

# EE2703 Assignment 6: Tubelight Simulation

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## 1 Introduction

In this assignment, we test python's ability to run simulating models by simulating a tube light. We do this by analysing the individual behaviour of each electron and plotting the necessary plots and graphs.

## 2 The Tube Light Model

We use a 1-Dimensional model of the tubelight. A uniform electric field is present, that accelerates electrons. Electrons are emitted by the cathode with zero energy, and accelerate in this field. When they get beyond a threshold energy  $E_0$ , they can drive atoms to excited states. The relaxation of these atoms results in light emission. In our model, we will assume that the relaxation is immediate. The electron loses all its energy and the process starts again.

Electrons reaching the anode are absorbed and lost. Each “time step”, an average of  $M$  electrons are introduced at the cathode. The actual number of electrons is determined by finding the integer part of a random number that is “normally distributed” with standard deviation of 2 (i.e  $M\sigma$ ).

## 3 Important Assumptions

1. position of electrons in the tubelight starts from  $x = 1$ , **Not  $x = 0$** .
2. the acceleration of an electron is  $1 \frac{m}{s^2}$ .
3. after an electron reaches  $x = n$ , it is considered to be out of the tubelight.
4. the time of update is  $1sec$ .
5. all collisions are considered to be **perfectly inelastic**.
6. every iteration, 5 electrons with a standard deviation of 2 are added.

## 4 The Assignment

### 4.1 Getting the input from the user

We require 5 inputs from the user, in the order of:  $n$   $M$   $n_k$   $u_0$   $p$   
where these parameters are defined below along with an example value:

```

n=100 # spatial grid size.
M=5 # number of electrons injected per turn.
nk=500 # number of turns to simulate.
u0=5 # threshold velocity.
p=0.25 # probability that ionization will occur

```

The input in the command prompt should look as follows:

```
(base) C:\Users\suma\Documents>python EE19B070_EE2703_tubelight.py 100 5 500 5 0.25
```

Figure 1: Visual representation of input

#### 4.1.1 Input Value Checks

The first 3 inputs are required to be integers while the last 2 can be floats. The code checks whether this is True and will throw out an error if it is False. This is done with a try except loop.

We get the input values from the user using import sys library and using the sys.argv command.

#### 4.1.2 Code

```

if len(sys.argv) != 6:
    print("There has to be 6 integer inputs")

inp = sys.argv
input_int = ['']*5

#checks if first 3 inputs are integers
for n in range(3):
    try:
        input_int[n] = int(sys.argv[n+1])
    except ValueError:
        print("Some of the first 3 inputs are NOT an integer, please check!!")
        break
    else:
        input_int[n] = int(sys.argv[n+1])

#checks if last 2 inputs are floating integers
for n in range(2):
    try:
        input_int[n+3] = float(sys.argv[n+4])
    except ValueError:
        print("Some of the last 2 inputs are NOT a rational number, please check
            !!")
        break
    else:
        input_int[n+3] = float(sys.argv[n+4])

```

## 4.2 Assigning the parameters

#### 4.2.1 Code

```

#assigning the values

n = input_int[0]
M = input_int[1]

```

```
nk = input_int[2]
u0 = input_int[3]
p = input_int[4]
```

### 4.3 Main code

Our code mainly consists of a repeating for block with code inside it.  
The code is give below, analysis of the code is also below.

```
#begin for loop
for i in range(1,nk): #iterate the block nk times
    ii = where(xx > 0)[0] #finds location of all existing electrons
    dx[ii] = u[ii] + 0.5 #gives us change in location
    xx[ii] = xx[ii] + dx[ii] #adding this change
    u[ii] = u[ii] + 1 #change in velocity due to the constant acceleration

    out_of_bounds = where(xx > n)[0] #finds location of all electrons out of
                                   #bounds
    xx[out_of_bounds] = 0 #initialising all the values of these electrons
                           #to 0
    u[out_of_bounds] = 0
    dx[out_of_bounds] = 0

    #defing the deviation of electrons added per turn
    Msig = 2

    #for the collisions
    kk = where(u > u0) #finds location of all electrons that can cause
                       #collisions
    ll=where(random.rand(len(kk[0]))<=p)[0]; #random distrinution of electrons
                                             #that collide
    kl=kk[0][ll];

    u[kl] = 0 #velocity of electrons after collision is 0
    xx[kl] = xx[kl] - dx[kl]*random.rand(len(kl)) #new electron position

    I.extend(xx[kl].tolist()) #adding all of these elctrons to our I vector
    m=int(random.rand()*Msig+M) #number of electrons added is 5 with a standard
                                #deviation of 2

    empty = where(xx == 0)[0]
    tt = min(len(empty),m) #checking which one is smaller

    xx[empty[:tt]] = 1 #setting the position of these elecrons to 1 and the rest
                       #of their values to 0
    u[empty[:tt]] = 0
    dx[empty[:tt]] = 0

    ii = where(xx > 0)[0] #again we find existing electrons and add it to X and V
    X.extend(xx[ii].tolist())
    V.extend(u[ii].tolist())
```

### 4.4 Analysing the code

#### 4.4.1 The For loop

loop is iterated nk times, so code is of the form:

```
for k=1:nk
    <<block>>
```

end

#### 4.4.2 Initialising the vectors

We initialise 3 vectors that we will use in our plots. These 3 vectors are:

```
Intensity of emitted light, I
Electron position, X
Electron velocity, V
```

#### 4.4.3 Finding existing electrons

We find these electrons by using the where command. This would be:

```
ii = where(xx > 0)
```

this command would give us all the positions of existing electrons.

#### 4.4.4 Random Function

the random function we used is

```
random.rand(length)
```

this code would give us a list of random numbers between 0 and 1 of given length.

#### 4.4.5 Number of electrons injected

The number of electrons injected is not always 5, but rather it is a bell curve with maximum probability being 5 and with a standard deviation of 2.

This function is the reason why we get similar but different graphs every time.

### 4.5 Plots

We have 4 plots per given set of parameters. We mainly use the *hist* command to plot them.

