-ESSENTIAL COMMON NAMES-

ALKANE

ALKENE

ALKYL HALIDE

4 ClCH=CCl₂ Westrosol or Triclean (Solvent)

ALCOHOL

5
$$CH_2 - OH$$
 Glycol or Ethylene Glycol $CH_2 - OH$

- 7 CH₂=CH-CH₂-OH Allyl Alcohol
- 8 CH₂=CH-OH Vinyl Alcohol

ETHER

9 C_6H_5 -O- CH_3 Anisole (Methyl Phenyl Ether)

KETONE

10 CH₃COCH₃ Acetone CARBOXYLIC ACID

12 HO
$$\operatorname{CH}_2$$
 COOH Glycolic Acid

Aceto Acetic Ester (AAE) or Ethyl Aceto Acetate

N-DERIVATIVES

18
$$NH_2 - C - NH_2$$
 Guanidine \parallel NH

AROMATIC COMPOUNDS

21
$$NH_2$$
 \longrightarrow SO_3H Sulphanilic acid

o-xylene

Terephthalic acid

m-xylene

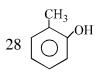
36 C₆H₅CHO

Nitrobenzene (oil of mirbane)

Anthranilic acid (o-aminobenzoic acid)

Orthanilic Acid

HETROCYCLIC COMPOUNDS

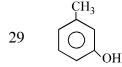


o-Cresol



Pyrrolidine

Benzaldehyde



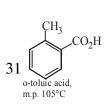
m-Cresol

38 N

Piperidine

$$30 C_6H_5CO_3H$$

Perbenzoic acid

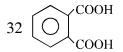


39 0

Tetrahydrofuran (THF)

Toluic acids

O CH₂-CH₂ Oxirane or Ethylene Oxide or



Phthalic acid

Quinuclidine

Oxo Cyclo Propane

Isophthalic acid

Aniline

SOME REAGENTS

- 43 Grignard's reagent RMgX
- 44 NBS N-Bromosuccinimide



POLAR PROTIC SOLVENTS

45 H –O–H Water

46 R-O-H Alcohol

OH

Phenol

48 CH₃-C-OH Acetic acid

49 HF Hydrogen Fluoride

50 NH₃ Ammonia

POLAR APROTIC SOLVENTS

51 DMS Dimethyl sulphide CH₃-S-CH₃

52 DMSO Dimethyl sulphoxide Me₂S=O

53 HMPT Hexamethylphosphoramide

or

HMPTA $O=P-(NMe_2)_3$

54 DMF Dimethyl formamide

 $\begin{array}{c} H-C-NMe_2 \\ \parallel \\ O \end{array}$

55 Crown ethers Cyclic polyethers

(12 - C - 4)

-DESIRABLE COMMON NAMES-

ALKANES

$$\begin{array}{ccc} 2 & -\mathrm{CH_2} - \mathrm{CH_2} - \mathrm{CH} - \mathrm{CH_3} & \text{Isopentyl Group} \\ & & | \\ & & \mathrm{CH_3} \end{array}$$

ALKENES

3
$$CH_3$$
– CH_2 – $CH=CH_2$ α –Butylene

4
$$CH_3$$
– CH = CH – CH_3 β –Butylene

5
$$CH_3 - C = CH_2$$
 Iso Butylene CH_3

ALKYNES

ETHER

$$CH_3CH(OCH_3)_2$$
 Methylal

ALDEHYDE

Glyoxalic acid

CHO

12
$$CH_3 - C - C - CH_3$$
 Dimethyl Glyoxal $\parallel \parallel$ O O

Methyl Glyoxal or Pyruvialdehyde

KETONE

$$14 \sum_{\text{CH}_3}^{\text{CH}_3} \text{C} = \text{CH} - \text{C}_{\text{C}} - \text{CH} = \text{C}_{\text{CH}_3}^{\text{CH}_3} \quad \text{Phorone}$$

15
$$CH_3$$
 $C = CH - C - CH_3$ Mesityl Oxide

CARBOXYLIC ACID

$$\begin{array}{ccc} 18 & {\rm C_6H_5-CH-COOH} & & {\rm Mendalic\ Acid} \\ & & | & & \\ & & {\rm OH} & & \\ \end{array}$$

ACID DERIVATIVES

$$\begin{array}{ccc} 23 & \mathrm{NH_2-C-C-NH_2} \\ & \parallel & \parallel \\ & \mathrm{O} & \mathrm{O} \end{array} \qquad \text{Oxanamide}$$

AROMATIC COMPOUNDS

$$\begin{array}{c}
N = N - O - SO_3^- Na^+ \\
24 O O + OH
\end{array}$$

Orange II

25
$$\langle \bigcirc \rangle$$
 N=N- $\langle \bigcirc \rangle$ NMe₂ Butter Yellow

(Cycloheptatrienolone)

(Cycloheptatrienone)

$$\alpha$$
-naphthol

$$\beta$$
-naphthol

33
$$H_2N$$
—N H_2 Benzidine

Saccharin (o-sulphobenzoic imide)

