

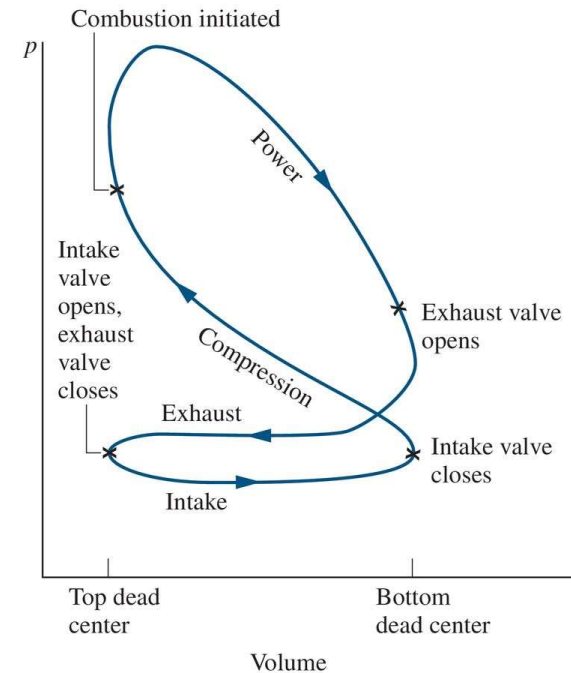
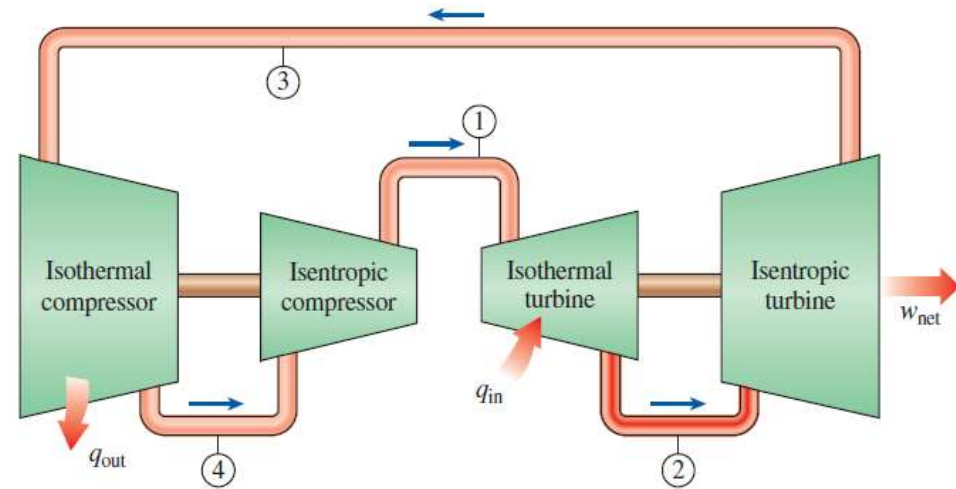
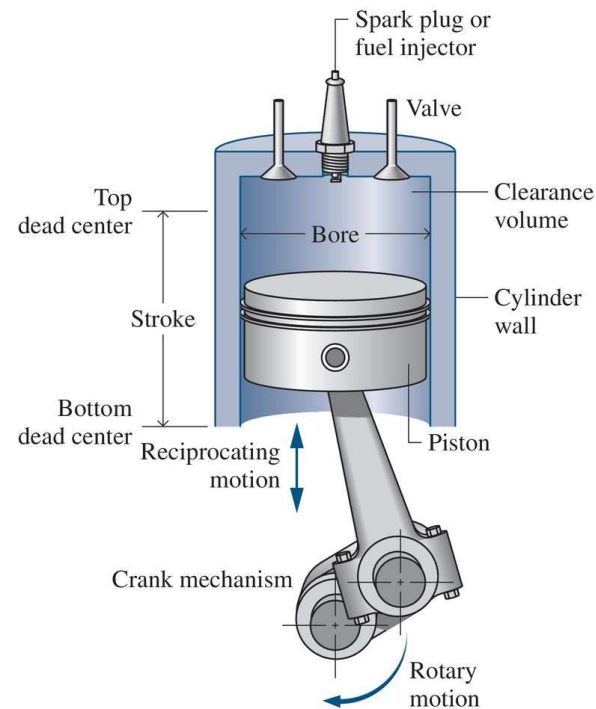
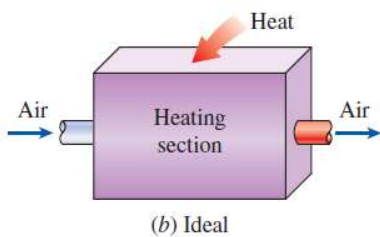
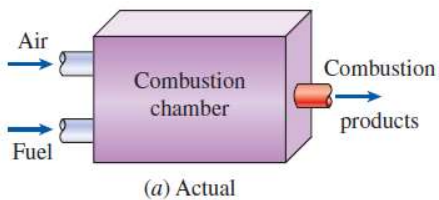
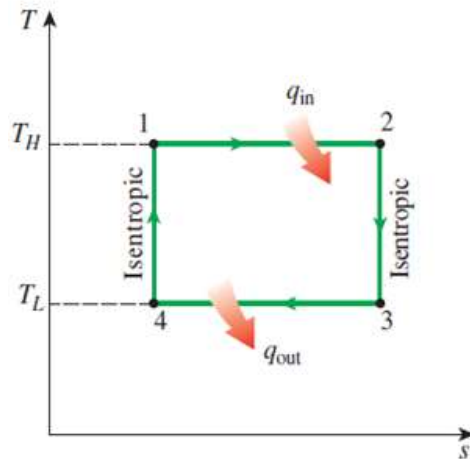
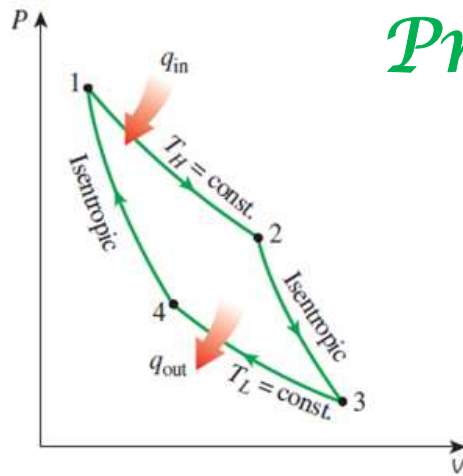
Otto-, Diesel-, Brayton- & Stirling-Cycles