#### 4.5.1 Histogram plot of X vs x

Here, X is the number of electrons.

```
#plot the number of electrons vs x
hist(X, bins = arange(0,101), rwidth=0.75, color = 'purple')
title('Number of Electrons vs $x$ with $u_0$=%0.3f and $p$=%0.3f'%(u0,p))
xlabel('$x$')
ylabel('Number of electrons')
show()
```

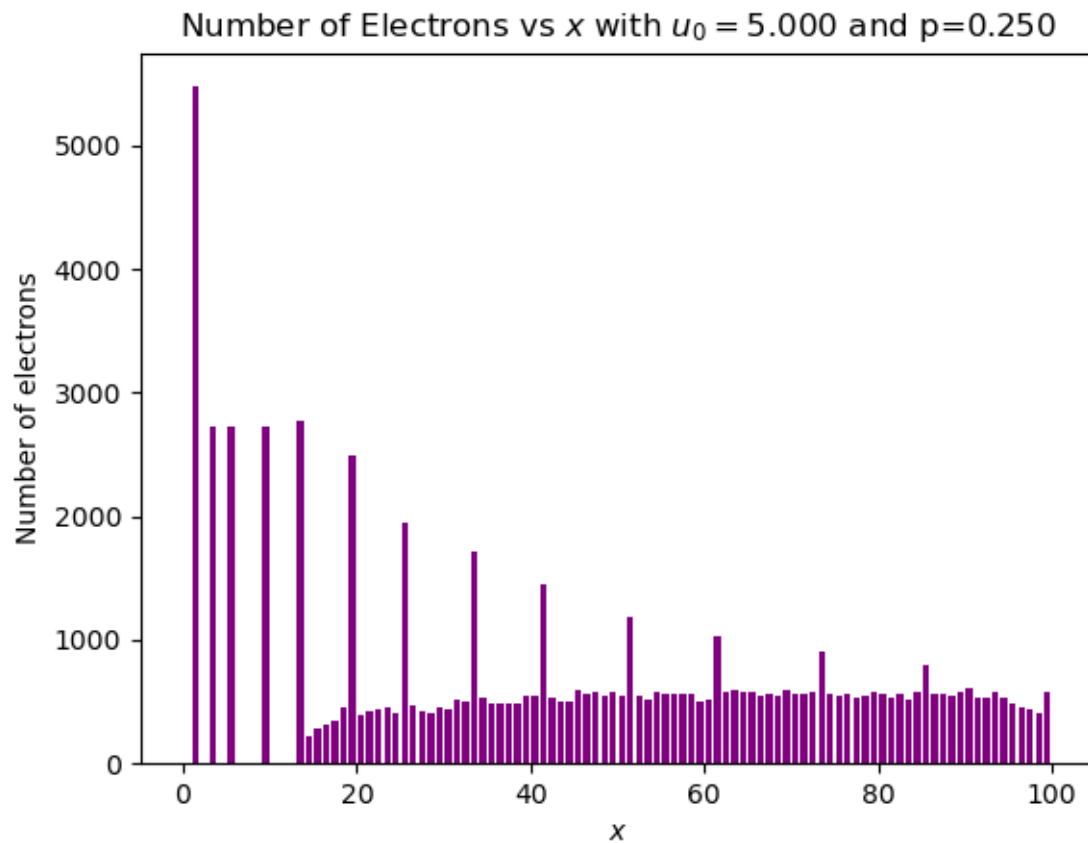


Figure 2: electron density vs position histogram plot for given values

#### 4.5.2 Histogram plot of I vs x

Here, I is the Intensity.

```
#plot the Intensity vs x
hist(I, bins = arange(0,101), rwidth=0.75, color = 'red')
title('Intensity vs $x$ with $u_0$=%0.3f and $p$=%0.3f'%(u0,p))
xlabel('$x$')
ylabel('Intensity')
show()
```

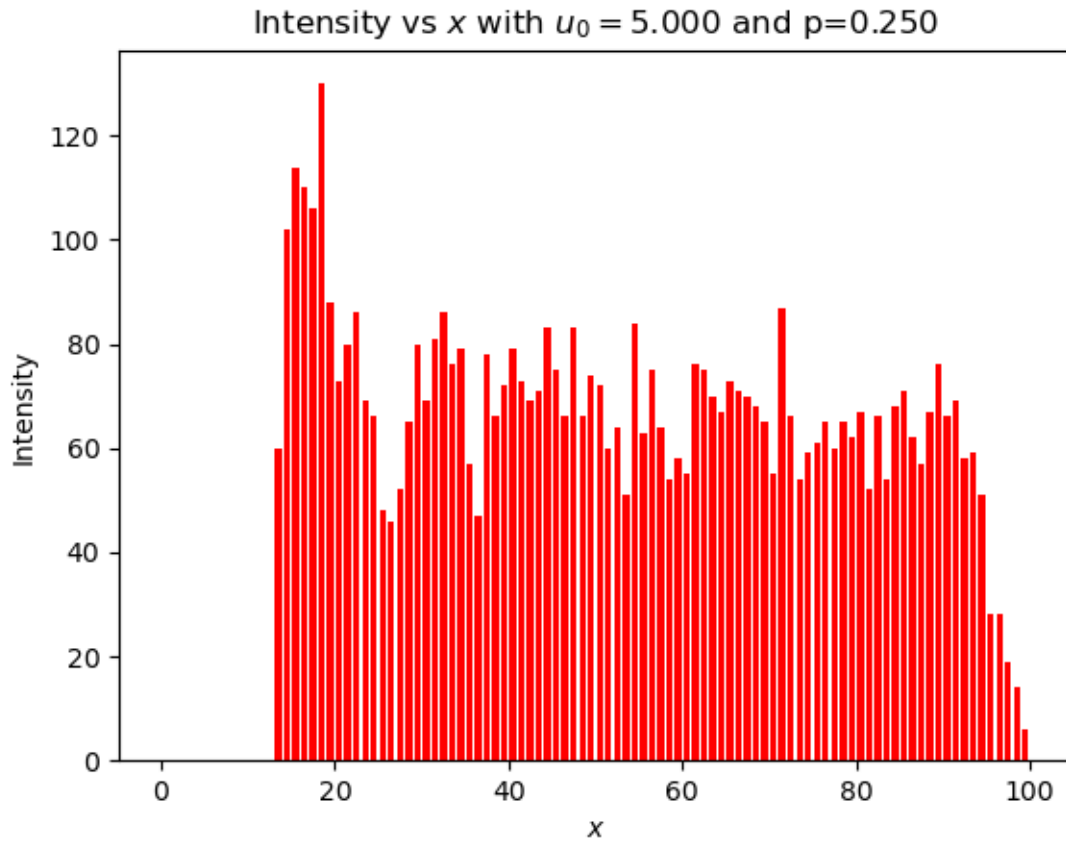


Figure 3: Intensity vs position histogram plot for given values

#### 4.5.3 Plot of V vs X

Here, V is the velocity. X is the x-axis and V is the y-axis. We use a **x**, cross marker at each plot point.

```
#electron phase plot
plot(X,V,'bx')
title('Electron phase plot with $u_0=%0.3f$ and $p=%0.3f$'%(u0,p))
xlabel('$x$')
ylabel('$v$')
show()
```

