

UNIVERSIDADE FEDERAL DE SANTA CATARINA CENTRO DE CIÊNCIAS FÍSICAS E MATEMÁTICAS PROGRAMA DE PÓS-GRADUAÇÃO EM FÍSICA

GRIFIM – Group of Interaction of Photons and Ions with Matter Supervision: Prof. Dr. Lucio Sartori Farenzena

ANALYSIS OF POLYMERIC FILMS DEGRADATION SUBJECTED TO IONIZING RADIATION

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OVERVIEW

WHAT DO WE DO?

WHY DO WE DO?

HOW DO WE DO?

PRELIMINARY RESULTS

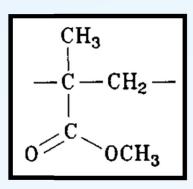
CONCLUSIONS

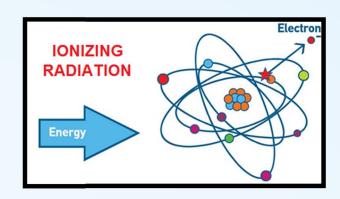


WHAT DO WE DO?

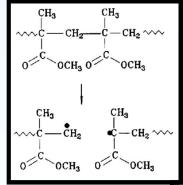
PMMA FILMS

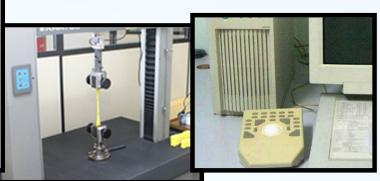
Poly(methyl methacrylate)





UV (H Ly-Alpha 10,2eV) H+, He+, C+ e N²⁺ (~10⁵-10⁶eV)





Chemical & physical changes

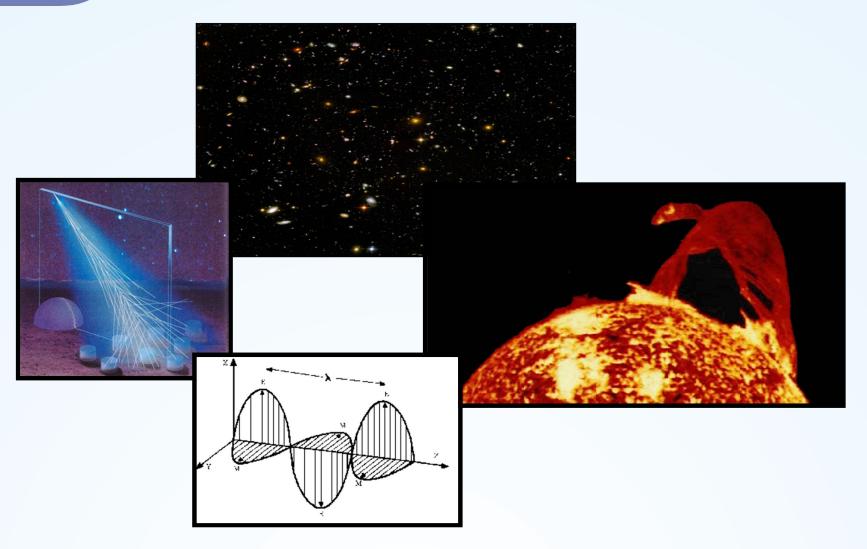
Heating, surface mass ejection, aesthetical changes





WHY DO WE DO?

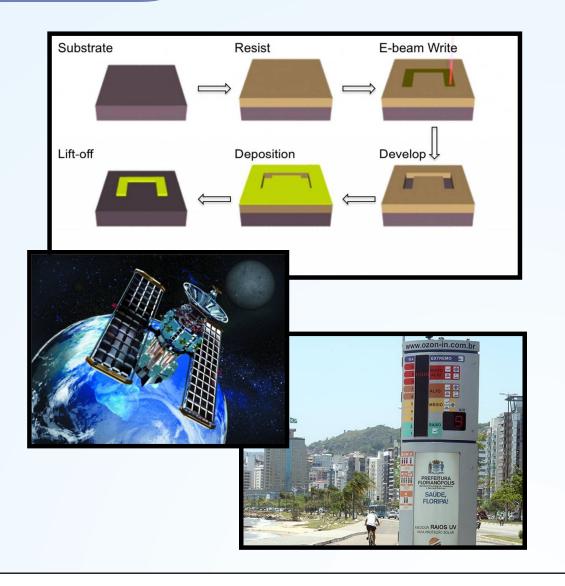
1. Natural Ionizing Radiation

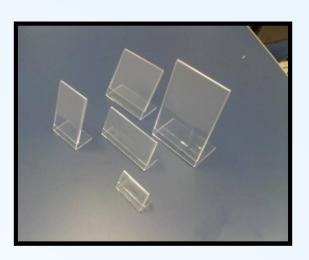




WHY DO WE DO?