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Indian Institute of Technology, Kanpur.

Previously: Gas power cycles



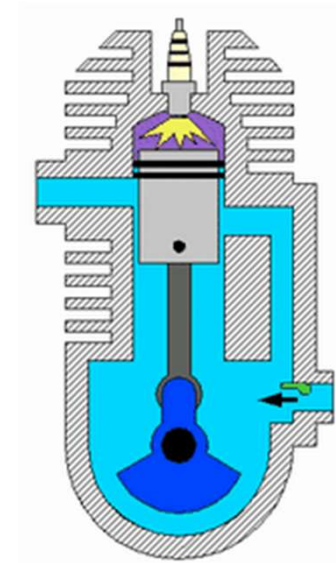
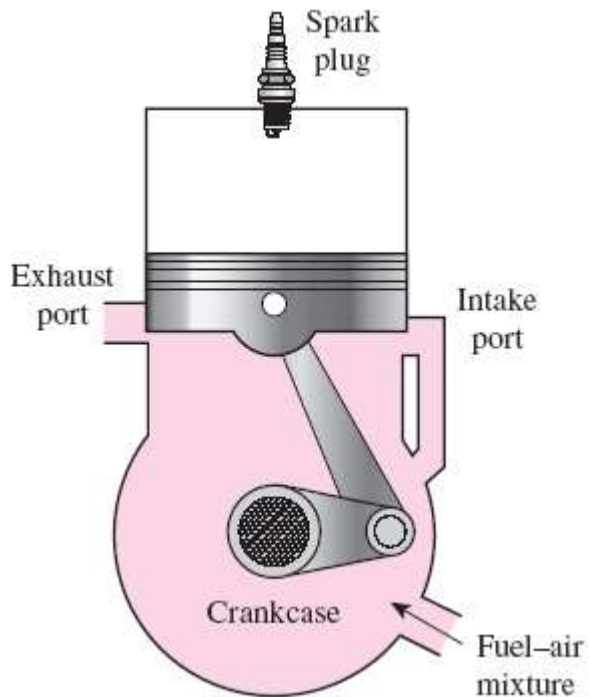
Two-Stroke Otto Cycle

Four-stroke cycle

1 cycle = 4 stroke = 2 revolution

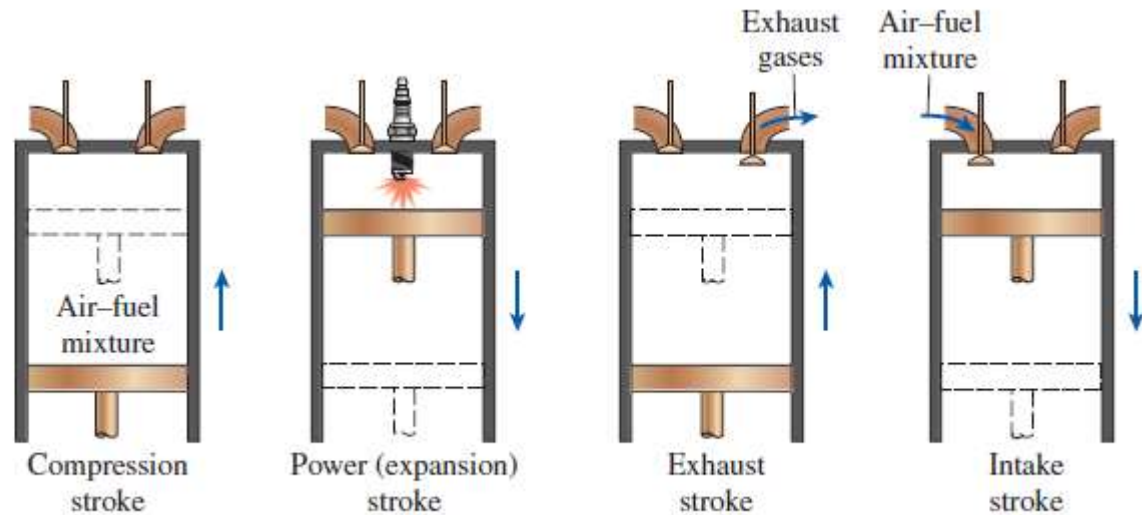
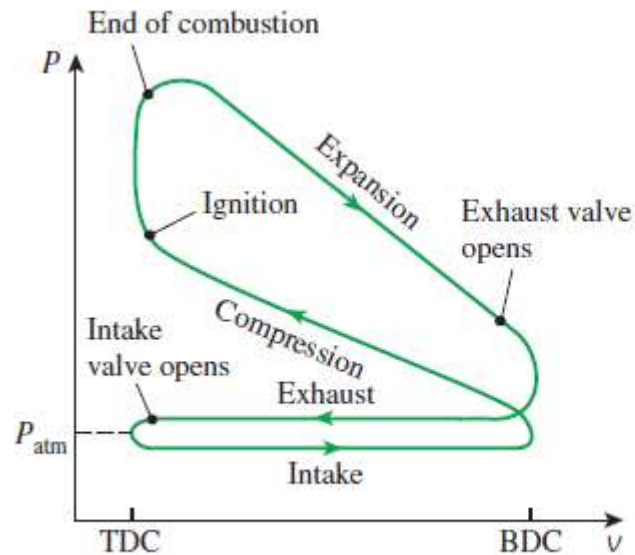
Two-stroke cycle