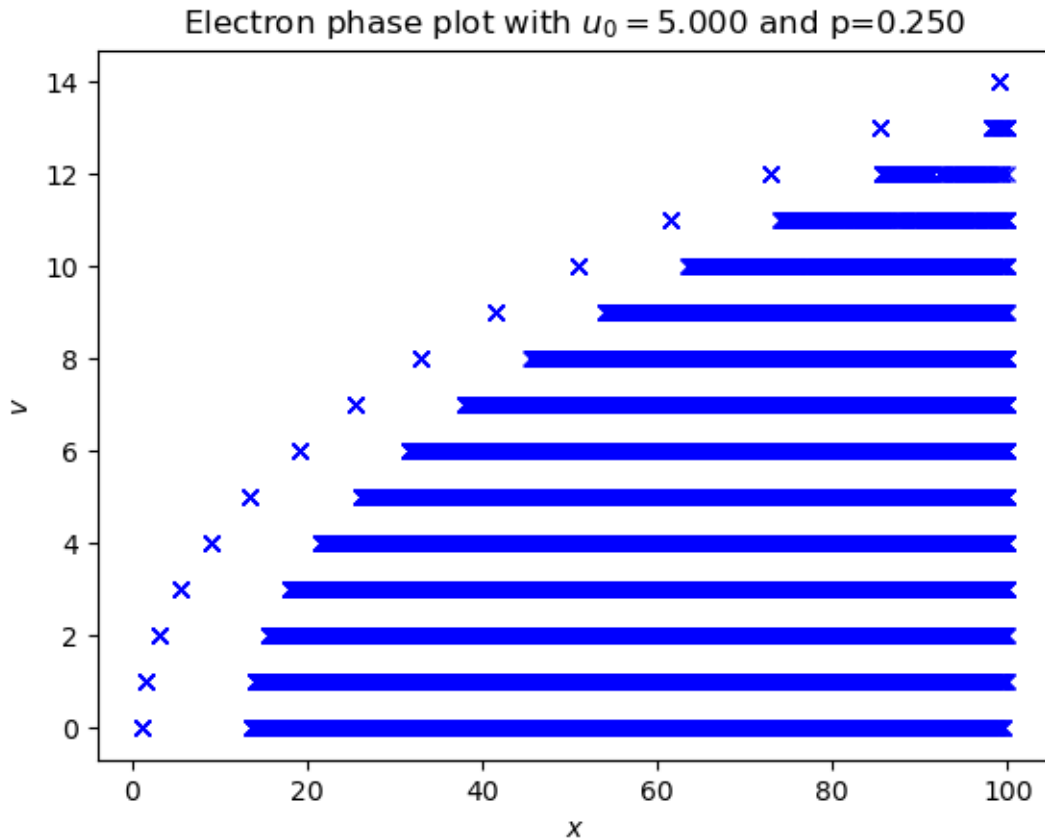


Figure 4: electron density vs velocity plot

#### 4.5.4 Intensity density plot

This plot is amazing, as it basically shows us how the real life light bulb would look like. We have used the *Greys\_r* colormap that indicates the bright and dark regions of the tubelight.

```
#plots the Intensity density plot
bins = arange(0,101)
hist, edges = np.histogram(I, bins)
hist=hist[newaxis,:]
extent=[bins.min(), bins.max(),0,1]
imshow(hist, aspect = "auto", cmap="Greys_r", extent=extent)
title('Intensity Density plot with $u_0=%0.3f$ and $p=%0.3f$'%(u0,p))
gca().set_yticks([])
colorbar()
show()
```

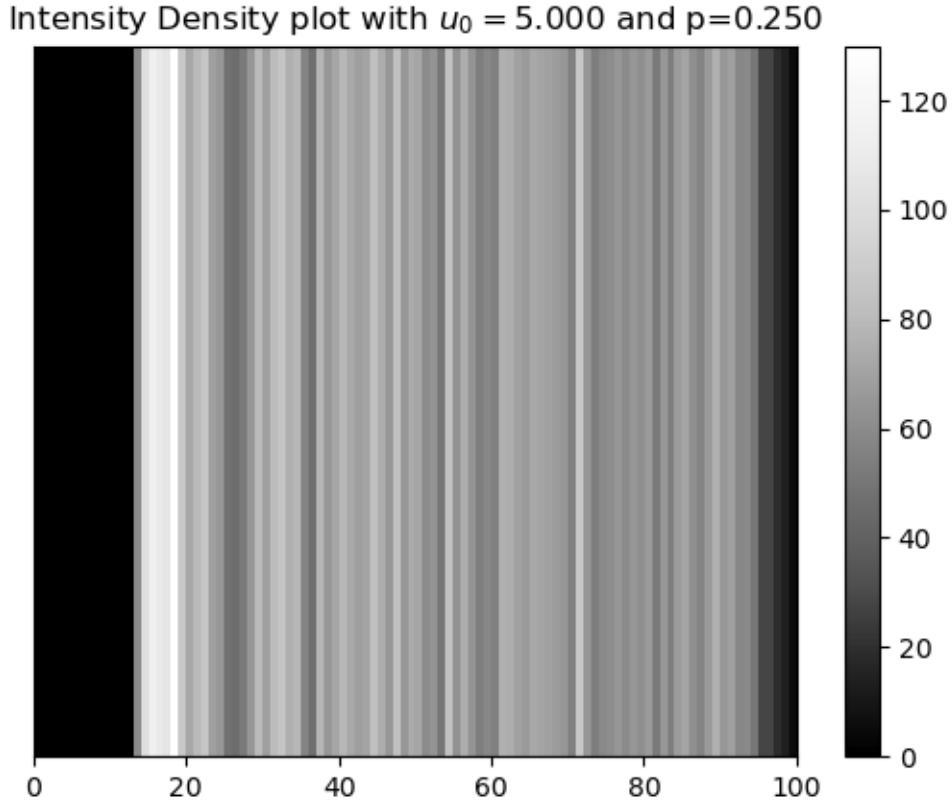


Figure 5: Intensity density plot

## 4.6 Answering Assignment Questions

### 4.6.1 Question1

Yes, we can actually get a better estimation for apparent  $dx$  using a **Random TIME** instead of a random number multiplied by  $dx$ . we then find this random time and find

$$dx'[kl] = u[kl] * tt + 0.5 * (tt)^2 \quad (1)$$

### 4.6.2 Question2

Yes, we can use the last  $ii$  as the initial  $ii$  of the next iteration.

To do this we, would need to define one loop of the for loop as a function that also returns the set  $ii$  as an output. We can define this set  $ii$  as an input of the function as well. So now, the output  $ii$  is used as the input  $ii$  for the next loop.

### 4.6.3 Question3

We see that we get more prominent black spaces when we increase  $u_0$  (thresh hold velocity) and decrease  $p$  (probability that ionization will occur).

We can see a prominent dark region for  $u_0 = 7$  and  $p = 0.25$ . (For visualisation, see the intensity density plot below).



