Infrared radiation of ozone in the mesosphere and lower thermosphere: energetic effects and remote sensing

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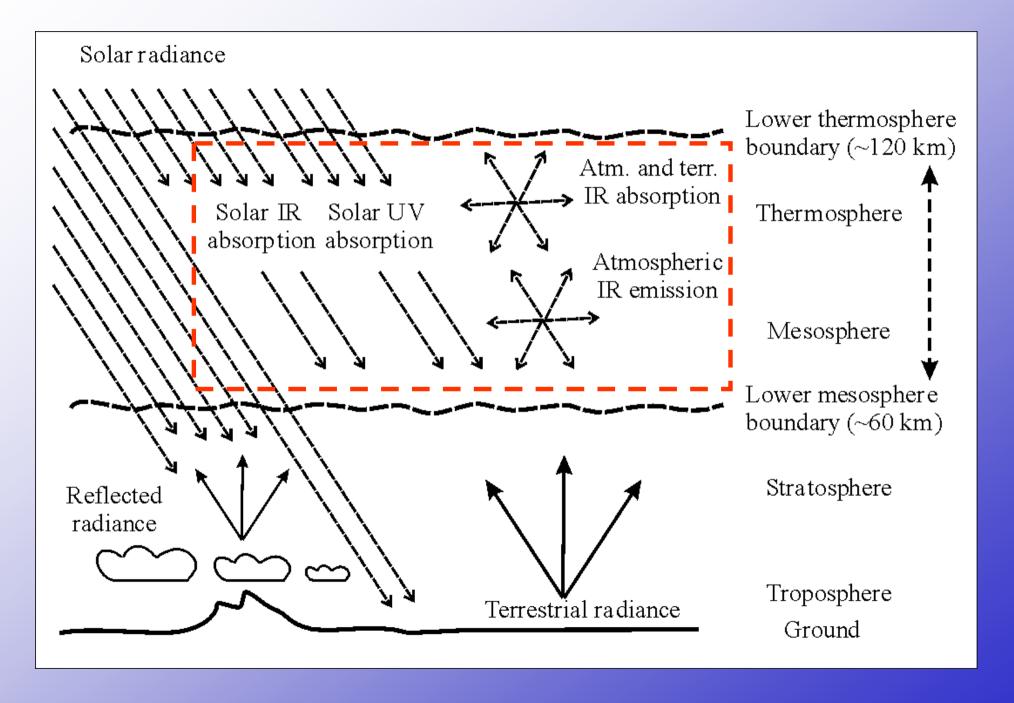
Ozone Workshop, October 2-4, 2013, Reims

MLT - why bother?

- MLT is a "gateway" between the lower atmosphere and space.
- MLT is sensitive to solar influence and to the inputs from below.
- Anthropogenic changes in greenhouse gases may change this input.
- Noctilucent clouds observed in polar MLT in the summer time are very sensitive to temperature changes and there are debates on their role as a "miner's canary" for global changes.
- MLT area absorption and emission in molecular bands affects the atmospheric observations of other areas.



Object of Study: IR radiance in the MLT



Cooling/heating rates

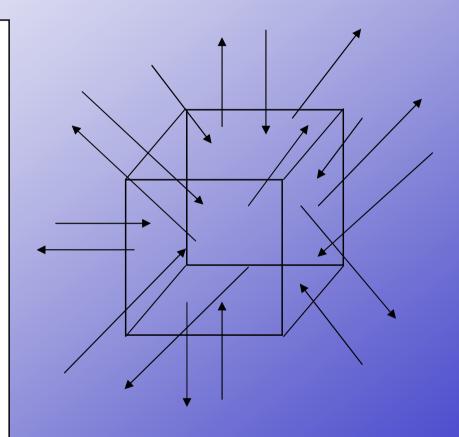
Radiative cooling/heating = radiative flux divergence

in W/m^3 :

