

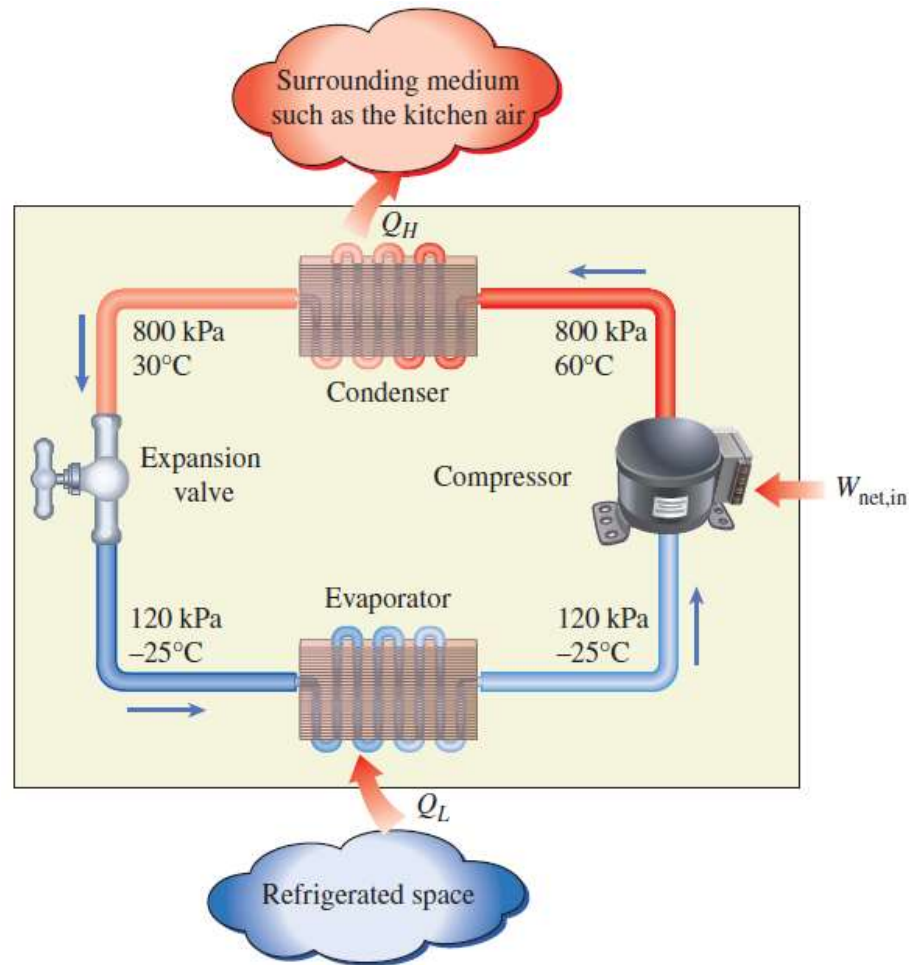
# *Clausius Statement, its equivalence to Kelvin-Planck Statement and the impossibility of perpetual motion machine*

**Raj Pala,**

[rpala@iitk.ac.in](mailto:rpala@iitk.ac.in)

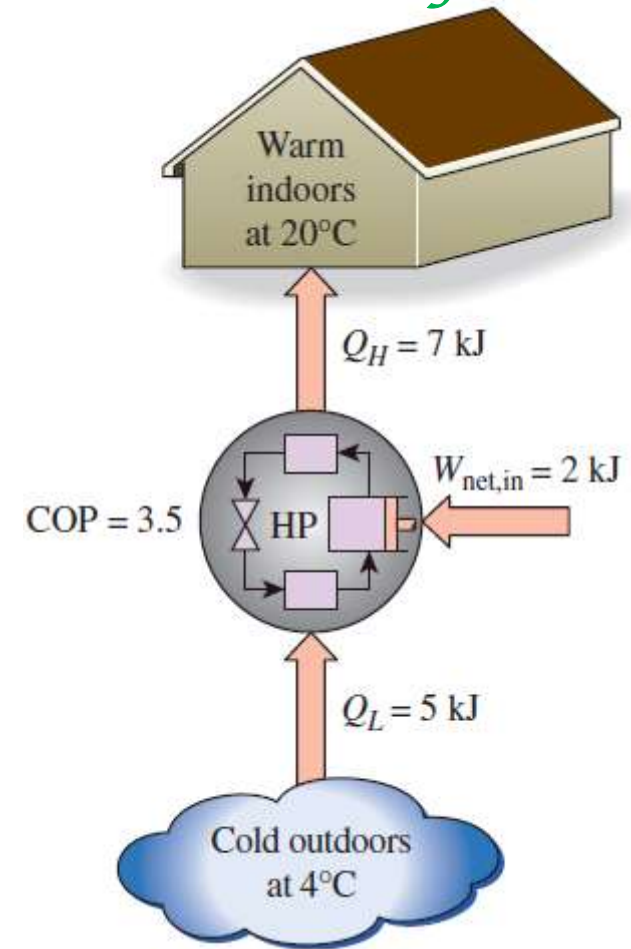
Department of Chemical Engineering,  
Associate faculty of the Materials Science Programme,  
Indian Institute of Technology, Kanpur.