1 cycle = 2 stroke = 1 revolution

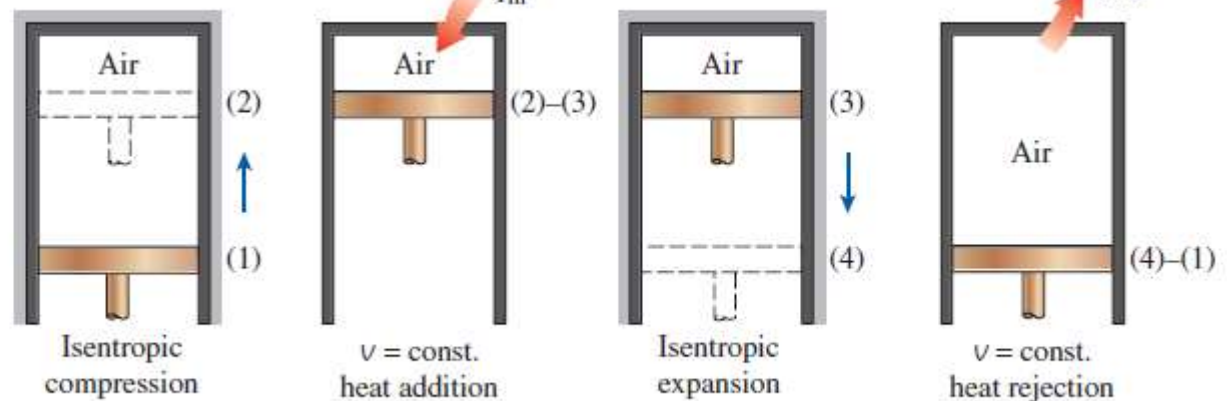
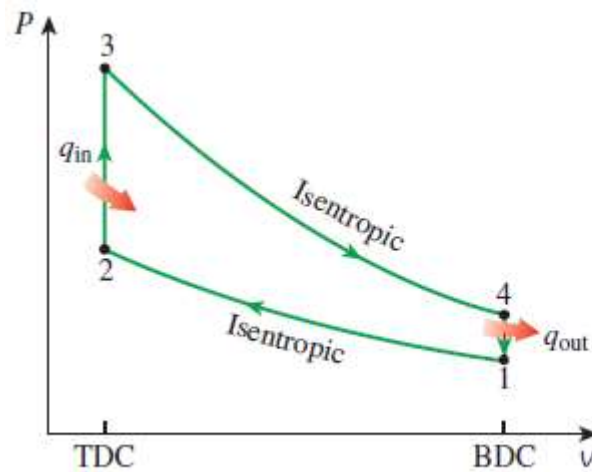


Otto Cycle & Spark-ignition engine (4-stroke)