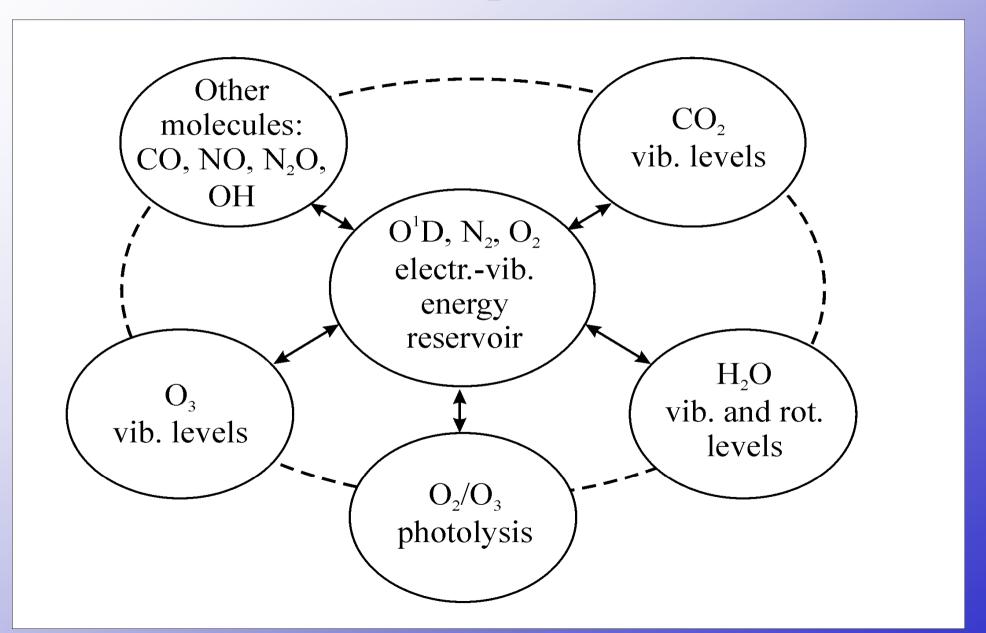
$$H(z) = \frac{1}{4\pi} \int_{\Omega}^{+\infty} \int_{-\infty}^{+\infty} \mu \frac{dI_{\mu\nu}(z)}{dz} dv$$

in K/day:

$$h(z) = H(z) \cdot \frac{24 \cdot 60 \cdot 60}{C_p(z) \rho(z)}$$



Energy Exchange Between Atmospheric Molecules



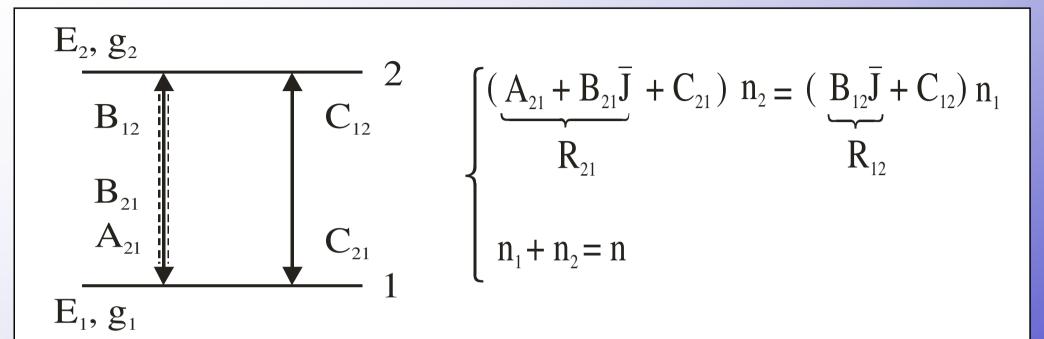
An important peculiarity of the MLT

- Infrared radiance absorption/emission corresponds to vibrational excitation/de-excitation of the molecules.
- To estimate the energetic characteristics of the given area or to interpret the infrared observations in MLT one needs to know the vibrational levels populations.

BUT !!!

- In the upper atmosphere, the collisions between the molecules are **not** frequent and the populations are **not** defined by local temperature.
- Breakdown of Local Thermodynamic Equilibrium (LTE).
- Special methodology is applied (non-LTE modeling).