## 4.7 Remaining Plots

### 4.7.1 For $u_0 = 5$ and $p = 0.5$

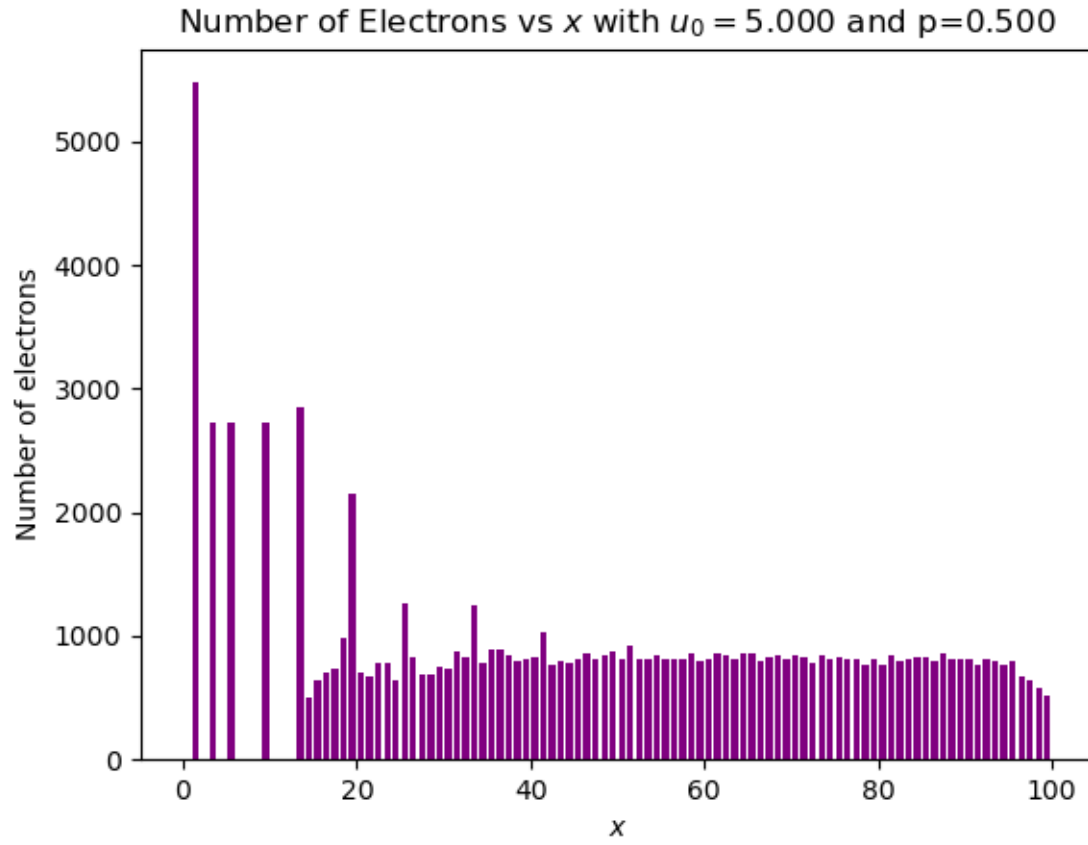


Figure 6: electron density vs position histogram plot for given values

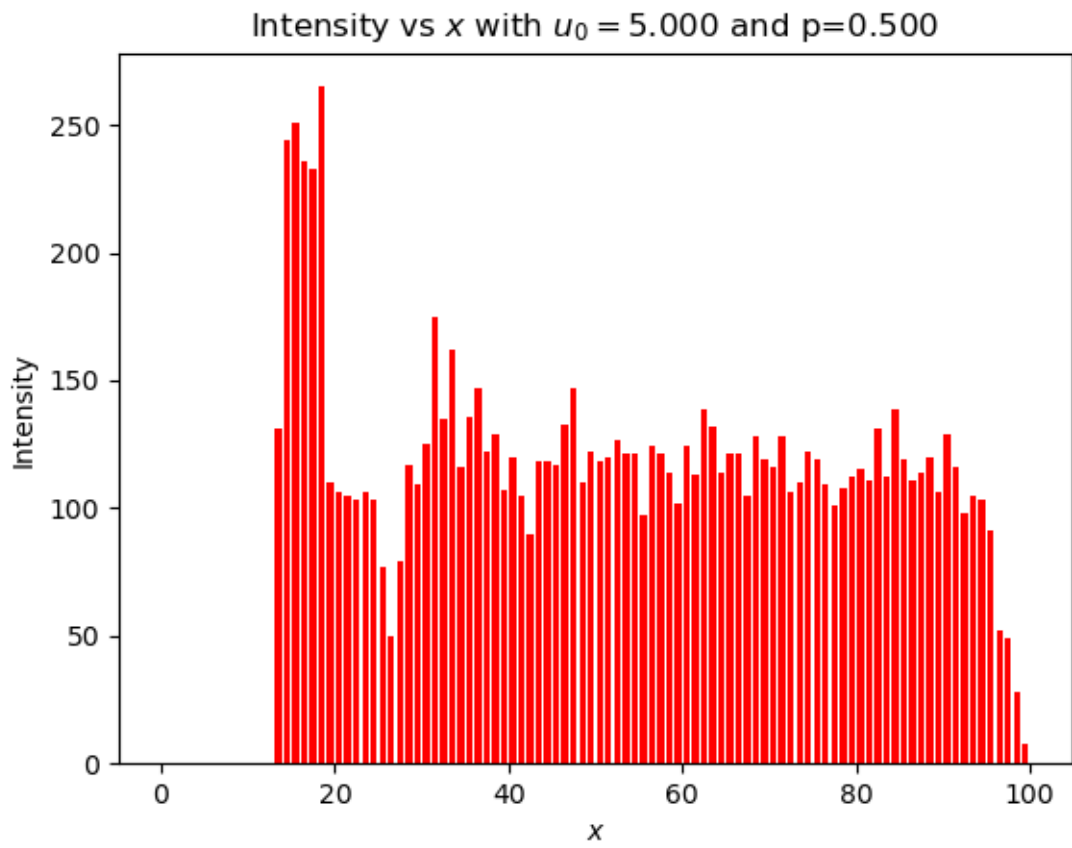


Figure 7: Intensity vs position histogram plot for given values

