

## N-Heterocycles

# Palladium-Catalyzed [4+2] and [5+2] Annulation for the Synthesis of Tetrahydroquinolines and 1,4-Benzoxazepines

Zhao-Fei Zhang,<sup>[a,b]</sup> Yan Zhang,<sup>[a]</sup> Chun-Lin Zhang,<sup>\*,[a]</sup> and Song Ye<sup>\*,[a,b]</sup>

**Abstract:** Palladium-catalyzed [4+2] and [5+2] annulations of propargyl carbonates with malonate-tethered anilines and 2-aminobenzyl alcohols were developed, giving the correspond-

ing tetrahydroquinolines and 1,4-benzoxazepines, respectively, in good to excellent yields.

Tetrahydroquinoline and benzoxazepine derivatives are ubiquitous structural motifs in numerous bioactive products and pharmaceuticals (Figure 1).<sup>[1]</sup> Consequently, significant attention has been devoted to the synthesis of these molecules. Many synthetic methods for tetrahydroquinolines have been developed, including hydrogenation of quinolines,<sup>[2]</sup> aza-Diels-Alder reactions,<sup>[3]</sup> and Reissert-type reactions.<sup>[4]</sup> Benzoxazepines can be accessed by Beckmann or Schmidt rearrangements,<sup>[5]</sup> transition-metal catalyzed cross-coupling.<sup>[6]</sup>

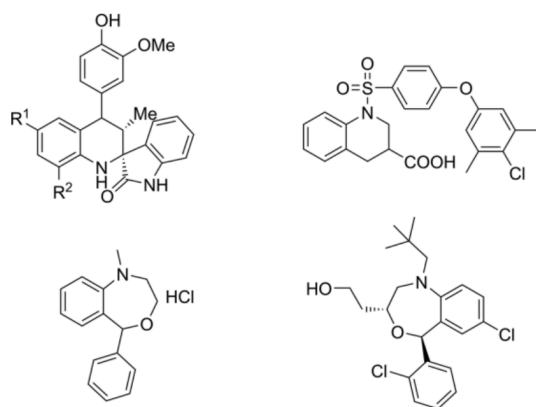
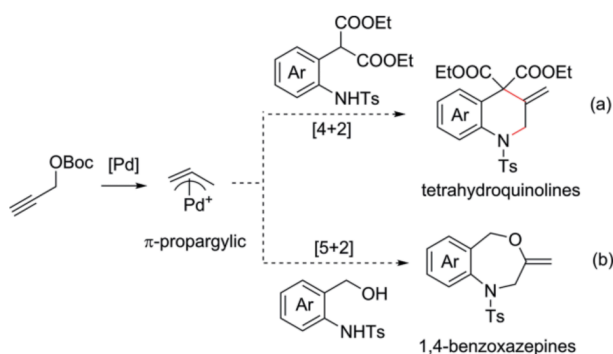


Figure 1. Selected biologically active tetrahydroquinolines and 1,4-benzazepines.

The palladium-catalyzed cyclization of propargylic compound,<sup>[7]</sup> particularly of propargylic carbonates,<sup>[8]</sup> via the key intermediate of  $\pi$ -propargyl palladium has been well developed for the synthesis of various cyclic compounds. In this paper, we

reported a palladium-catalyzed [4+2] annulation of propargylic carbonates with malonate-tethered anilines<sup>[9]</sup> to give tetrahydroquinoline (Scheme 1, reaction a). The corresponding [5+2] annulation of propargylic carbonates with 2-amino-benzyl alcohols to give 1,4-benzoxazepines was also realized (Scheme 1, reaction b).



Scheme 1. Palladium catalyzed annulations for the synthesis of tetrahydroquinolines and 1,4-benzoxazepines.

