

# Major factors to be considered during air sampling

- Constituent or pollutant
- Sampling rate
- Sampling time

# Types of air monitoring

- **Personal sampling**
  - Undertaken within breathing zone of operative
    - As close to mouth and nose as practicable (within 200 mm)
    - Inlet to the sampling head shouldn't be obstructed
- **Portable monitors**
  - Can be moved during sampling
- **Stationary sampling**
  - Fixed during sampling

# Air Quality Measurement

- Measurements of air quality generally fall into:
  - **Source sampling** - Emission from a source is measured
  - **Ambient Air Quality** - Measures the quality of air in a particular place.
  - **Industrial Hygiene sampling** - for testing the air quality inside of factories and places of work
  - **Indoor sampling** - to study the quality of air in living spaces

# Types of measurement/sampling

- Active sampling
- Passive sampling
- Electrochemical methods
- Optical analyzers

# Types of Ambient monitoring Stations

Station type	Description
Pedestrian exposure station	In congested areas, surrounded by buildings with many pedestrians and average traffic flow > 10000 vehicles/day.
Downtown neighbor hood exposure stations	In central business districts but not congested areas, and less high rise buildings. The average traffic flow < 500 vehicles/day. Eg. parks, landscapes areas etc.
Residential population exposure station	In the residential or sub-urban areas but not in central business districts. Should be more than 100 m away from any street.
Mesoscale stations	At appropriate height to collect meteorological and air quality data at upper elevation; main purpose to collect the trend of data variations not human exposure.
Non-urban stations	In remote non-urban areas with no traffic and industrial activity. Mainly for background concentrations.

***Location of all stations but mesoscale stations should be 0.5 m from curve and 2.5 to 3.5 m above the ground. For mesoscale stations it could be at even higher elevations.***

# Active sampling

- Pumped Sampling
- Analytes are physically pulled into the sampler using an air sampling pump
- Concentrating the analytes on some sort of media (adsorbents, or filters)

# Passive sampling

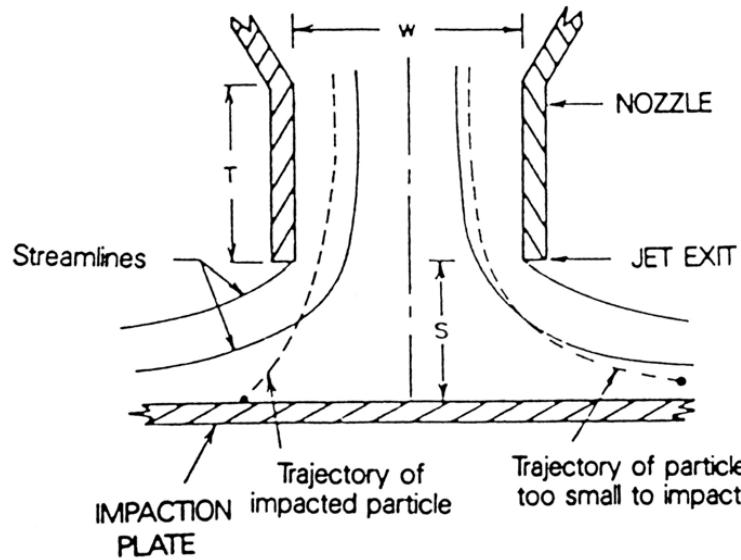
- Diffusive sampling
- Analytes enter the sampler unassisted due to molecular diffusion.
- Unlike active sampling, passive samplers don't require electricity.
- The adsorbed analytes can be desorbed off the adsorbent by solvent or thermal desorption.