2. Ionizing Radiation and PMMA





lifetime of satellites coverage

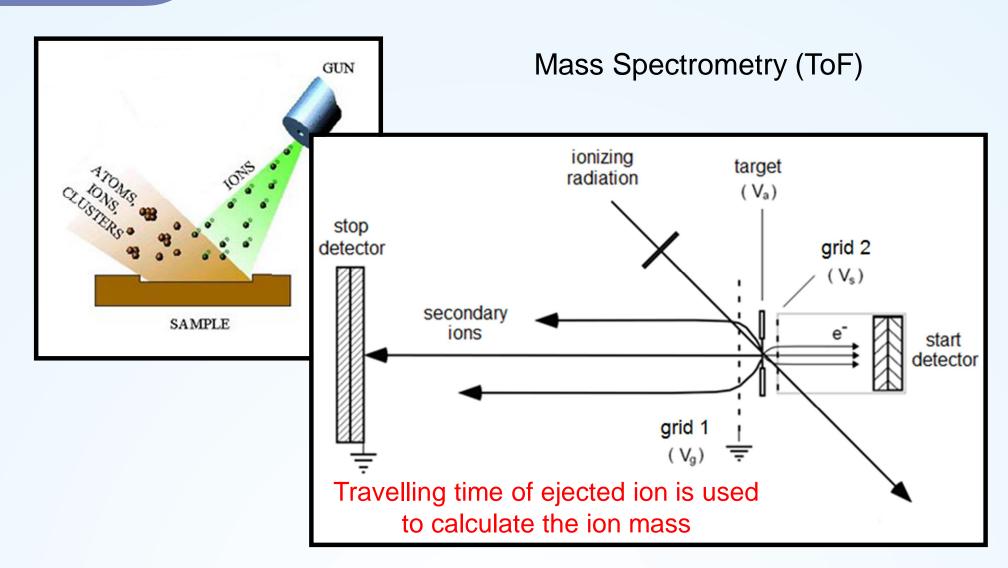
Polymer degradation subjected to environment radiation

Optical applications and lithography



HOW DO WE DO?

1. Surface Mass Ejection (Desorption)



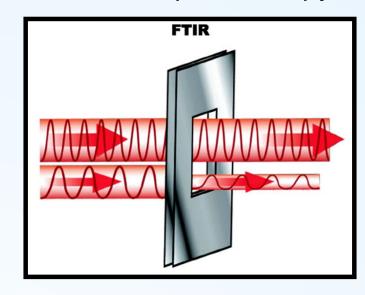


HOW DO WE DO?

2. Chemical Changes (Degradation)

n !			
Polymer	Abbreviation	Monomer	Туре
Polyethylene	PE	-{CH ₂ -CH ₂ } _n	I
Polystyrene	PS	CH ₂ -CH n	I
Polyvinyl chloride	PVC	-{cH₂-cH} <u>-</u> a	I
Polytetrafluorethylene	PTFE		п
Poly(methacrylic acid)	PMAA	СН ₃ ————————————————————————————————————	п
Poly(methyl methacrylate)	PMMA	$ \begin{array}{c} CH_3 \\ -CH_2-C-\\ -C \\ OCH_3 \end{array} $	П
Poly(maleic acid)	PMA	$ \begin{array}{c} -\left\{CH_{2}-CH\right\}_{n}\\ O=C\\ OCH_{3} \end{array} $	I