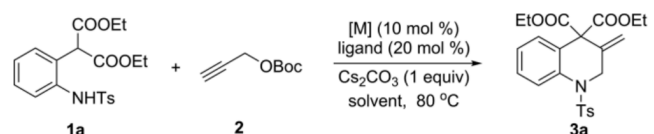
The model reaction of malonate-tethered aniline **1a** and propargyl carbonate **2a** was investigated under palladium catalysis (Table 1). We were happy to find the desired [4+2] annulation product **3a** was formed in 62 % NMR yield when the reaction was carried out using 10 mol-% Pd<sub>2</sub>dba<sub>3</sub>·CHCl<sub>3</sub> and 20 mol-% DPPF as the catalyst and ligand at 80 °C in toluene (entry 1). A series of bisphosphine ligands were then screened (entries 2–7) and DPPF performed best to give **3a** in 92 % yield (entry 3). Lowering the reaction temperature to 40 °C decreased the yield (entry 8). Screening of the solvent revealed that 1,4-dioxane was the best of choice, giving **3a** in near-quantitative yield (entries 8–10). The excellent yield (>99 %) was kept when the loading of palladium catalyst was reduced to 3 mol-% (entry 11). Further reducing of the loading of Pd catalyst to 1 mol-% resulted in decreased yield (70 %, entry 12). Other typical metal complexes, such as Pd(OAc)<sub>2</sub>, [Pd(η<sup>3</sup>-C<sub>3</sub>H<sub>5</sub>)Cl]<sub>2</sub> and Ni(cod)<sub>2</sub> could not give better results (entries 13–15). Control experiments revealed that the reaction did not occur in the absence of palladium or ligand (entries 16–17).

[a] Dr. Z.-F. Zhang, Y. Zhang, C.-L. Zhang, S. Ye  
Beijing National Laboratory for Molecular Sciences, CAS Key Laboratory of Molecular Recognition and Function, Institute of Chemistry, Chinese Academy of Sciences, 4  
100190 Beijing, China  
E-mail: zhangchunlin@iccas.ac.cn  
songye@iccas.ac.cn

[b] Dr. Z.-F. Zhang, S. Ye  
University of Chinese Academy of Sciences,  
Beijing 100049, China

Supporting information and ORCID(s) from the author(s) for this article are available on the WWW under <https://doi.org/10.1002/ejoc.202000262>.

Table 1. Optimization of the reaction conditions.<sup>[a]</sup>



Entry	[M]	Ligand	Solvent	Yield [%] <sup>[b]</sup>
1	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPF	toluene	62
2	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPE	toluene	89
3	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPP	toluene	92
4	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPB	toluene	80
5	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	Xantphos	toluene	24
6	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPEphos	toluene	31
7	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	XPhos	toluene	24
8 <sup>[c]</sup>	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPP	toluene	70
9	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPP	THF	98
10	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPP	dioxane	> 99
11 <sup>[d]</sup>	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPP	dioxane	> 99 (98 <sup>[e]</sup> )
12 <sup>[f]</sup>	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	DPPP	dioxane	70 <sup>[e]</sup>
13 <sup>[d]</sup>	Pd(OAc) <sub>2</sub>	DPPP	dioxane	37
14 <sup>[d]</sup>	[Pd(η <sup>3</sup> -C <sub>3</sub> H <sub>5</sub> )Cl] <sub>2</sub>	DPPP	dioxane	70
15 <sup>[d]</sup>	Ni(cod) <sub>2</sub>	DPPP	dioxane	10
16	/	DPPP	dioxane	0
17	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub>	/	dioxane	0

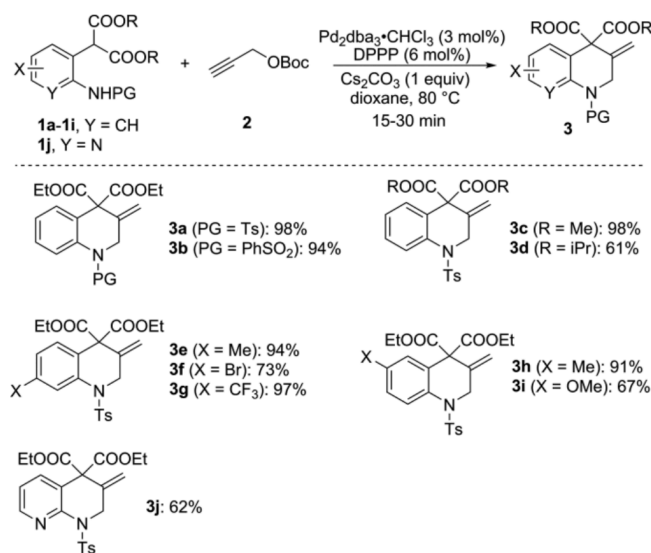
[a] Reaction condition: **1a** (0.2 mmol, 83.0 mg), **2a** (0.26 mmol, 40 μL), Cs<sub>2</sub>CO<sub>3</sub> (0.2 mmol), Pd<sub>2</sub>dba<sub>3</sub>·CHCl<sub>3</sub> (0.02 mmol), ligand (0.04 mmol), N<sub>2</sub> atmosphere. [b] NMR yield using 1,3,5-trimethoxybenzene as an internal standard. [c] The reaction was conducted at 40 °C. [d] 3 mol-% [M] and 6 mol-% ligand were used. [e] Isolated yield. [f] 1 mol-% [M] and 2 mol-% ligand were used.