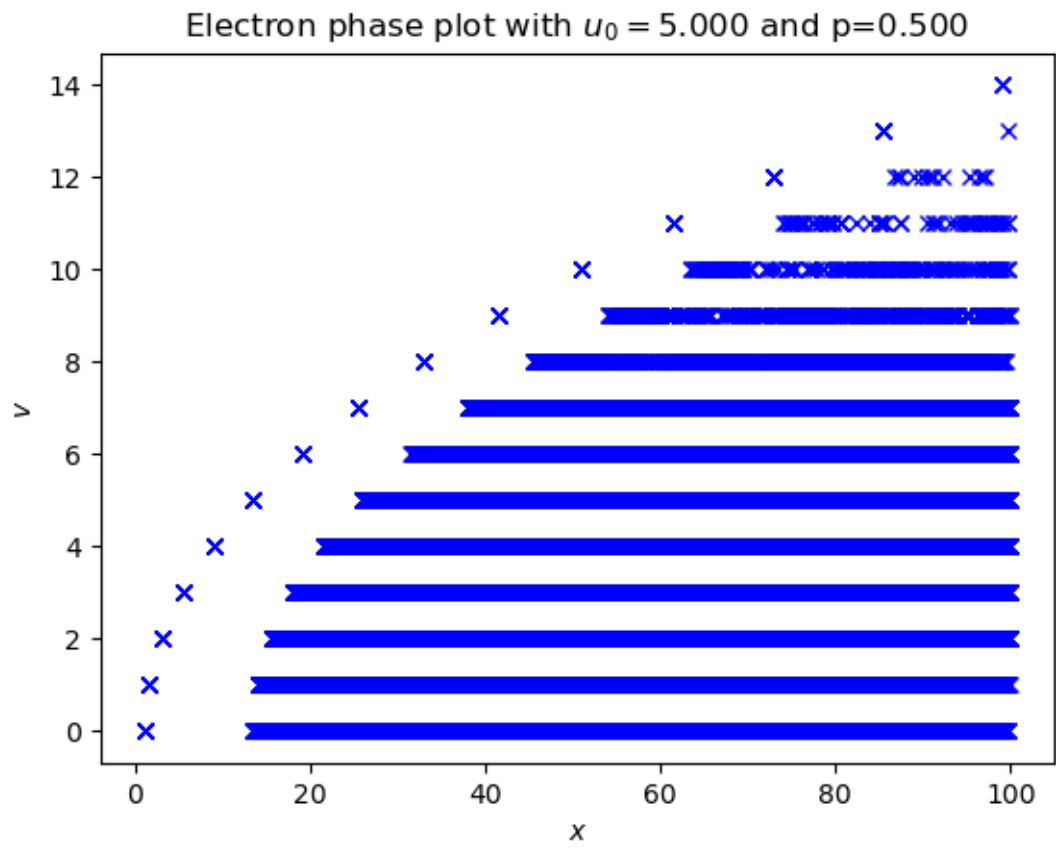


Figure 8: electron density vs velocity plot

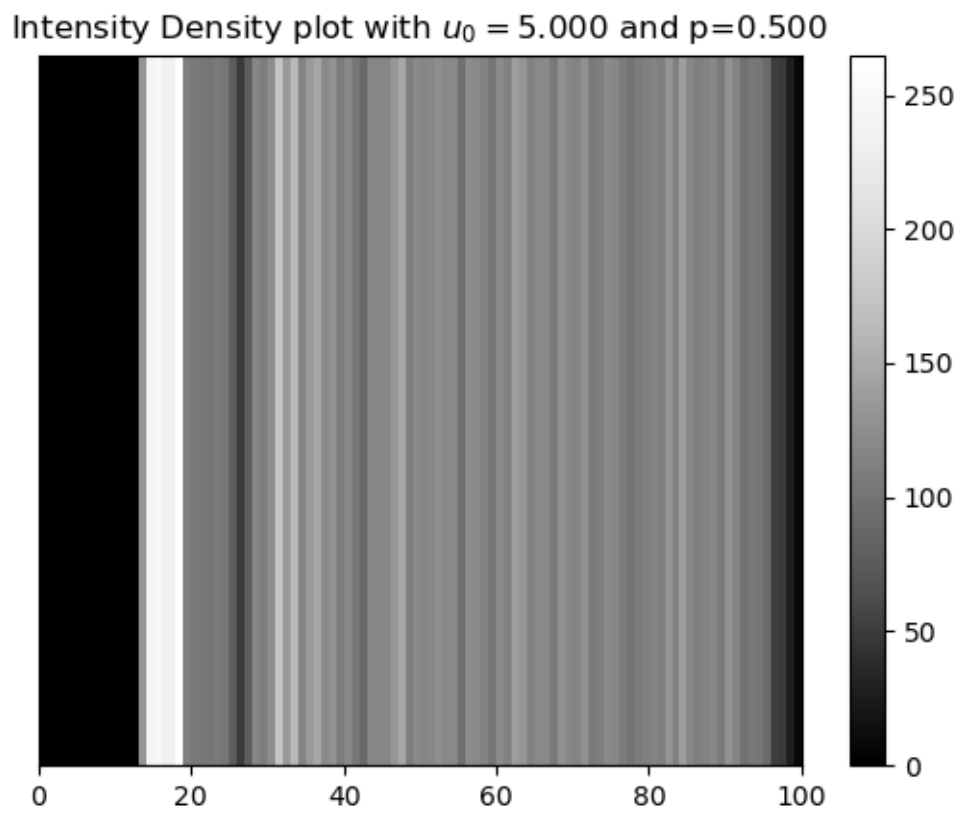


Figure 9: Intensity density plot

4.7.2 For  $u_0 = 7$  and  $p = 0.5$

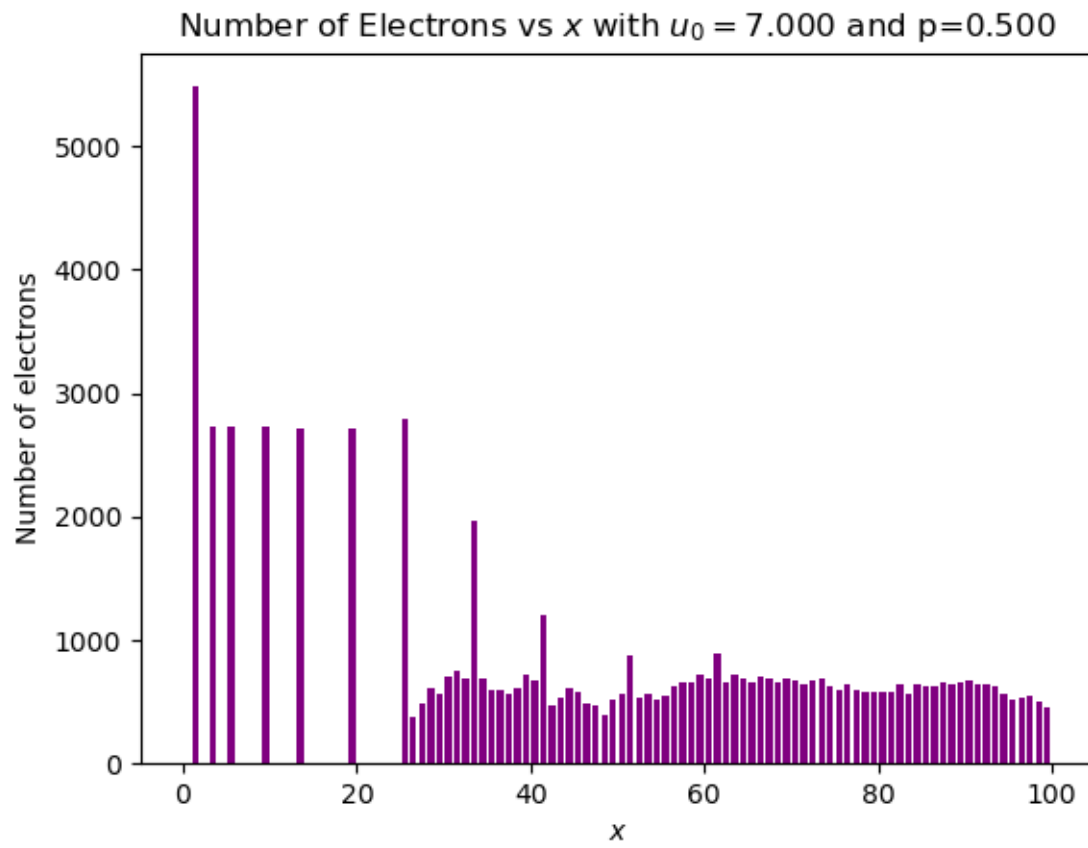