## Previous Lecture: Refrigeration & Heat Pumps



$$COP_R = \frac{\text{Desired output}}{\text{Required input}} = \frac{Q_L}{W_{net,in}}$$

$$COP_R = \frac{Q_L}{Q_H - Q_L} = \frac{1}{Q_H/Q_L - 1}$$

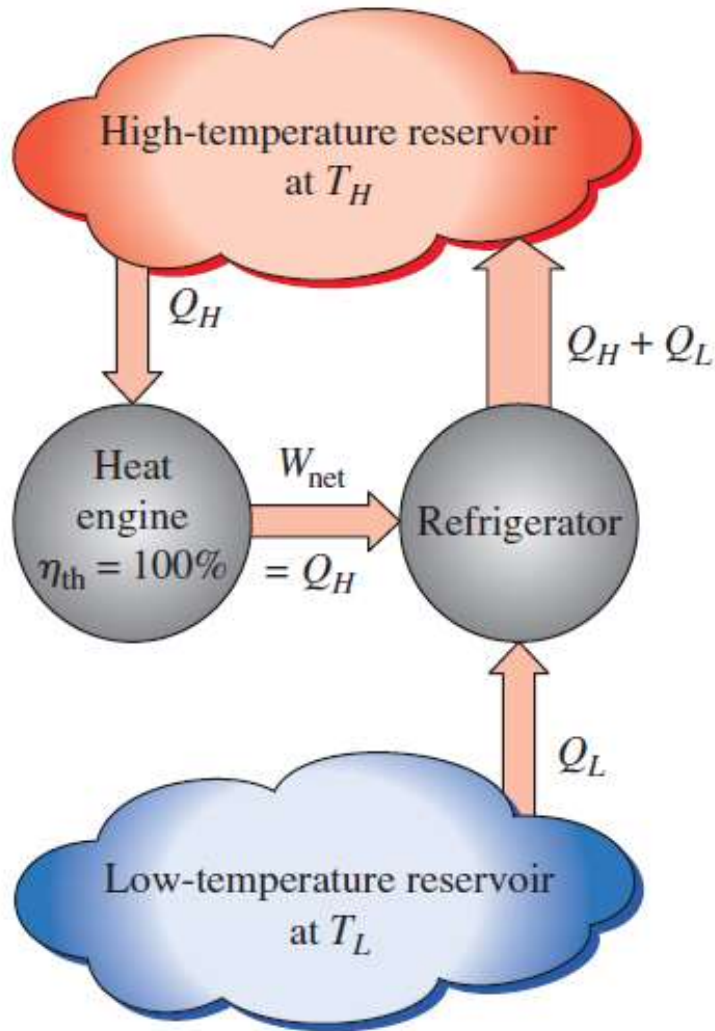


$$COP_{HP} = \frac{\text{Desired output}}{\text{Required input}} = \frac{Q_H}{W_{net,in}}$$

