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# RESEARCH ARTICLE

# Modelling the physiological status of yeast during wine fermentation enables the prediction of secondary metabolism

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#### **Abstract**

Saccharomyces non-cerevisiae yeasts are gaining momentum in wine fermentation due to their potential to reduce ethanol content and achieve attractive aroma profiles. However, the design of the fermentation process for new species requires intensive experimentation. The use of mechanistic models could automate process design, yet to date, most fermentation models have focused on primary metabolism. Therefore, these models do not provide insight into the production of secondary metabolites essential for wine quality, such as aromas. In this work, we formulate a continuous model that accounts for the physiological status of yeast, that is, exponential growth, growth under nitrogen starvation and transition to stationary or decay phases. To do so, we assumed that nitrogen starvation is associated with carbohydrate accumulation and the induction of a set of transcriptional changes associated with the stationary phase. The model accurately described the dynamics of time series data for biomass and primary and secondary metabolites obtained for various yeast species in single culture fermentations. We also used the proposed model to explore different process designs, showing how the addition of nitrogen could affect the aromatic profile of wine. This study underlines the potential of incorporating yeast physiology into batch fermentation modelling and provides a new means of automating process design.

# INTRODUCTION

The winemaking sector faces new challenges. In recent years, concerns about the impact of alcohol consumption on health and road safety have increased (Poznyak & Rekve, 2015). Additionally, alcohol negatively affects the perception of the complexity of wine flavours. Therefore, the wine industry is actively seeking ways to produce wines with a lower alcohol concentration, ideally, without compromising taste and consumer's acceptance or increasing production costs (Varela et al., 2015).

Yeasts play a critical role in responding to these challenges. Winemakers count on hundreds of commercially available *Saccharomyces cerevisiae* starters that differ in aroma profiles (Scott et al., 2021). However, all of them generate similar ethanol yields, translating into comparable concentrations of ethanol in wine. Alternatives may emerge from exploring yeast biodiversity (Hittinger et al., 2018; Steensels & Verstrepen, 2014).

Cold-tolerant species of the *Saccharomyces* genus, namely, *S. kudriavzevii* and *S. uvarum*, and their hybrids with *S. cerevisiae*, have shown promising physiological properties (Pérez-Torrado et al., 2018). Recent research explored

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optimal growth temperatures (López-Malo et al., 2013), resistance to ethanol (Lairón-Peris et al., 2020) or aroma production (Gamero et al., 2013; Pérez et al., 2022) in fermentations led by these species. These studies show a general decrease in ethanol production and an increase in glycerol (Querol et al., 2018). Although these results reveal great potential for these species to produce wines with a lower ethanol content, the design of the fermentation process requires intensive experimentation.

The use of mechanistic models combined with exometabolome data throughout fermentation could systematize the process design. Models can not only facilitate the selection of the right species for a particular product, but also the automatic design of common interventions such as nitrogen addition to deal with slow fermentations (Bisson & Butzke, 2000). Therefore, the possibility of predicting and optimizing wine fermentation is of utmost importance.

Recently, wine fermentation modelling has received considerable attention (see the review by Miller & Block, 2020). Many authors adopt the macroscopic scale modelling approach to model the dynamics of biomass, relevant nutrients (sugars and yeast assimilable nitrogen, YAN), and products such as ethanol (Coleman et al., 2007; Cramer et al., 2002). These models often rely on the Monod equation to describe the rate of biomass growth as a function of a limiting substrate (e.g. YAN or sugars).

However, available macroscopic models do not account for the five phases observed in batch fermentation, that is, the lag phase, exponential growth, growth under nitrogen limitation, stationary phase and decay phase (Henriques & Balsa-Canto, 2021). In addition, they generally focus on primary metabolism (hexose uptake and ethanol formation), which does not necessarily overlap with the kinetics of aroma production.

Recently, Henriques et al. (2021) suggested an ad hoc approach to account for these phases in a dynamic genome-scale model. In this formulation, different fermentation phases were associated with the production or consumption of different extracellular metabolites. The limitation of this approach is that it lacks a mechanistic connection between the limiting substrate (nitrogen) and the duration of each phase.

In the present work, we formulate a continuous model that can accurately describe fermentation phases. To this end, we extend the model proposed by Henriques and Balsa-Canto (2021) assuming that nitrogen starvation is associated not only with carbohydrate accumulation but also with the induction of a set of transcriptional changes associated with the stationary phase (Rossignol et al., 2003). The final model accounts for the dynamics of yeast biomass and primary metabolites (hexoses, YAN and ethanol) and the production of secondary metabolism compounds (higher alcohols, acetate esters, etc.). The model was calibrated using experimental data obtained during natural must fermentation by four cold-tolerant *Saccharomyces* strains in individual cultures:

S. cerevisiae (ScT73), S. kudriavzevii (SkCR85) and two strains of the species S. uvarum (SuBMV58 and SuCECT12600) (Minebois et al., 2020). The calibrated model successfully explained the data, showing its ability to automatically recover all fermentation phases while accurately capturing the dynamics of aroma production. Remarkably, the model shows relevant qualitative and quantitative differences between the various strains. From a process engineering point of view, the most interesting difference found was that glycerol production and higher alcohol were modulated, at least to a certain extent, by the duration of the adaptation period to nitrogen starvation and that this period appears to be strain dependent.

We finally used the model to explore different scenarios of YAN availability and supplementation showing how final composition varies not only with supplementation but with its timing. Although these particular results require additional experimental verification, since the model accurately describes the dynamics of fermentation with four strains of three different species (*S. cerevisiae*, *S. uvarum* and *S. kudriavzevii*), we are confident in its potential to design and control alcoholic fermentation with other species.

# EXPERIMENTAL PROCEDURES

# **Experimental data**

The experimental data were obtained in a previous study by the authors and reported in Minebois et al. (2020). Here, four yeast strains belonging to the species S. cerevisiae, S. uvarum and S. kudriavzevii were considered: the commercial strain T73 (Lalvin T73 from Lallemand), S. cerevisiae (ScT73); the commercial strain BMV58 (SuBMV58, Velluto BMV58 from Lallemand), the noncommercial strain CECT12600 and the natural isolate S. kudriavzevii CR85 (SkCR85). The time-course experimental data were obtained from fermentation assays performed in Merseguera grape must with added sucrose in order to reach a probable alcoholic grade of 12.5%. All data were derived from three independent biological replicates in 500 ml, temperature-controlled bioreactors set at 25°C. The data-sets include dry weight (DW), hexoses, organic acids, the main fermentative byproducts, yeast assimilable nitrogen (YAN) and volatile compounds at 10 sampling times during fermentation.