HETROCYCLIC COMPOUNDS

Hexa-methylenetetramine or Urotropine

41
$$CH_2$$
 CH_2 Oxitane

$$\begin{array}{ccc} 42 & \mathrm{CH_3} - \mathrm{C} - \mathrm{NH_2} & & \mathrm{Amidine} \\ & & || & & \\ & & \mathrm{NH} & & \end{array}$$

SOME 43 LAH	REAGENTS Lithium aluminium	51 TEL		Tetra ethyl lead
	hydride:LiAIH ₄	52 Gillman	n's reagent	$\rm R_2CuLi/[R_2Cu]^-Li^+$
44 SBH	Sodium borohydride NaBH ₄	53 Tollen'	s reagent	alk. sol. of AgNO ₃
45 PCC	Pyridinium chlorochromate	54 Fehling	s's reagent	alk. sol. of CuSO ₄
	$\left[\begin{array}{c} \bigodot \\ \bigodot \\ \bigvee_{N \oplus} \end{array} \right] \operatorname{CrO_3Cl}$	55 Hinsberg's reagent		CH ₃ SO ₂ -Cl
	H J		SOME	GROUPS
46 Raney Nickel	Ni–Al alloy	56 Ts	Tosyl	$CH_3 = \left(\begin{array}{c} O \\ S \\ O \end{array} \right)$
47 Wilkinson's catalyst	Tris(Triphenylphosphine) chlororhodium (I) (PPh ₃) ₃ RH [⊕] Cl ^Θ	57 Ms	Mesyl	O
48 Bayer's reagent	1% dil. alkaline	58 Ac	Acyl	CH ₃ -C-
, ,	aq.sol. of KMnO ₄	59 Bs	Brosyl	O
49 Braddy's reagent2,4 DNP50 Liemieux reagent	$H_2N-NH O_2$ O_2 O_2 O_3 O_4	60 Tf	Triflate	CF ₃ -S-O

NOMENCLATURE OF ORGANIC COMPOUND AND COMMON NAMES

EXERCISE # O-I

- Q.1 How many 1°carbon atom will be present in a simplest open chain hydrocarbon having two 3° and one 2° carbon atom?
 - (A) 3

- (B)4
- (C) 5
- (D) 6

NC0001

- Q.2 Alicyclic compounds are:
 - (A) Aromatic compounds

(B) Aliphatic cyclic compounds

(C) Heterocyclic compounds

(D) None of the above

NC0002

- Q.3 How many 1°, 2°, 3° C atoms does 1, 3, 5-Trimethyl cyclohexane have?
 - (A) 3, 6, 0
- (B) 3, 4, 2
- (C) 0, 3, 6
- (D) 3, 3, 3

NC0003

- Q.4 The compound which has one isopropyl group is:
 - (A) 2,2,3,3-Tetramethyl pentane
- (B) 2,2-Dimethyl pentane

(C) 2,2,3-Trimethyl pentane

(D) 2-Methyl pentane

NC0004

- Q.5 Which of the following is the first member of ester homologous series?
 - (A) Ethyl ethanoate

(B) Methyl ethanoate

(C) Methyl methanoate

(D) Ethyl methanoate

NC0005

- Q.6 A group closely related compounds which can be expressed by a general formula & in which two consecutive members differ by 14 in their molecular masses is called
 - (A) a heterogeneous series

(B) a homologous series

(C) a homogeneous series

(D) a electrochemical series

NC0006

Number of secondary carbon atoms present in the above compounds are respectively:

- (A) 6,4,5
- (B)4,5,6
- (C) 5,4,6
- (D) 6,2,1

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- Q.8 The molecular formula of the first member of the family of alkenynes and its name is given by the set
 - (A) C_3H_6 , Alkene

(B) C_5H_6 , Pent-1-en-3-yne

(C) C_6H_8 , Hex-1-en-5-yne

(D) C₄H₄, Butenyne

NC0008

Q.9 Which of the following is a heterocyclic compound:

$$(A) | HC = CH$$

$$HC = CH$$

$$HC = COOE$$

$$HC = CH$$
 CH_2
 CH_2
 CH_2
 CH_3
 CH_2
 CH_3
 CH_3

NC0009

Q.10 The correct IUPAC name of the compound
$$CH_3 - CH_2 - C = C - CH - C - CH_2 - CH_2 - CH_3 : C_2H_5$$

- (A) 5-Ethyl-3, 6-dimethyl non-3-ene
- (B) 5-Ethyl-4, 7-dimethyl non-3-ene
- (C) 4-Methyl-5, 7-diethyl oct-2-ene
- (D) 2,4-Ethyl-5-methyl oct-2-ene