$$(q_{in} - q_{out}) + (w_{in} - w_{out}) = h_{exit} - h_{inlet}$$

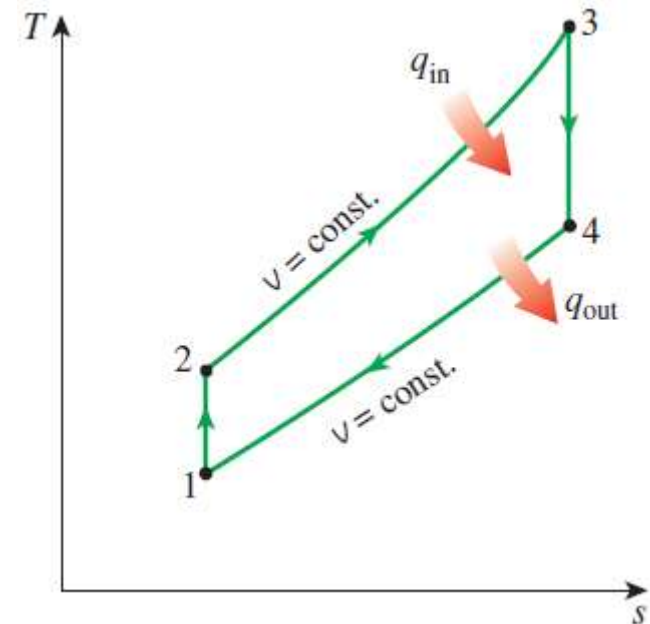
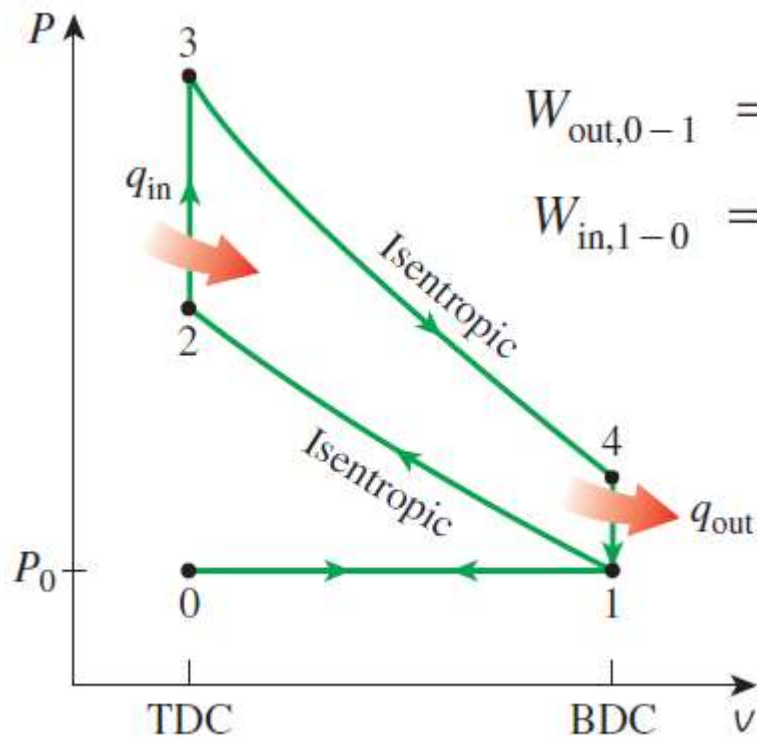
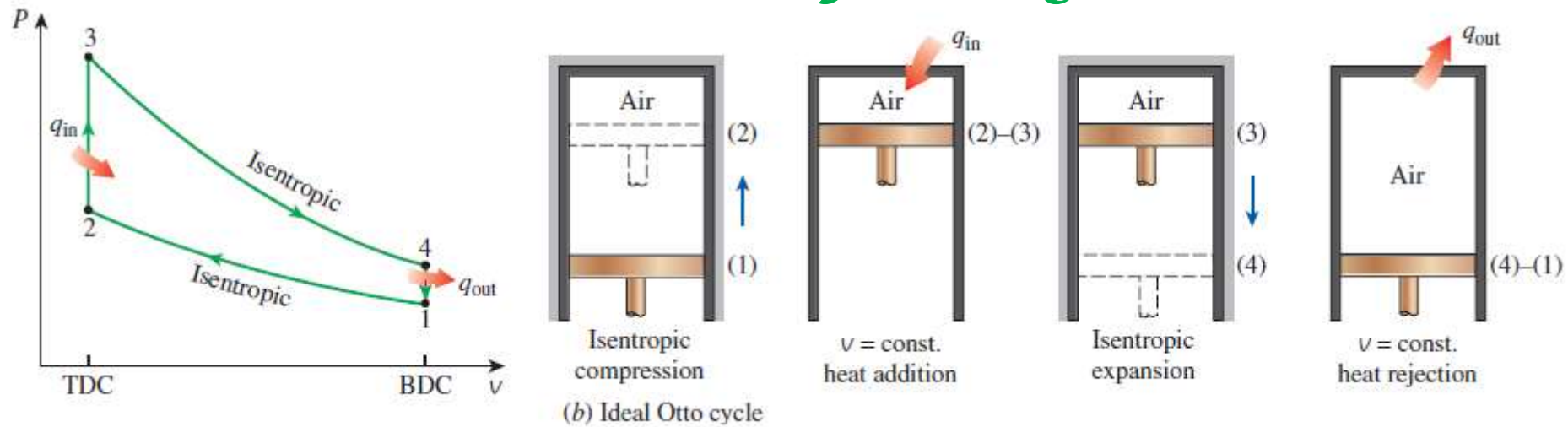


(a) Actual four-stroke spark-ignition engine



(b) Ideal Otto cycle

Ideal Otto Cycle engine



Analysis of Otto Cycle

- 1-2 Isentropic compression
- 2-3 Constant-volume heat addition
- 3-4 Isentropic expansion
- 4-1 Constant-volume heat rejection

