

Lecture 12

$e, \rho_h, \text{ atoms}$ waves particles \rightarrow photoelectric effect
↳ conduction in metals

Drude's conductivity:

$$\sigma = \frac{ne^2}{m} \tau = (Nee) \cdot \left(\frac{e\tau}{m} \right)$$

relaxation time

no. of carrier charge

mobility

$$v_d = \mu E$$

↳ mobility { scaling factor / proportionality const }

$T \Rightarrow$ statistical thermodynamic variable

$\tau \Rightarrow$ mean time to scatter

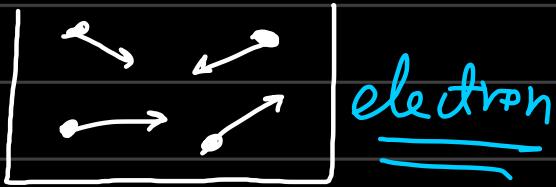
$\mu \Rightarrow$ also mean value

$v_d \Rightarrow$ mean drift velocity

$$R_H = \frac{1}{N_c e} \quad \{ \text{Hall coefficient} \}$$

$$\boxed{E_y = R_H J_k B} \quad \text{transverse potential}$$

$$\sigma(\omega) = \frac{\sigma_0}{1 - i\omega\tau}$$



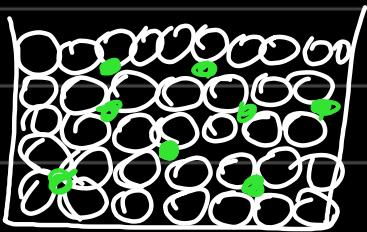
ω_p = plasma frequency

$$\omega_p = \frac{n e^2}{m \epsilon} \quad \text{(intrinsic material property)}$$



$\omega < \omega_p \Rightarrow \text{decay}$

$\omega > \omega_p \Rightarrow \text{transmits.}$



hole

Vacancy movement can be tracked.

} describes transparency of materials to EM radiation.

The time-independent SF eqn:

Free particle: energy is continuous

Particle in a box: energy is quantized:

$$E = \frac{\hbar^2 k^2}{2m} \quad k = \frac{2\pi n}{a}$$

$$= \frac{\hbar^2}{4\pi^2} \times \frac{4\pi^2 n^2}{2ma^2} = \underline{\underline{\frac{\hbar^2 n^2}{2ma^2}}}$$

$$\text{Hydrogen atom: } V = -\frac{kZe^2}{r} = -\frac{Ze^2}{4\pi\epsilon_0 r}$$

$$\Psi(r, \theta, \phi) = R(r) \Theta(\theta) \underline{\Phi}(\phi)$$

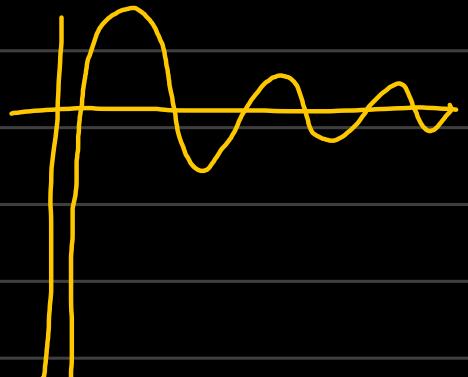
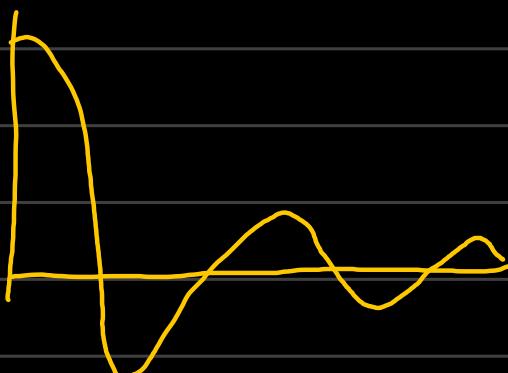
$$\frac{\partial^2 \underline{\Phi}}{\partial \phi^2} = -m^2 \underline{\Phi}$$

$$\underline{\Phi}(\phi) = e^{im\phi}$$

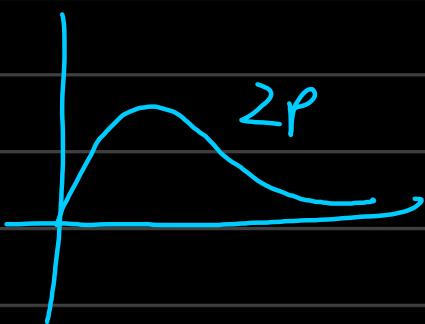
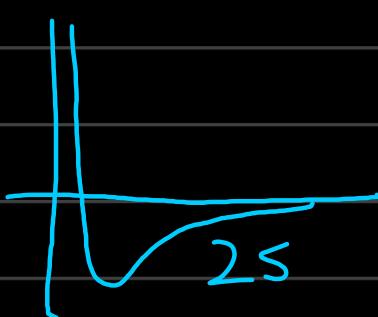
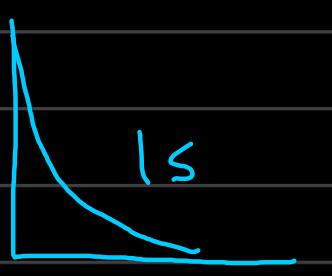
$$u(r) = A r J_\ell(kr) + B r N_\ell(kr)$$

\downarrow
Bessel's func

\curvearrowright
Neumann func.



