

1st Law of Thermodynamics

★ How? [Conduction (heat through slab)
Convection (mixing)
Radiation (emission & absorption of light)]
↳ Power \propto (Temp.)⁴ ↑

"Stefan-Boltzmann Law"

Ideal (Carnot) "engines"

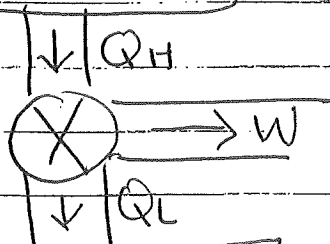
↑ Substitute Temps. (in K) for heats (Q's)

1832 dead

Kelvin \approx 1848 rediscovered

EX. One of the most efficient heat engines ever built is a coal fired steam turbine electric power plant in Ohio River valley.

$$T_H = 1870^\circ\text{C} = 2143\text{K}$$



$$T_L = 430^\circ\text{C} = 703\text{K}$$

$$1^{\text{st}}: Q_H = W + Q_L$$

$$\epsilon = \frac{\text{get}}{\text{pay}} = \frac{W}{Q_H} = \frac{Q_H - Q_L}{Q_H} = 1 - \frac{Q_L}{Q_H}$$

What is its max. efficiency?

$$\epsilon_{\text{carnot}} = 1 - \frac{T_L}{T_H} = 1 - \frac{703}{2143} = 0.67 \quad (67\%)$$

What power is delivered if the plant absorbs $1.4 \times 10^5 \text{ J}$ per second from the hot reservoir? (NOTE: The actual efficiency of the plant is 42%)

$$\epsilon = 0.42 = \frac{W}{Q_H} \quad Q_H \leftarrow 1.4 \times 10^5 \text{ J}$$

Can you find the heat expelled?
 Q_L

$$\therefore W = 0.42 * Q_H = 58800 \text{ J}$$

"per second"

$$\underline{58.8 \text{ kW}} \quad \text{Answer.}$$

EX. 1 A freezer has a C.O.P. = 6.3.
It is advertised as using 457 kW.hr of energy per year.

a. On average, how much energy does it use per day?

$$\text{Power} = \frac{\text{Energy}}{\text{time}} \Rightarrow \text{Energy} = \underset{\substack{\uparrow \\ \text{kW} \times \text{hr}}}{\text{Power}} \times \underset{\uparrow}{\text{time}}$$

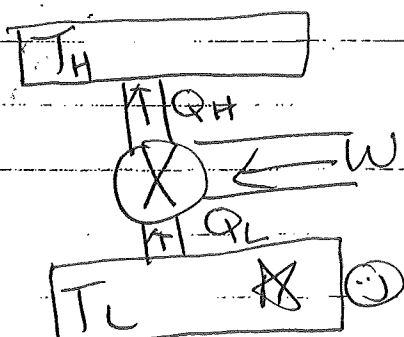
$$457 \frac{\text{kW} \cdot \text{hr}}{\text{yr}} \times \frac{1000 \text{ W}}{\text{kW}} \times \frac{1 \text{ yr}}{365 \text{ days}} = 1252.1 \frac{\text{W} \cdot \text{hr}}{\text{day}}$$



$$\Rightarrow 1252.1 \frac{\text{J}}{\text{s}} \times (\text{hr}) \times \frac{3600 \text{ s}}{\text{hr}} = 4.51 \times 10^6 \frac{\text{J}}{\text{day}}$$



b. how much heat is removed from the freezer in a single day?

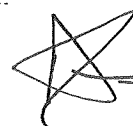


$$1^{\text{st}}: Q_L + W = Q_H$$

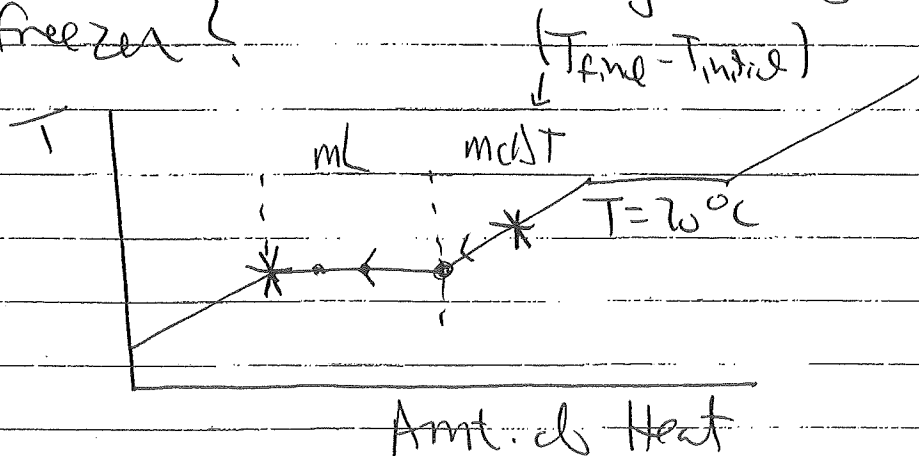
$$\text{C.O.P.} = \frac{\text{get}}{\text{pay}} = \frac{Q_L}{W} = 6.3$$