$$q_{in} = u_3 - u_2 = c_v(T_3 - T_2)$$

$$q_{out} = u_4 - u_1 = c_v(T_4 - T_1)$$

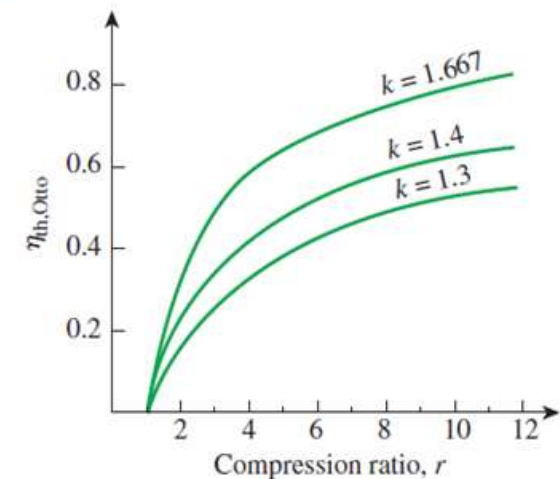
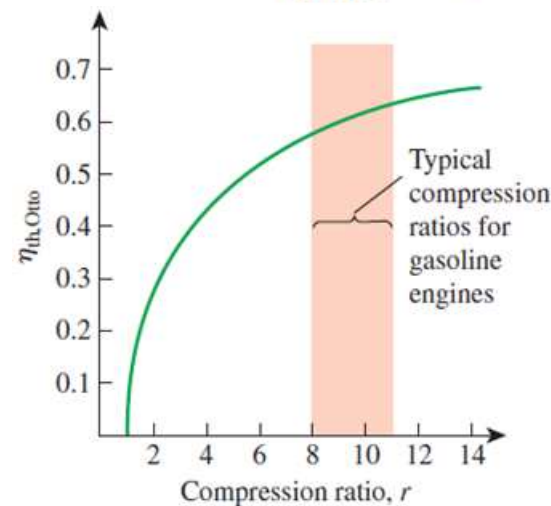
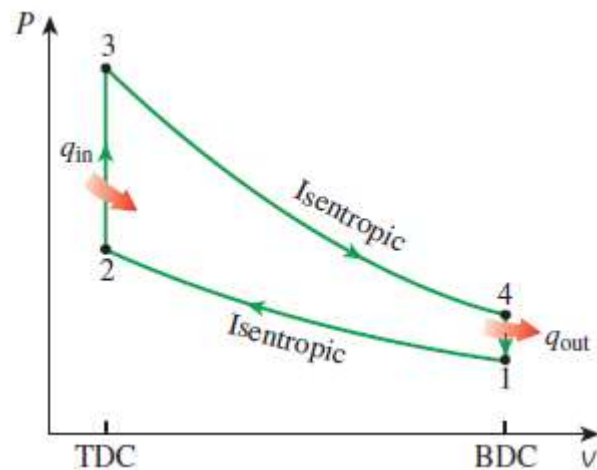
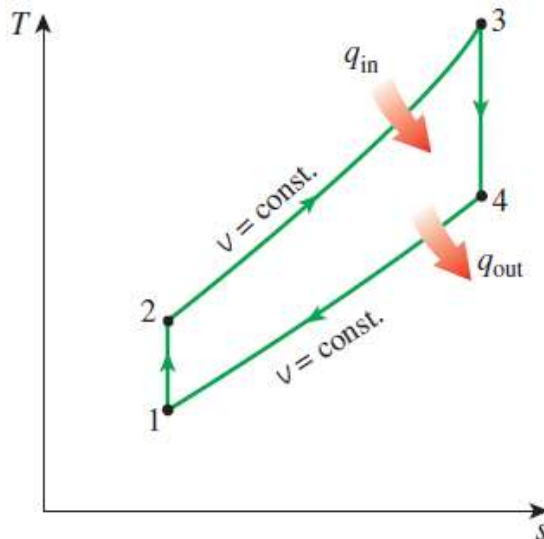
$$\eta_{th,Otto} = \frac{w_{net}}{q_{in}} = 1 - \frac{q_{out}}{q_{in}} = 1 - \frac{T_4 - T_1}{T_3 - T_2} = 1 - \frac{T_1(T_4/T_1 - 1)}{T_2(T_3/T_2 - 1)}$$

Isoentropic gas results for $1 \rightarrow 2$ & $3 \rightarrow 4$

$$\frac{T_1}{T_2} = \left(\frac{v_2}{v_1}\right)^{k-1} = \left(\frac{v_3}{v_4}\right)^{k-1} = \frac{T_4}{T_3}$$