NC0010

Q.11 The IUPAC name of
$$CH = CH - CHCH_2CH_3$$
 is: CH_3

- (A) 1-Cyclohexyl-3-methyl pent-1-ene
- (B) 3-Methyl-5-cyclohexyl pent-1-ene
- (C) 1-Cyclohexyl-3-ethyl but-1-ene
- (D) 1-Cyclohexyl-3,4-dmethyl but-1-ene

NC0011

(A) But-2-ene-2,3-diol

- (B)Pent-2-ene-2,3-diol
- (C) 2-Methylbut-2-ene-2,3-diol
- (D) Pent-3-ene-3,4-diol

NC0012

- Q.13 IUPAC name of CH_2 =CH CN is:
 - (A) Ethenenitrile
- (B) Vinyl cyanide
- (C) Cyono ethene
- (D) Prop-2-enenitrile

NC0013

Q.14 The IUPAC name of
$$CH_3 CH_2 - N - CH_2 CH_3$$
 is:

- (A) N-Methyl-N-ethyl ethanamine
- (B) Diethyl methanamine
- (C) N-Ethyl-N-methyl ethanamine
- (D) Methyl diethyl ethanamine

NC0014

E

- Q.15 The IUPAC name of acetyl acetone is:
 - (A) Pentane-2,5- dione

(B)Pentane -2,4-dione

(C) Hexane-2,4-dione

(D)Butane-2,4-dione

NC0015

- Q.16 When vinyl & allyl are joined each other, we get
 - (A) Conjugated alkadiene

(B) cumulative alkadiene

(C) Isolated alkadiene

(D) Allenes

NC0016

True statement for the above compounds is:

- (A) (a) is phenol while (b) is alcohol
- (B) Both (a) and (b) are primary alcohol
- (C) (a) is primary and (b) is secondary alcohol (D) (a) is secondary and (b) is primary alcohol

NC0017

- Q.18 The IUPAC name of the following structure (CH₂)C.C.C.(CH₂)CH(CH₂) is:
 - (A) 3-Methylhex-4-yn-2-ene

(B) 3-Methylhex-2-en-4-yne

(C) 4-Methylhex-4-en-4-yne

(D) All are correct

NC0018

- Q.19 The IUPAC name of the following structure is [CH₃CH(CH₃)], C(CH₂CH₃)C(CH₃) C(CH₂CH₃),
 - (A) 3,5-Diethyl-4,6-dimethyl-5-[1-methylethyl]hept-3-ene
 - (B) 3,5-Diethyl-5-isopropyl-4,6-dimethylhept-2-ene
 - (C) 3,5-Diethyl-5-propyl-4,6-dimethylhept-3-ene
 - (D) None of these

NC0019

Q.20 The correct IUPAC name of $CH_3 - CH_2 - C - COOH$ is:

(A) 2-Methyl butanoic acid

(B) 2-Ethylprop-2-enoic acid

(C) 2-Carboxybutene

(D) None of the above

NC0020

- Q.21 The correct IUPAC name of 2-ethylpent-3-yne is:
 - (A) 3-Methyl hex-4-yne

(B) 4-Ethyl pent-2-yne

(C) 4-methyl hex-2 yne

(D) None of these

Q.22 All the following IUPAC names are correct except:

- (A) 1-Chloro-1-ethoxy propane
- (B) 1-Amino-1-ethoxypropane

(C) 1-Ethoxy-2-propanol

(D) 1-Ethoxy-1-propanamine

NC0022

Q.23 The IUPAC name of the compound CH₃CH = CHCH=CHC=CCH₃ is:

(A)Octa-4,6-diene-2-yne

(B) Octa-2,4-diene-6-yne

(C) Oct-2-yne-4,6-diene

(D) Oct-6-yne-2,4-diene

NC0023

(A) 3-Cyclohexanol Propyne

- (B) 3-[3-Hydroxy Cyclohexyl] Propyne
- (C) 3-Propynyl Cyclohexanol
- (D) 3-(2-propynyl) Cyclohexanol

NC0024

Q.25 The IUPAC name of β -ethoxy- α -hydroxy propionic acid (trivial name) is:

- (A) 1,2-Dihydroxy-1-oxo-3-ethoxy propane
- (B) 1-Carboxy-2-ethoxy ethanol
- (C) 3-Ethoxy-2-hydroxy propanoic acid
- (D) All above

NC0025

Q.26 As per IUPAC rules, which one of the following groups, will be regarded as the principal functional group?