# Particulate Matter Sampling

- Gravimetric based
- Light scattering based

## Working principles of gravimetric PM samplers

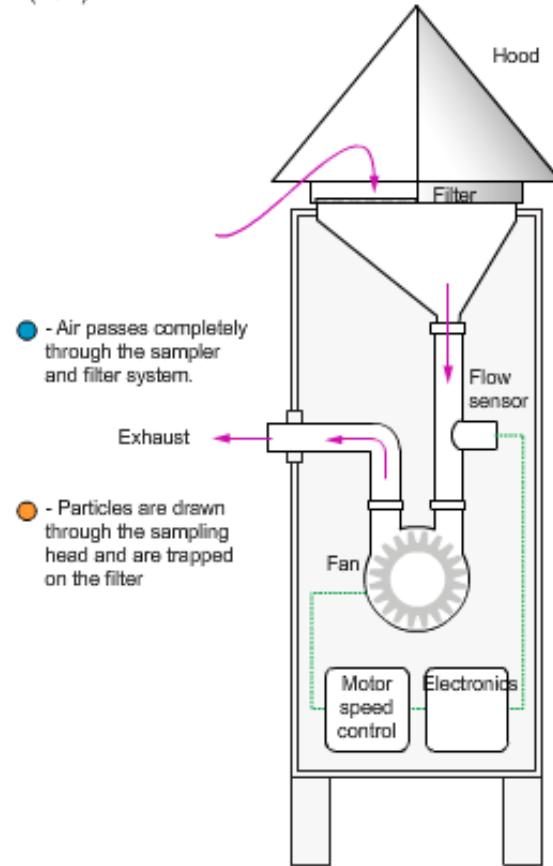
- Samplers work on the principle of Inertia
- Sub classified into
  - Impactors



- Jet of aerosol impinge on flat plate.
- Particles larger than cut size slips across the streamline and impacts the plate
- Smaller particles follow the streamline

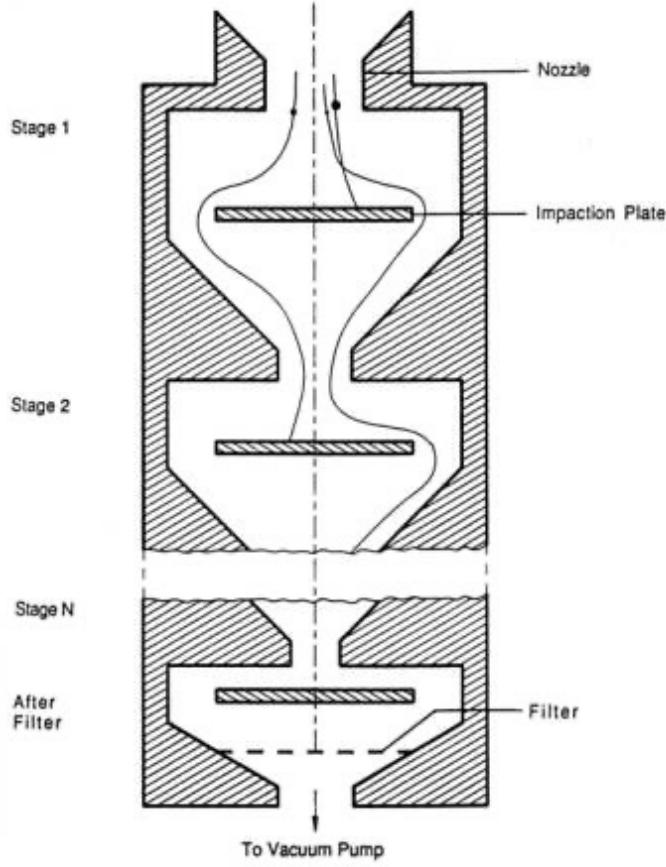
# High Volume Sampler

High volume sampler  
for Total Suspended Particulates  
(TSP)