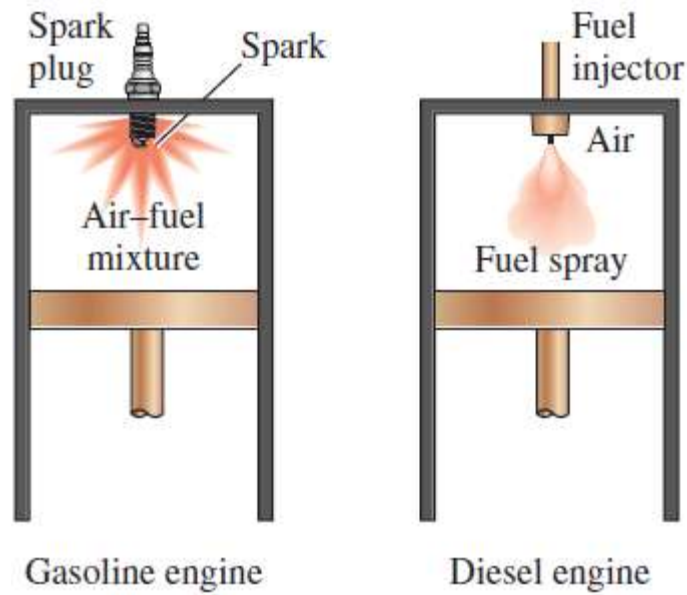
$$r = \frac{V_{max}}{V_{min}}$$

$$\eta_{th,Otto} = 1 - \frac{1}{r^{k-1}}$$

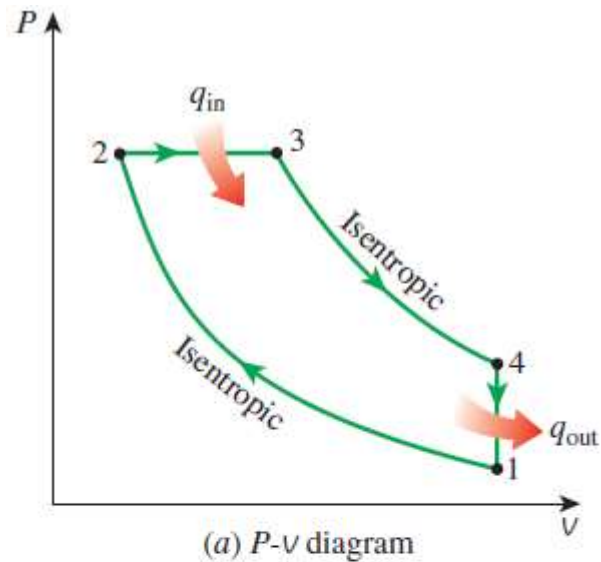


Compression ratio & autoignition/engine knock

Diesel engine & compression-ignition engines



Analysis of Diesel-engines



$$q_{in} - w_{b,out} = u_3 - u_2 \rightarrow q_{in} = P_2(v_3 - v_2) + (u_3 - u_2)$$

$$= h_3 - h_2 = c_p(T_3 - T_2)$$

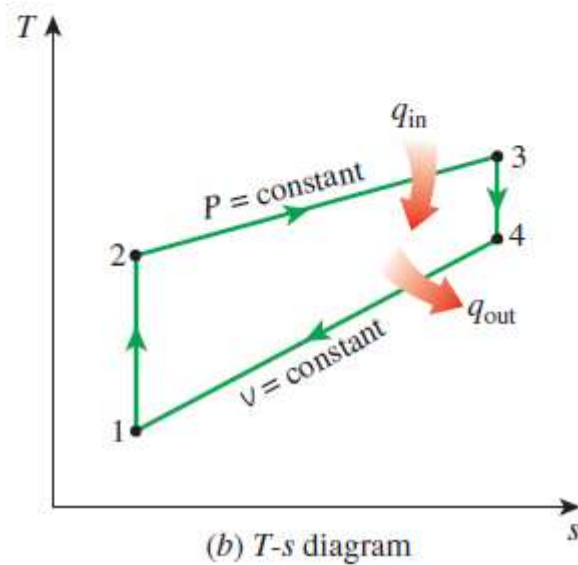
$$-q_{out} = u_1 - u_4 \rightarrow q_{out} = u_4 - u_1 = c_v(T_4 - T_1)$$

$$\eta_{th,Diesel} = \frac{w_{net}}{q_{in}} = 1 - \frac{q_{out}}{q_{in}} = 1 - \frac{T_4 - T_1}{k(T_3 - T_2)} = 1 - \frac{T_1(T_4/T_1 - 1)}{kT_2(T_3/T_2 - 1)}$$

Isoentropic gas results for $1 \rightarrow 2$ & $3 \rightarrow 4$

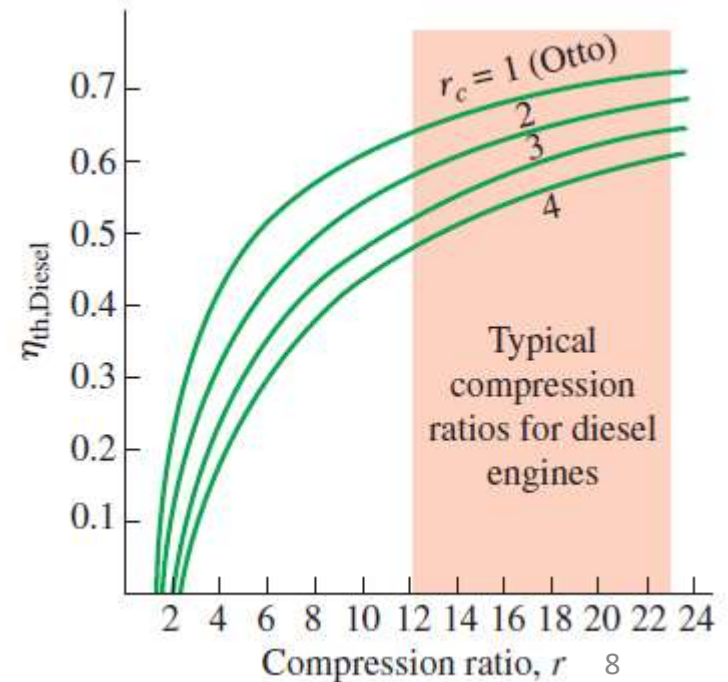
$$r_c = \frac{V_3}{V_2}$$

$$\eta_{th,Diesel} = 1 - \frac{1}{r^{k-1}} \left[\frac{r_c^k - 1}{k(r_c - 1)} \right]$$

