{CH_3}{C}}-Br$: 346K.

d)  : 453 / 446 / 444 K.

e) n-butane : 273K.

f) Methoxy ethane : 281K

g) Propanal : 322K.

h) Acetone : 329K.

i) Propan-1-ol : 370K

j) $nC_4H_9-NH_2$: 351K.

k) $(C_2H_5)_2NH$: 329K.

l) $C_2H_5N(CH_3)_2$: 310.5K

m) $C_2H_5CH(CH_3)_2$: 301K.

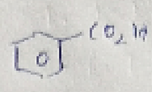
n) nC_4H_9OH : 390K.

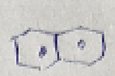
o) CH_3Cl : -24.2°C

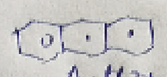
p) CH_2Cl_2 : 39.6°C

q) $CHCl_3$: 61.2°C

r) CCl_4 : 76.72°C

 (Benzoic acid) : 249.2°C

 : Naphthalene : 218°C

 : Anthracene : 340°C

CH_3-O-CH_3 : (Dimethyl ether) : 34.6°C

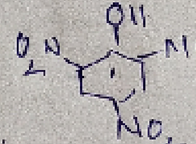
$CH_3-\underset{O}{\overset{\|}{C}}-NH_2$: (Acetamide) : 221.3°C

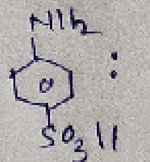
$CH_3-\underset{O}{\overset{\|}{C}}-OH$: (Acetic Acid) : 118°C

CH_3-NH_2 : methyl amine : -6.3°C

$(CH_3)_2NH$: Dimethyl amine : 7°C

$(CH_3)_3N$: Trimethyl amine 2.9°C

 : Picric Acid. 300°C.

 : 183K.

CH_4	111K	C_6H_{14}	342K
C_2H_6	184K	C_7H_{16}	371K
C_3H_8	231K	C_8H_{18}	399K
C_4H_{10}	273K.	C_9H_{20}	423K.
C_5H_{12}	309K.		