# Mathematical methods

The model consisted of 26 ordinary differential equations depending on 36 parameters ( $\theta$ ). The dry weight of biomass, carbohydrate and protein, and all metabolites are collected in a vector of state variables  $x(t) \in X \subset \mathbb{R}^{n_X}$ , which is the unique solution of the set of ordinary nonlinear differential equations:

Moreover, given an experiment with  $n_o$  observables and  $n_s^o$  sampling times per observable  $o, \mathbf{y}^o \in Y \subset \mathbb{R}^{n_s^o}$  regards the vector of  $n_s^o$  discrete time measurements as follows:

$$\mathbf{y}^{o}(t_{s}^{o}, \mathbf{u}, \theta) = \mathbf{g}^{o}(\mathbf{x}(\mathbf{u}, \theta, t_{s}^{o}), \theta, t);$$
  $s = 1, \dots, n_{s}^{o}$ 

where  $t_s^o$  refers to the sth sampling time for the observable o. Thus every experimental (measured) data point is denoted as yms and, similarly, the corresponding model predictions are denoted as  $y_s^o$ . Note that in our case, all states included in the model except the carbohydrate and protein content in the biomass, were observed during the experiments.

To identify the best model we used an iterative identification procedure (Vilas et al., 2018) which consisted of the following steps: (1) parameter estimation, to compute those parameter values that minimize the distance between model predictions and data; (2) practical identifiability analysis through the Fisher information matrix to compute confidence intervals for the parameters and to detect correlated parameters; (3) model reduction, to improve practical identifiability while ensuring goodness of fit. The procedure is implemented in the AMIGO2 toolbox (Balsa-Canto et al., 2016) devoted to the identification of biological systems models. The input files for simulation and parameter estimation are publicly available on the AMIGO2 website: https://sites.google.com/site/amigo 2toolbox/examples.

Parameter estimation was formulated as a nonlinear optimization problem to find the unknown model parameters that minimize the distance between the predicted values of the model and the available data. The AMIGO2 toolbox (Balsa-Canto et al., 2016) was used to solve the parameter estimation problem. In particular, we selected the least squares approach. The problem is formulated as follows:

Find  $\theta$  to minimize:

$$J(\theta) = \sum_{o=1}^{n_o} \sum_{s=1}^{n_o^s} [y_s^o(\theta) - y m_s^o]^2$$
 (3)

subject to the model equations and algebraic constraints defining the parameters feasible region  $\theta \in \Theta \subset \mathbb{R}^{n_{\theta}}$ .

Model equations were solved using CVODES (Hindmarsh et al., 2005) while the global optimizer enhances Scatter Search (eSS, Egea et al., 2009) was selected to solve the parameter estimation problem due to its well-recognized efficiency and robustness.

Practical identifiability was analysed using the Fisher information matrix and the correlation matrix. The Fisher Information Matrix (F) (Walter & Pronzato, 1997) reads as follows:

$$\mathcal{F} = \underbrace{\mathcal{E}}_{V_{m}\mid\theta^{*}} \left\{ \left[ \frac{\partial J(\theta)}{\partial \theta} \right] \left[ \frac{\partial J(\theta)}{\partial \theta} \right]^{T} \right\}$$
(4)

where E represents the expectation for the optimal value of the parameters  $\theta^*$ .

To compute the confidence intervals we used the Crammèr-Rao inequality, which establishes, under certain assumptions on the number of data and nonlinearity of the model, that the covariance matrix may be approximated by the inverse of the Fisher information matrix ( $C \leq \mathcal{F}$ ) (Walter & Pronzato, 1997). The confidence intervals on the parameters are defined as follows:

$$\pm t_{\alpha/2}^{\gamma} \sqrt{C_{ii}} \tag{5}$$

where  $t_{\alpha/2}^{\gamma}$  is given by Student's *t*-distribution,  $\gamma = N_{\rm d} - \eta$  degrees of freedom and  $(1-\alpha)$  100% is the confidence interval selected, typically 95%.

From the covariance matrix it is also possible to compute the correlation matrix:

$$Cr_{ij} = \frac{C_{ij}}{\sqrt{C_{ii}C_{jj}}} \quad i = 1, \dots, n_{\theta} \quad j = 1, \dots, n_{\theta}$$
 (6)

in such a way that two parameters  $(p_i, p_i)$  are completely uncorrelated if  $Cr_{ij} = 0$  or completely correlated if  $Cr_{ij} = \pm 1.$ 

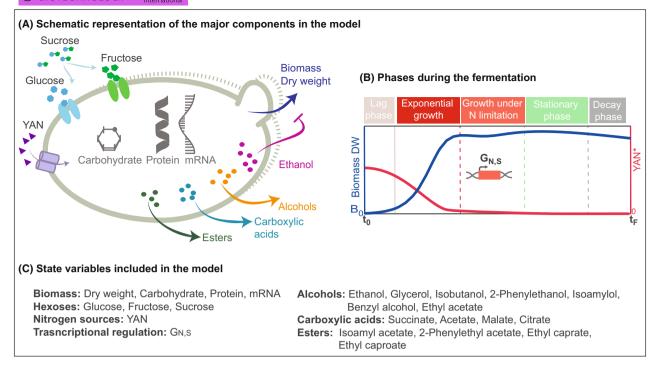
Model reduction was achieved by fixing or removing parameters according to their influence in the model predictions, that is, those with the lowest sensitivity value. In addition, we reduced the number of parameters when a high correlation was detected between pairs of parameters. The quality of fit of reduced models was compared to that of the original model to guarantee a satisfactory goodness-of-fit.

# RESULTS

# Modelling biomass formation

Figure 1 presents a schematic representation of the model, the phases that we have considered during modelling and all the compounds incorporated into the model. To model biomass growth, we considered that once inoculated into the grape must, cells need a temporary period to adjust to the new environment (the

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Model details. (A) shows a schematic representation of the main components included in the model; (B) shows the phases observed during fermentation: the lag phase, exponential growth, growth under nitrogen limitation, stationary phase and decay phase, which are included in the model; (C) presents all variables included in the model, including those related to biomass production and biomass composition, substrates, and products.

lag phase). Exponential growth occurs during the first hours after the lag phase, when assimilable nitrogen (YAN) is abundant. Under YAN depletion, yeasts begin to accumulate carbohydrates, producing a substantial increase in the dry weight biomass content.

The yeast growth rate was described using two different terms. The first term describes exponential growth, and this phase lasts until the assimilable nitrogen sources (YAN) are practically depleted. The second term corresponds to a secondary increase in the dry weight of the biomass as a result of the accumulation of carbohydrates, where (X) the dry weight of the biomass in (g/L), and its dynamics are as follows:

$$\frac{dX}{dt} = \log(t) \cdot \left(\mu_{N} + \mu_{C}\right) \tag{7}$$

The time-dependent function lag(t) represents the lag phase modelled here as proposed by Baranyi and Roberts (1994):

$$lag(t) = \frac{a_0}{a_0 + (1 - a_0) \cdot e^{-\mu_{\text{maxN}} \cdot t}}$$
 (8)

where  $a_0$  is a parameter bounded between 0 and 1 and represents the physiological state of the inoculum. If  $a_0$  is small ( $a_0 \ll 1$ ), then the lag phase is large. If  $a_0 = 1$ , then there is no lag phase.