LTE and non-LTE: two-level atom

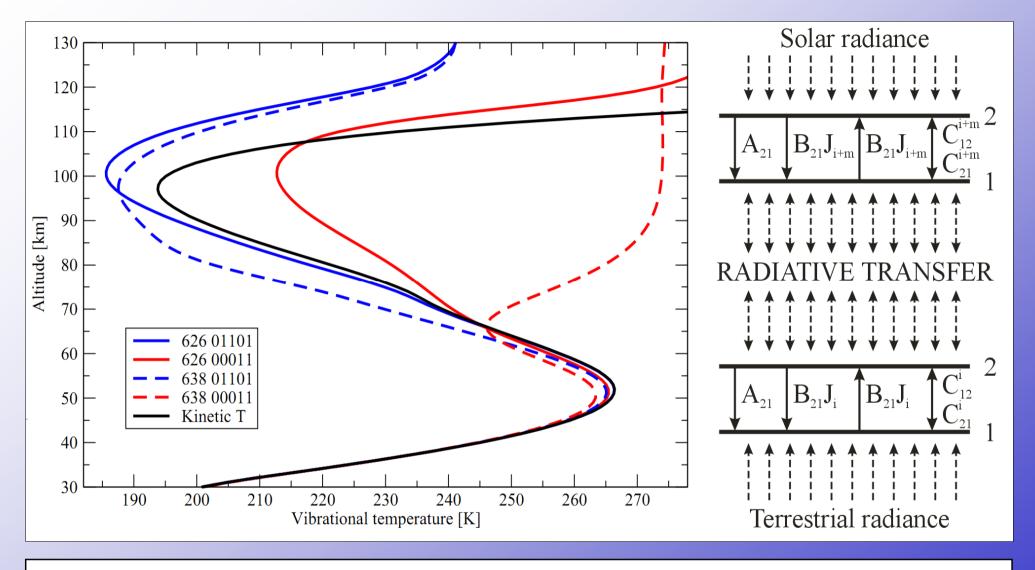


In the lower atmosphere:

In the upper atmosphere:

Collisions are less frequent, and the populations are not defined by local temperature anymore = non-LTE

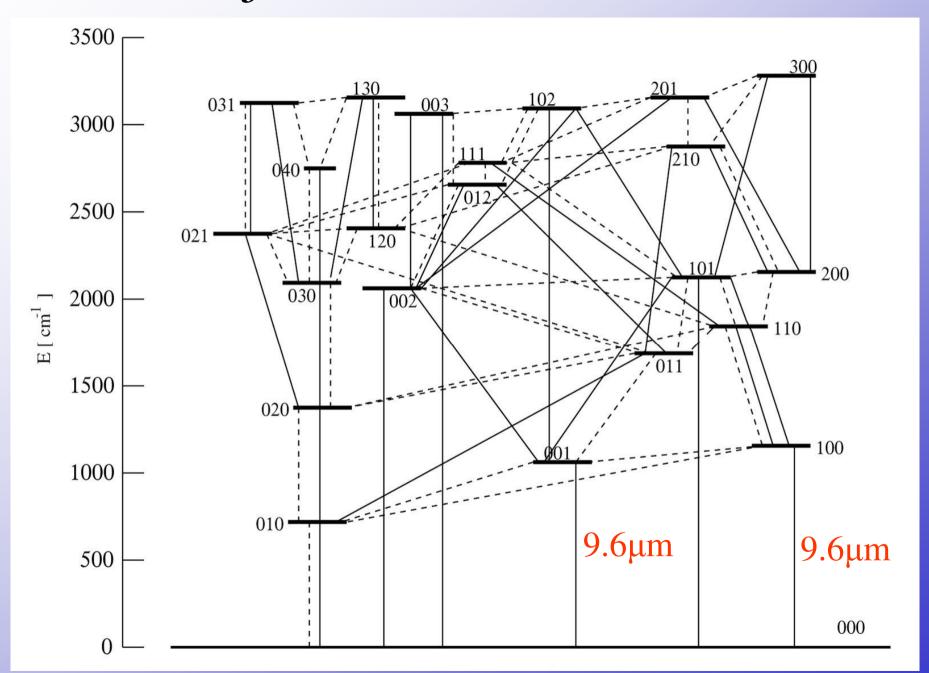
Non-LTE populations: vibrational temperatures



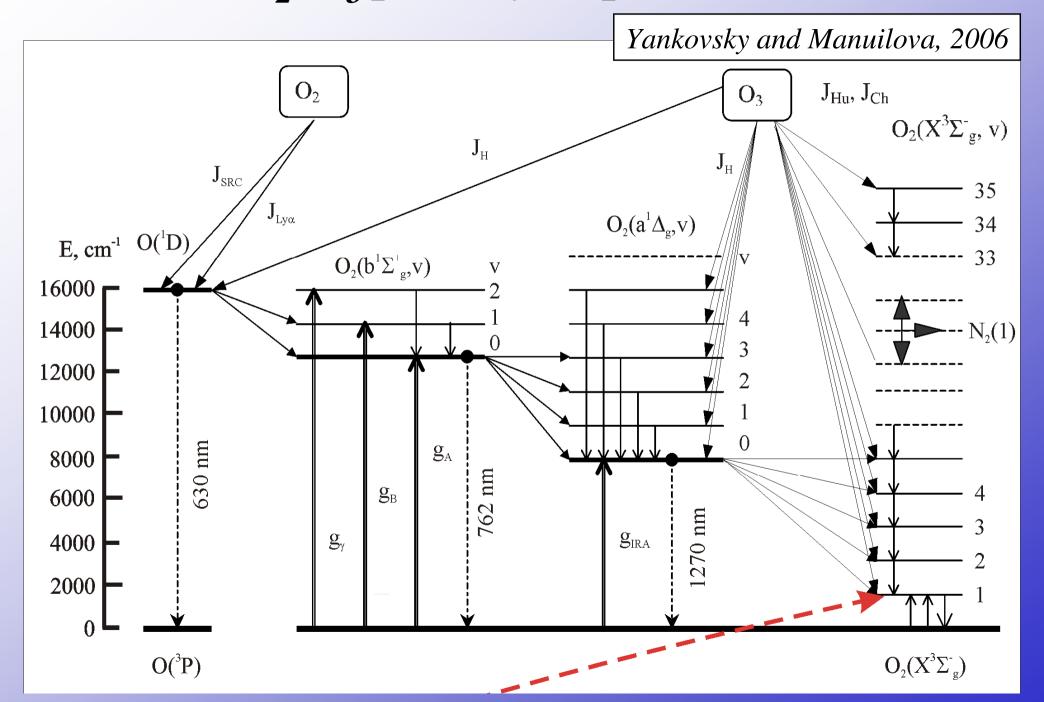
$$\frac{n_l}{n_0} = \frac{g_l}{g_0} \cdot e^{-(E_l - E_0)/kT_{vib}}$$