- (A) —C≡C—
- (B) —OH
- (C) —C—
- (D) -C H

NC0026

Q.27 The IUPAC name of the compound

- is
- (A) 4-Prop-1-enyl hepta-1,6-diene
- (B) 4-Propylidene hepta-1,6-diene
- (C) 4-Propenyl hepta-1,6-diene
- (D) 4-[Prop-2-enyl] hepta-1,6-diene

NC0027

Q.28 The IUPAC name of the given compound is:

- (A) 1,1-Dimethyl-3-hydroxy cyclohexane
- (B) 3,3-Dimethyl-1-hydroxy cyclohexane
- (C) 3,3-Dimethylcyclohexanol
- (D) 1,1-Dimethylcyclohexan-3-ol

- (A) 2-Chloro-4-N-ethylpentanoic acid
- (B) 2-Chloro-3-(N,N-diethyl amino)-propanoic acid
- (C) 2-Chloro-2-oxo diethylamine
- (D) 2-Chloro-2-carboxy-N-ethyl ethane

Ph Q.30 The IUPAC name of the compound is $CH_3 - CH - \dot{C}H - NH_2$ CH₃

- (A) 1-Amino-1-phenyl-2-methyl propane
- (B) 2-Methyl-1-phenyl propan-1-amine
- (C) 2-Methyl-1-amino-1-phenyl propane
- (D) 1-Isopropyl-1-phenyl methyl amine

NC0030

Q.31 Which of the following compound is wrongly named?

- (A) CH₃CH₂CH₂CHCOOH
- 2-Chloro pentanoic acid
- (B) $CH_3C \equiv CCHCOOH$ ĊH,

- 2-Methyl hex-3-enoic acid
- (C) CH₃CH₂CH=CHCOCH₃
- Hex-3-en-2-one
- (D) $CH_3 CHCH_2CH_2CHO$; CH_3
- 4-Methyl pentanal

NC0031

Q.32 The correct IUPAC name of the following compound is:

$$\begin{aligned} \mathbf{O} &= \mathbf{C} - \mathbf{C} \mathbf{H}_2 - \mathbf{C} \mathbf{H} - \mathbf{C} \mathbf{H} \mathbf{O} \\ & | & | \\ \mathbf{O} \mathbf{H} & \mathbf{H} - \mathbf{C} = \mathbf{O} \end{aligned}$$

- (A) 3,3-Diformylpropanoic acid
- (B) 3-Formyl-4-oxo-butanoic acid
- (C) 3,3-Dioxo propanoic acid
- (D) 3,3-Dicarbaldehyde propanoic acid

NC0032

Q.33 The correct IUPAC name of compound $CH_3 - CH_2 - C - CH - CHO$ is :

(A) 2-Cyano-3-oxopentanal

- (B) 2-Formyl-3-oxopentanenitrile
- (C) 2-Cyanopentane-1,3-dione
- (D) 1,3-Dioxo-2-cyanopentane

Q.34 IUPAC name of compound

$$\begin{array}{c} \text{CH}_2\text{-CH}_3\\ \text{CH}_3\text{-CH-CH-CH}_3\\ \text{CH}_2\text{-CH}_3 \end{array}$$

(A) 2, 3-diethyl butane

- (B) 2-ethyl-3-methyl pentane
- (C) 3-methyl-2-ethyl pentane
- (D) 3,4-dimethyl hexane

NC0034

Q.35 The IUPAC name of compound
$$CH_3$$
 – C – CH – CH – CH – CH_3 is: CH_3 CHO

- (A) 3,5-Dimethyl-4-Formyl pentanone
- (B) 1-Isopropyl-2-methyl-4-oxo butanal
- (C) 2-Isopropyl-3-methyl-4-oxo pentanal
- (D) None of the above

NC0035

$$\begin{array}{c|c} HO-C=O & CH_3 \\ \hline Q.36 \text{ The IUPAC name of compound } CH_3-C=C-C-H \text{ is :} \\ \hline & | & | \\ NH_2 CI \end{array}$$

- (A) 2-Amino-3-chloro-2-methylpent-2-enoic acid
- (B) 3-Amino-4-chloro-2-methylpent-2-enoic acid
- (C) 4-Amino-3-chloro-2-methylpent-2-enoic acid
- (D) All of the above

NC0036

Q.37 The IUPAC name of the structure is:

- (A) 3-Amino-2-formyl butane-1, 4-dioic acid
- (B) 3-Amino-2, 3-dicarboxy propanal
- (C) 2-Amino-3-formyl butane-1, 4-dioic acid
- (D) 1-Amino-2-formyl succinic acid

NC0037

Q.38 One among the following is the correct IUPAC name of the compound

$$\begin{matrix} & & H \\ | & | \\ \text{CH}_3\text{CH}_2 - \text{N} - \text{CHO} \end{matrix}$$

(A) N-Formyl aminoethane

(B) N-Ethyl formyl amine

(C) N-Ethyl methanamide

(D) Ethylamino methanal

Q.39 The IUPAC name of the structure is:

- (A) 1,2-Dimethyl-Cyclohexane
- (B) 1,6-Dimethyl-Cyclohexene
- (C) 1,2-Dimethyl-Cyclohex-2-ene
- (D) 2,3-Dimethyl-Cyclohexane

NC0039

Q.40 The IUPAC name of C_6H_5CH =CH-COOH is :

(A) Cinnamic acid

- (B) 1-Phenyl-2-carboxy ethane
- (C) 3-Phenyl prop-2-enoic acid
- (D) Dihydroxy-3-phenyl propionic acid