Nitrogen sources were considered the limiting nutrient, and we used Monod kinetics to model growth associated with cell division:

$$\mu_{N} = X \cdot \mu_{\text{maxN}} \cdot \phi_{N} \tag{9}$$

where  $\mu_{\text{maxN}}$  (h<sup>-1</sup>) is the maximum specific growth rate and  $\phi_N$  is the Monod equation that describes the relationship between substrate availability and growth. Here,  $\phi_{N}$ is written as:

$$\phi_{N} = \frac{N}{N + k_{N}} \tag{10}$$

where N is the extracellular YAN in (g/L) and  $k_N(g/L)$  is the parameter of the Monod equation.

We also included a mechanism for carbohydrate accumulation ( $\mu_{\rm C}$ ). This secondary growth rate ( $\mu_{\rm C}$ ) is associated with nitrogen starvation. Therefore,  $\mu_{C}$  reads as follows:

$$\mu_{C} = \phi_{N,S} \cdot X \cdot \Phi_{C} \cdot \left(\Theta_{C} - \frac{X_{C}}{X}\right) \tag{11}$$

where  $\theta_{\rm C}$  represents a target value for the ratio of carbohydrates content in biomass  $\binom{X_{\rm C}}{X}$ , and its impact is controlled by the value of  $\Phi_{\rm C}$  (h^-1);  $\phi_{\rm N,S}$  is a term associated with the activation of secondary growth:

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$$\phi_{N,S} = 1 - \frac{N \cdot (1 + k_{sC})}{N + k_{sC}}$$
 (12)

 $k_{\rm sC}$  regulates the nitrogen concentration needed to induce carbohydrate accumulation and the transcriptional changes associated with nitrogen starvation.

We divided biomass into carbohydrates, proteins and mRNA. Their dynamics are described by the following ordinary differential equations:

$$\frac{dX_{C}}{dt} = \lambda_{C} \cdot \mu_{N} + \mu_{C} - k_{D} \cdot X_{C} \cdot \phi_{ED}$$
 (13)

$$\frac{dX_{P}}{dt} = \lambda_{P} \cdot \mu_{N} - k_{D} \cdot X_{P} \cdot \phi_{ED}$$
 (14)

$$\frac{dX_{\text{mRNA}}}{dt} = (1 - \lambda_{\text{C}} - \lambda_{\text{P}}) \cdot \mu_{\text{N}} - k_{\text{D}} \cdot X_{\text{mRNA}} \cdot \phi_{\text{ED}}$$
 (15)

where  $\lambda_{\rm C}$  and  $\lambda_{\rm P}$  refer to the percentages of carbohydrates and proteins in biomass respectively; and  $k_{\rm D} \cdot \phi_{\rm ED}$  represents the cellular decay due to the production of ethanol (Eth):

$$\phi_{\text{ED}} = \frac{\text{Eth}}{\text{Eth} + k_{\text{Ed.}}} \tag{16}$$

Finally, we model the consumption of the YAN assuming that it is proportional to  $\mu_N$ :

$$\frac{dN}{dt} = -\left(1 - \lambda_{C}\right) \cdot \frac{\mu_{N}}{Y_{x/N}} \tag{17}$$

where  $Y_{x/N}$  ( $g_{DW}/g_{N}$ ) is the yield of nitrogen to mRNA and protein biomass.

# Modelling fermentation rate

The hexoses needed for fermentation come from sucrose, glucose and fructose. Sucrose is broken down into glucose and fructose. The transport of hexoses can be expressed as:

$$v_{\text{Glx}} = -\log(t) \cdot X_{\text{P}} \cdot v_{\text{max,Glx}} \cdot \frac{\text{Glx}}{\text{Glx} + \text{ks}_{\text{Glx}}} \cdot \phi_{\text{Eth}}$$
 (18)

$$v_{F} = -\log(t) \cdot X_{P} \cdot v_{\text{max,F}} \cdot \frac{F}{F + ks_{F}} \cdot \phi_{\text{Eth}}$$
 (19)

where  $v_{\text{Glx}}$  and  $v_{\text{F}}$  are the glucose and fructose uptake rates respectively;  $v_{\text{max,Glx/F}}$  ( $\frac{g/L}{h \cdot g_{\text{DW}}/L}$ ) is the maximum rate of hexose transport; and  $ks_{\text{Glx/F}}$  (g/L) is the Michaelis–Menten constant. We assumed that hexose uptake is inhibited by the presence of ethanol (Hjersted et al., 2007):

$$\phi_{\mathsf{Eth}} = \frac{1}{1 + \frac{\mathsf{Eth}}{k_{\mathsf{E}_i}}} \tag{20}$$

where  $k_{E_i}$  (g/L) refers to the strength of the inhibitory effect.

Taking these factors into account, the glucose and fructose dynamics are as follows:

$$\frac{dGlx}{dt} = v_{Glx} + \frac{1}{2} \frac{dSucrose}{dt}$$
 (21)

$$\frac{dF}{dt} = v_F + \frac{1}{2} \frac{dSucrose}{dt}$$
 (22)

where sucrose follows the mass action kinetics. Equations (15) and (16) describe the glucose and fructose dynamics. These equations account for the effect of sucrose degradation  $\frac{\mathrm{dSucrose}}{\mathrm{d}t}$  and glucose/fructose transport into the cell  $v_{\mathrm{Glx/F}}$ . Ethanol inhibition  $(\phi_{\mathrm{Eth}})$  occurs in hexose transport, slowing the dynamics in the presence of high ethanol concentrations.

We assumed that the fermentation rate is proportional to the total protein content ( $X_p$ ), in agreement with (Henriques & Balsa-Canto, 2021). We also assume that oxygen is not present during fermentation. The oxygen measured was lower than 0.088 g/L for all four species.

# Modelling the production of extracellular metabolites

During alcoholic fermentation, glucose and fructose are metabolized through central carbon metabolism into various metabolites such as ethanol, acetate, glycerol or succinate. The excretion of these metabolites was assumed to be proportional to the consumption of hexoses. For example, ethanol production is as follows:

$$\frac{\text{dEth}}{\text{d}t} = -Y_{\text{Eth}} \cdot (v_{\text{Glx}} + v_{\text{F}}) \tag{23}$$

where  $Y_{Eth}(g_{Eth}/g_{hexose})$  is the yield of ethanol production.