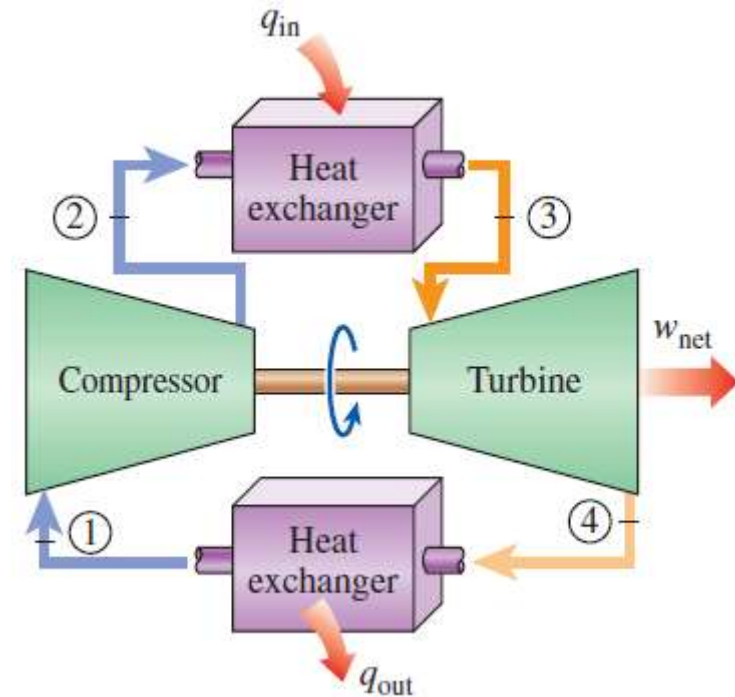
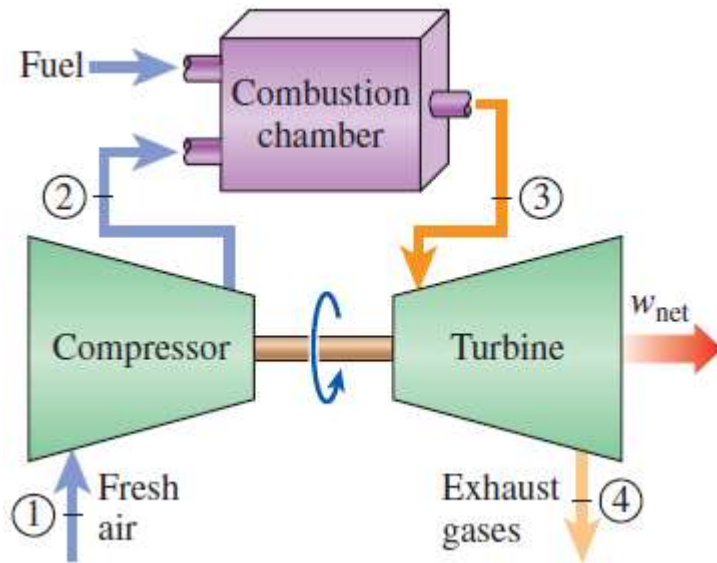


$$\eta_{th,Otto} = 1 - \frac{1}{r^{k-1}}$$

$$\eta_{th,Otto} > \eta_{th,Diesel}$$

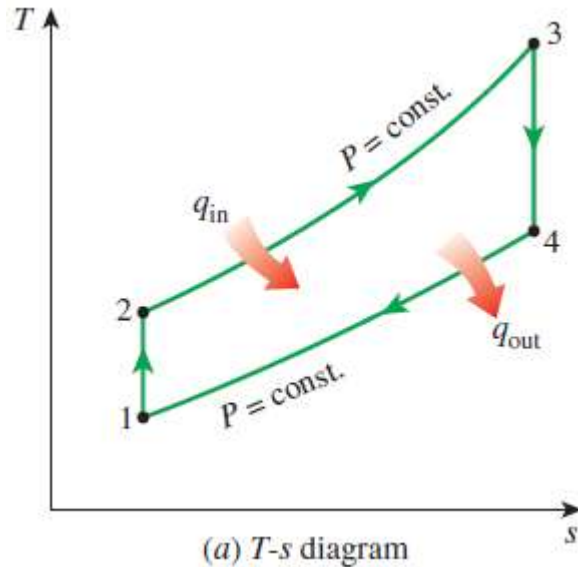


Brayton Cycle



- 1-2 Isentropic compression (in a compressor)
- 2-3 Constant-pressure heat addition
- 3-4 Isentropic expansion (in a turbine)
- 4-1 Constant-pressure heat rejection

