

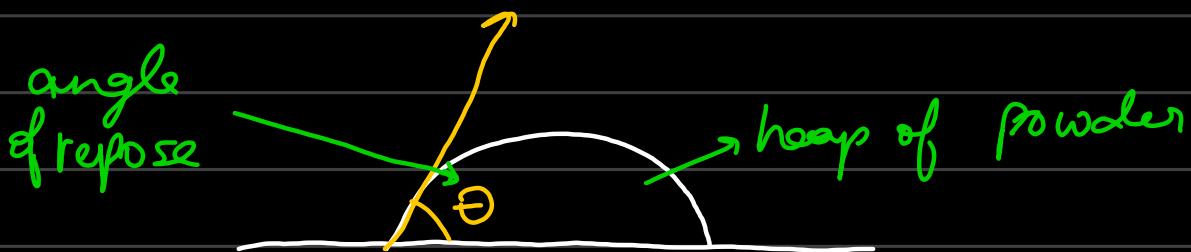
Lecture 15

Power Processing

Agglomeration:

↳ agglomerated particle would always reduce packing density because of irregular shape.

→ Cohesive forces causes agglomeration.



$\theta > 45^\circ \Rightarrow$ high cohesive forces
high agglomeration tendency

θ low \Rightarrow good flowability,
 $\{0-20^\circ\}$ low cohesive force
low agglomeration tendency

→ De-Agglomeration can be accomplished by
a combination of drying, milling &
surface treatments.

Functionalised surface: surface treatment

Q) Calculate X_c that would yield best packing fraction, if it is known that smaller particles pack in loose random packing and larger particles acquire dense random packing, when required.

$$PF_{\text{fine}} = 0.6$$

$$PF_{\text{coarse}} = 0.64$$

$$PF_{\text{app}} = PF_c + PF_f (1 - PF_c)$$

$$\begin{aligned} PF_{\text{app}} &= 0.64 + 0.6 (0.34) \\ &= 0.844 \end{aligned}$$

$$PF_{\text{app}} = PF_f + X_c (1 - PF_f)$$

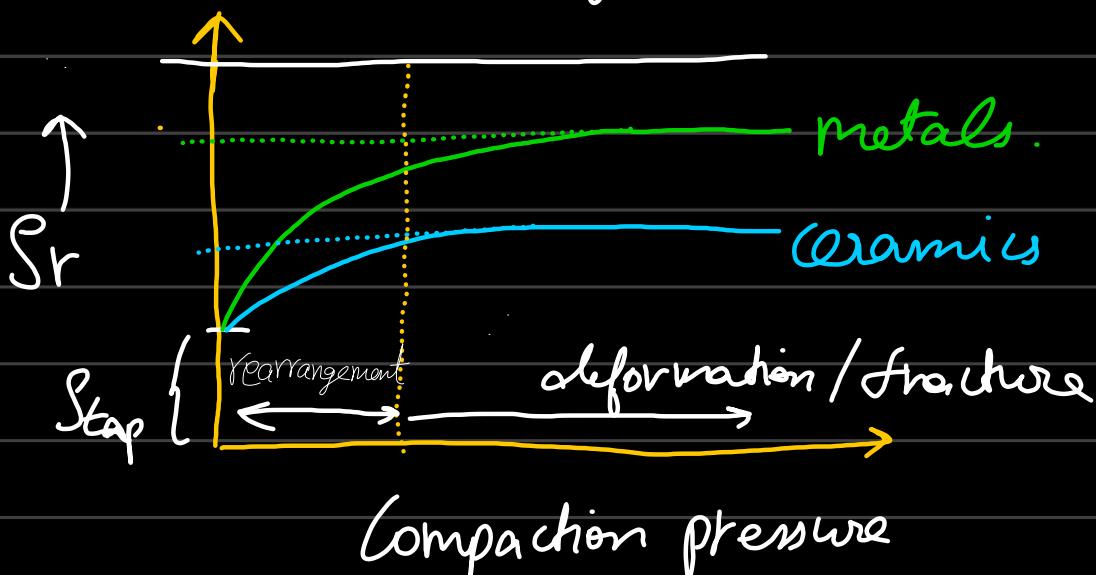
$$0.844 = 0.6 \times X_c (0.4)$$

$$0.244 = X_c \times 0.4$$

$$\boxed{X_c = 0.61}$$

Powder Compaction:

Stages in Compaction: tapping, rearrangement,
↳ 3rd stage = fracture deformation.
in case of ceramics.