We previously observed (Henriques et al., 2021) that the production of higher alcohols began earlier than that of ethanol or glycerol. In addition, we observed that neither glycerol nor higher alcohols were produced until the end of fermentation. To accommodate these observations, we assumed that the production of a set of metabolites, such as higher alcohols (HA) and acetate esters (EST), was produced by low nitrogen as follows:

$$\frac{dHA}{dt} = -Y_{HA} \cdot (v_{GIX} + v_{F}) \cdot (1 - G_{N,S})$$
 (24)

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$$\frac{\text{dEST}}{\text{d}t} = -Y_{\text{EST}} \cdot (v_{\text{Glx}} + v_{\text{F}}) \cdot (1 - G_{\text{N,S}})$$
 (25)

where  $Y_{HA}$  ( $g_{HA}$  /  $g_{hexose}$ ) and  $Y_{EST}$  ( $g_{EST}$  /  $g_{hexose}$ ) correspond to yields of a specific higher alcohol or ester respectively and G<sub>NS</sub> represents the set of transcriptional changes induced by nitrogen starvation responsible for inhibiting the production of higher alcohols and glycerol. Specifically, we considered isoamyl alcohol, isobutanol, 2-phenyl ethanol (PEA) and benzyl alcohol belonging to the higher alcohol taxonomy and ethyl acetate, isobutyl acetate, isoamyl acetate, 2-phenylethyl acetate, ethyl caprate and ethyl caproate in the case of esters. G<sub>N.S</sub> intends to characterize the experimental observations by Rossignol et al. (2003). In their work, the authors observed that during fermentation of a synthetic medium that mimicked a natural must, growth arrest was caused by nitrogen exhaustion, and subsequent entry into the stationary phase triggered significant transcriptional reprogramming. To represent this transition using a limited number of parameters, we describe  $G_{N,S}$  with the following ordinary differential equation:

$$\frac{\mathsf{d}G_{\mathsf{N},\mathsf{S}}}{\mathsf{d}t} = \left(\phi_{\mathsf{N},\mathsf{S}} - G_{\mathsf{N},\mathsf{S}}\right) \cdot \tau_{G_{\mathsf{N},\mathsf{S}}} \tag{26}$$

The production of succinate and acetate differs in fermentations led by *S. cerevisiae* and *S. uvarum*. Noticeably, none of these compounds could be described using the formulas described above for ethanol, glycerol, higher alcohols or acetate esters. In the case of succinate, its production was well described by a later start of production associated with nitrogen starvation and was not inhibited by the transcriptional changes associated with nitrogen starvation:

$$\frac{\text{dSuccinate}}{\text{d}t} = -Y_{\text{Succ}} \cdot \phi_{\text{N,S}} \cdot (v_{\text{Glx}} + v_{\text{F}}) \quad (27)$$

where  $Y_{\rm Succ}$  ( $g_{\rm succinate}/g_{\rm hexose}$ ) is the yield of succinate produced from hexose.

Regarding acetate, as noted in Henriques and Balsa-Canto (2021), *S. uvarum* produces acetate during growth and consumes it during the stationary phase. Therefore, we divided acetate metabolism into two separate fractions:

$$\frac{\text{dAcetate}}{\text{d}t} = v_{\text{ACE,P}} + v_{\text{ACE,C}} \tag{28}$$

where  $v_{\rm ACE,P}$  describes acetate production and  $v_{\rm ACE,C}$  its consumption. In our model, acetate production was inhibited by nitrogen starvation and represented by the following equation:

$$v_{\text{ACE,P}} = -Y_{\text{ACE,P}} \cdot (1 - \phi_{\text{N,S}}) \cdot (v_{\text{Glx}} + v_{\text{F}})$$
 (29)

where  $Y_{ACE,P}(g_{acetate}/g_{hexose})$  is the yield of acetate produced from hexose. In contrast, acetate consumption is activated by nitrogen starvation and is described by mass action:

$$v_{\text{ACE,C}} = -k_{\text{ACE,C}} \cdot X_{\text{P}} \cdot \phi_{\text{N,S}} \cdot \text{Acetate} \cdot (1 - G_{\text{N,S}})$$
 (30)

where  $k_{\text{ACE,C}}$  (h<sup>-1</sup>. g<sub>protein</sub><sup>-1</sup>) is the mass action parameter. To account for the fact that acetate was not fully consumed in the analysed experimental data, its consumption was inhibited by  $G_{\text{N.S.}}$ 

# Modelling carboxylic acid consumption

As shown in Figure S1, citrate and malate were consumed during growth. Here, we describe the time course trajectories of these carboxylic acids (CA<sub>C</sub>) with mass action:

$$\frac{dCA_{C}}{dt} = -k_{CA_{C}} \cdot (1 - \phi_{N,S}) \cdot CA_{C}$$
 (31)

where  $k_{\rm CA_C}$  (h<sup>-1</sup>) is the mass action parameter and  $(1-\phi_{\rm N,S})$  enforces that consumption is only possible before nitrogen starvation.

# Model calibration, identifiability analysis and model reduction

The original model consisted of 26 ordinary differential equations depending on 36 parameters. We calibrated the model to time series data of biomass, substrates and products obtained for fermentations led by *S. cerevisiae* T73 (ScT73), *S. kudriavzevii* CR85 (SkCR85), *S. uvarum* BMV58 (SuBMV58) and CECT12600 (SuCECT12600).

Next, we computed the Fisher information matrix and confidence intervals for the parameters using the Crammér–Rao inequality. We found that although the proposed model produced a very good adjustment to the experimental data, the confidence intervals (Table S1) were significantly wide because of practical identifiability issues (see the Supporting Information for further details). Using the information from the confidence intervals, we decided to reduce the model in such a way that the less influencing parameters were removed. The final structure of the model remained the same for all species.

Regarding the parameters associated with biomass production, we found that  $k_{\rm s,C}$  generally tended to be very small (often below  $1\times 10^{-3}$ ) and was associated with high uncertainty in all strains. Consequently, we set this parameter to  $1\times 10^{-4}$  for all strains. From a physiological point of view, fixing this parameter at such a low value ensures that the effects associated with  $k_{\rm s,C}$  require nitrogen depletion. Furthermore, we

fixed parameters corresponding to biomass composition using the values reported by (Schulze et al., 1996) for nitrogen-limited batch fermentation (59% protein and 12% mRNA). Additionally, the parameter that accounts for cell decay due to ethanol ( $K_{\rm Ed,}$ ) was highly correlated with the cell decay rate ( $K_{\rm D}$ ) (Figures S3–S6 and Table S2). Since both parameters showed large confidence intervals and were highly correlated, we decided to remove  $\phi_{\rm ED}$  from the model. Finally, with the exception of *S. kudriavzevii*, we found that the model could explain the data without the need to introduce a lag phase. Consequently, in the cases of *S. cerevisiae* and *S. uvarum* strains, we set  $a_0$  to 1, which in practice means that there is no lag phase for these strains.