Figure 10: electron density vs position histogram plot for given values

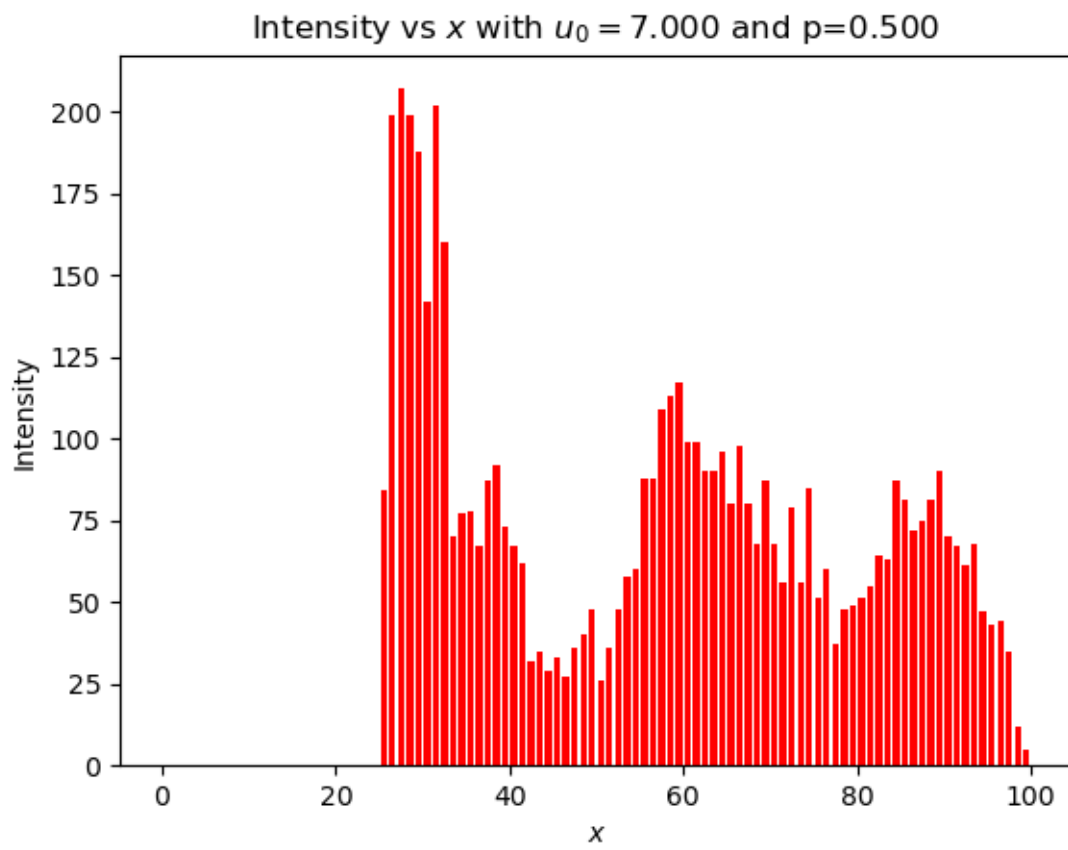


Figure 11: Intensity vs position histogram plot for given values

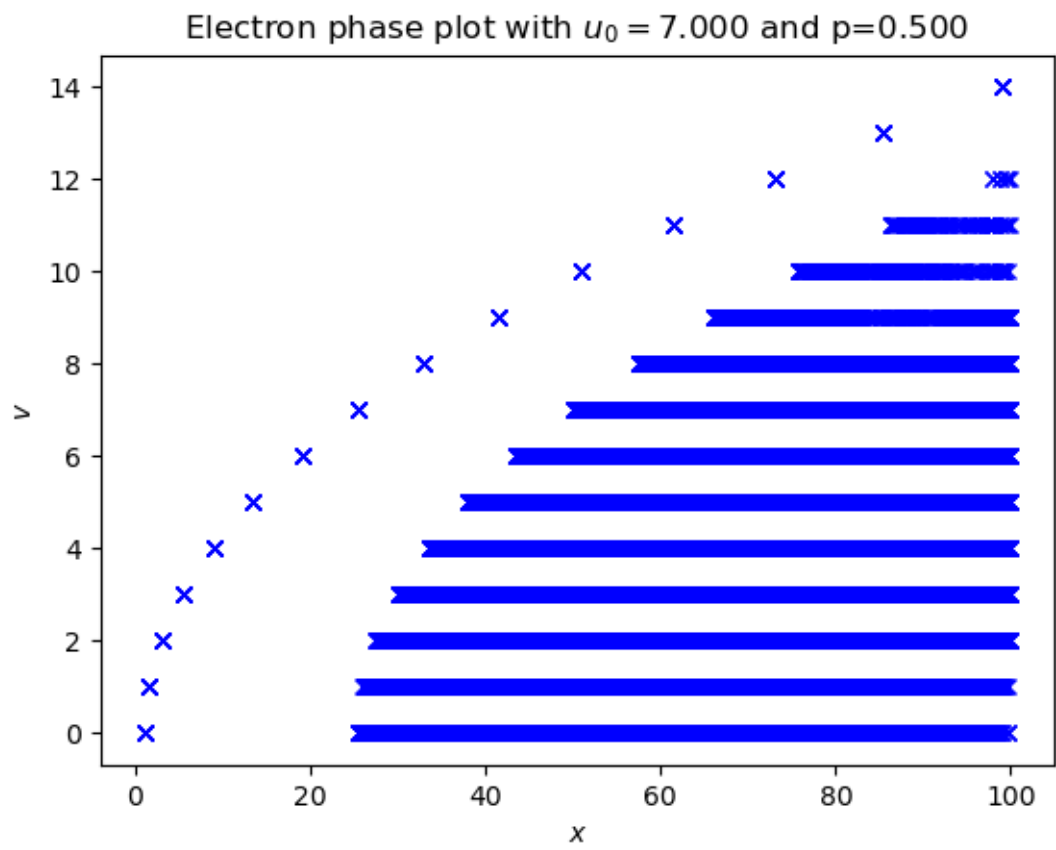


Figure 12: electron density vs velocity plot