With the optimized condition in hand, a series of malonate-tethered anilines was then tested for the palladium catalyzed [4+2] annulation reaction (Scheme 2). It was found that the *N*-benzenesulfonyl group worked as well as the *N*-tosyl group, giving the corresponding tetrahydroquinoline (**3b**) in high yield. Besides diethyl, dimethyl malonate-tethered anilines worked as well (**3c**), while diisopropyl malonate resulted in some decreased yield (**3d**). Both electron-donating groups and electron-withdrawing groups were well tolerated on the anilines, giving the corresponding products (**3e–3i**) in good to high yields. Interestingly, the malonate-tethered 2-aminopyridine (**1j**) worked as well to give the corresponding tetrahydronaphthyridine **3j** in good yield, which is a core structure for various bioactive compounds.<sup>[10]</sup>

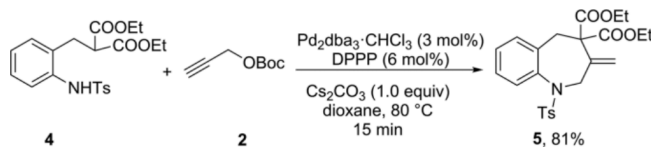
The [5+2] annulation of malonate-tethered aniline **4** with one more carbon linkage was also carried out in the standard conditions, which afforded the desired benzazepine **5** in 81 % yield (Scheme 3).

The investigation was continued by using 2-amino benzyl alcohol **6** as the 1,5-bisnucleophile for the reaction, which led to synthesis of the corresponding 1,4-benzoxazepines (**7a–7e**) in good yields (Scheme 4).<sup>[11]</sup>

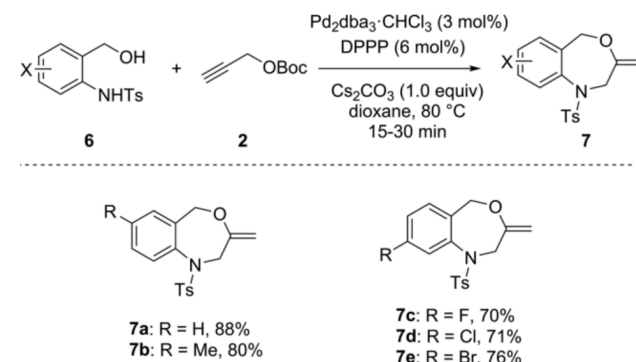
In addition, the reaction of benzene-1,2-diamine and ethane-1,2-diamine worked well for the reaction, giving the corre-



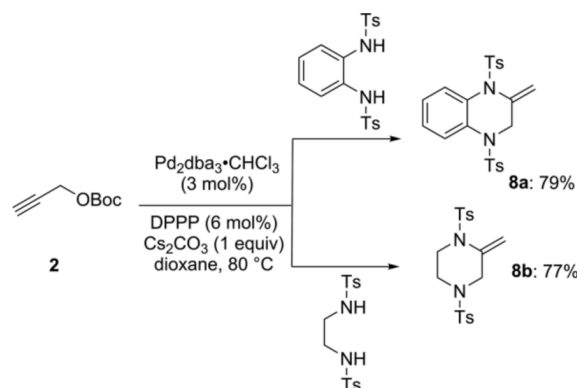
Scheme 2. Synthesis of tetrahydroquinolines **3**.



Scheme 3. Synthesis of benzazepine **5**.



Scheme 4. Synthesis of 1,4-benzoxazepines **7**.