On the basis of the identifiability analysis, we were also able to eliminate terms associated with fermentation rate. In the initial model, the maximum specific uptake rate parameters ( $v_{\text{max,Glx}}$  and  $v_{\text{max,F}}$ ) and the substrate affinity parameters ( $k_{\text{s,Glx}}$  and  $k_{\text{s,F}}$ ) were highly correlated for glucose and fructose (Figures S3–S6). To address this issue, we assumed a unique maximum uptake rate for both transport rates. Similarly, the parameter that accounts for the inhibition of transport due to ethanol ( $k_{\text{E}_i}$ ) and the hexose affinity parameters ( $k_{\text{s,Glx}}$  and  $k_{\text{s,F}}$ ) were highly correlated in all strains, and we were able to eliminate  $\phi_{\text{Eth}}$  without substantial impacts on the quality of the model.

The final model depends on 33 parameters to be estimated from the experimental data. Their optimal values and the corresponding confidence intervals are reported in Tables 1 and 2. Figures 2 and 3 present

the predictions of the model for biomass and variables associated with growth and fermentation rates. Despite the extensive reduction of the model, it can still satisfactorily explain the dry weight, sugar uptake and YAN trajectories. The median  $R^2$  was 0.95 for all observables and all strains. It should be noted that the model captured well the dynamics of glycerol, acetate and higher alcohols, such as 2-phenylethanol and isoamyl alcohol, all of which are relevant to the quality of the wine.

# Differences in primary and secondary metabolism

As expected, *S. cerevisae* T73 was the highest producer of ethanol in terms of both final concentration and estimated yield ( $Y_{\rm eth} = 0.47 \cdot g_{\rm eth} \cdot g_{\rm glucose}^{-1}$ ). In contrast, as shown in Figure 2, ScT73 presented the lowest final glycerol concentration. Interestingly, for T73, the estimated yield for glycerol production ( $Y_{\rm Gly} = 0.047 \cdot g_{\rm glycerol} \cdot g_{\rm glucose}^{-1}$ ) was higher than those of SuBMV58 (0.043) and SuCECT12600 (0.043). This result is explained by the dynamics of  $G_{\rm N,S}$  and modulated by the parameter  $\tau_{\rm G_{\rm N,S}}$ , which was estimated to be between 2.3 and 7.6 times higher in ScT73 than in the other strains. Therefore, to a large extent, differences in glycerol production were influenced by how long after nitrogen depletion the cells continued to produce glycerol. Furthermore, the same pattern of early inhibition of higher alcohols by  $G_{\rm N,S}$  was also able to explain the formation of aromas such as 2-phenylethanol and

TABLE 1 Biomass and nutrients uptake-related parameters: optimal values and associated confidence intervals.

	ScT73		SuBMV58		SuCECT12600		SkCR85	
Par	Value	$\theta(\%)$	Value	θ(%)	Value	$\theta(\%)$	Value	θ(%)
$\Phi_{C}$	$2.82 \times 10^{-1}$	512	$1.61 \times 10^{-1}$	136	$1.61 \times 10^{-2}$	78.4	1.65	783
$Y_{x/N}$	13.70	12	5.47	30.3	9.01	11.49	9.53	17.6
$\mu_{maxN}$	$5.00 \times 10^{-1}$	17.50	$2.43 \times 10^{-1}$	22.4	$4.35 \times 10^{-1}$	21.9	$1.02 \times 10^{-1}$	8.1
$\Theta_{C}$	$4.05 \times 10^{-1}$	23	$7.90 \times 10^{-1}$	31.6	1	55.7	$4.69 \times 10^{-1}$	22.9
k <sub>N</sub>	$2.90 \times 10^{-1}$	28.90	$9.22 \times 10^{-2}$	54.5	$2.74 \times 10^{-1}$	39.5	$4.24 \times 10^{-2}$	43.7
kd	$5.07 \times 10^{-3}$	19.1	$8.01 \times 10^{-3}$	123	$9.20 \times 10^{-3}$	40.4	$6.37 \times 10^{-3}$	21.2
v <sub>max,Glx</sub>	1.20	14.60	3.85	59.8	2.08	28.5	1.88	28.9
k <sub>s,Glx</sub>	37.85	21.1	51.07	128	55.47	47.8	46.95	45.2
$k_{s,F}$	$1.40 \times 10^{2}$	16.9	$1.69 \times 10^{2}$	92.3	$1.95 \times 10^{2}$	39.6	$2.00 \times 10^{2}$	35.2
k <sub>sucrose</sub>	$3.38 \times 10^{-2}$	0.9	$3.06 \times 10^{-2}$	0.802	$3.61 \times 10^{-2}$	0.273	$2.14 \times 10^{-2}$	1.23
$ au_{G_{N,S}}$	$1.12 \times 10^{-1}$	19.1	$1.48 \times 10^{-2}$	52	$2.45 \times 10^{-2}$	36.3	$4.80 \times 10^{-2}$	26.1
<b>a</b> <sub>0</sub>	1	-	1	-	1	-	$2.18 \times 10^{-2}$	63.2
$\lambda_{C}$	0.29	-	0.29	-	0.29	-	0.29	-
$\lambda_{P}$	0.59	-	0.59	-	0.59	-	0.59	-
k <sub>sC</sub>	$1.00 \times 10^{-4}$	_	$1.00 \times 10^{-4}$	_	$1.00 \times 10^{-4}$	-	$1.00 \times 10^{-4}$	_

Note: Confidence intervals marked with the symbol – correspond to the parameters whose values were fixed during the optimization. The parameter descriptions can be found in Section 3.

Parameters related to the production and consumption of products: Optimal values and associated confidence intervals.

