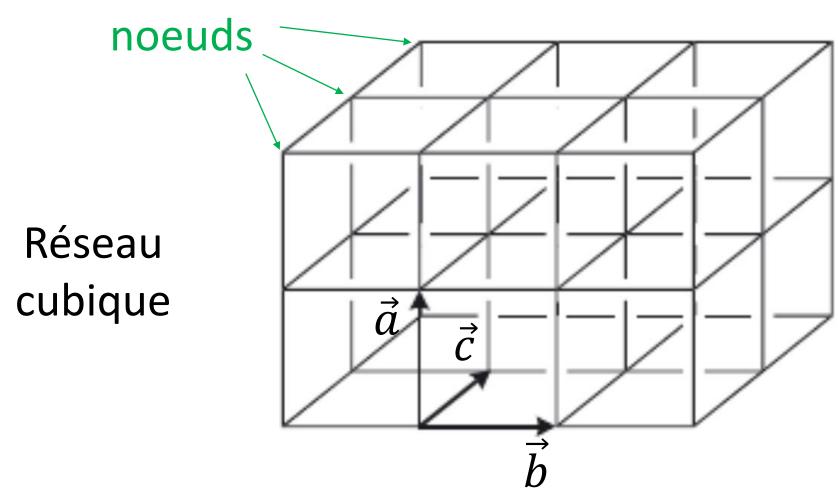
### LC17: Solides cristallins

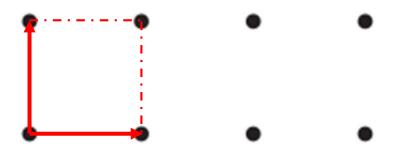
#### Réseau et noeuds

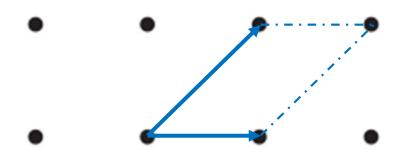


Bruno FOSSET, Jean-Bernard BAUDIN et Frédéric LAHITETE. Chimie tout-en-un PCSI.Dunod, 2016.

## Illustration du concept de maille (2D)

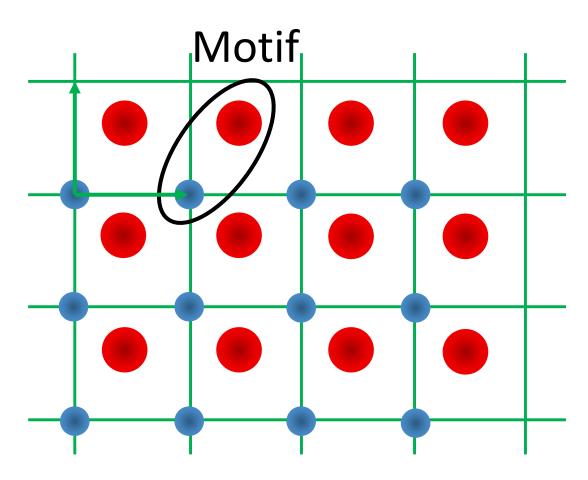
Deux exemples de mailles pour un réseau carré