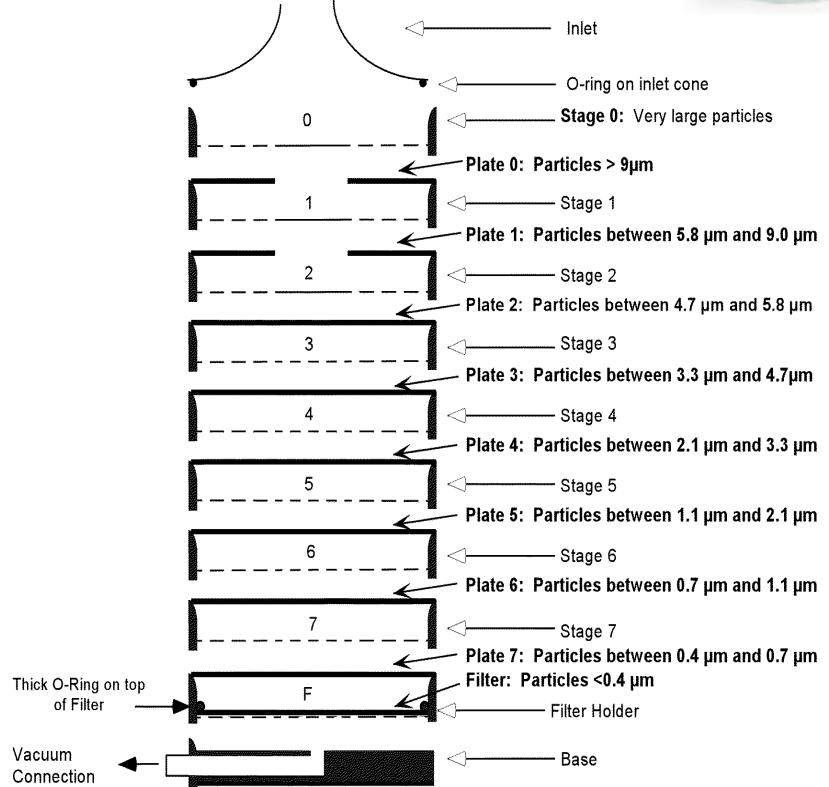


<https://www.qld.gov.au/environment/pollution/monitoring/air-pollution/samplers>

# Cascade Impactor



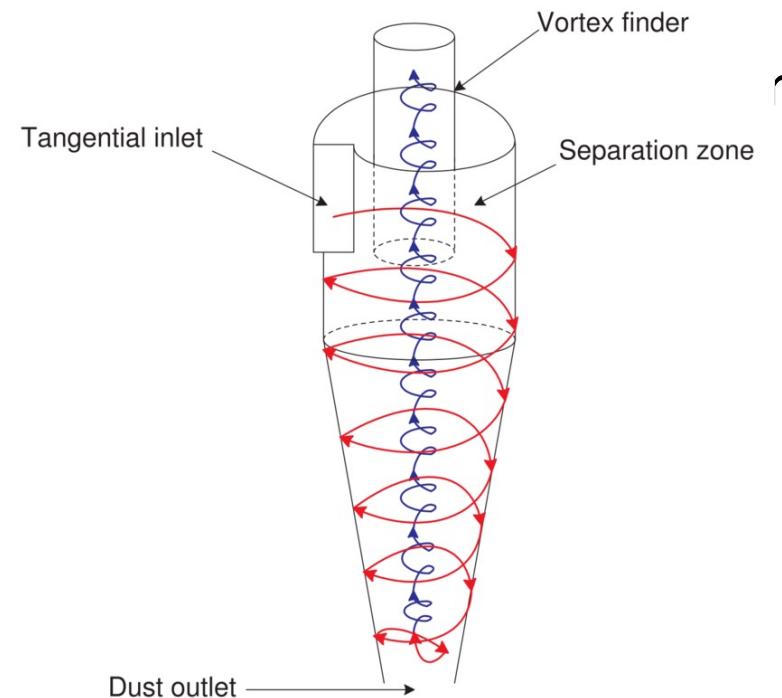
Schematic Diagram of Cascade Impactor.