$$\therefore Q_L = 6.3 \times W = 28.4 \times 10^6 \text{ J}$$

day



c.) What is the maximum amt. of water @ 20°C that can be converted to ice @ 0°C in a single day by this freezer?



$$Q_{\text{removed from water}} = m C_{\text{water}} (0 - 20) - m L_f$$

$$= -m (20 C_{\text{water}} + L_f) = -417.8 \times 10^3 \text{ m}$$

\uparrow 4190 $\frac{\text{J}}{\text{kgK}}$ \uparrow 334 $\times 10^3 \frac{\text{J}}{\text{kg}}$

Must be REMOVED

To the question: In 1 day, can remove $28.4 \times 10^6 \text{ J}$

$$\therefore 28.4 \times 10^6 = 417.8 \times 10^3 (m)$$

$$m = 68.1 \text{ kg}$$

Answer.

Ch. 21

All matter has mass.

Some matter has additional property of charge.

Protons $\Rightarrow +1.6 \times 10^{-19}$ Coulombs (C)

Electrons $\Rightarrow -1.6 \times 10^{-19}$ C SI unit

Opposites attract, likes repel.

In solids, protons are "locked" in nucleus.
Electrons can move.



Easy to come by in Nature ☺

Static Electricity ☺

DEMO w/ FOILS (Electroscope)

"Charging by Induction"

Triboelectric Series (separation of charge by rubbing)

positive end
(lose electrons)

air
human skin
asbestos
rabbit fur
acetate
glass
mica
human hair
nylon
wool
cat fur
lead
silk
aluminum
paper
(0) cotton (0)
steel
wood
amber
wax
vulcanized rubber
mylar
copper & nickel
brass & silver
synthetic rubber
gold & platinum
sulfur
acetate & rayon
celluloid
polyester
polystyrene
orlon

Ancient !

Ben Franklin 1747(?) "positive" and "negative", single fluid, all sorts of stuff!!
1752 lightening is electrical discharge (lightening rod)
1753 Copely Medal, Royal Society ,MA Yale & Harvard

J.J. Thompson

1897 Electron

Ernst Rutherford

1911 Nucleus (coined word 'proton')

James Chadwick

1932 Neutron

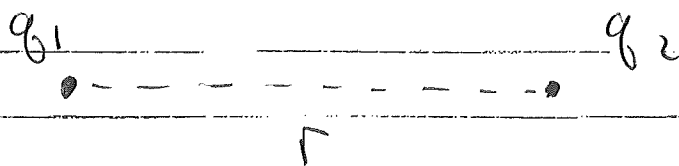
Charles Coulomb (French) 1736-1806 $K = 8.99 \times 10^9$
1785 $F \propto \frac{q_1 q_2}{r^2}$ $= \frac{1}{4\pi\epsilon_0}$

Cavendish $\frac{1}{4\pi\epsilon_0}$
Not published
1731-1810

acrylic
rubber balloon
saran
cellophane tape
polyurethane
polyethylene
polypropylene
polyvinylchloride
silicon
teflon
silicone rubber
(gain electrons)
negative end

Faraday 1844
"Lines of Force"

Coulomb's Law



$$|\vec{F}_{q_1 q_2}| = \frac{k |q_1| |q_2|}{r^2} \quad || \quad k = 8.99 \times 10^9 \frac{Nm^2}{C^2}$$

Assign direction !!!