Analysis of Brayton Cycle



$$(q_{\text{in}} - q_{\text{out}}) + (w_{\text{in}} - w_{\text{out}}) = h_{\text{exit}} - h_{\text{inlet}}$$

$$q_{\text{in}} = h_3 - h_2 = c_p(T_3 - T_2)$$

$$q_{\text{out}} = h_4 - h_1 = c_p(T_4 - T_1)$$

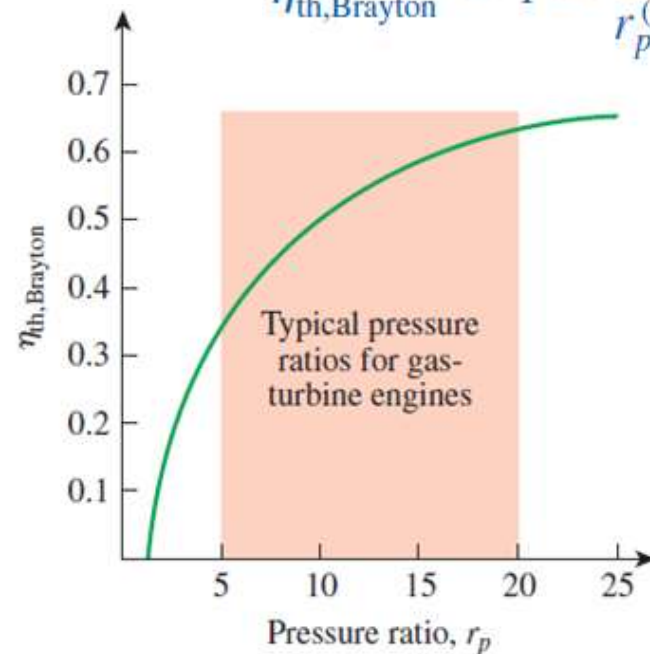
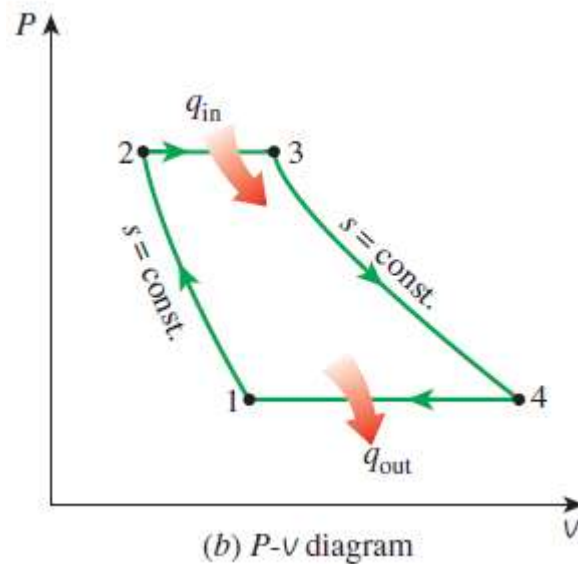
$$\eta_{\text{th,Brayton}} = \frac{w_{\text{net}}}{q_{\text{in}}} = 1 - \frac{q_{\text{out}}}{q_{\text{in}}} = 1 - \frac{c_p(T_4 - T_1)}{c_p(T_3 - T_2)} = 1 - \frac{T_1(T_4/T_1 - 1)}{T_2(T_3/T_2 - 1)}$$

Isoentropic gas results for $1 \rightarrow 2$ & $3 \rightarrow 4$

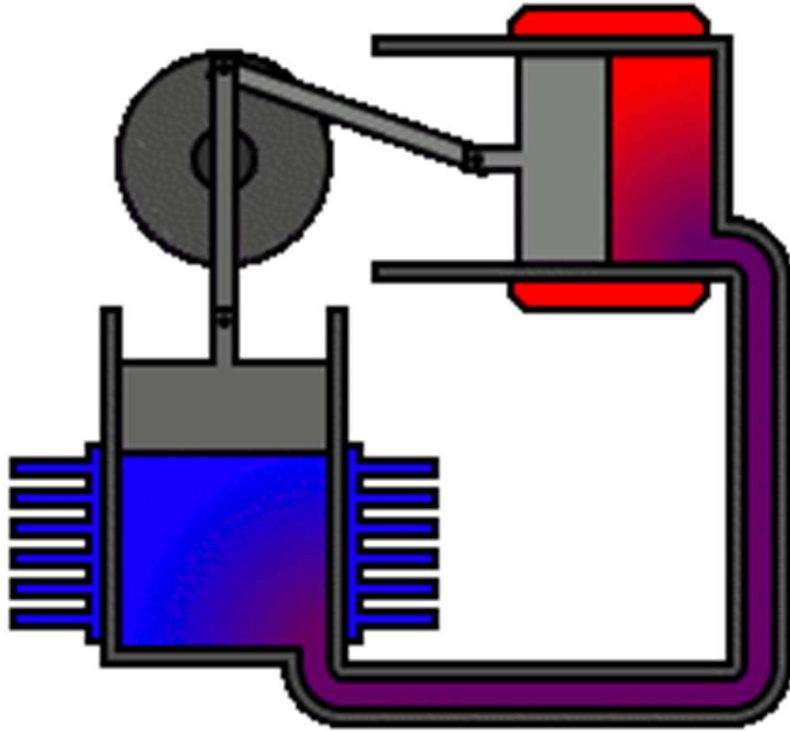
$$\frac{T_2}{T_1} = \left(\frac{P_2}{P_1}\right)^{(k-1)/k} = \left(\frac{P_3}{P_4}\right)^{(k-1)/k} = \frac{T_3}{T_4}$$

$$r_p = \frac{P_2}{P_1}$$

$$\eta_{\text{th,Brayton}} = 1 - \frac{1}{r_p^{(k-1)/k}}$$



Solar Powered Stirling Engine



https://en.wikipedia.org/wiki/Solar-powered_Stirling_engine
https://en.wikipedia.org/wiki/Stirling_engine

What's next?

- Vapor power cycle