Scheme 5. Synthesis of piperazines.

sponding piperazines **8a** and **8b**, respectively, in good yields (Scheme 5).<sup>[8h]</sup>

The structure of compound **3c** was determined by X-ray analysis of its single crystal (Figure 2).<sup>[12]</sup>

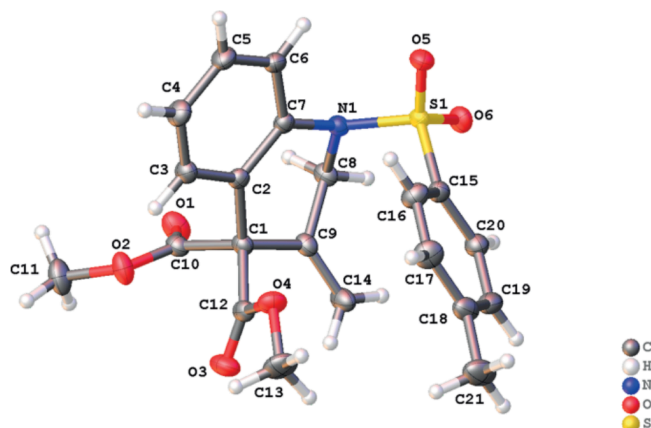
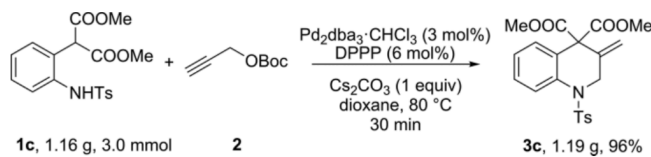


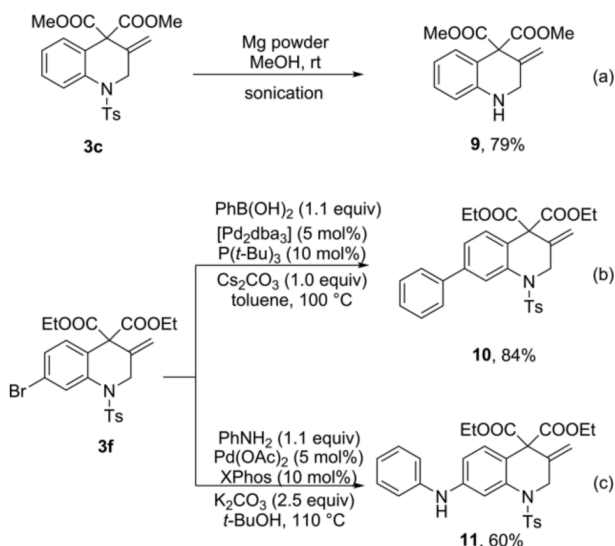
Figure 2. X-ray structure of **3c**.

The reaction could be easily scaled up to 3.0 mmol, giving the tetrahydroquinoline **3c** in 1.19 g, 96 % yield (Scheme 6).



Scheme 6. Scale up reaction.

The resulted highly functionalized tetrahydroquinolines afford many possible further chemical transformations. For example, the *N*-tosyl group in compound **3c** could be removed by Mg/MeOH (Scheme 7, reaction a). Compound **3f** could readily undergo the cross-coupling reactions with phenylboronic acid and aniline to give the corresponding products **10** and **11**, respectively, in good to high yields.<sup>[13]</sup>



Scheme 7. Further chemical transformations.

In summary, the palladium-catalyzed [4+2] and [5+2] annulations of propargyl carbonate with malonate-tethered anilines and 2-aminobenzyl alcohol were developed, giving the corresponding tetrahydroquinolines and 1,4-benzoxazepines, respectively, in good to high yields. The reaction features readily available starting materials, efficient construction of six- and seven-membered heterocycles with potential bioactivities.

## Acknowledgments

This work was supported by Beijing National Laboratory for Molecular Sciences (BNLMS-CXXM-202003), the National Natural Science Foundation of China (Nos 21521002, 21831008).