 T_{vib} = T_{kin} : LTE T_{vib} \neq T_{kin} : non-LTE

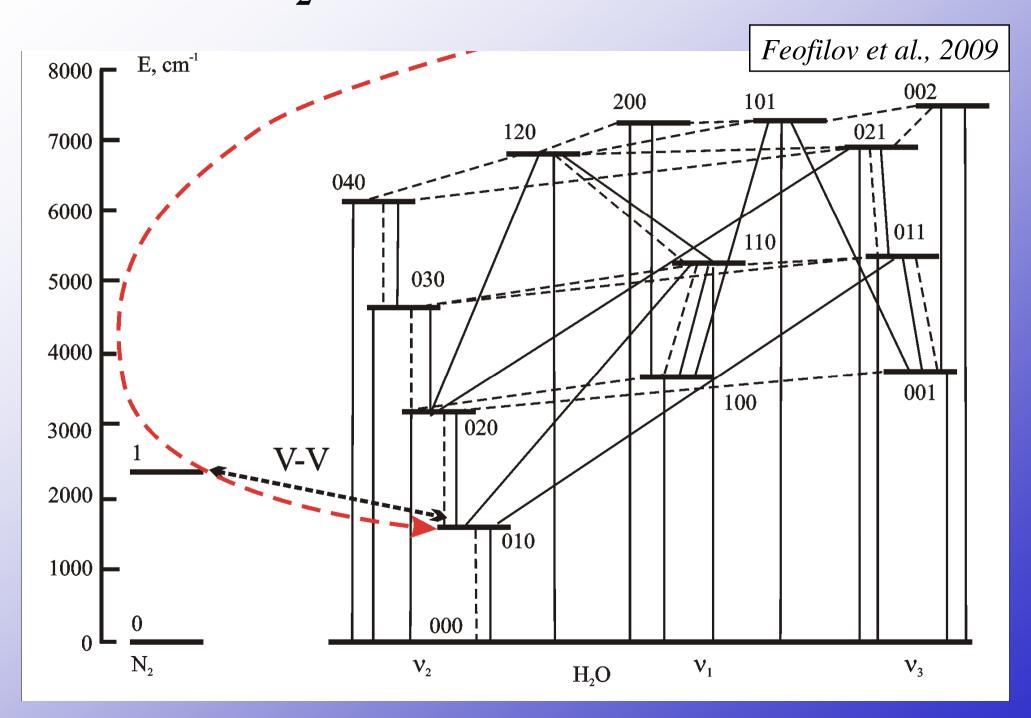
O₃ levels and transitions



O₂/O₃ photolysis products



H₂O levels and transitions



Energy exchange processes for O₃

$$O_{3}(v_{1}, v_{2}, v_{3}) + M(N_{2}, O_{2}, O) \stackrel{k_{int1}}{\longleftrightarrow} O_{3}(v_{1} - 1, v_{2}, v_{3} + 1) + M$$

$$O_{3}(v_{1}, v_{2}, v_{3}) + M(N_{2}, O_{2}, O) \stackrel{k_{int2}}{\longleftrightarrow} O_{3}(v_{1} - 1, v_{2} + 1, v_{3}) + M$$

$$O_{3}(v_{1}, v_{2}, v_{3}) + M(N_{2}, O_{2}, O) \stackrel{k_{int3}}{\longleftrightarrow} O_{3}(v_{1}, v_{2} + 1, v_{3} - 1) + M$$