Opposites attract
Likes repel

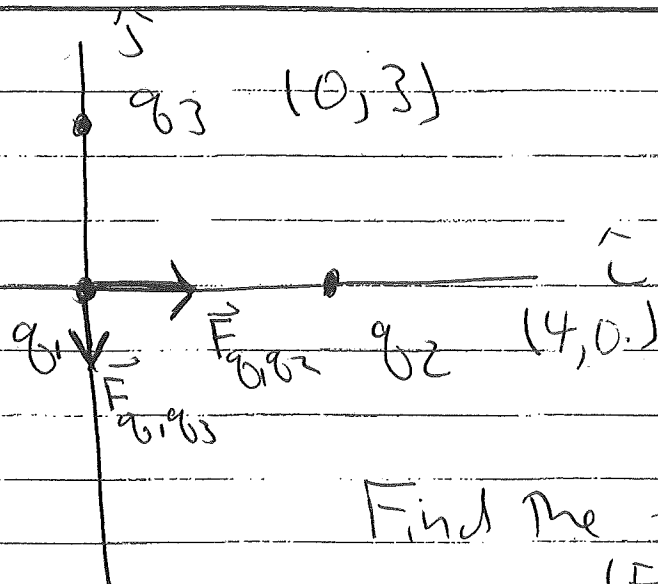
$$k = \frac{1}{4\pi\epsilon_0}$$

↑
permittivity
of free
space

Recall Newton: Interaction
characterized by a
single force experienced from
two different perspectives.

Electric force \Rightarrow "Spooky" || Action at a distance

Ex. 1



SI units

$$\begin{cases} q_1 = -5 \text{ C} \\ q_2 = +10 \text{ C} \\ q_3 = -8 \text{ C} \end{cases}$$

Find the force on q_1 .
(Electr)

Charges are "nailed down" 😊

$\vec{F}_{q_1q_2}$ is force on q_1 due to q_2 .

$$|\vec{F}_{q_1q_2}| = \frac{k|q_1||q_2|}{r_{12}^2} = \frac{8.99 \times 10^9 (5)(10)}{4^2}$$

$$= 2.81 \times 10^{10} \text{ newtons}$$

$$\vec{F}_{q_1q_2} = +2.81 \times 10^{10} \hat{x}$$

"Superposition"

$\vec{F}_{q_1q_3}$ is force on q_1 due to q_3 .

$$|\vec{F}_{q_1q_3}| = \frac{k|q_1||q_3|}{r_{13}^2} = 4.0 \times 10^{10} \text{ newtons}$$

$$\vec{F}_{q_1q_3} = -4 \times 10^{10} \hat{y}$$

These are NOT notes. They are a visual aid(20%) for a verbal explanation(80%).

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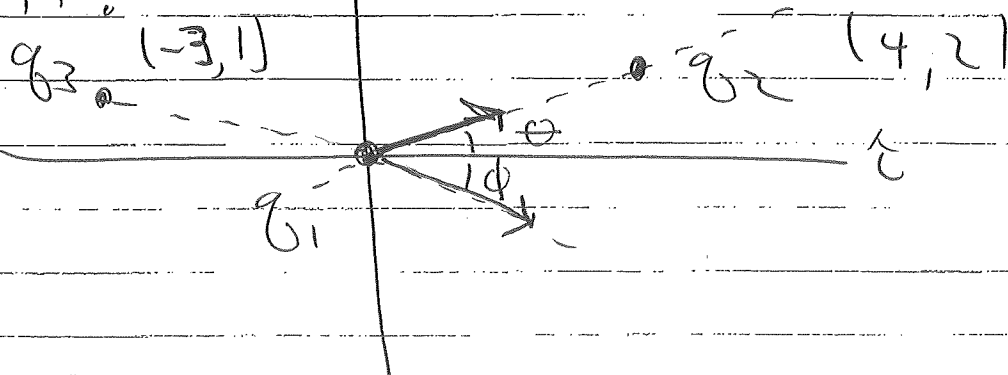
$$\vec{F}_{on\ q_1} = \vec{F}_{q_1q_2} + \vec{F}_{q_1q_3} = 2.81 \times 10^{10} \hat{i} - 4 \times 10^{10} \hat{j}$$



$\vec{F}_{on\ q_1}$

(OR) report as a
magnitude and direction

What if?



q_3

q_2

q_1

A Paradigm Shift :

q_1

q_2

$$|\vec{F}_{q_1q_2}| = \frac{k|q_1||q_2|}{r^2}$$

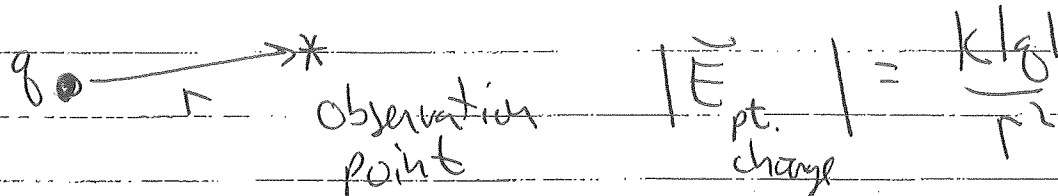
Spooky
Force

Hmmmm... Suppose q_1 disturbs all
the space around it. *

q_2 then interacts w/ that "disturbed"
space.