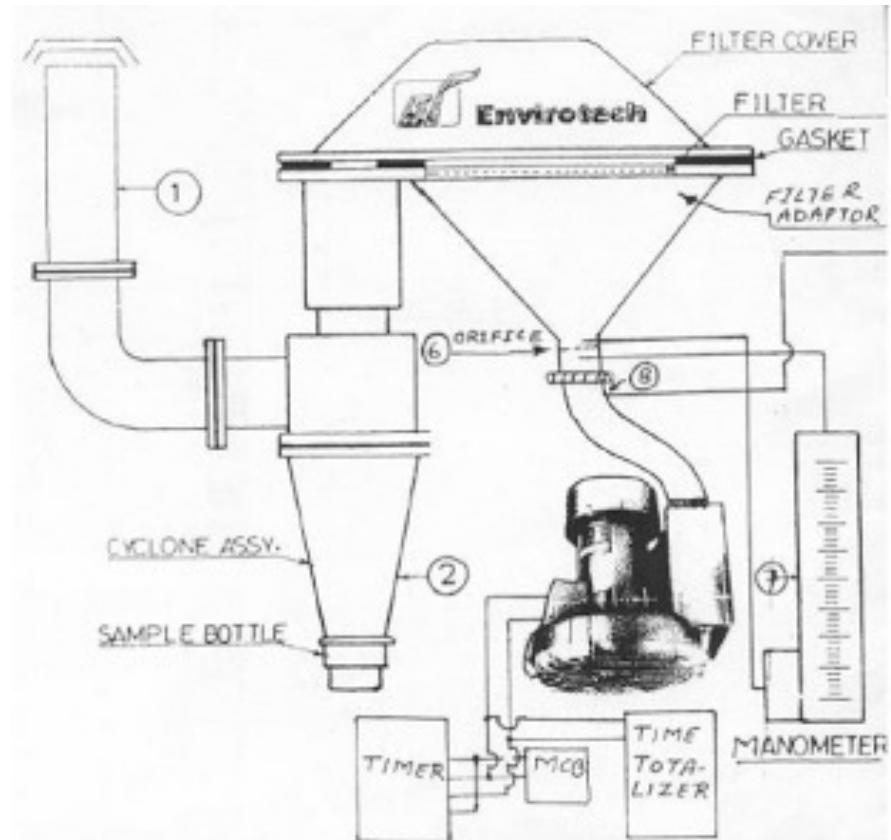
# Working principles of gravimetric PM samplers

## □Cyclone

- Jet of air impinges tangentially on inner surface
- Swirls downward in a cyclonic fashion and into a conical section
- Air reverses direction and spirals upward to exit at the upper end
- Particles larger than cut size and later collects in grit pot.

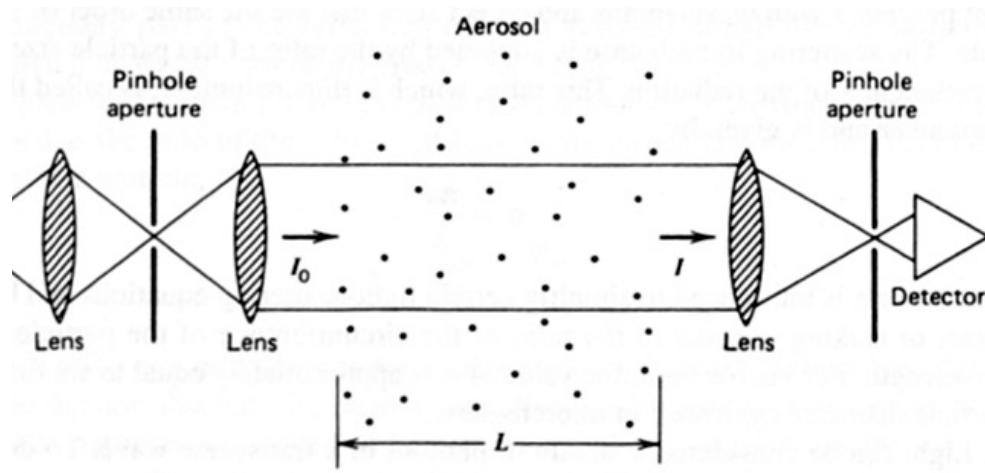


# Respirable dust sampler



SCHEMATIC DIAGRAM OF HIGH VOLUME SAMPLER.