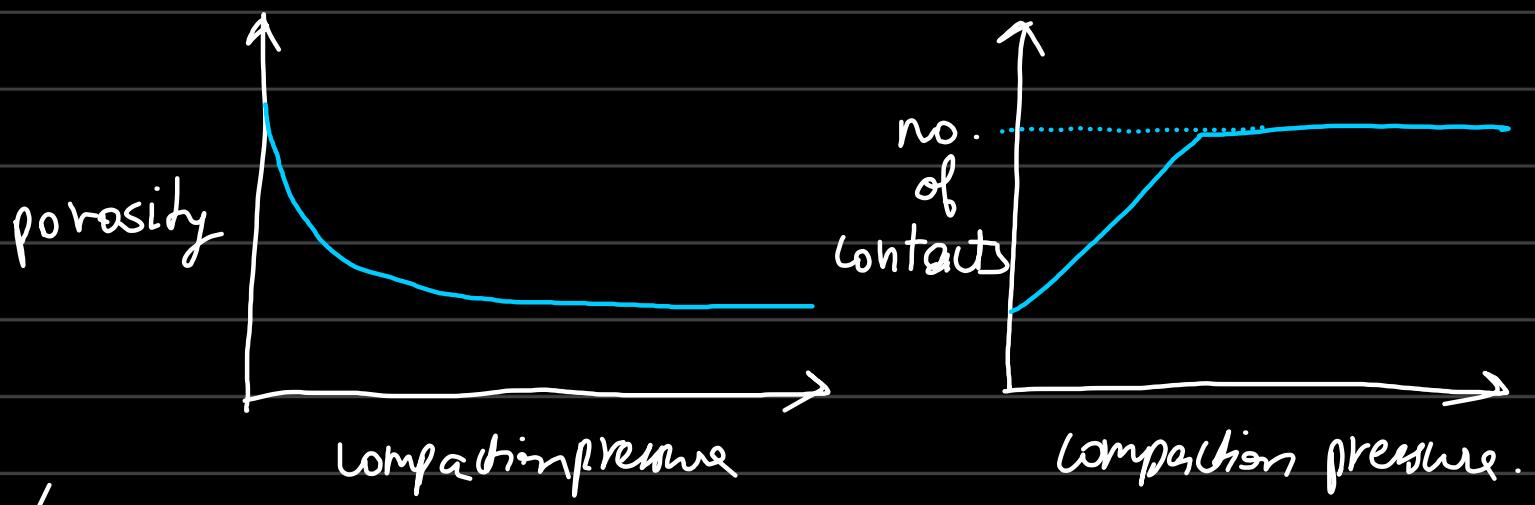


→ Saturation (decrease in slope)
due to Work Hardening

→ Hot/Warm Compaction:

↳ dynamic recovery & dynamic recrystallization can happen which soften and allow increase in strain.

→ not popular as high surface area allows for rapid oxidation.

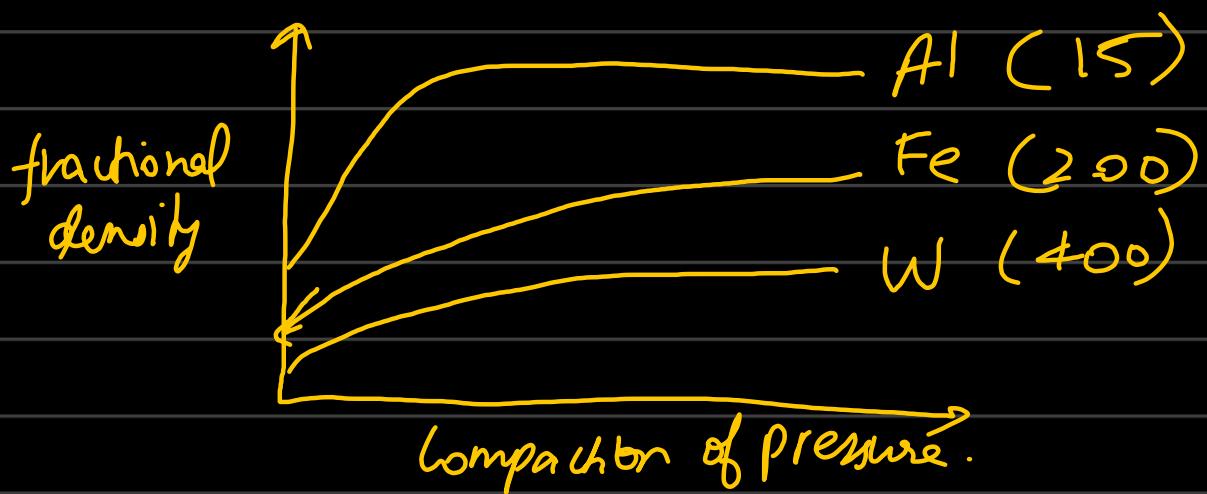


↳ Typical green density :