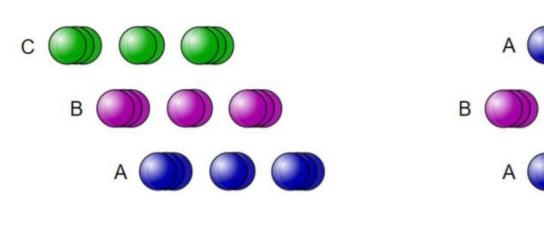


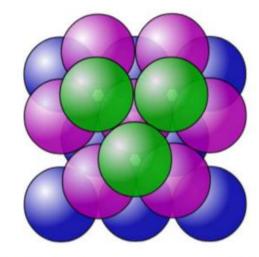
### Réseau + motif = cristal

Réseau carré

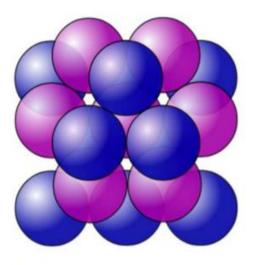


### **Empilements compacts**



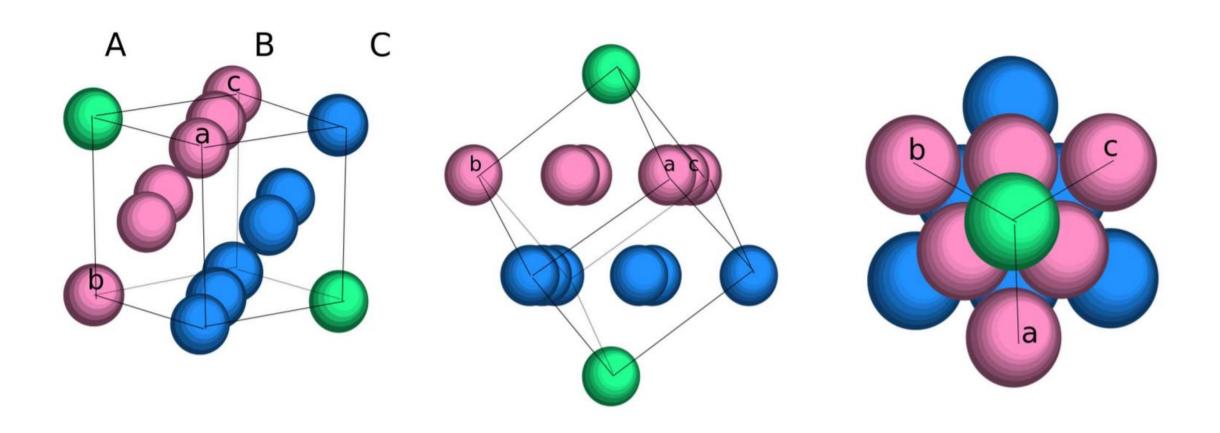


Cubique à faces centrées ABC

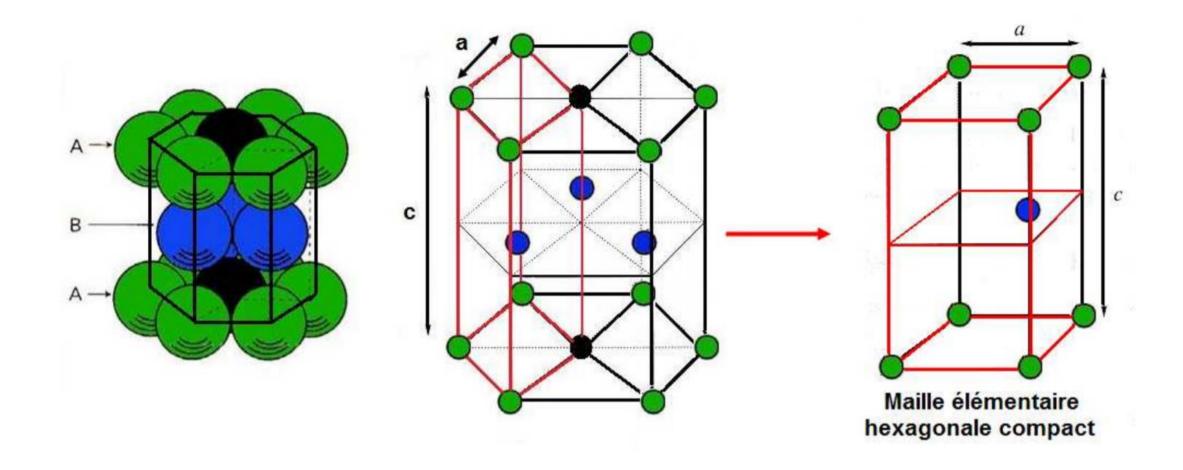


Hexagonal compact ABA

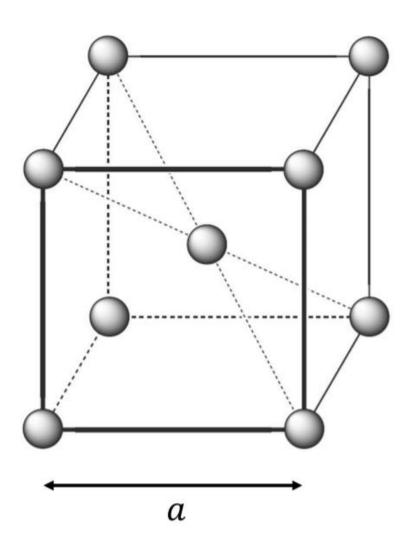
## Structure cubique face centrée



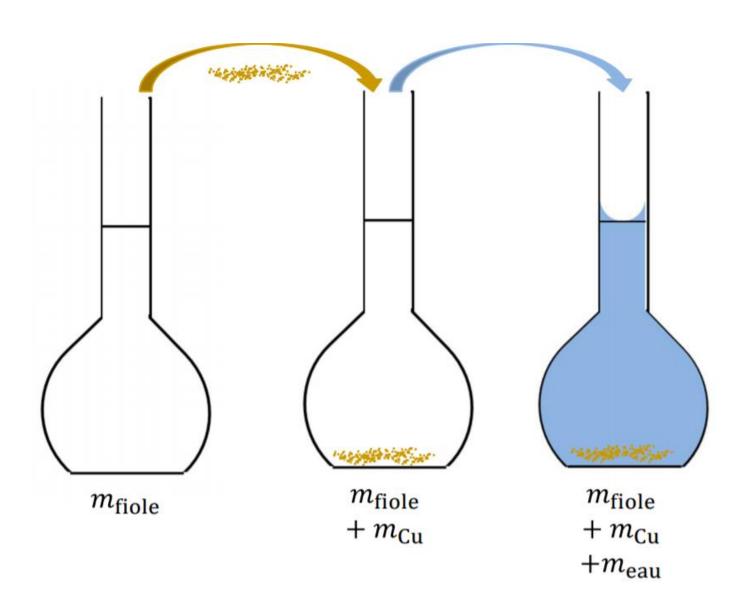