NC0040

Q.41 The IUPAC name of
$$\mathrm{BrCH}_2$$
 – CH – CO – CH_2 – $\mathrm{CH}_2\mathrm{CH}_3$ is:
$$\begin{array}{c} | \\ \mathrm{CONH}_2 \end{array}$$

- (A) 2-Bromomethyl-3-oxohexanamide
- (B) 1-Bromo-2-amino-3-oxohexane
- (C) 1-Bromo-2-amino-n-propyl ketone
- (D) 3-Bromo-2-propyl propanamide

NC0041

Q.42 IUPAC name will be
$$CH_2 - CH - CH_2$$

(A) 1,2,3-Tricyano propane

(B) Propane-1,2,3- trinitrile

(C) 1,2,3-Cyano propane

(D) Propane-1,2,3-tricarbonitrile

NC0042

- (A) 3-Carbonyl methoxy -5- Ethanoyl oxy cyclohexanoic aicd
- (B) 3-Ethanoyl oxy -5- Methoxy carbonyl cyclohexane carboxylic acid
- (C) 5-Ethanoyl oxy -5- Methoxy carbonyl cyclohexanoic aicd
- (D) 3-Methoxy carbonyl -5- Ethanoyl oxy cyclohexane carboxylic aicd

NC0043

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Q.44 The IUPAC name of
$$CH_3 - C - O - CH_2 - C - OH$$
 is:

(A) 1-Acetoxy acetic acid

(B) 2-Acetoxy ethanoic acid

(C) 2-Ethanoyloxyacetic acid

(D) 2-Ethanoyloxyethanoic acid

NC0044

Q.45
$$CH_3 - O - C - CH_2 - COOH$$
O

The correct IUPAC systematic name of the above compound is:

(A) 2-Acetoxy ethanoic acid

- (B) 2-Methoxy carbonyl ethanoic acid
- (C) 3-Methoxy formyl ethanoic acid
- (D) 2-Methoxy formyl acetic acid

NC0045

Q.46 The IUPAC name of
$$\begin{array}{|c|c|}\hline & OH \\ \hline & CH_3 \end{array}$$
 is

- (A) 3-Methyl cyclobut-1-ene-2-ol
- (B) 4-Methyl cyclobut-2-ene-1-ol
- (C) 4-Methyl cyclobut-1-ene-3-ol
- (D) 2-Methyl cyclobut-3-ene-1-ol

NC0046

Q.47 The IUPAC name of
$$O_2N$$
—CHO is OCH_3

- (A) 2-Methoxy-4-nitro benzaldehyde
- (B) 4-Nitro anisaldehyde
- (C) 3-Methoxy-4-formyl nitro benzene
- (D) 2-Formyl-4-nitro anisole

NC0047

Q.48 The IUPAC name of compound

$$\begin{array}{c} \mathrm{H_{3}COOC-CH-COOCH_{3}} \\ | \\ \mathrm{CH_{2}OH} \end{array}$$

- (A) 2-(Hydroxy methyl) methyl propanedioate (B) Methyl-2-(hydroxy methyl) propanedioate
- (C) 2-(Hydroxy methyl) dimethyl propanedioate
- (D) None of these

- (A) 2-Formyl ethanoic propanoic Anhydride (B) 2-Oxo-propanoic prop-2-enoic Anhydride
- (C) Prop-2-enoic-2-formyl propanoic Anhydride (D) 2-Formyl ethanoic prop-2-enoic Anhydride

$$$\rm CH_2-CHO$$$
 Q.50 The IUPAC name of $\rm OHC-CH_2-CH_2-CH-CH_2-CHO$ is:

- (A) 4,4-Di(formylmethyl) butanal
- (B) 2-(Formylmethyl) butane-1, 4-dicarbaldehyde

(C)Hexane-3-acetal-1, 6-dial

(D) 3-(Formylmethyl) hexane-1, 6-dial

NC0050

Q.51 The IUPAC name of
$$COOC_2H_5$$
 is :

- (A) 2-Chlorocarbonyl ethylbenzoate
- (B) 2-Carboxyethyl benzoyl chloride
- (C) Ethyl-2-(chlorocarbonyl)benzoate
- (D) Ethyl-1-(chlorocarbonyl)benzoate

NC0051

Q.52 The IUPAC name of
$$\bigcirc$$
 C – CH $_3$ is:

(A) Phenyl ethanone

(B) Methyl phenyl ketone

(C) Acetophenone

(D) Phenyl methyl ketone

NC0052

Q.53 Structural formula of isopropyl methanoate is:

(B)
$$H-C-O-CH_2-CH-CH_3$$

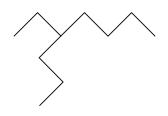
O CH_3

(C)
$$CH_3 - C - O - CH_2 - CH_2$$
 (D) $H - C - O - CH - CH_3$ O CH_3

EXERCISE # O-II

Give the IUPAC names for each of the following:

Q.1



Q.9

NC0062

NC0054

NC0063

NC0064

NC0055

NC0065

NC0056



NC0066

NC0057

NC0067

Q.5

Q.15

Q.17

Q.6

NC0058

NC0059

NC0068

NC0069

Q.7 ОН

NC0070

OH Q.8

NC0060 Q.18

NC0071

Q.19
$$\bigcirc$$
 CH₂

Q.29
$$C$$
-OC₂H₅

$$Q.20 \qquad \stackrel{CH_3}{\frown} ^{C_2H_5}$$

NC0076

NC0078

NC0079

Q.21
$$CH_3$$
 $CH(CH_3)_2$

Q.22
$$CH_2CH_2CH_2CH_3$$

NC0085

NC0084

Q.33
$$CH_3$$
 $CHCH_2CH_3$ NH_2

NC0087

Q.35
$$CH_3$$
 $C-OCH_3$ $C-OCH_3$

NC0088

NC0089

NC0090

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NC0104

Q.45
$$\bigcirc$$
 COOC₂H₅

NC0094 Q.50
$$\longrightarrow$$
 CH₂ -C-CH₂-CH-CH₃ NC0103

NC0098 Q.54
$$\begin{array}{c} \begin{array}{c} \begin{array}{c} \text{NO}_2 \\ \\ \text{COCH}_2\text{--CH--CH=-CH---CH}_3 \end{array} \end{array}$$

EXERCISE # S-I

- Q.1 Which of the following pairs have absence of carbocyclic ring in both compounds?
 - (A) Pyridine, Benzene

(B) Benzene, Cyclohexane

(C) Cyclohexane, Furane

(D) Furane, Pyridine

NC0109

- Q.2 The commercial name of trichloroethene is:
 - (A) Westron
- (B) Perclene
- (C) Westrosol
- (D) Orlone

NC0110

- Q.3 A substance containing an equal number of primary, secondary and tertiary carbon atoms is:
 - (A) Mesityl Oxide
- (B) Mesitylene
- (C) Maleic acid
- (D) Malonic acid

NC0111

Q.4 The IUPAC name of the compound Glycerine $CH_2 - CH - CH_2$ is:

- (A) 1,2,3-Tri hydroxy propane
- (B) 3-Hydroxy pentane-1,5-diol

(C) 1,2,3-Hydroxy propane

(D) Propane-1,2,3-triol

NC0112

- Q.5 Which of the following is crotonic acid:
 - (A) CH₂= CH-COOH

(B) C₆H₅-CH=CH-COOH

(C) CH₃-CH=CH-COOH

(D) CH - COOH

NC0113

- Q.6 The group of heterocylic compounds is:
 - (A) Phenol, Furane

(B) Furane, Thiophene

(C) Thiophene, Phenol

(D) Furane, Aniline

Q.7 Column - I

(Common Name)

- (A) Isooctane
- (B) Neopentane
- (C) Ethylidene chloride (Geminal dihalide)
- (D) Ethylene Dichloride (Vicinal dihalide)

Q.8 Column - I (Common Name)

(A) Acetone

- (B) Acetaldehyde
- (C) Crotonaldehyde
- (D) Acrolein

Column - II

(Structural formula)

(Q)
$$CH_3 - C - CH - CH - CH_3$$

 $CH_3 - CH_3 - CH_3$

(R)
$$CH_3 - CH_2 - CH_2 - CH_2 - CH_2 - CH_3 - CH_3$$

 CH_3

(T)
$$CH_3$$
- CH
 CI

NC0115

Column - II

(Structural formula)

(P)
$$CH_2 = CH - C - H$$

(Q)
$$CH_3 - C - CH_3$$

(R)
$$CH_3 - CH = CH - C - H$$

Q.9 Column - I

(Common Name)

(C)
$$H_2N - CH_2 - COOH$$

Column - II

(Structural formula)

- (P) Lactic acid (In milk)
- (Q) Glyoxal
- (R) Glyceraldehyde
- (S) Glycine
- (T) Glycerol

NC0117

Q.10 **Column - I**

(Common Name)

- (A) Fumaric acid
- (B) Adipic acid
- (C) Maleic acid
- (D) Tartaric acid

Column - II

(Structural formula)

$$\begin{array}{ccc} & \text{HO}-\text{CH}-\text{COOH} \\ & & \mid \\ & \text{HO}-\text{CH}-\text{COOH} \end{array}$$

$$(Q) \begin{array}{c} HC-COOH \\ \parallel \\ HOOC-CH \end{array}$$

$$\begin{array}{ccc} & H-C-COOH \\ & \parallel \\ & H-C-COOH \end{array}$$

(S) COOH(CH₂)₄COOH

Q.11 Column - I

(Common Name)



Column - II

(Structural formula)

- (P) Pyrrole
- (Q) Furan
- (R) Thiophene
- (S) Indol
- (T) Pyridine

NC0119

Q.12 Column - I

(Common Name)