# Light Scattering method

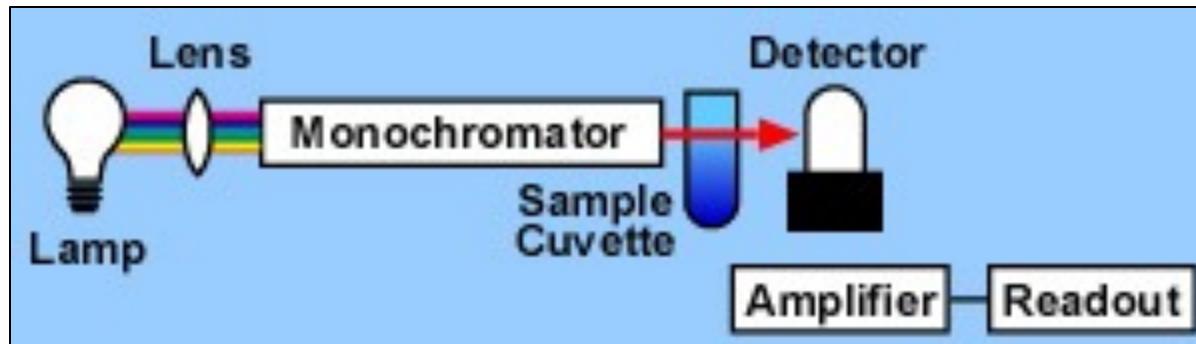


Aethalometer

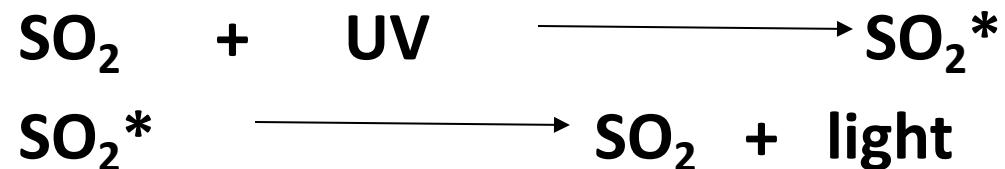
# Measurement of gases

## Sulfur Dioxide

- UV Fluorescence
  - Air sample drawn into a scrubber chamber and then on into an irradiation chamber where it is exposed to UV light
- $\text{SO}_2$  absorbs in 190-230nm
- The amount of fluorescent radiation is directly proportional to the concentration of  $\text{SO}_2$



# Sulfur Dioxide



# Measurement of gases

## Oxides of Nitrogen

- **Chemiluminescence**
  - Passing sample over a catalyst to convert  $\text{NO}_2$  to NO
  - Suitable for ambient air containing NOx (NO and  $\text{NO}_2$ ) at levels less than 1 mL/m<sup>3</sup>
  - Reaction of NO with ozone in a dark enclosed chamber to produce light
  - Provided the ozone is present in excess the light output is directly proportional to the concentration of NO
- $\text{NO} + \text{O}_3 \longrightarrow \text{NO}_2^* + \text{O}_2$
- $\text{NO}_2^* \longrightarrow \text{NO}_2 + h\nu \text{ (light)}$

# Measurement of gases

## Ozone

- Either by chemiluminescence methods or direct reading UV detectors.
- Chemiluminescence method-Sample drawn into a mixing chamber mixed with a stream of Ethene - causes a chemiluminescent reaction and the subsequent emitted light at about 430nm
- Direct reading UV method - stream of gas in the sample is drawn through a flow cell where it is irradiated with UV light at 254nm

# Measurement of gases

## Carbon Monoxide

- Non-dispersive infra red devices
- Sample through a flow cell in the instrument where it is irradiated with infrared radiation

# Impingers

**Impingers** are glass bubble tubes designed for the collection of airborne particles into a liquid medium.

- When using an air sampler, a known volume of air bubbles is pumped through the glass tube that contains a liquid specified in the method.
- The liquid is then analyzed to determine airborne concentrations.

Pollutant	Solution
$\text{SO}_2$	Tetrachloromercurate
$\text{NO}_2$	Arsenite
$\text{NH}_3$	Sulfuric acid
$\text{H}_2\text{S}$	Cadmium hydroxide
$\text{Cl}_2$	Methylorange



Glass Impinger