Hammett Analyzer User Guide

Main Screen

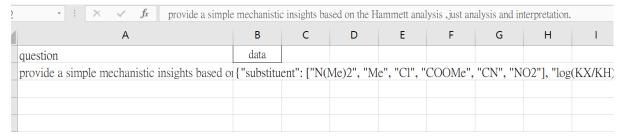
On the main screen, you will see two functionalities: "Upload PDF and Question File" and "Hammett Plot Generation."



Upload PDF and Question Flie

In the "Upload PDF and Question File" function, you can perform question answering and analysis. The "Select PDF Document" option allows you to upload a chemistry paper PDF for the program to read and learn from. Below that, the "Select Question Excel File" option allows you to upload an XLSX file containing questions for the program to answer. The program will analyze the questions based on the provided paper, and the format of the XLSX file should be as follows:

Example XLSX File

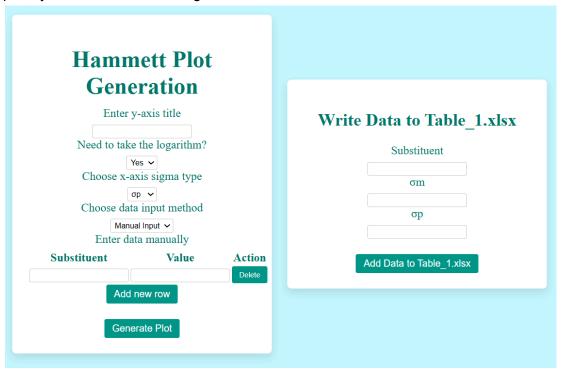


- The first column must have the header "question" and you can input multiple questions to be analyzed below it.
- If your questions are accompanied by data, you can create a second column with the header "data" and input the corresponding data below.

After uploading, click "Upload" and wait 1–2 minutes to download the program's response.

Hammett Plot Generation

In the "Hammett Plot Generation" function, you can generate and analyze a Hammett plot based on the provided substituents and corresponding data. After clicking "Generate the plot," you will see the following screen:



On the left side under "**Hammett Plot Generation**," you can generate a plot and perform analysis by completing the following fields:

Enter y-axis title

Enter the y-axis label for the plot.

Need to take the logarithm?

Choose whether the y-axis data needs to be logarithmic.

Choose x-axis sigma type

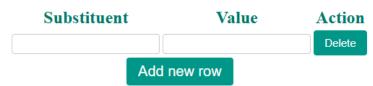
Select whether the x-axis data should use the substituent's σp or σm . (The currently recorded substituents and their corresponding σp and σm values in the program are sourced from the "Table 1.Hammett and Modified Swain-Lupton Constants" in the paper "A Survey of Hammett Substituent Constants and Resonance and Field Parameters" by Corwin, Hansch, A. Leo, and R. W. Taft.)

Choose data input method

Choose how to upload the data:

1.Manual Input

Enter data manually



- If you choose this option, you can input the data manually in the fields below. Enter the substituent in the "Substituent" field and the corresponding y-axis data in the "Value" field.
- Click "Add new row" to add more data input rows. Click "Delete" to remove a row.
- After clicking "Generate Plot," wait 1–2 minutes for the results.

2. Upload File

- If you choose this option, you can directly upload an XLSX file containing substituents and the corresponding y-axis data.
- Note: The format of the XLSX file must be as follows:

	А	В	
1	substituent	value	
	NMe2	3.11	
3	OMe	1.21	
4	Me	1.16	
5	Н	1.00	
6	C1	0.79	
7	COOMe	0.96	
8	CF3	0.80	
9			
10			
11			

- The first column must have the header "substituent" and you can input multiple substituents below it.
- The second column must have the header "value" and you can input the corresponding y-axis data for each substituent below.
- After uploading, click "Generate Plot" and wait 1–2 minutes for the results.

Error Handling

If you encounter a "KeyError" after clicking "Generate Plot," it may indicate that the format of the substituent (e.g., XXX as an example) is incorrect, or the substituent might not exist in the table. For example, if the table contains "COOMe" but you input "CO2Me," a KeyError will occur.