- (A) p-Cresol
- (B) p-Xylene
- (C) Resorcinol
- (D) Quinol
- (E) Catechol

Column - II

(Structural formula)

$$(P) \quad \bigcirc OH \\ OH$$

OH

Q.13 Which of the following is not correctly matched:

(A) Lactic acid

$$\begin{array}{c} \mathrm{CH_3-CH-COOH} \\ \mathrm{OH} \end{array}$$

(B) Tartaric acid

(C) Pivaldehyde

CH₃C(CH₃)₂CHO

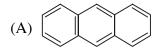
(D) Iso-octane

 $\begin{array}{c} \operatorname{CH_3} - \operatorname{CH} - \operatorname{CH_2} - \operatorname{CH_2} - \operatorname{CH_2} - \operatorname{CH_2} - \operatorname{CH_3} \\ \operatorname{CH_3} \end{array}$

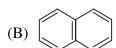
NC0121

Q.14 Column - I

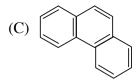
Column - II



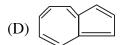
(P) Phenanthrene



(Q) Anthracene



(R) Azulene



(S) Napthalene

EXERCISE # JEE-ADVANCED & MAINS

Q.1 The IUPAC name of the compound having the formula is:

[JEE 1984]

$$\begin{array}{c} CH_3 \\ H_3C - C - CH = CH_2 \\ CH_3 \end{array}$$

- (A) 3,3,3-trimethyl-1-propene
- (B) 1,1,1-trimethyl-2-propene
- (C) 3,3-dimethyl-1-butene
- (D) 2,2-dimethyl-3-butene

NC0123

Q.2 Write the IUPAC name of $CH_3CH_2CH = CH \cdot COOH$

[JEE 1986]

NC0124

- Q.3 The IUPAC name of the compound $CH_2=CH-CH(CH_3)_2$ is:
 - (A) 1,1-dimethyl-2-propene
- (B) 3-methyl-1-butene

(C) 2-vinyl propane

(D) None of the above

[JEE 1987]

NC0125

Q.4 The number of sigma and pi-bonds in 1-butene 3-yne are:

[JEE 1989]

(A) 5 sigma and 5 pi

(B) 7 sigma and 3 pi

(C) 8 sigma and 2 pi

(D) 6 sigma and 4 pi

NC0126

Q.5 Write I.U.P.A.C name of following:

(b)
$$H_3C-N-CH-CH_2CH_3$$

 CH_3 C_2H_5

[JEE 1990]

Me = methyl group

NC0127

Q.6 Write IUPAC name of succinic acid.

[JEE 1994]

- Q.7 The IUPAC name of C_6H_5COCl is
 - (A) Benzoyl chloride

- (B) Benzene chloro ketone
- (C) Benzene carbonyl chloride
- (D) Chloro phenyl ketone
- [JEE 2006]

Q.8 The IUPAC name of the following compound is

[JEE 2009]

- (A) 4-Bromo-3-cyanophenol
- (B) 2-Bromo-5-hydroxybenzonitrile
- (C) 2-Cyano-4-hydroxybromobenzene
- (D) 6-Bromo-3-hydroxybenzonitrile

NC0130

Q.9 The IUPAC name(s) of the following compound is(are):

[JEE 2017]

(A) 4-methylchlorobenzene

- (B) 4-chlorotoluene
- (C) 1-chloro-4-methylbenzene
- (D) 1-methyl-4-chlorobenzene

NC0131

10. The IUPAC name of the following compound is :

[JEE Mains On_line

2018]

- (A) 4-methyl-3-ethylhex-4-ene
- (B) 4,4-diethyl-3-methylbut-2-ene
- (C) 3-ethyl-4-methylhex-4-ene
- (D) 4-ethyl-3-methylhex-2-ene

JEE-Chemistry ALLEN

ΔN	181	N	FI	R-	K	F۱	7
	131			~		_	

Q.1	В	Q.2	В	Q.3	D	Q.4	D	Q.5	\mathbf{C}	Q.6	В	Q.7	A
Q.8	D	Q.9	A	Q.10	A	Q.11	A	Q.12	В	Q.13	D	Q.14	\mathbf{C}
Q.15	В	Q.16	\mathbf{C}	Q.17	D	Q.18	В	Q.19	A	Q.20	В	Q.21	\mathbf{C}
Q.22	В	Q.23	В	Q.24	D	Q.25	C	Q.26	D	Q.27	D	Q.28	C
Q.29	В	Q.30	В	Q.31	В	Q.32	В	Q.33	В	Q.34	D	Q.35	C
Q.36	В	Q.37	C	Q.38	C	Q.39	В	Q.40	C	Q.41	A	Q.42	D
Q.43	В	Q.44	D	Q.45	В	Q.46	В	Q.47	A	Q.48	В	Q.49	D
Q.50	D	Q.51	C	Q.52	A	Q.53	D						