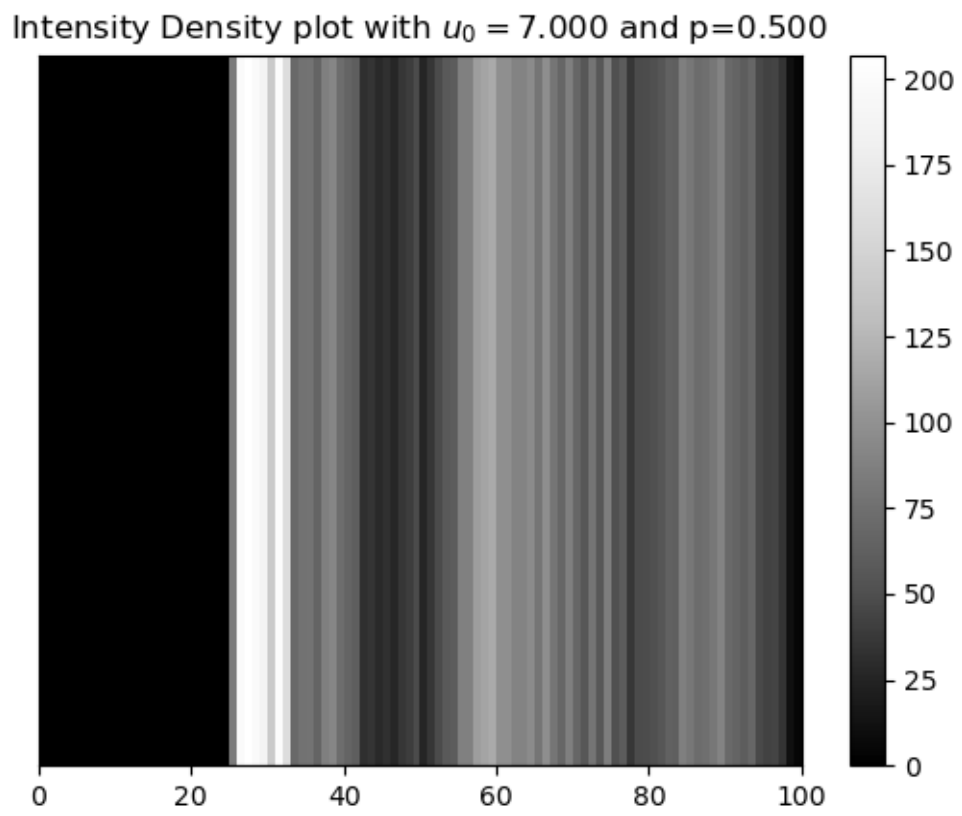


Figure 13: Intensity density plot



4.7.3 For  $u_0 = 2$  and  $p = 0.75$

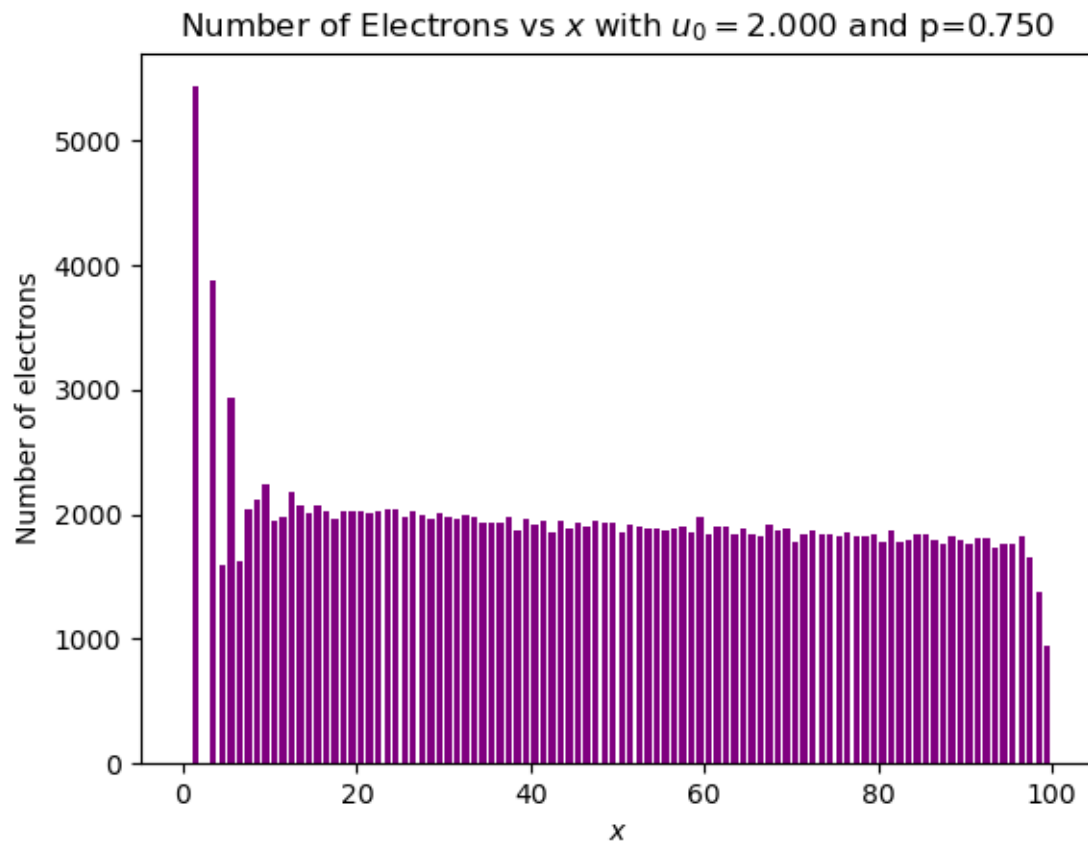


Figure 14: electron density vs position histogram plot for given values