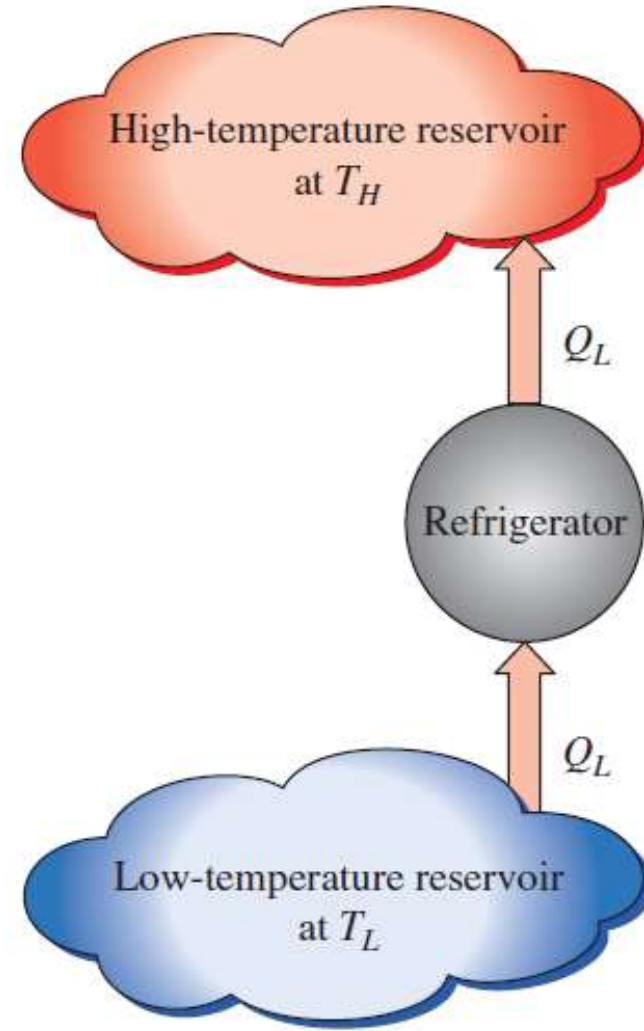
## *Clausius Statement of 2<sup>nd</sup> Law of TD*

“It is impossible to construct a device that operates in a cycle and produces no effect other than the transfer of heat from a lower-temperature body to a higher-temperature body.”