EXERCISE # O-II

	EXERCISE # O-II							
Q.1	4-Ethyl octane	Q.17	1,2-epoxy propane					
Q.2	3-Ethyl-2,4-dimethyl pentane	Q.18	1,3,4-trimethyl cyclobutene					
Q.3	5-Methyl hepta-1,3,6-triene	Q.19	Methylene cyclohexane					
Q.4	Hepta-1,5-dien-3-yne	Q.20	1-ethyl-2-methylcyclopentane					
Q.5	2-Isopropyl-4-methyl pent-1-ene	Q.21	1-methyl-3-(methyl ethyl) cyclohexane					
or	4-Methyl-2-(methyl ethyl) pent-1-ene		or 1-isopropyl-3-methylcyclohexane					
		Q.22	Butyl cyclohexane					
Q.6	3-Methoxypropene	Q.23	Isopropylidenecyclopentane					
Q.7	1-Hydroxybut-3-en-2-one	Q.23	or 1-methyl ethylidene cyclopentane					
Q.8	2-Ethylbut-2-en-1-ol	Q.24	3-Bromo-4-cyclopropyl cyclopentane carboxylic acid					
Q.9	3-nitroprop-2-en-1-ol							
Q.10	4 hydrayyhay 5 an 1 yn 2 ana	Q.25	Cyclopent-2-en-1-one					
Q.10	4-hydroxyhex-5-en-1-yn-3-one	Q.26	1-(3-butenyl) cyclopentene					
Q.11	4,6-Bis-[1,1-Dimethyl ethyl] Nonane	Q.27	1,2-diethenyl cyclohexene					
Q.12	2-Formyl pentane nitrile	Q.27	1,2-diethenyr cyclonexene					
_		Q.28	1-cyclohexyl-1-propanone					
Q.13	2,2,6,7-tetramethylocatane	Q.29	Ethyl cyclohexanecarboxylate					
Q.14	3-Ethyl-4,6-dimethyloctane	Q.29	Ethyreyclonexanecarboxyrate					
0.15		Q.30	4-Bromo-2-ethyl cyclopentanone					
Q.15	5-Methyl cyclohexa-1,3-diene	Q.31	3-(hydroxymethyl)-5-methylheptanal					
Q.16	4-Ethyl Pent-4-en-2-amine	•						

Q.32	2-Bromo-6-oxocyclohexanecarbaldehyde	Q.43	8-chloro bicyclo(4,2,0) oct-2-ene
Q.33	5-amino-6-(1-methyl propyl)		2-cyclopenten-1-ol
	cyclo hex-2-enol	Q.45	Ethyl-2-oxo cyclo pentane carboxylate
Q.34	2-bromo-2-methyl cyclopentanone	Q.46	2-Formyl Benzoic acid
Q.35	Methyl-2-methoxy-6-methyl-3- cyclohexene		3-Mthyl Benzoic acid
	carboxylate	Q.48	Cyclohex-2-en-1,4-dione
Q.36	Bicylo(2,2,1)heptane	Q.49	2-ethynyl cyclohexanol
Q.37	9-methyl bicyclo(4,2,1) nonane	Q.50	4-chloro-1-cyclopentyl pentane-2-one
Q.38	Bicyclo [3,2,2] Non-6-one	Q.51	1-Amino methyl-2-ethyl cyclohexanol
Q.39	spiro(4,5) decane	Q.52	4-isopropyl -1-propyl cyclohexene or 4-(methyl ethyl)-1-propyl cyclohexene
Q.40	2-Methyl Benzoyl Chloride	Q.53	2-(2-oxo-cyclohexyl) propanoic acid
Q.41	1,3,3-Trimethyl cyclohexene	Q.54	3-ethoxy-1(1-nitrocyclohexyl)-hex-4-en-1-one
Q.42	Bicyclo(2,2,1) heptane	Q.55	1,3-diphenyl-1,4-pentadiene

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EXERCISE # S-I

Q.1

Q.2

0.3

Q.4

В

D

Q.5

Q.6 В

Q.7

(A) Q, (B) S, (C) T, (D) P

 \mathbf{C}

Q.8

(A) Q, (B) S, (C) R, (D) P

 \mathbf{C}

Q.9

(A) Q, (B) R, (C) S, (D) P

Q.10 (A) Q, (B) S, (C) R, (D) P

Q.11 (A) T, (B) P, (C) S, (D) Q

Q.12 (A) R, (B) S, (C) T, (D) Q, (E) P

Q.13 D

Q.14 (A) Q, (B) S, (C) P, (D) R

EXERCISE # JEE-ADVANCED & MAINS

Q.1 \mathbf{C}

Q.2 $CH_3 - CH_2 - CH = CH - COOH$

2-pentene-1-oic acid and or 2-pentenoic acid

Q.3 В **Q.4** B

Q.5 (a) 5,6-diethyl-3-methyl-dec-4-ene

(b) N,N, 3-trimethyl-3-pentanamine

Q.6 Butane-1,4-dioic acid **Q.7** \mathbf{C} **Q.8** B **Q.9** B,C

10. D

Important Notes				