## Structure hexagonal compacte



# Exemple d'empilement non compact : la structure cubique centrée



### Mesure du paramètre de maille du cuivre

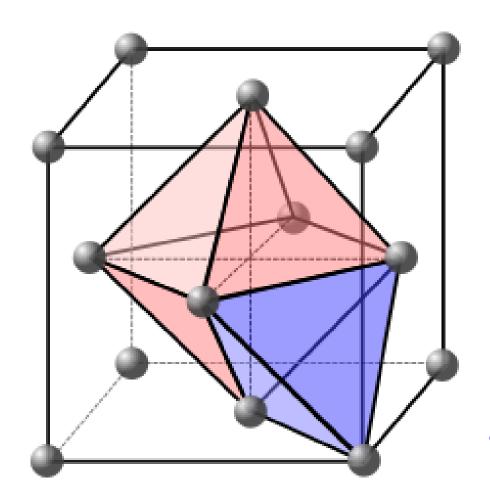


$$V_{\rm Cu} = V_{\rm fiole} - \frac{m_{\rm eau}}{\rho_{\rm eau}}$$

$$\rho_{\mathrm{Cu}} = \frac{m_{\mathrm{Cu}}}{V_{\mathrm{Cu}}}$$

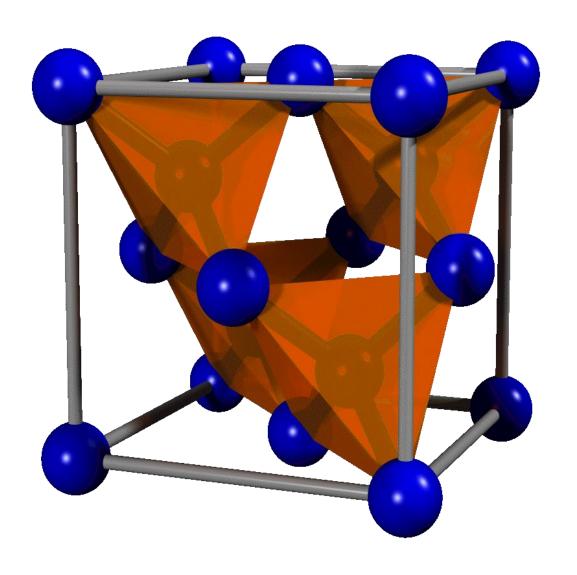
