# ANNEX 8A.1

# PREFIXES, UNITS AND ABBREVIATIONS, STANDARD EQUIVALENTS

# Annex 8A.1 Prefixes, units and abbreviations, standard equivalents

### Prefixes and multiplication factors

Multiplication Factor	Abbreviation	Prefix	Symbol	
1 000 000 000 000 000	$10^{15}$	peta	P	
1 000 000 000 000	$10^{12}$	tera	T	
1 000 000 000	$10^{9}$	giga	G	
1 000 000	$10^{6}$	mega	M	
1 000	$10^{3}$	kilo	k	
100	$10^{2}$	hecto	h	
10	$10^{1}$	deca	da	
0.1	10-1	deci	d	
0.01	10-2	centi	c	
0.001	10-3	milli	m	
0.000 001	10-6	micro	μ	

#### Units and abbreviations

cubic metre	$m^3$
hectare	ha
gram	g
tonne	t
Joule	J
degree Celsius	°C
calorie	cal
year	yr
capita	cap
gallon	gal
dry matter	d.m.
kilogram	kg
pound	lb
atmosphere	atm
Pascal	Pa
hour	h
Watt	W

#### Units and abbreviations, and standard equivalents

1 tonne of oil equivalent (toe)	1 toe	1 x 10 <sup>10</sup> calories	1 x 10 <sup>10</sup> cal
1 ktoe		41.868 terajoules	41.868 TJ
1 short ton	1 sh t	0.9072 tonne	0.9072 t
1 tonne	1 t	1.1023 short tons	1.1023 sh t
1 tonne	1 t	1 megagram	1 Mg
1 kilotonne	1 kt	1 gigagram	1 Gg
1 megatonne	1 Mt	1 teragram	1 Tg
1 gigatonne	1 Gt	1 petagram	1 Pg
1 kilogram	1 kg	2.2046 pounds	2.2046 lb
1 hectare	1 ha	10 <sup>4</sup> square meters	$10^4  \text{m}^2$
1 calorie <sub>IT</sub>	1 cal <sub>IT</sub>	4.1868 Joules	4.1868 J
1 atmosphere	1 atm	101.325 kilopascal	101.325 kPa
1 gram	1 g	0.002205 pounds	0.00205 lb
1 pound	1 lb	453.6 gram	453.6 g
1 terajoule	1 TJ	2.78 x 10 <sup>5</sup> kilowatt hour	2.78 x 10 <sup>5</sup> kWh
1 kilowatt hour	1 kWh	3.6 x 10 <sup>6</sup> Joules	3.6 x 10 <sup>6</sup> J

## Formulae for chemical compounds

Gas
Carbon dioxide
Methane
Nitrous oxide
Hydrofluorocarbons
Perfluorocarbons
Sulphur hexafluoride
Nitrogen trifluoride
Trifluoromethyl sulphur pentafluoride
Chlorofluorocarbons
HFC-23
HFC-32
HFC-41
HFC-125
HFC-134
HFC-134a
HFC-143
HFC-143a
HFC-152
HFC-152a
HFC-161
HFC-227ea
HFC-236cb
HFC-236ea