**Keywords:** Tetrahydroquinolines · 1,4-Benzoxazepines · Annulation · Palladium catalysis · Propargyl carbonates

- [1] a) H. L. Yale, B. Beer, J. Pluscec, E. R. Spitzmiller, *J. Med. Chem.* **1970**, *13*, 713; b) J. Bonjoch, D. Solé, *Chem. Rev.* **2000**, *100*, 3455; c) F. Zhu, Y. Du, J. Chen, B. Chen, Y. Zhu, X. Zhai, S. Xu, W. Zhou, *Chromatographia* **2009**, *69*, 1315; d) V. Sridharan, P. A. Suryavanshi, J. C. Menéndez, *Chem. Rev.* **2011**, *111*, 7157; e) L. Chen, P. T. Wilder, B. Drennen, J. Tran, B. M. Roth, K. Chesko, P. Shapiro, S. Fletcher, *Org. Biomol. Chem.* **2016**, *14*, 5505; f) Z.-L. Zang, S. Karnakanti, S. Zhao, P. Hu, Z. Wang, P.-L. Shao, Y. He, *Org. Lett.* **2017**, *19*, 1354.
- [2] a) W.-B. Wang, S.-M. Lu, P.-Y. Yang, X.-W. Han, Y.-G. Zhou, *J. Am. Chem. Soc.* **2003**, *125*, 10536; b) M. Rueping, A. P. Antonchick, T. Theissmann, *Angew. Chem. Int. Ed.* **2006**, *45*, 3683; *Angew. Chem.* **2006**, *118*, 3765; c) Z.-Y. Han, H. Xiao, X.-H. Chen, L.-Z. Gong, *J. Am. Chem. Soc.* **2009**, *131*, 9182; d) T. Wang, L.-G. Zhuo, Z. Li, F. Chen, Z. Ding, Y. He, Q.-H. Fan, J. Xiang, Z.-X. Yu, A. S. C. Chan, *J. Am. Chem. Soc.* **2011**, *133*, 9878; e) X.-F. Cai, W.-X. Huang, Z.-P. Chen, Y.-G. Zhou, *Chem. Commun.* **2014**, *50*, 9588; f) Á. Vivancos, M. Beller, M. Albrecht, *ACS Catal.* **2018**, *8*, 17.
- [3] a) D. Keck, S. Vanderheiden, S. Bräse, *Eur. J. Org. Chem.* **2006**, 4916; b) G. Dagousset, J. Zhu, G. Masson, *J. Am. Chem. Soc.* **2011**, *133*, 14804.
- [4] a) M. Takamura, K. Funabashi, M. Kanai, M. Shibasaki, *J. Am. Chem. Soc.* **2001**, *123*, 6801; b) Z. Sun, S. Yu, Z. Ding, D. Ma, *J. Am. Chem. Soc.* **2007**, *129*, 9300.
- [5] G. L. Grunewald, V. H. Dahanukar, P. Ching, K. R. Criscione, *J. Med. Chem.* **1996**, *39*, 3539.
- [6] a) B. H. Yang, S. L. Buchwald, *Org. Lett.* **1999**, *1*, 35; b) Z. Xu, K. Li, R. Zhai, T. Liang, X. Gui, R. Zhang, *RSC Adv.* **2017**, *7*, 51972.
- [7] a) J.-R. Labrosse, P. Lhoste, D. Sinou, *Org. Lett.* **2000**, *2*, 527; b) S. Wang, L. Zhang, *J. Am. Chem. Soc.* **2006**, *128*, 8414; c) S. Wang, L. Zhang, *J. Am. Chem. Soc.* **2006**, *128*, 14274; d) H.-P. Bi, X.-Y. Liu, F.-R. Gou, L.-N. Guo, X.-H. Duan, X.-Z. Shu, Y.-M. Liang, *Angew. Chem. Int. Ed.* **2007**, *46*, 7068; *Angew. Chem.* **2007**, *119*, 7198; e) L.-N. Guo, X.-H. Duan, X.-Y. Liu, J. Hu, H.-P. Bi, Y.-M. Liang, *Org. Lett.* **2007**, *9*, 5425; f) N. Marion, S. P. Nolan, *Angew. Chem. Int. Ed.* **2007**, *46*, 2750; *Angew. Chem.* **2007**, *119*, 2806; g) H. Ohno, A. Okano, S. Kosaka, K. Tsukamoto, M. Ohata, K. Ishihara, H. Maeda, T. Tanaka, N. Fujii, *Org. Lett.* **2008**, *10*, 1171; h) A. Furstner, *Chem. Soc. Rev.* **2009**, *38*, 3208; i) L.-N. Guo, X.-H. Duan, Y.-M. Liang, *Acc. Chem. Res.* **2011**, *44*, 111.
- [8] a) M. Yoshida, M. Fujita, T. Ishii, M. Ihara, *J. Am. Chem. Soc.* **2003**, *125*, 4874; b) M. Yoshida, M. Higuchi, K. Shishido, *Org. Lett.* **2009**, *11*, 4752; c) M. Yoshida, C. Sugimura, K. Shishido, *Org. Lett.* **2011**, *13*, 3482; d) M. Yoshida, S. Ohno, K. Namba, *Angew. Chem. Int. Ed.* **2013**, *52*, 13597; *Angew. Chem.* **2013**, *125*, 13842; e) M. Yoshida, C. Sugimura, *Tetrahedron Lett.* **2013**, *54*, 2082; f) T. D. Montgomery, A. E. Nibbs, Y. Zhu, V. H. Rawal, *Org. Lett.* **2014**, *16*, 3480; g) A. E. Nibbs, T. D. Montgomery, Y. Zhu, V. H. Rawal, *J. Org. Chem.* **2015**, *80*, 4928; h) T. D. Montgomery, V. H. Rawal, *Org. Lett.* **2016**, *18*, 740; i) L. Ding, R.-D. Gao, S.-L. You, *Chem. Eur. J.* **2019**, *25*, 4330; j) X.-L. Min, X.-R. Xu, Y. He, *Org. Lett.* **2019**, *21*, 9188; k) A. Kawase, H. Omura, T. Doi, H. Tsukamoto, *Chem. Lett.* **2019**, *48*, 1402.