	ScT73		SuBMV58		SuCECT12600		SkCR85	
Par	Value	θ(%)	Value	θ(%)	Value	θ(%)	Value	θ(%)
Y <sub>Eth</sub>	$4.71 \times 10^{-1}$	1.16	$4.32 \times 10^{-1}$	4.88	$4.57 \times 10^{-1}$	2.82	$4.67 \times 10^{-1}$	2.9
Y <sub>Glycerol</sub>	$4.71 \times 10^{-2}$	4.13	$4.28 \times 10^{-2}$	7.16	$4.29 \times 10^{-2}$	4.2	$5.78 \times 10^{-2}$	5.28
Y <sub>Ace</sub>	$2.63 \times 10^{-3}$	14.4	$4.13 \times 10^{-3}$	37.1	$2.28 \times 10^{-3}$	49.8	$1.39 \times 10^{-2}$	7.83
kc <sub>Ace</sub>	$1.04 \times 10^{-2}$	44.8	$3.53 \times 10^{-2}$	51.2	$6.56 \times 10^{-2}$	24.6	$1.50 \times 10^{-2}$	37.7
kc <sub>Citrate</sub>	$1.09 \times 10^{-2}$	15.6	$4.83 \times 10^{-2}$	38.6	$1.52 \times 10^{-2}$	20	$1.41 \times 10^{-2}$	21
kc <sub>Malate</sub>	$1.88 \times 10^{-3}$	16.5	$1.72 \times 10^{-2}$	39.1	$4.67 \times 10^{-3}$	19.1	$7.44 \times 10^{-3}$	20.7
Y <sub>Succinate</sub>	$3.49 \times 10^{-3}$	7.39	$3.48 \times 10^{-2}$	31.9	$1.29 \times 10^{-2}$	19.8	$7.45 \times 10^{-3}$	9.99
Y <sub>Lactate</sub>	$1.24 \times 10^{-3}$	4.01	$1.13 \times 10^{-3}$	11	$1.35 \times 10^{-3}$	5.19	$9.37 \times 10^{-4}$	5.44
Y <sub>EthylAcetate</sub>	$1.07 \times 10^{-3}$	10.4	$9.11 \times 10^{-4}$	8.15	$9.53 \times 10^{-4}$	3.95	$9.13 \times 10^{-4}$	9.81
Y <sub>IsobutylAcetate</sub>	$4.36 \times 10^{-6}$	17.9	$2.75 \times 10^{-6}$	5.43	$2.68 \times 10^{-6}$	3.02	$1.62 \times 10^{-6}$	8.93
Y <sub>Isobutanol</sub>	$5.22 \times 10^{-4}$	10.8	$3.89 \times 10^{-4}$	9.43	$3.44 \times 10^{-4}$	3.26	$3.61 \times 10^{-4}$	9.46
Y <sub>IsoamylAcetate</sub>	$1.80 \times 10^{-4}$	9.68	$9.05 \times 10^{-5}$	10.2	$9.54 \times 10^{-5}$	3.29	$2.50 \times 10^{-5}$	9.34
Y <sub>Isoamylalcohol</sub>	$4.57 \times 10^{-3}$	8.73	$2.59 \times 10^{-3}$	8.12	$2.01 \times 10^{-3}$	4.42	$2.12 \times 10^{-3}$	10.1
Y <sub>PhenylethylAc</sub>	$3.38 \times 10^{-5}$	8.3	$9.90 \times 10^{-5}$	10.6	$6.67 \times 10^{-5}$	3.53	$1.63 \times 10^{-5}$	6.91
Y <sub>Phenylethanol</sub>	$1.37 \times 10^{-3}$	7.87	$4.65 \times 10^{-3}$	7.66	$2.64 \times 10^{-3}$	4.88	$9.78 \times 10^{-4}$	12.7
Y <sub>EthylCaproate</sub>	$3.06 \times 10^{-5}$	6.62	$1.08 \times 10^{-5}$	6.24	$1.06 \times 10^{-5}$	3.95	$1.94 \times 10^{-5}$	8.98
Y <sub>EthylCaprate</sub>	$3.82 \times 10^{-5}$	18.6	$7.14 \times 10^{-5}$	6.69	$5.58 \times 10^{-5}$	9.56	$4.58 \times 10^{-5}$	4.48
Y <sub>Benzyl</sub> Alcohol	$8.42 \times 10^{-5}$	25.6	-	-	$1.01 \times 10^{-4}$	6.86	$1.93 \times 10^{-4}$	4.89

Note: Parameters marked with  $Y(g/g_{\text{hexose}})$  refer to the yield production of a specific higher alcohol or ester. Parameters marked with  $k(h^{-1}, g_{\text{nonterin}}^{-1})$  are related to the mass action kinetics. Parameter values marked with the symbol - correspond to those cases in which available data were scarce or too noisy to estimate the corresponding model parameters.

isoamyl alcohol, highlighted by the good R<sup>2</sup> square values obtained (between 0.86 and 0.98).

In the proposed model, succinate production was initiated at the beginning of nitrogen depletion. As shown in Figure 3, the onset of succinate flux is well captured by the model ( $R^2$  between 0.82 and 0.94). Interestingly, the model could not accurately explain the data in the case of ScT73 ( $R^2 = 0.69$ ). Given that, compared to other species, ScT73 shows much lower succinate production, possibly because its production has different causal explanations in fermentations of S. cerevisiae and the cold-tolerant strains analysed here.

The model was also able to capture the dynamics of acetate production and consumption. SkCR85 produced and consumed the largest amounts of acetate. In contrast, ScT73 produced a similar amount of acetate compared to SuBMV58 or SuCECT12600 but consumed only a very small fraction at the beginning of nitrogen depletion, which is in line with the small estimate for  $k_{ACE,C}$  (1.04x10<sup>-2</sup> · h<sup>-1</sup> · g<sub>DW</sub><sup>-1</sup>) and the previously mentioned rapid onset of  $G_{N,S}$ .

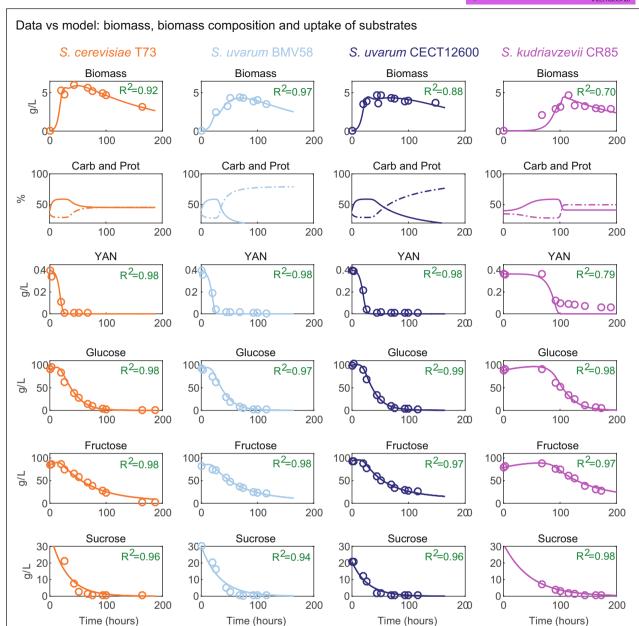
Regarding the fit to the data for higher alcohols and esters, the median  $R^2$  values over all measured

compounds were 0.93 for ScT73, 0.92 for SuBMW58, 0.95 for SuCECT12600 and 0.91 for SkCR85. The fits to the data can be seen in Figure 2 and Figures S1 and S2. The yields are well identified from the data with confidence intervals ≤ 25% for most compounds (Table 2). It should be noted that their values are generally quite low, with orders of magnitude ranging from  $10^{-6}$  and  $10^{-3}$ . When comparing the performance of strains in terms of these metabolic products, ScT73 produces the highest amounts of isobutanol, isoamyl alcohol, isobutyl acetate, isoamyl acetate, and ethyl caproate; SuBMV58 produces the highest amount of phenylethanol, phenylethyl acetate and ethyl caprate.