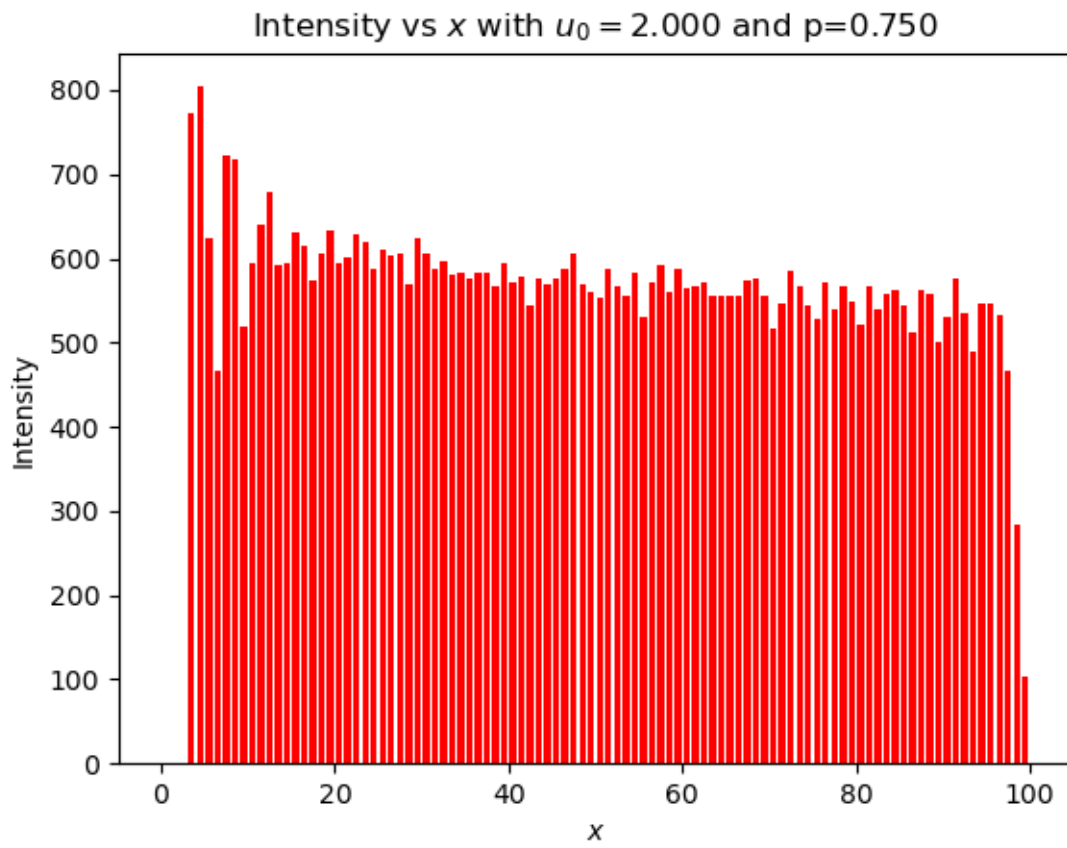


Figure 15: Intensity vs position histogram plot for given values

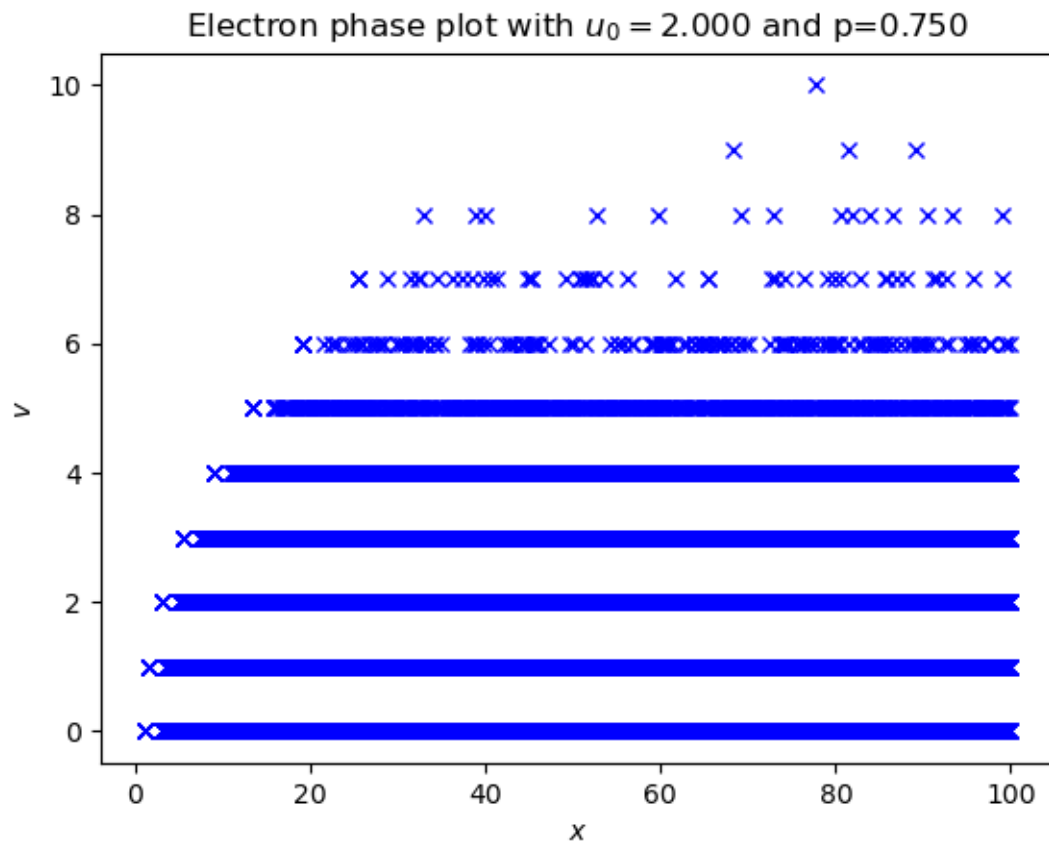


Figure 16: electron density vs velocity plot

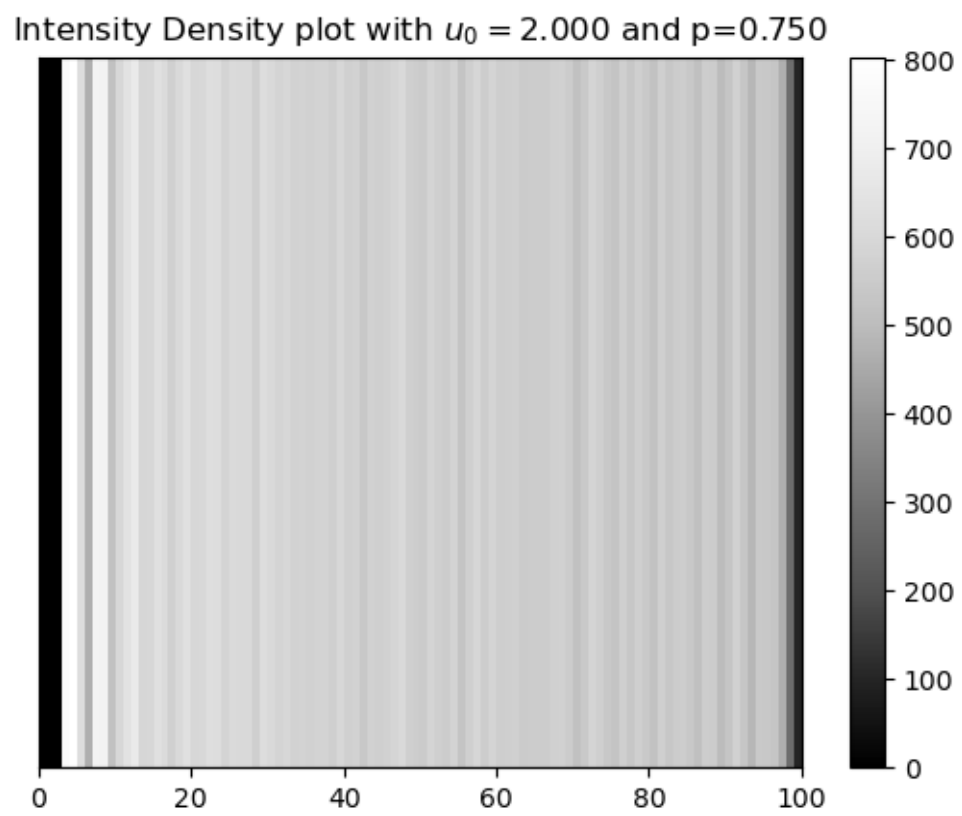


Figure 17: Intensity density plot