KeyError

KeyError: 'XXX'

Solution



Use the "Write Data to Table_1.xlsx" function on the right to input the missing substituent and the corresponding op and om values. This will write the data directly into Table_1.xlsx. You can then return to the "Hammett Plot Generation" function, upload the data, and the program will recognize the newly added substituent for analysis.

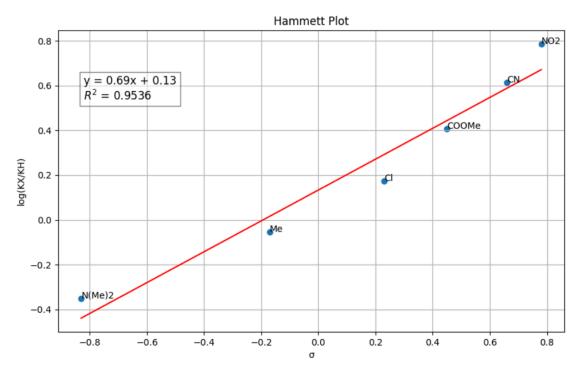
Example Input/Output

Below are sample input data, output data, and their formats.

Example data

substituent	log(KX/KH)
N(Me)2	-0.351
Me	-0.053
CI	0.173
COOMe	0.406
CN	0.6128
NO2	0.786

Example output chart



Example output answer

Question: provide simple mechanistic insights based on the Hammett analysis ,just analysis and interpretation.

Data: {"substituent": ["N(Me)2", "Me", "CI", "COOMe", "CN", "NO2"], "log(KX/KH)": ["-0.351", "-0.053", "0.173", "0.406", "0.6128", "0.786"], " σ ": [-0.83, -0.17, 0.23, 0.45, 0.66, 0.78], " ρ ": 0.6894303517439718, " ρ ": 0.9535601549580636}

Answer: ## Analysis of Hammett Plot Data and Mechanistic Insights

Your data clearly shows a strong linear correlation between the logarithm of the rate constant ratio (log k_X/k_H) and the Hammett substituent constant (σ) for a series of substituted aromatic compounds. This is strong evidence for a **Hammett relationship**, suggesting that electronic effects of the substituents play a significant role in the reaction mechanism.

Here's a breakdown:

- * **Positive ρ value (ρ = 0.69):** This indicates that the reaction is accelerated by **electron-withdrawing groups (EWGs)** and decelerated by electron-donating groups (EDGs). The positive ρ value suggests a build-up of negative charge in the transition state, which is stabilized by EWGs.
- * **Magnitude of ρ (0.69):** The ρ value's magnitude reflects the reaction's sensitivity to electronic effects. A value of 0.69 suggests a moderate sensitivity, meaning the reaction rate is noticeably affected by the electronic nature of the substituents but not to an extreme degree.
- * **High R² value (0.95):** This confirms a strong linear correlation between log k_XH</sub> and σ, supporting the validity of the Hammett relationship for this reaction.
- **Mechanistic Insights**

Based on the Hammett analysis, we can propose some mechanistic insights:

- 1. **Transition State with Negative Charge Build-up:** The positive ρ value suggests a transition state where negative charge accumulates near the reaction center. This is consistent with a mechanism where the reaction proceeds through an intermediate or transition state with increased electron density.
- 2. **Possible Rate-Determining Step:** The reaction likely involves a step where electron withdrawal from the aromatic ring stabilizes the transition state, potentially making it the rate-determining step. This could involve, for example, nucleophilic attack on the aromatic ring or formation of a carbocation intermediate.
- **Further Investigation**

To gain a more complete understanding of the mechanism, additional experiments are recommended:

- * **Varying the reaction conditions:** Investigating the effect of solvent polarity, temperature, and other reaction parameters on the ρ value can provide further insights into the transition state structure.
- * **Studying other substituents:** Expanding the range of substituents, particularly those with different steric effects, can help separate electronic and steric contributions to the reaction rate.

* **Isotopic labeling:** Employing isotopic labeling studies can help pinpoint the exact bond-forming and bond-breaking events in the rate-determining step.

By combining the Hammett analysis with additional experimental data, you can develop a comprehensive understanding of the reaction mechanism.