# Formulae for chemical compounds (Continued)

	• • • • • • • • • • • • • • • • • • • •
Chemical formula	Gas
CF <sub>3</sub> CH <sub>2</sub> CF <sub>3</sub>	HFC-236fa
CH <sub>2</sub> FCF <sub>2</sub> CHF <sub>2</sub>	HFC-245ca
CHF <sub>2</sub> CH <sub>2</sub> CF <sub>3</sub>	HFC-245fa
CF <sub>3</sub> CH <sub>2</sub> CF <sub>2</sub> CH <sub>3</sub>	HFC-365mfc
CF <sub>3</sub> CHFCHFCF <sub>2</sub> CF <sub>3</sub>	HFC-43-10mee
CF <sub>3</sub> OCHF <sub>2</sub>	HFE-125
CHF <sub>2</sub> OCHF <sub>2</sub>	HFE-134
CH <sub>3</sub> OCF <sub>3</sub>	HFE-143a
CF <sub>3</sub> CHClOCHF <sub>2</sub>	HCFE-235da2
CF <sub>3</sub> CF <sub>2</sub> OCH <sub>3</sub>	HFE-245cb2
CF <sub>3</sub> CH <sub>2</sub> OCHF <sub>2</sub>	HFE-245fa2
CHF <sub>2</sub> CF <sub>2</sub> OCH <sub>3</sub>	HFE-254cb2
CF <sub>3</sub> CF <sub>2</sub> CF <sub>2</sub> OCH <sub>3</sub>	HFE-347mcc3
CHF <sub>2</sub> CF <sub>2</sub> CH <sub>2</sub> OCHF <sub>2</sub>	HFE-356pcf3
CHF <sub>2</sub> CF <sub>2</sub> OCH <sub>2</sub> CH <sub>3</sub>	HFE-374pc2
C <sub>4</sub> F <sub>9</sub> OCH <sub>3</sub>	HFE-7100
$C_4F_9OC_2H_5$	HFE-7200
CHF <sub>2</sub> OCF <sub>2</sub> OC <sub>2</sub> F <sub>4</sub> OCHF <sub>2</sub>	H-Galden 1040x
CHF <sub>2</sub> OCF <sub>2</sub> OCHF <sub>2</sub>	HG-10
CHF <sub>2</sub> OCF <sub>2</sub> CF <sub>2</sub> OCHF <sub>2</sub>	HG-01
CF <sub>4</sub>	Perfluoromethane
$C_2F_6$	Perfluoroethane
$C_3F_8$	Perfluoropropane
$C_4F_{10}$	Perfluorobutane
c-C <sub>4</sub> F <sub>8</sub>	Perfluorocyclobutane
$C_5F_{12}$	Perfluourpentane
$C_6F_{14}$	Perfluorohexane
c-C <sub>3</sub> F <sub>6</sub>	Perfluorocyclopropane
CF <sub>3</sub> CHFOCF <sub>3</sub>	HFE-227ea
CF <sub>3</sub> CHFOCHF <sub>2</sub>	HFE-236ea2
CF <sub>3</sub> CH <sub>2</sub> OCF <sub>3</sub>	HFE-236fa
CHF <sub>2</sub> CH <sub>2</sub> OCF <sub>3</sub>	HFE-245fa1
CF <sub>3</sub> CH <sub>2</sub> OCH <sub>3</sub>	HFE-263fb2
CF <sub>3</sub> CF <sub>2</sub> OCF <sub>2</sub> CHF <sub>2</sub>	HFE-329mcc2
CF <sub>3</sub> CF <sub>2</sub> OCH <sub>2</sub> CF <sub>3</sub>	HFE-338mcf2
CF <sub>3</sub> CF <sub>2</sub> OCH <sub>2</sub> CHF <sub>2</sub>	HFE-347mcf2
CF <sub>3</sub> CHFCF <sub>2</sub> OCH <sub>3</sub>	HFE-356mec3
CHF <sub>2</sub> CF <sub>2</sub> CF <sub>2</sub> OCH <sub>3</sub>	HFE-356pcc3
CHF <sub>2</sub> CF <sub>2</sub> OCH <sub>2</sub> CHF <sub>2</sub>	HFE-356pcf2
CF <sub>3</sub> CF <sub>2</sub> CH <sub>2</sub> OCH <sub>3</sub>	HFE-365mcf3

CF <sub>3</sub> I	Trifluoroiodomethane
$\mathrm{CH_{2}Br_{2}}$	Dibromomethane
CHCl <sub>3</sub>	Chloroform
CH <sub>3</sub> Cl	Chloromethane
CH <sub>2</sub> Cl <sub>2</sub>	Dichloromethane
СО	Carbon monoxide
$NO_X$	Nitrogen oxides
NMVOC	Non-methane volatile organic compound
$SO_2$	Sulphur dioxide
$NH_3$	Ammonia