## *Violation of $\mathcal{K}$ - $\mathcal{P}$ 's statement implies violation of Clausius*



(a) A refrigerator that is powered by a 100 percent efficient heat engine



(b) The equivalent refrigerator

## *Violation of Clausius' statement implies violation of K-P...*

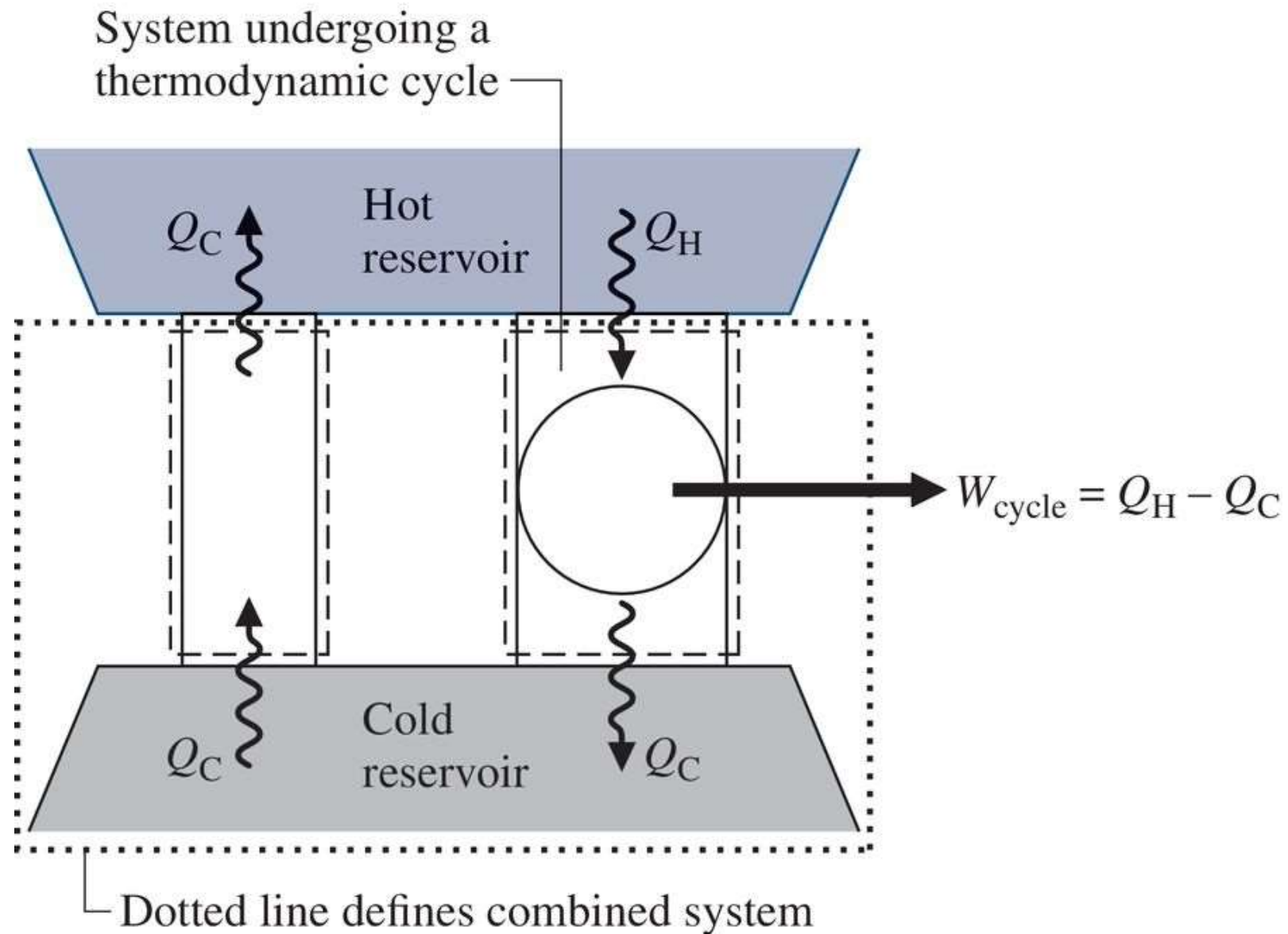
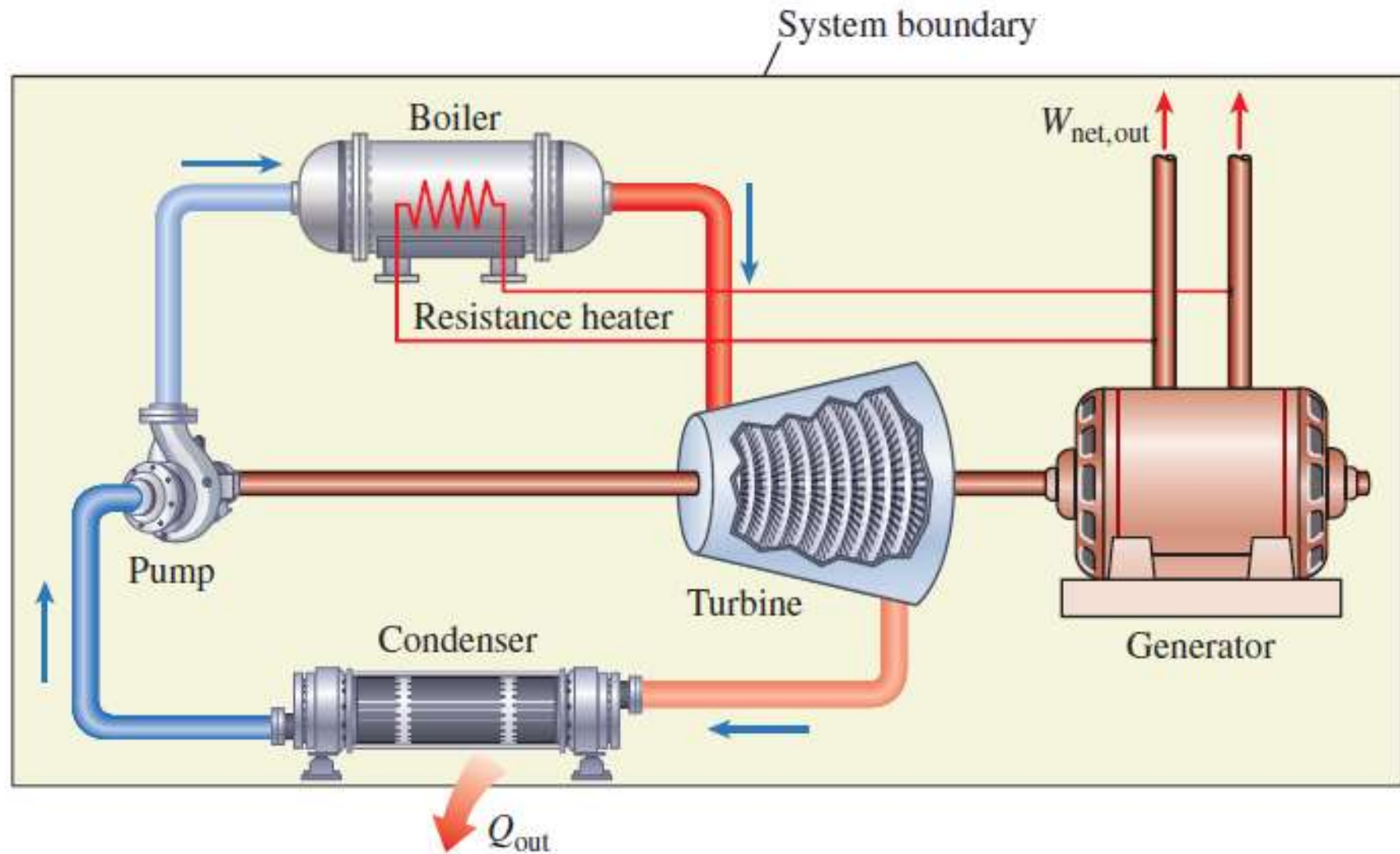


