

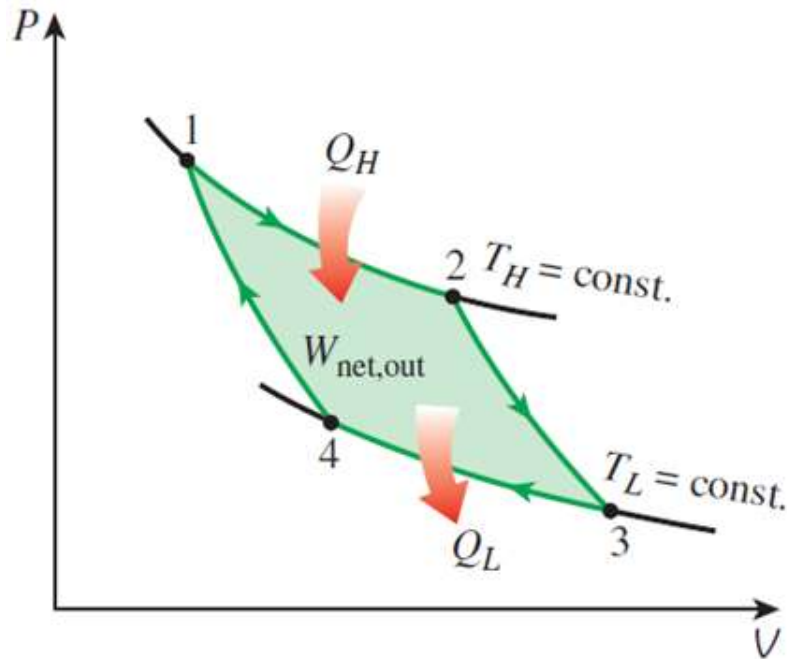
Clausius Inequality

Raj Pala,

rpala@iitk.ac.in

Department of Chemical Engineering,
Associate faculty of the Materials Science Programme,
Indian Institute of Technology, Kanpur.

Previous lecture: Entropy as a State Function!



$$\frac{Q_H}{T_H} + 0 - \frac{Q_L}{T_L} + 0 = 0$$

$$\sum_i \frac{Q_i}{T_i} = 0$$

$$\sum_i U_i = 0; \oint U = 0; U \text{ is a State Function}$$

$$\left(\frac{Q_H}{Q_L} \right)_{\text{rev}} = \frac{T_H}{T_L}$$

$$\sum_i \frac{Q_i}{T_i} = 0; S_i = \frac{Q_i}{T_i}; \sum_i S_i = 0; \oint S = 0; S \text{ is a State Function!!!}$$

Fig: TD-Cengel & Boles

Clausius Inequality



$$\text{All cycles: } \oint \frac{\delta Q}{T} \leq 0$$

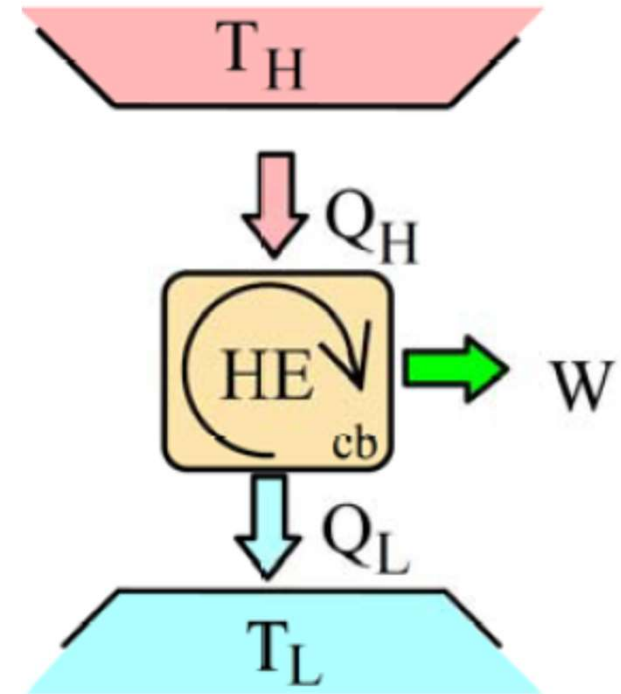
= reversible processes;