# Exploring the modulation of aroma production with nitrogen supplementation

We then used the model to predict the role of nitrogen supplementation in the production of relevant metabolites. We considered two different scenarios: first, in which we modified the initial concentration of YAN (the results are shown in Figure 4) and second,

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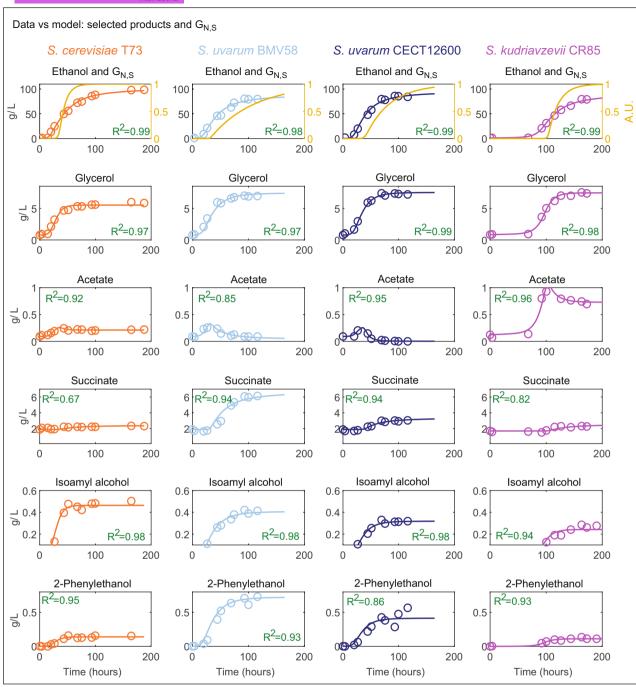


Fit of the model to growth-related time course data for batch fermentations led by four strains of three species of the Saccharomyces genus. Figures show the dynamics of biomass, YAN, glucose, fructose and sucrose. The figures also show the percentage of carbohydrates (dashed line) and proteins (solid line) in the biomass, as predicted by the model.

in which YAN was supplemented later during fermentation (the results are shown in Figure 5). In general, the supplementation of YAN accelerates the dynamics of growth, being this effect more evident for S. cerevisiae and S. uvarum strains. Early supplementation further enhances this effect (Figure S11). We estimated the uncertainty associated to model predictions using the uncertainty associated to the experimental error (Figures S7-S10).

Figure 4 shows a screening comparison between all species by the time at which fermentation ended (residual glucose is 2 g/L). As expected, the final

concentration of the metabolites was directly correlated with the initial nitrogen concentration, with the exception of the production of acetate, succinate, and carboxylic acids. In these cases, in which the metabolites are consumed at the beginning of nitrogen depletion, low nitrogen supplementation produced higher amounts of the metabolite at the end of the process. For example, as shown in Figure 4, ScT73 produced 0.221 ± 0.042 g/L of acetate when using 200 mg, while a final concentration of 0.190 ± 0.036 g/L was achieved for high supplementation (500 mg). This pattern is also followed in the production of citrate and malate for

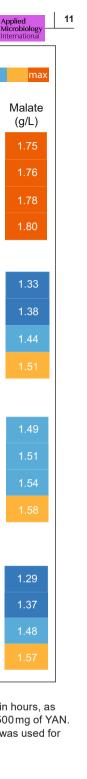


**FIGURE 3** Fit of the model to fermentation time course data for batch fermentations led by four strains of three species of the *Saccharomyces* genus. Figures show the dynamics of the production of ethanol, glycerol, acetate, succinate, isoamyl alcohol and 2-phenylethanol.

SuBMV58 and SuCECT12600. In contrast, no significant differences were observed in acetate production in SkCR85 when changing the initial nitrogen conditions (0.711  $\pm$  0.075 g/L for 500 mg, - 0.709  $\pm$  0.075 g/L for 400 mg, -0.715  $\pm$  0.076 g/L for 300 mg and -0.723  $\pm$  0.077 g/L for 200 mg).

The model predicts that ScT73 will produce wines with more ethanol but with lower amounts of glycerol than the other strains under all conditions tested.

SuBMV58 will produce less ethanol than the wild strain SuCECT12600. SkCR85 would produce wine containing between  $9.95\pm0.94\%$  and  $9.98\pm0.94\%$  of alcohol and would be the most glycerol-producing strain  $(9.072\pm0.420~g/L$  except for nitrogen-limited media). Interestingly, the available YAN at the beginning of fermentation does not significantly affect the final amount of ethanol but greatly improves the efficiency of the process. The results revealed that we can obtain similar



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**FIGURE 4** Predictions of the model for various initial nitrogen conditions. The figure shows the duration of fermentation in hours, as well as the final production of the relevant compounds. Four different initial values were considered that ranged from 200 to 500 mg of YAN. Dark blue, corresponds to the minimum production, while orange, corresponds to the maximum production. The same scale was used for all strains to facilitate the comparison.

amounts of ethanol but reduce the duration of fermentation between a minimum of 10% and a maximum of 47% by increasing the initial amount of YAN by 100 mg.

Simulations show similar behaviour in isobutanol production for all species and under all tested conditions. In contrast, there are differences in the production of 2-phenylethanol and isoamyl alcohol. SuBMV58 produced the maximum amount of 2-phenylethanol (0.76  $\pm$  0.15 g/L for 500 mg of YAN) followed by SuCECT12600 with a maximum of (0.44  $\pm$  0.07 g/L for 500 mg of YAN). ScT73 and SkCR85 produced similar

amounts of 2-phenylethanol, with ScT73 being slightly more efficient, while ScT73 produced more isoamyl alcohol than the other species under all tested conditions (ScT73:  $0.55 \pm 0.15$  g/L; SuBMV58  $0.43 \pm 0.10$  g/L; SuCECT1600:  $0.34 \pm 0.04$  g/L; SkCR85:  $0.30 \pm 0.09$  g/L for 500 mg of YAN).

We have also considered the scenario in which the initial amount of YAN is the one found in the natural must (approximately 300 mg) and we supplemented the medium with 200 mg later in the process. We compared the additions in the exponential phase and during the

FIGURE 5 Predictions of the model for various nitrogen supplementations. The figure shows the duration of fermentation in hours, as well as the final production of the relevant compounds. The figure shows the time in hours at which the YAN was added: during the exponential growth phase (15h for ScT73, SuBMV58 and SuCECT12600, and 76h for SkCR85) and during the transition to YAN exhaustion (24h for ScT73, SuBMV58 and SuCECT12600, and 92h for SkCR85). The initial YAN corresponded to 0.307 g/L. Dark blue corresponds to the minimum production, while orange corresponds to the maximum production compared between different species.

transition to YAN exhaustion. The results reveal that supplementation later in the process tends to prolong the duration of fermentation, up to  $10\,\%$  in the case of ScT73. The glycerol yield was also slightly larger (Figure S11). Higher differences were observed in the final amounts of succinate, around the  $4\,\%$  increase for the *S. uvarum* strain when supplemented in the exponential phase. In contrast, supplementation in the exponential phase decreased the final amount of acetate, up to  $50\,\%$  in the case of SuCECT12600. The final malate and citrate concentrations were also favoured by an early supplementation.