### Sites tétraédriques et octaédriques

Site octaédrique

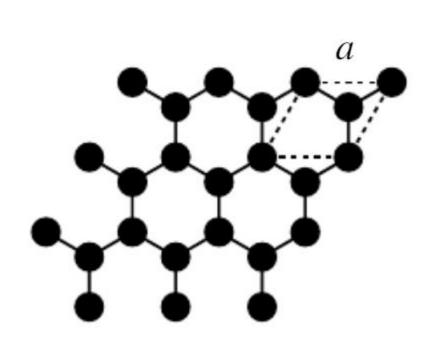


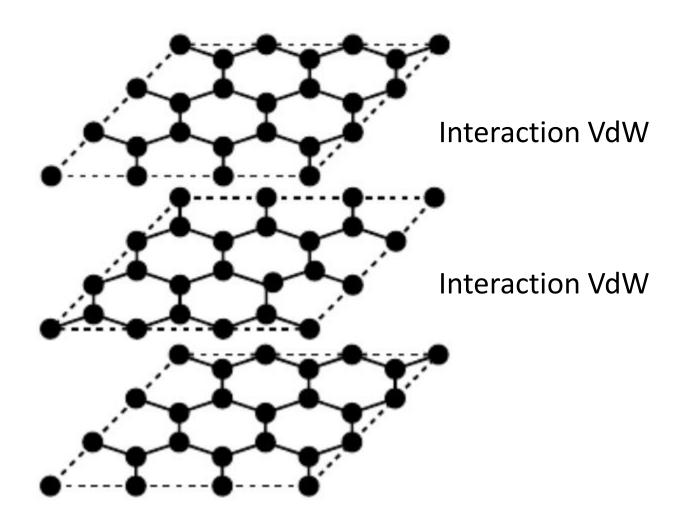
Site tétraédrique

#### Le carbone diamant : un cristal covalent

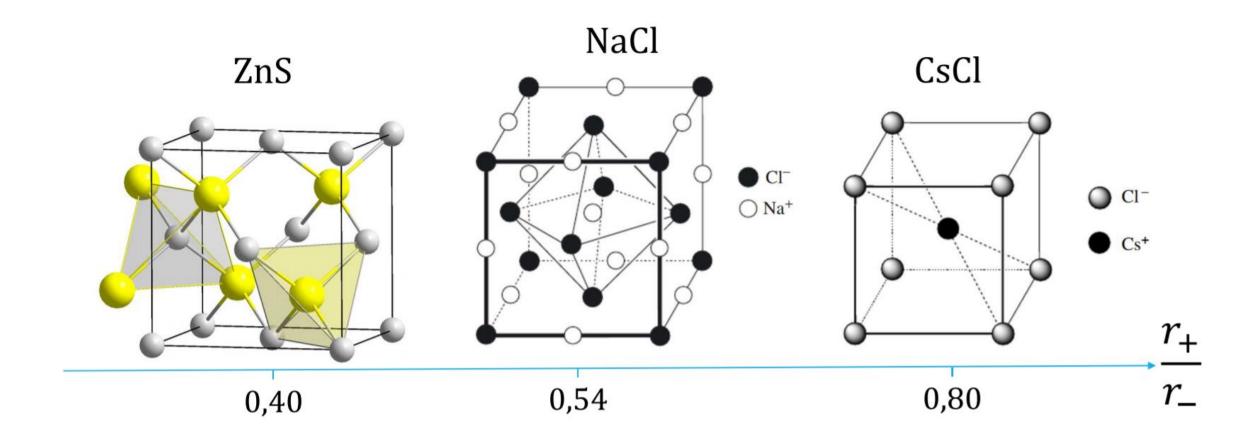


# La carbone graphite

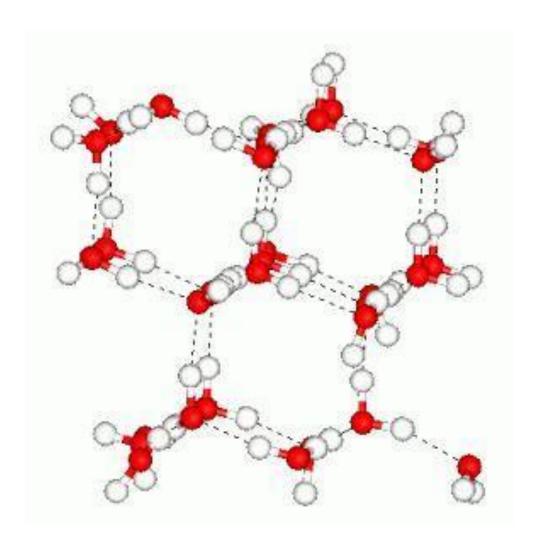




# **Evolution de la structure avec le rapport des rayons des ions**



# Un cristal moléculaire : la glace I



--- Liaisons hydrogène

