

Publications relating to hybrid halide perovskite solar cells from the Walsh Materials Design (WMD) research group

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Reviews and Perspectives

1. Butler, K. T., Frost, J. M. & Walsh, A. Ferroelectric materials for solar energy conversion: photoferroics revisited. *Energy Environ. Sci.* **8**, 838–848 (2015).
2. Walsh, A. Principles of chemical bonding and band gap engineering in hybrid organic–inorganic halide perovskites. *J. Phys. Chem. C* **119**, 5755–5760 (2015).
3. Frost, J. M. & Walsh, A. What is moving in hybrid halide perovskite solar cells? *Acc. Chem. Res.* **58**, 528–535 (2015).

Electronic Structure and Optical Properties

4. Brivio, F., Walker, A. B. & Walsh, A. Structural and electronic properties of hybrid perovskites for high-efficiency thin-film photovoltaics from first-principles. *APL Mater.* **1**, 042111 (2013).
5. Brivio, F., Butler, K. T., Walsh, A. & Van Schilfgaarde, M. Relativistic quasiparticle self-consistent electronic structure of hybrid halide perovskite photovoltaic absorbers. *Phys. Rev. B - Condens. Matter Mater. Phys.* **89**, 155204 (2014).
6. Leguy, A. *et al.* Experimental and theoretical optical properties of methylammonium lead halide perovskites. *Nanoscale* **8**, 6317 - 6327 (2016).

Molecular Disorder and Ferroelectricity

7. Frost, J. M. *et al.* Atomistic origins of high-performance in hybrid halide perovskite solar cells. *Nano Lett.* **14**, 2584–2590 (2014).
8. Frost, J. M., Butler, K. T. & Walsh, A. Molecular ferroelectric contributions to anomalous hysteresis in hybrid perovskite solar cells. *APL Mater.* **2**, 081506 (2014).
9. Weller, M. T., Weber, O. J., Frost, J. M. & Walsh, A. Cubic perovskite structure of black formamidinium lead iodide, α -[HC(NH₂)₂]PbI₃, at 298 K. *J. Phys. Chem. Lett.* **6**, 3209–3212 (2015).
10. Leguy, A. M. A. *et al.* The dynamics of methylammonium ions in hybrid organic–inorganic perovskite solar cells. *Nat. Commun.* **6**, 7124 (2015).
11. Grancini, G. *et al.* Role of microstructure in the electron–hole interaction of hybrid lead halide perovskites. *Nat. Photonics* **7**, 695 (2015).

12. Bakulin, A. A. *et al.* Real-time observation of organic cation reorientation in methylammonium lead-iodide perovskites. *J. Phys. Chem. Lett.* **6**, 3663–3669 (2015).

Defect Chemistry, Alloys & Ion Transport

13. Eames, C. *et al.* Ionic transport in hybrid lead iodide perovskite solar cells. *Nat. Commun.* **6**, 7497 (2015).
14. Walsh, A., Scanlon, D. O., Chen, S., Gong, X. G. & Wei, S.-H. Self-regulation mechanism for charged point defects in hybrid-halide perovskites. *Angew. Chemie Int. Ed.* **54**, 1791–1794 (2015).
15. Brivio, F., Caetano, C. & Walsh, A. Thermodynamic Origin of Photoinstability in the $\text{CH}_3\text{NH}_3\text{Pb}(\text{I}_{1-x}\text{Br}_x)_3$ Hybrid Halide Perovskite Alloy. *J. Phys. Chem. Lett.* **7**, 1083 – 1087 (2016).

Phonons and Vibrational Spectra

16. Brivio, F. *et al.* Lattice dynamics and vibrational spectra of the orthorhombic, tetragonal, and cubic phases of methylammonium lead iodide. *Phys. Rev. B* **92**, 144308 (2015).

Solar Cells and Contacts

17. Murray, A. *et al.* Modular design of SPIRO-OMeTAD analogues as hole transport materials in solar cells. *Chem. Commun.* **51**, 8935–8938 (2015).
18. Butler, K. T., Frost, J. M. & Walsh, A. Band alignment of the hybrid halide perovskites $\text{CH}_3\text{NH}_3\text{PbCl}_3$, $\text{CH}_3\text{NH}_3\text{PbBr}_3$ and $\text{CH}_3\text{NH}_3\text{PbI}_3$. *Mater. Horiz.* **2**, 228–231 (2015).
19. Butler, K. T., Kumagai, Y., Oba, F. & Walsh, A. Screening procedure for structurally and electronically matched contact layers for high-performance solar cells: hybrid perovskites. *J. Mater. Chem. C* **4**, 1149–1158 (2016).

Inorganic Halide Perovskites

20. da Silva, E. L., Skelton, J. M., Parker, S. C. & Walsh, A. Phase stability and transformations in the halide perovskite CsSnI_3 . *Phys. Rev. B* **91**, 144107 (2015).
21. Protesescu, L. *et al.* Nanocrystals of Cesium Lead Halide Perovskites (CsPbX_3 , X = Cl, Br, and I): Novel Optoelectronic Materials Showing Bright Emission with Wide Color Gamut. *Nano Lett.* **15**, 3692–3696 (2015).

Pb-free Materials

22. Yang, R. X., Butler, K. T. & Walsh, A. Assessment of Hybrid Organic-Inorganic Antimony Sulfides for Earth-Abundant Photovoltaic Applications. *J. Phys. Chem. Lett.* **6**, 5009–5014 (2015).

23. Ganose, A. M., Butler, K. T., Walsh, A. & Scanlon, D. O. Relativistic electronic structure and band alignment of BiSI and BiSeI: candidate photovoltaic materials. *J. Mater. Chem. A* **4**, 2060–2068 (2016).
24. Butler, K. T.; McKechnie, S.; Azarhoosh, P.; van Schilfgaarde, M.; Scanlon, D. O.; Walsh, A. Quasi-Particle Electronic Band Structure and Alignment of the V-VI-VII Semiconductors SbSI, SbSBr, and SbSeI for Solar Cells. *Appl. Phys. Lett.* **2016**, *108*, 112103.