Comparing the results obtained with the results previously obtained for initial supplementation (Figure 4, 0.5 g/L), we observe that the final amount of ethanol and succinate tends to increase when supplementing at the beginning of fermentation.

# DISCUSSION

To the best of our knowledge, this is the first work to propose a kinetic model to predict secondary

metabolism of *Saccharomyces* yeast in wine. Here, we developed a model based on experimental data, previous modelling efforts, and evidence from the literature. More specifically, we combined a previous model of biomass formation (Henriques & Balsa-Canto, 2021) with a mechanism that simulates the impact of transcriptional reprogramming (Rossignol et al., 2003) on metabolic fluxes in nitrogen-starved cells, in particular on the synthesis of fermentation byproducts such as glycerol and aromas.

The introduction of this mechanism ( $G_{N,S}$ ) is based on the earlier finding that, unlike ethanol, there was no higher alcohol production during the decay phase (Henriques et al., 2021). Until recently, the most prevalent view was that higher alcohol production in fermented beverages was mainly associated with the catabolism of exogenous amino acids. However, this is in contradiction to previous results showing higher alcohol formation without nitrogen sources (Stevens, 1961). Recently, Crépin et al. (2017) added new information on this question, showing that a significant fraction of the higher alcohols and ester produced during alcoholic

fermentation come from the catabolism of amino acids synthesized de novo, that is, sugar metabolism.

Regarding biomass formation, we divided growth into two components ( $\mu_N$  and  $\mu_C$ ). The first is associated with the availability of the limiting substrate (YAN) and the second with the accumulation of carbohydrates when nitrogen becomes scarce. Although the formation of new protein and mRNA is strictly dependent on YAN (required for the synthesis of amino acids), storage carbohydrates, such as glycogen, can be transiently formed from extracellular hexoses in the absence of nitrogen (Schulze et al., 1996; Varela et al., 2004). Thus, during early growth  $(\mu_N \gg \mu_C)$ , biomass composition remains relatively stable, with mRNA and protein covering a considerable fraction. In contrast, when low nitrogen is available, growth is characterized by a reduced contribution of  $\mu_N$  and increased  $\mu_{C}$ . Although this strategy adds complexity to the model, it is important to consider the fact that biomass can increase after nitrogen depletion to accurately describe the experimental data. Although this model has been show to nicely capture biomass composition data in two independent data sets (Henriques & Balsa-Canto, 2021; Schulze et al., 1996; Varela et al., 2004), a possible contribution from the mobilization of arginine storages (Crépin et al., 2014), in some experimental conditions, cannot be ruled out.

We performed a parameter identifiability analysis and found that the model has wide confidence intervals, particularly in the parameters associated with biomass formation and fermentation rate. To address this problem, we simplified the model. The finding that the parameters associated with hexose transport (V<sub>max.H</sub> and  $k_{s,H}$ ) are highly correlated is relevant because it probably applies previous modelling efforts. Pizarro et al. (2007) used parameter values obtained for hexose transporters in vitro and information on the concentrations of multiple hexose transporters. Therefore, this alternative would only be applicable to S. cerevisiae, for which detailed kinetic studies have been performed on the transporters. Although poor identifiability could limit the predictive capabilities of inferred models (Chis et al., 2016; Lira-Parada et al., 2021), in the future, these issues could be addressed with a careful experimental design (Balsa-Canto et al., 2008), maximizing the information content of the parameters.

Important differences were found between the species analysed. For example, the dynamics of  $G_{N,S}$  were fast in S. cerevisiae and comparatively slower in S. kudriavzevii and especially in S. uvarum. These differences are interesting since, if confirmed, they could be exploited to tune the aromatic profiles of wines.

Recently, Godillot et al. (2022) investigated the aroma composition of wines with varying initial nitrogen (amino acids with diphosphate ammonia), added nitrogen and the timing of nitrogen addition. This recent dataset offers the possibility to validate or invalidate

some of the modelling assumptions described in this work. In this respect, consistent with the observations of Godillot et al. (2022) our model predicts that the addition of nitrogen results in an increase in the final amount of PEA. However, the same authors did not observe the effect on the production of isoamyl alcohol that we predicted with our model. Furthermore, the authors noted a negative linear effect on the initial nitrogen, which is in contrast to our model predictions. A possible explanation for this finding could be the fact that the model, despite the good fit in terms of the  $R^2$ score, is not perfectly synchronized with the isoamyl alcohol production data. In fact, for ScT73 and SkCR85, isoamyl alcohol production appears to begin early compared to the data, while the inverse pattern can be observed for S. uvarum strains (Figure 3). Taking into account the results of amino acid consumption obtained by Crépin et al. (2012), it seems likely that these patterns in higher alcohol behaviour are caused by the activation of genes under nitrogen catabolite repression by the SPS system (Ljungdahl, 2009), leading to the spillover of excess higher alcohols in the absence of exogenous amino acids.

# CONCLUSIONS

In this work we propose a continuous model that incorporates the physiological status of yeast to improve the description of the fermentation process and, in particular, of glycerol and higher alcohol formation. The strength of our model is that it is able to accurately capture the multiple phases of fermentation by introducing a simple representation of regulatory mechanisms. Including the role of preferential nitrogen sources (ammonia, glutamate and alanine) and relating them to the activation of genes leading to amino acid degradation genes should motivate future research on the detailed modelling of the dynamics of aromagenesis.

We expect that the model could also be applied to non-Saccharomyces yeasts by recalibrating the parameters involved with a similar data setting. This opens new opportunities for the design of novel single culture fermentation processes.

# **AUTHOR CONTRIBUTIONS**

Artai R. Moimenta: Formal analysis (equal); investigation (equal); visualization (equal); writing – original draft (equal). David Henriques: Formal analysis (equal); investigation (equal); supervision (equal); visualization (equal); writing – original draft (equal). Romain Minebois: Data curation (equal); investigation (equal); writing – review and editing (equal). Amparo Querol: Conceptualization (equal); data curation (equal); funding acquisition (equal); investigation (equal); writing – review and editing (equal). Eva Balsa-Canto:

Conceptualization (lead); formal analysis (equal); funding acquisition (equal); investigation (equal); methodology (lead); supervision (equal); visualization (equal); writing – original draft (equal); writing – review and editing (lead).

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# **CONFLICT OF INTEREST STATEMENT**

The authors declare no conflict of interest.

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#### SUPPORTING INFORMATION

Additional supporting information can be found online in the Supporting Information section at the end of this article.

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