Intra-molecular energy transfer

$$O_3(v_1, v_2, v_3) + M(N_2, O_2, O) \stackrel{k_{VT1}}{\leftrightarrow} O_3(v_1, v_2 - 1, v_3) + M$$

$$O_3(v_1, v_2, v_3) + M(N_2, O_2, O) \stackrel{k_{VT2}}{\leftrightarrow} O_3(v_1 - 1, v_2, v_3) + M$$

$$O_3(v_1, v_2, v_3) + M(N_2, O_2, O) \stackrel{k_{VT3}}{\leftrightarrow} O_3(v_1, v_2, v_3 - 1) + M$$

Vibrationaltranslational exchange

$$\begin{split} O_3(000) + N_2(\nu = 1) & \overset{k_{VV1}}{\leftrightarrow} O_3(200) + N_2(\nu = 0) + 130 \text{ cm}^{-1} \\ O_3(000) + O_2(\nu = 1) & \overset{k_{VV2}}{\leftrightarrow} O_3(100) + O_2(\nu = 0) + 456 \text{ cm}^{-1} \\ O_3(000) + O_2(\nu = 1) & \overset{k_{VV3}}{\leftrightarrow} O_3(001) + O_2(\nu = 0) + 517 \text{ cm}^{-1} \\ O_3(102) + O_2(\nu = 0) & \overset{k_{VV4}}{\leftrightarrow} O_3(000) + O_2(\nu = 2) - 4.9 \text{ cm}^{-1}. \end{split}$$

Vibrationalvibrational exchange

$$O_2 + O(^3P) + M \rightarrow O_3(v_1, v_2, v_3) + M,$$

Three-body recombination

Plus emission / absorption of radiation

Three-body recombination and nascent population model

A hybrid of models

Manuilova et al (1998) kinetics for lower states, by collisions with N_2 , O_2 , O^3P

Quasi-continuum
Is treated
in terms
of rate constants

A) Gil-Lopez et al., (2005): Zero surprisal model of nascent population, $E_{\rm D\,I}$ is the dissociation energy

$$P(v) = (1 - E(v)/E_D)^{1.5} / \sum_{v} (1 - E(v)/E_D)^{1.5}$$

- B) Kaufmann et al. (2006): Zero surprisal model or all excitation goes on 00v3, with v3=3,5 or 8
- C) Fernandez et al. (2009, 2010): single level nascent population at O_3 (006), assuming that about 70% of recombination energy goes to vibrations

An inconsistency between O_3 retrieved from 9.6 um and simulated/measured 4.8 um radiance

Kaufmann et al, 2006:

The rate of stretching to bending mode transition 001,100-> 010 K_{D2} must be 3-4 lower than well known measured value

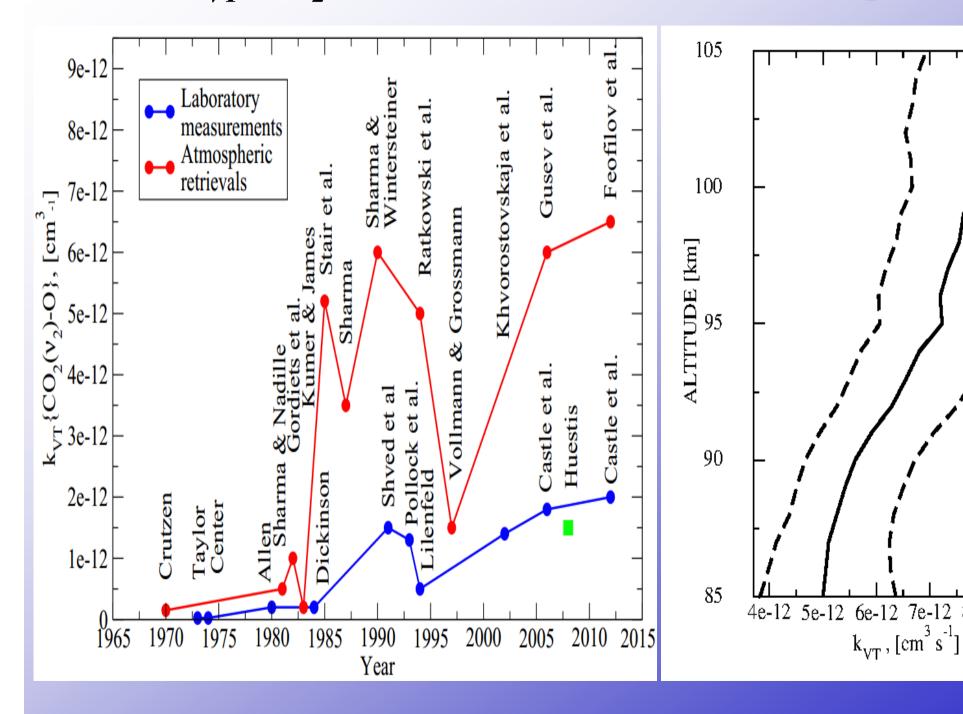
4.8 microns emission from 200, 101, 002 for known O3 was 2-3 times lower than measured for ~ 75 km