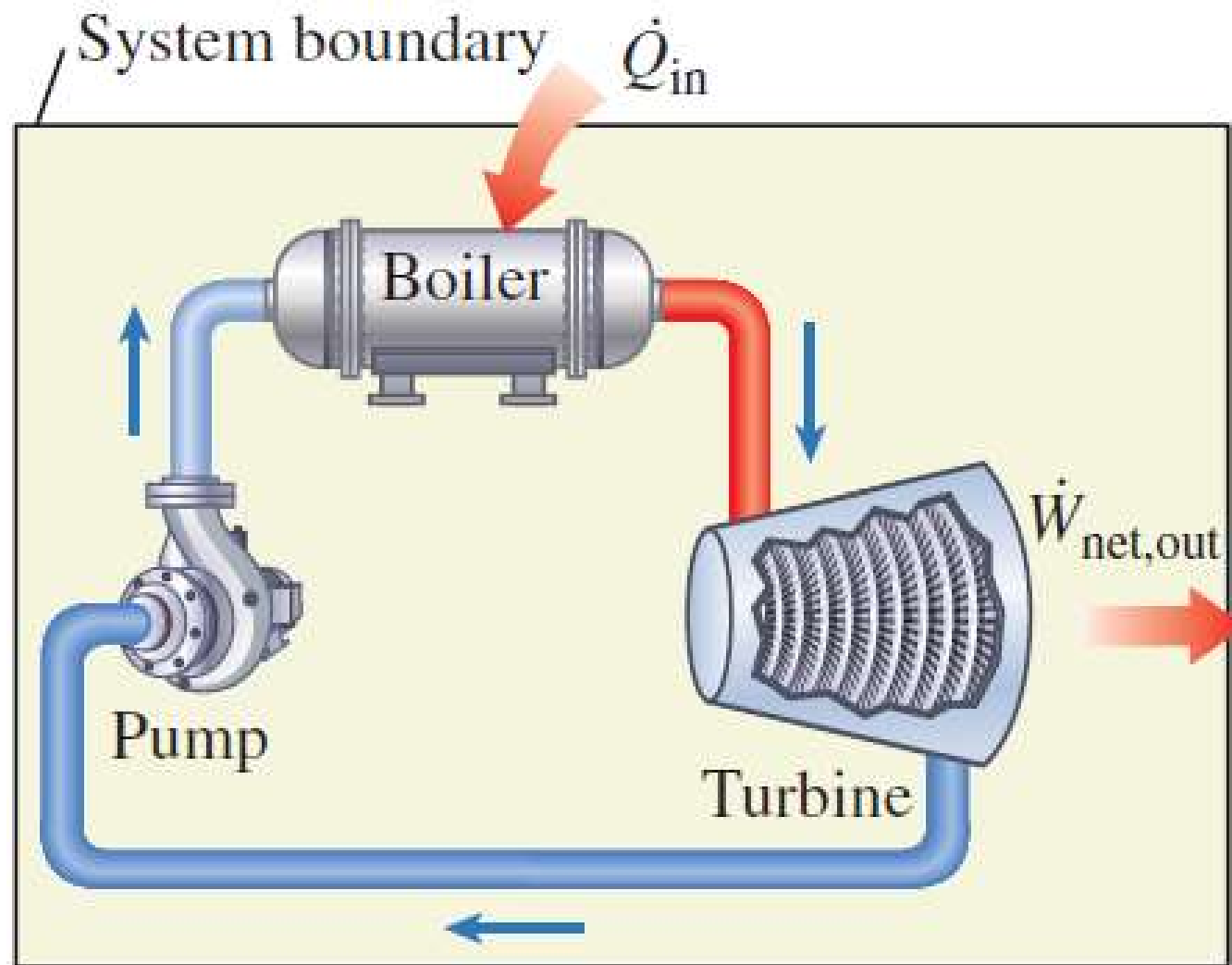
Fig: Cengel & Boles: Moran, Shapiro, Boettner, Bailey

## *Perpetual motion machine violating 1<sup>st</sup> TD law*





*Perpetual motion machine violating 2<sup>nd</sup> TD law*



# *What's next?*

- Reversible & irreversible processes