- [9] a) G. Kang, M. Yamagami, S. Vellalath, D. Romo, *Angew. Chem. Int. Ed.* **2018**, *57*, 6527; *Angew. Chem.* **2018**, *130*, 6637; b) D. Sinou, B. Kryczka, P. Lhoste, N. Dominczak, *Lett. Org. Chem.* **2006**, *3*, 371.
- [10] a) M. A. Seefeld, W. H. Miller, K. A. Newlander, W. J. Burgess, W. E. DeWolf, P. A. Elkins, M. S. Head, D. R. Jakas, C. A. Janson, P. M. Keller, P. J. Manley, T. D. Moore, D. J. Payne, S. Pearson, B. J. Polizzi, X. Qiu, S. F. Rittenhouse, I. N. Uzinskas, N. G. Wallis, W. F. Huffman, *J. Med. Chem.* **2003**, *46*, 1627; b) J. Wang, M. J. Breslin, P. J. Coleman, M. E. Duggan, C. A. Hunt, J. H. Hutchinson, C.-T. Leu, S. B. Rodan, G. A. Rodan, L. T. Duong, G. D. Hartman, *Bioorg. Med. Chem. Lett.* **2004**, *14*, 1049; c) T.-g. Nam, C. L. Rector, H.-y. Kim, A. F. P. Sonnen, R. Meyer, W. M. Nau, J. Atkinson, J. Rintoul, D. A. Pratt, N. A. Porter, *J. Am. Chem. Soc.* **2007**, *129*, 10211.
- [11] a) M. Yar, E. M. McGarrigle, V. K. Aggarwal, *Org. Lett.* **2009**, *11*, 257; b) S. M. Lynch, L. Tafesse, K. Carlin, P. Ghatak, B. Shao, H. Abdelhamid, D. J. Kyle, *Bioorg. Med. Chem. Lett.* **2015**, *25*, 48.
- [12] CCDC 1994587 (for **3c**) contains the supplementary crystallographic data for this paper. These data are provided free of charge by the joint Cambridge Crystallographic Data Centre and Fachinformationszentrum Karlsruhe Access Structures service [www.ccdc.cam.ac.uk/structures](http://www.ccdc.cam.ac.uk/structures).
- [13] a) C. Liu, M. Szostak, *Angew. Chem. Int. Ed.* **2017**, *56*, 12718; *Angew. Chem.* **2017**, *129*, 12892; b) Y. Wang, C. Zheng, S.-L. You, *Angew. Chem. Int. Ed.* **2017**, *56*, 15093; *Angew. Chem.* **2017**, *129*, 15289.

Received: February 28, 2020