Fernandez et al., 2010:

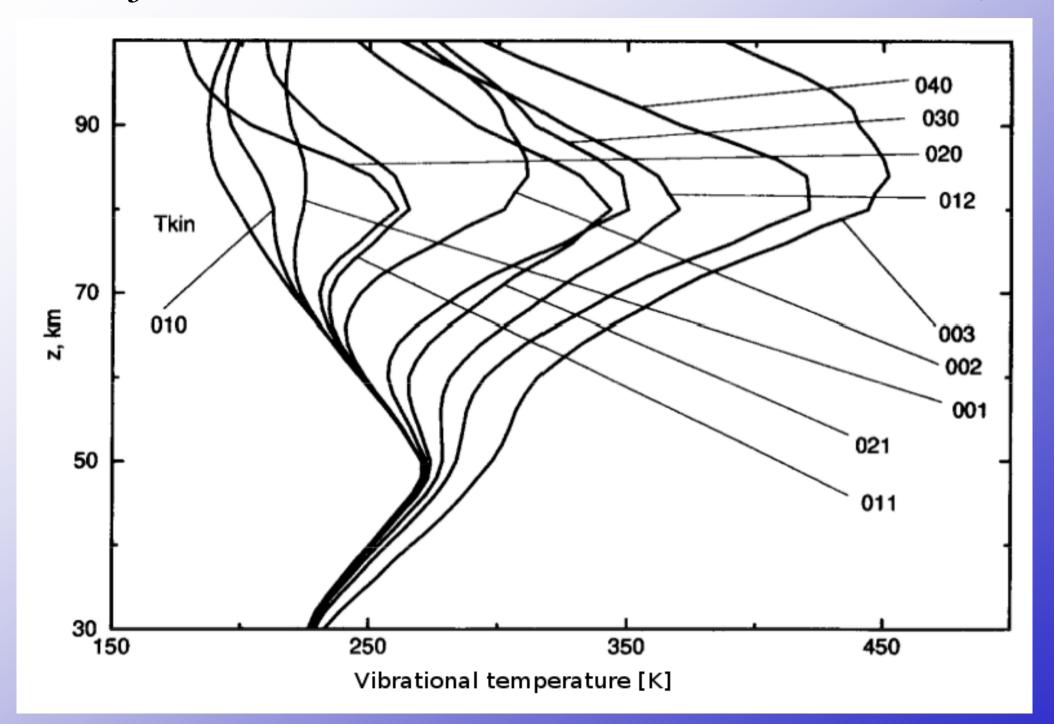
new bending to stretching mode transitions, for lower levels 020->100,001 SSH estimated K_{2D} For higher levels – harmonic oscillator rule