Infrared Spectroscopy



$$A(\lambda) = -log\left(\frac{I(\lambda)}{I_0(\lambda)}\right)$$

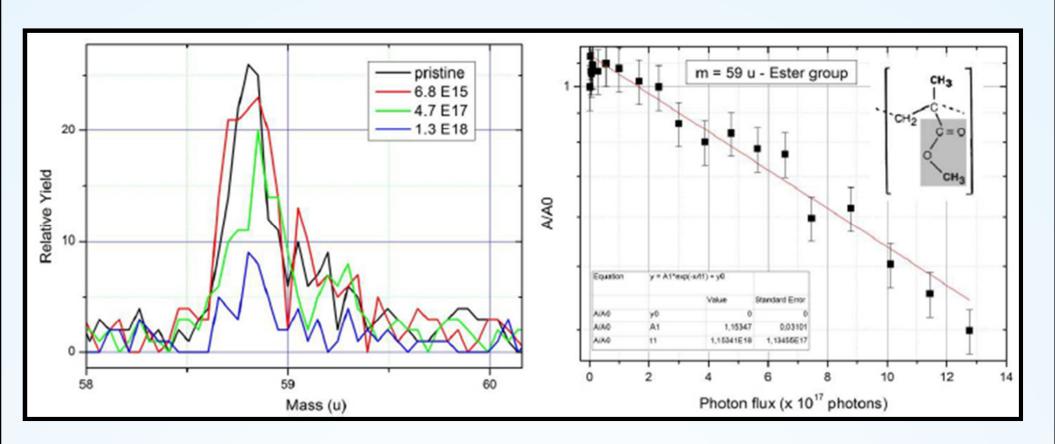
Beer-Lambert Law for absorbance

TYPE I Reticulation TYPE II Degradation



PRELIMINARY RESULTS

Surface Mass Ejection

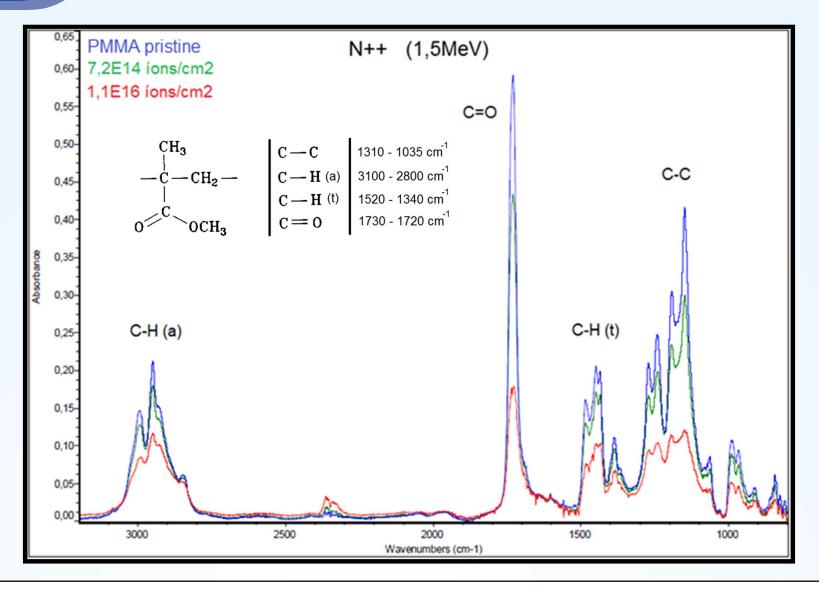


Ester group mass signal on spectrum has exponential decay with radiation dose



PRELIMINARY RESULTS

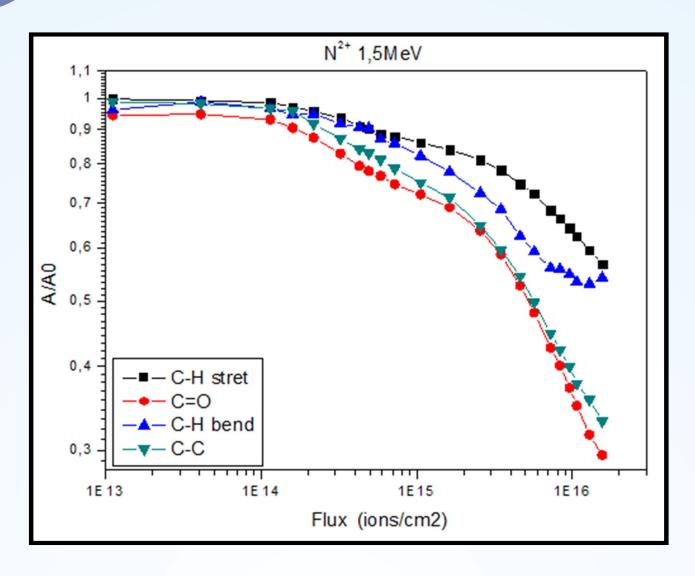
Chemical Changes





PRELIMINARY RESULTS

Chemical Changes





CONCLUSIONS

PMMA is degrading polymer when subjected to ionizing radiation

Degradation increases with the energy deposited by each projectile (depending on the electronic *stopping power*)

The degradation process can be analyzed with two complementary analytical techniques: FTIR (bulk information) and ToF (ejection information)

These two techniques can lead us to have a better understanding about degradation routes on polymers subjected to ionizing radiation



END