< irreversible processes

Reversible heat engines

$$\oint \delta Q = Q_H - Q_L \geq 0$$

$$\oint (1/T) \delta Q = \frac{Q_H}{T_H} - \frac{Q_L}{T_L} = 0$$



Irreversible: Less Work \rightarrow More Heat rejection

- Irreversible/**A**ctual HE: $W_{ac} < W_{rev} \Rightarrow Q_{L ac} > Q_L$

$$\oint \delta Q = Q_H - Q_L \geq 0$$

$$\oint \delta Q = Q_H - Q_{L ac} \geq 0$$

$$\oint (1/T) \delta Q = \frac{Q_H}{T_H} - \frac{Q_L}{T_L} = 0$$

$$\oint (1/T) \delta Q = \frac{Q_H}{T_H} - \frac{Q_{L ac}}{T_L} < 0$$

Refrigerator

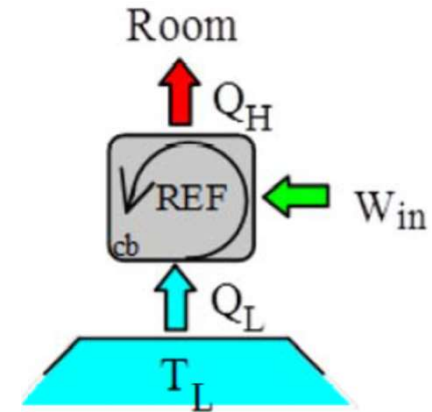
$$W_{ac} > W_{rev} \Rightarrow Q_{H ac} > Q_H$$

$$\oint \delta Q = Q_L - Q_H \leq 0$$

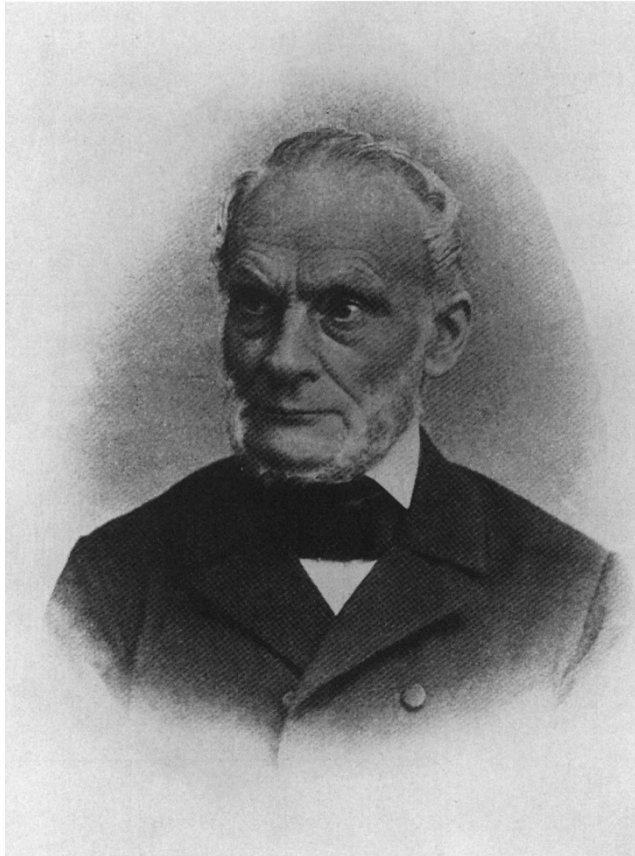
$$\oint \delta Q = Q_L - Q_{H ac} \leq 0$$

$$\oint (1/T) \delta Q = \frac{Q_L}{T_L} - \frac{Q_H}{T_H} = 0$$

$$\oint (1/T) \delta Q = \frac{Q_L}{T_L} - \frac{Q_{H ac}}{T_H} < 0$$



Clausius Inequality



$$\text{All cycles: } \oint \frac{\delta Q}{T} \leq 0$$

= reversible processes;

< irreversible processes

What's next?

- More & more Entropy!