$k_{VT}\{CO_2-O\}$ measurements: lab vs atmosphere

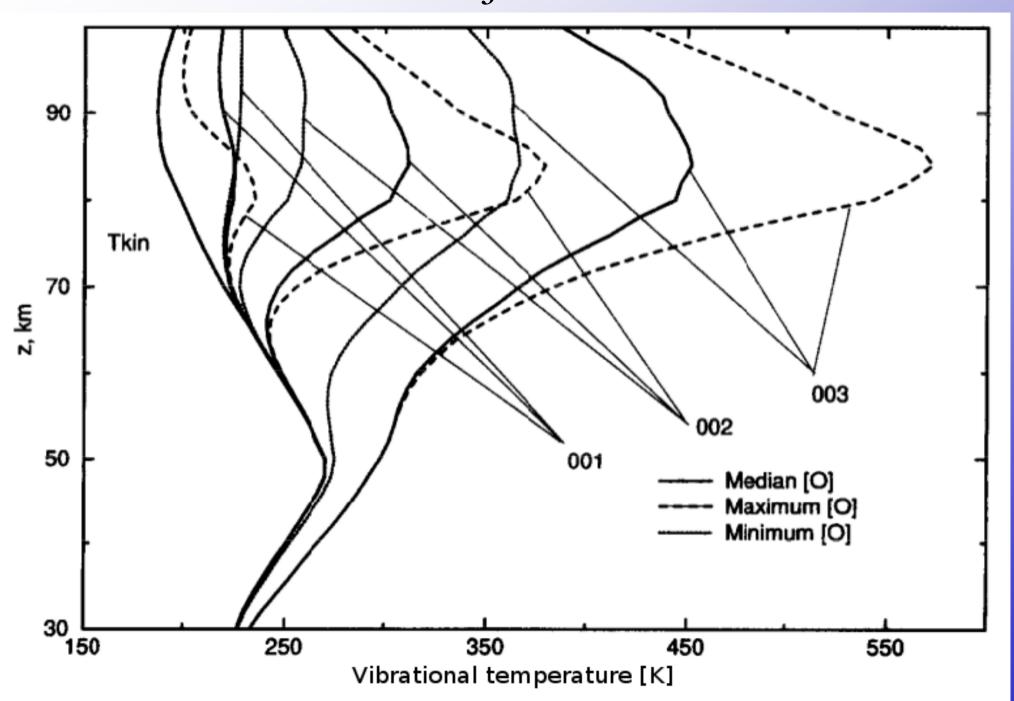
7e-12 8e-12 9e-12



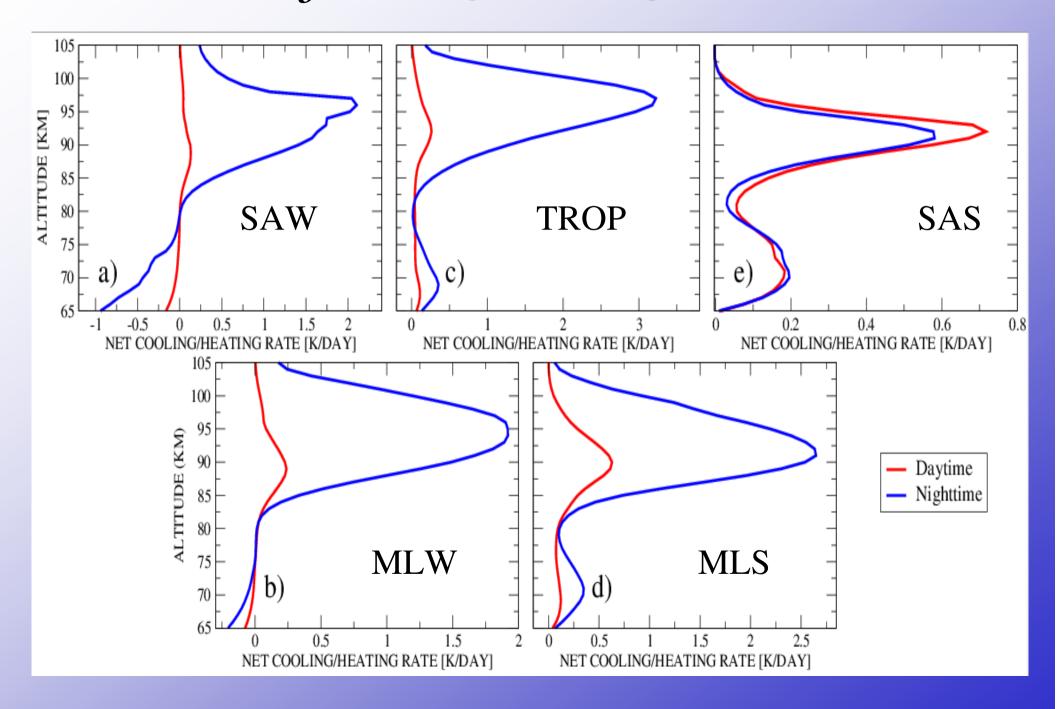
 O_3 vibrational level populations – midlatitude day