Metals > 85%

Ceramics < 60%

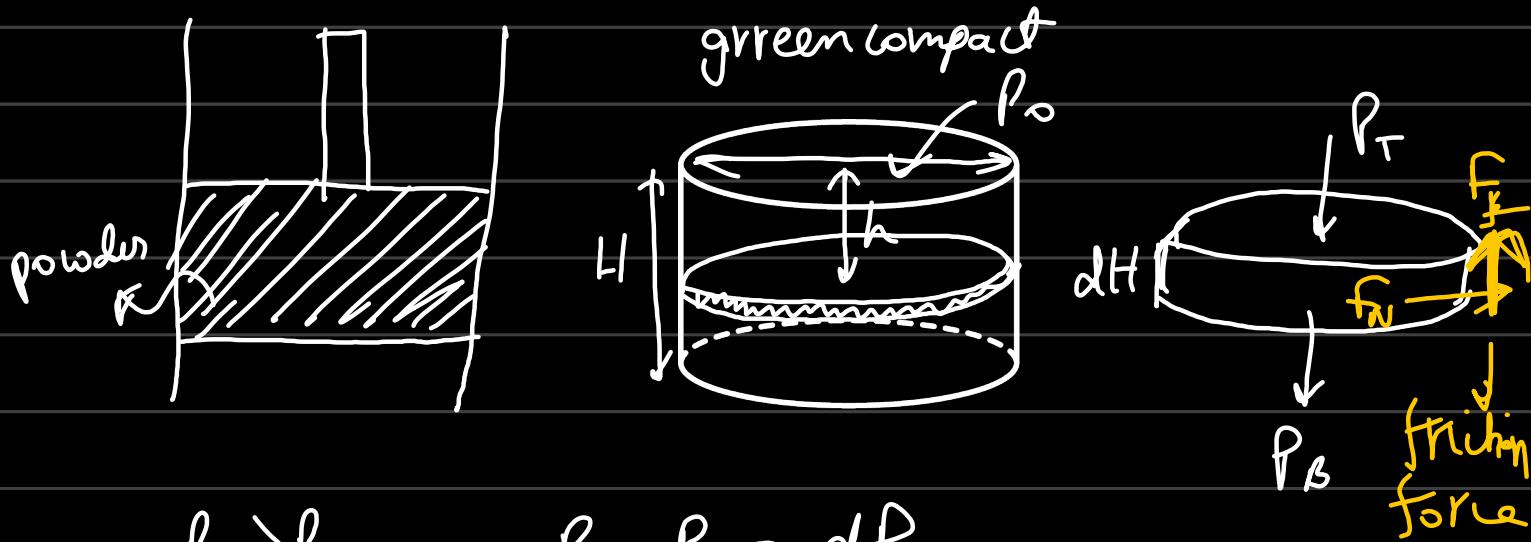
- * Al - Softest (fcc)
- Steel → Hard (bcc)
- Mg → Brittle (hcp)



Hardness & Melting Temperature.

Friction Analysis

- friction can cause varying pressure across the pattern along the depth & radius.
- friction decides spatial distribution of pressure.



$$P_T > P_B \Rightarrow P_T - P_B = dP$$

$$F_f = \mu F_N$$

↳ coefficient of restitution.

$$F_N = \pi (P_T) (\pi D) (dH)$$

$$F_f = \mu \pi P_T \pi D dH$$

Apply force balance:

$$A(P_T - P_B) + \mu f_N = 0$$

$$\Delta P = -\mu f_N$$

$$\frac{\pi D^2}{4} dP = -\mu Z P_T \pi D dh$$

$$dP = - \frac{4 \mu Z P_T dh}{D}$$

$$\ln \left(\frac{P}{P_0} \right) = - \frac{4 \mu Z h}{D}$$

$$P = P_0 e^{-\frac{4 \mu Z h}{D}}$$

decaying of pressure.

→ not acceptable as
for low pressure
more shrinkage
takes place.

→ tapered compact formed
after sintering.

