Publications relating to hybrid halide perovskite solar cells from the Walsh Materials Design (WMD) Group at the University of Bath

Updated: Tuesday, March 8, 2016

Reviews and Perspectives

- 1. Butler, K. T., Frost, J. M. & Walsh, A. Ferroelectric materials for solar energy conversion: photoferroics revisited. *Energy Environ. Sci.* **8**, 838–848 (2015).
- 2. Walsh, A. Principles of chemical bonding and band gap engineering in hybrid organic–inorganic halide perovskites. *J. Phys. Chem. C* **119**, 5755–5760 (2015).
- 3. Frost, J. M. & Walsh, A. What is moving in hybrid halide perovskite solar cells? *Acc. Chem. Res.* (2016). doi:10.1021/acs.accounts.5b00431

Electronic Structure and Optical Properties

- 1. Brivio, F., Walker, A. B. & Walsh, A. Structural and electronic properties of hybrid perovskites for high-efficiency thin-film photovoltaics from first-principles. *APL Mater.* **1**, 042111 (2013).
- 2. Brivio, F., Butler, K. T., Walsh, A. & Van Schilfgaarde, M. Relativistic quasiparticle self-consistent electronic structure of hybrid halide perovskite photovoltaic absorbers. *Phys. Rev. B Condens. Matter Mater. Phys.* **89**, 155204 (2014).
- 3. Leguy, A. *et al.* Experimental and theoretical optical properties of methylammonium lead halide perovskites. *Nanoscale* (2016). doi:10.1039/C5NR05435D

Molecular Disorder and Ferroelectricity

- 1. Frost, J. M. *et al.* Atomistic origins of high-performance in hybrid halide perovskite solar cells. *Nano Lett.* **14**, 2584–2590 (2014).
- 2. Frost, J. M., Butler, K. T. & Walsh, A. Molecular ferroelectric contributions to anomalous hysteresis in hybrid perovskite solar cells. *APL Mater.* **2**, 081506 (2014).
- 3. Weller, M. T., Weber, O. J., Frost, J. M. & Walsh, A. Cubic perovskite structure of black formamidinium lead iodide, α-[HC(NH₂)₂]PbI₃, at 298 K. *J. Phys. Chem. Lett.* **6**, 3209–3212 (2015).
- 4. Leguy, A. M. A. *et al.* The dynamics of methylammonium ions in hybrid organic-inorganic perovskite solar cells. *Nat. Commun.* **6,** 7124 (2015).
- 5. Grancini, G. *et al.* Role of microstructure in the electron–hole interaction of hybrid lead halide perovskites. *Nat. Photonics* **7,** 695 (2015).
- 6. Bakulin, A. A. *et al.* Real-time observation of organic cation reorientation in methylammonium lead-iodide perovskites. *J. Phys. Chem. Lett.* **6,** 3663–3669 (2015).

Defect Chemistry, Alloys & Ion Transport

- 1. Eames, C. *et al.* Ionic transport in hybrid lead iodide perovskite solar cells. *Nat. Commun.* **6,** 7497 (2015).
- 2. Walsh, A., Scanlon, D. O., Chen, S., Gong, X. G. & Wei, S.-H. Self-regulation mechanism for charged point defects in hybrid-halide perovskites. *Angew. Chemie Int. Ed.* **54**, 1791–1794 (2015).
- 3. Brivio, F., Caetano, C. & Walsh, A. Thermodynamic Origin of Photoinstability in the CH3NH3Pb(I1-xBrx)3 Hybrid Halide Perovskite Alloy. *J. Phys. Chem. Lett.* (2016). doi:10.1021/acs.jpclett.6b00226

Phonons and Vibrational Spectra

1. Brivio, F. *et al.* Lattice dynamics and vibrational spectra of the orthorhombic, tetragonal, and cubic phases of methylammonium lead iodide. *Phys. Rev. B* **92**, 144308 (2015).

Solar Cells and Contacts

- 1. Murray, A. *et al.* Modular design of SPIRO-OMeTAD analogues as hole transport materials in solar cells. *Chem. Commun.* **51,** 8935–8938 (2015).
- 2. Butler, K. T., Frost, J. M. & Walsh, A. Band alignment of the hybrid halide perovskites CH₃NH₃PbCl₃, CH₃NH₃PbBr₃and CH₃NH₃PbI₃. *Mater. Horiz.* **2**, 228–231 (2015).
- 3. Butler, K. T., Kumagai, Y., Oba, F. & Walsh, A. Screening procedure for structurally and electronically matched contact layers for high-performance solar cells: hybrid perovskites. *J. Mater. Chem. C* **4,** 1149–1158 (2016).

Inorganic Halide Perovskites

- 1. da Silva, E. L., Skelton, J. M., Parker, S. C. & Walsh, A. Phase stability and transformations in the halide perovskite CsSnI₃. *Phys. Rev. B* **91**, 144107 (2015).
- 2. Protesescu, L. *et al.* Nanocrystals of Cesium Lead Halide Perovskites (CsPbX₃, X = Cl, Br, and I): Novel Optoelectronic Materials Showing Bright Emission with Wide Color Gamut. *Nano Lett.* **15**, 3692–3696 (2015).

Pb-free Materials

- 1. Yang, R. X., Butler, K. T. & Walsh, A. Assessment of Hybrid Organic-Inorganic Antimony Sulfides for Earth-Abundant Photovoltaic Applications. *J. Phys. Chem. Lett.* **6,** 5009–5014 (2015).
- 2. Ganose, A. M., Butler, K. T., Walsh, A. & Scanlon, D. O. Relativistic electronic structure and band alignment of BiSI and BiSeI: candidate photovoltaic materials. *J. Mater. Chem. A* **4**, 2060–2068 (2016).