Radial wavefunctions:



$$\text{A} \quad E(3p) > E(3s)$$

\downarrow
due to Orbital Angular momentum.
 \hookrightarrow contributes an additional energy.

Angular momentum:

$$\hat{L} \Psi = L \Psi$$

angular momentum
operator.

$$L = \hbar \sqrt{l(l+1)}$$

$$L_z = m_e L$$

→ Selection rules:

↳ Electronic transition. $\Delta l = \pm 1$

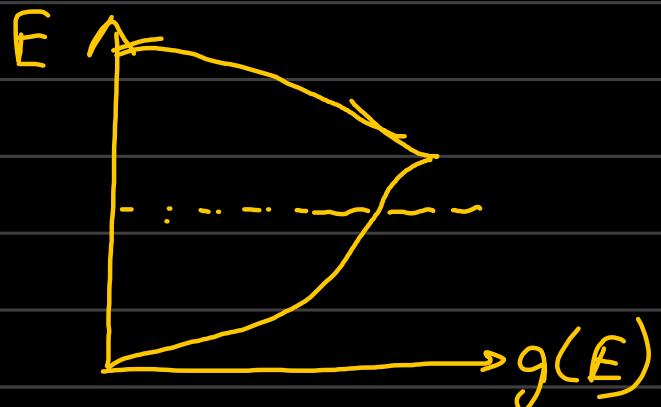
Angular momentum
is conserved

~~$S \rightarrow d$~~
 $S \rightarrow p$

Density of States:

$$D = \frac{1}{V} \frac{dN}{dE}$$

no.
energy levels
(eV cm^{-3})

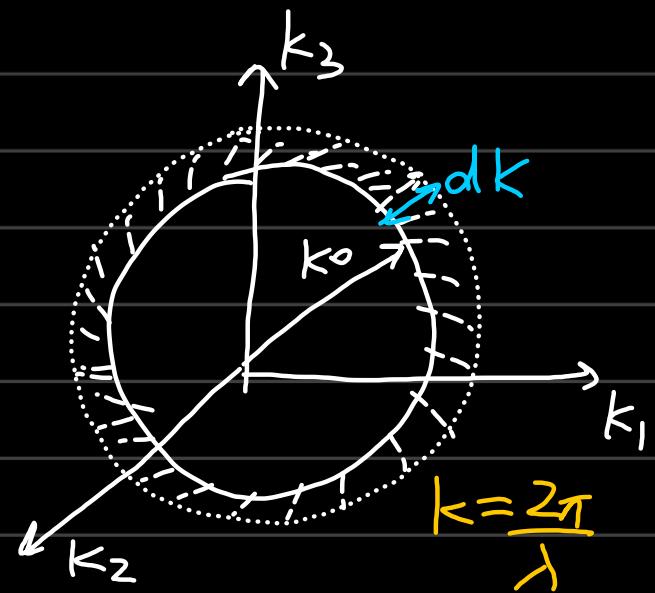


→ Anti-symmetric energy addition has higher no. of configurations, hence accounts for higher density of states.

$$\rightarrow E = \frac{\hbar^2 k^2}{2m} \quad p = \hbar k \quad ; \quad E = \frac{p^2}{2m}$$

$$dE = \frac{\hbar k}{m} dk$$

$$E_0 = \frac{\hbar^2 k_0^2}{2m}$$



$$E_0 \rightarrow E_0 + dE$$

$$k_0 \rightarrow k_0 + dk$$

$N = ?$ {How many electrons are added?}

$$k = \frac{n\pi}{a} \quad ; \quad \Delta k = \frac{\pi}{a}, \quad \Delta V_k = \left(\frac{\pi}{a}\right)^3$$

smallest possible

No. of states added: $E_0 \rightarrow E_0 + dE \rightarrow k_0 + dk$

$$dN = \frac{\text{vol. increment}}{\text{vol. discreteness}}$$

$$= \frac{4\pi k_0^2 dk}{\left(\frac{\pi}{a}\right)^3} = a^3 \times \frac{4}{\pi^2} k_0^2 dk$$

$$dN = V \times \frac{4}{\pi^2} \times k_0^2 \times \frac{dE m}{\hbar^2 k_F}$$

$$\frac{dN}{V dE} = \frac{4 k_0 m}{\pi^2 \hbar^2} = \underline{\text{Density of states (D)}}$$