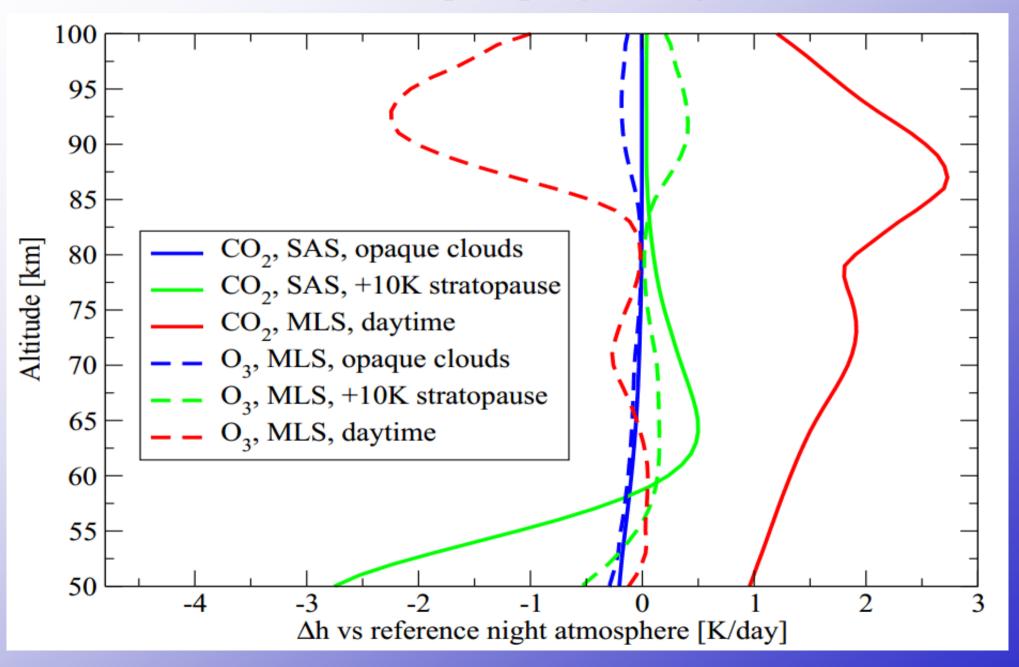
Sensitivity of v₃-levels to atomic O



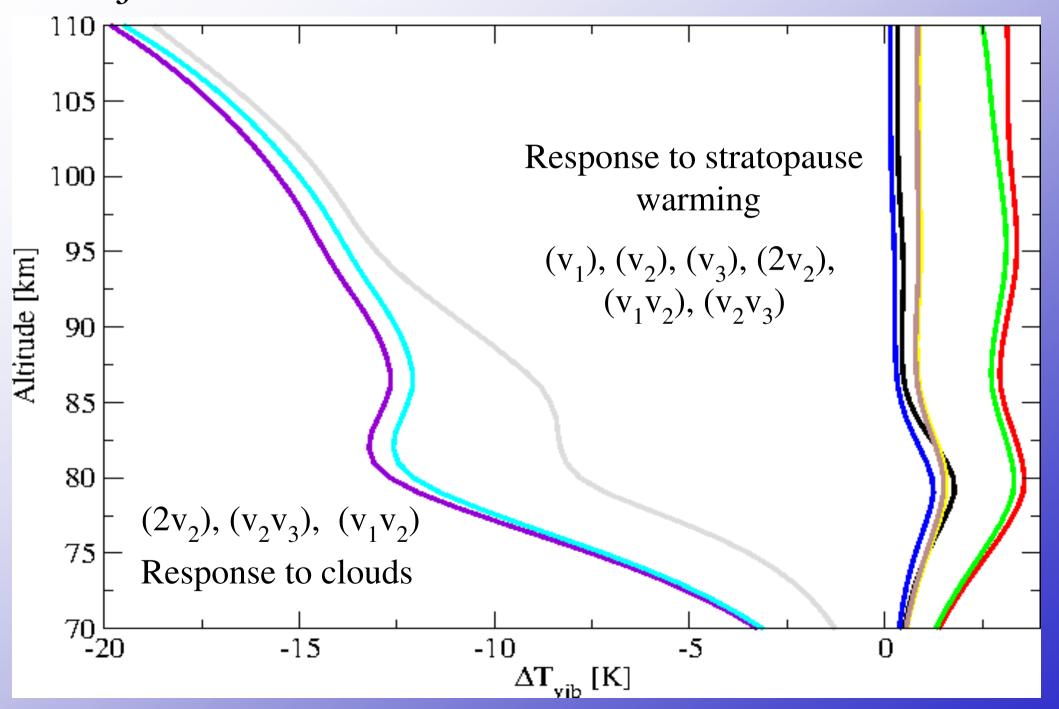
O₃ cooling/heating rates



MLT sensitivity to clouds, stratopause T, and solar pumping changes



O₃ vibr. levels, sensitive to changes in other layers



Take home messages

- \vee O₃ is an important component of the MLT
- \sim O₃(9.6µm) IR cooling/heating in 65-105 km altitude range is -1...+3 K/day
- Energetic effects of direct radiative coupling with tropo- and stratosphere are less than 1 K/day
- Changes in lower atmosphere affect vibrational levels, which are pumped by radiation coming in optically thin lines.
- Adequate model of nascent population of O₃ molecule formed in three-body recombination is still required
- Collisional rate calculations are needed.