Characterize that disturbance with a
"VECTOR FIELD" ...

Define: the electric field produced
by a point charge

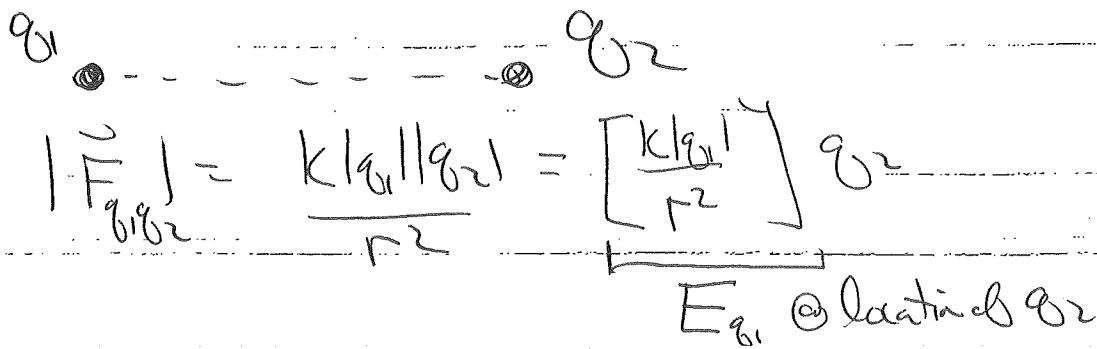


$$| \vec{E}_{\text{pt. charge}} | = \frac{k|q_1|}{r^2}$$

Radially away from positive charges.
Radially toward negative charges.

Michael Faraday, "Lines of Force", ~1844

Revisit:



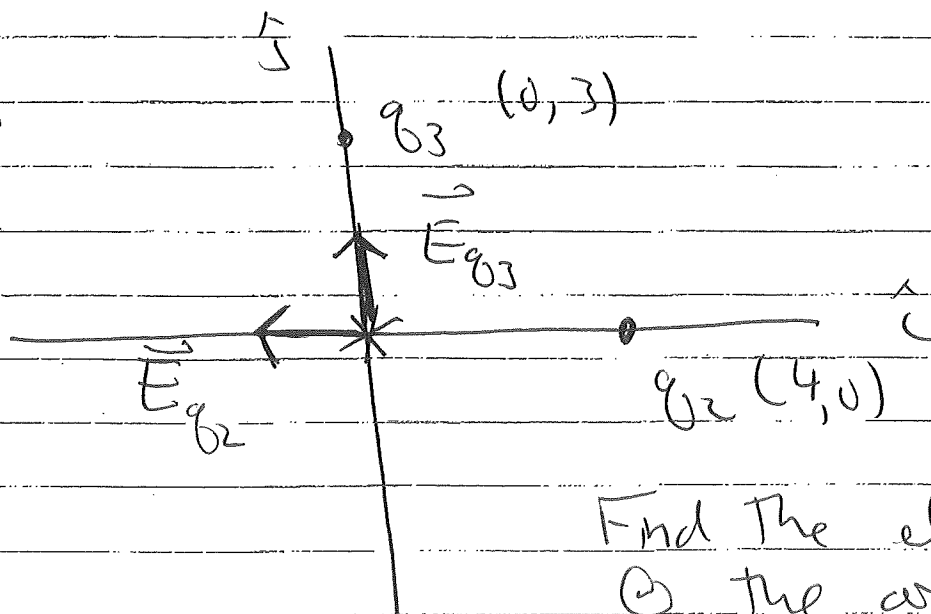
$$| \vec{F}_{q_1 q_2} | = \frac{k|q_1||q_2|}{r^2} = \left[\frac{k|q_1|}{r^2} \right] q_2$$

$E_{q_1} \text{ @ location of } q_2$

$$\vec{F} = q \vec{E}$$



EX.



$$\begin{cases} q_2 = 10 \text{ C} \\ q_3 = -8 \text{ C} \end{cases}$$

Find The electric field
① the origin.

$$\vec{E}_{@ \text{origin}} = \vec{E}_{q_2 @ \text{origin}} + \vec{E}_{q_3 @ \text{origin}}$$

$$\vec{E}_{q_2 @ \text{origin}} = - \frac{k|q_2|}{4^2} \hat{i} = -5.62 \times 10^9 \hat{i}$$

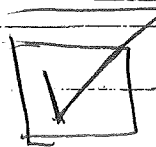
$$\vec{E}_{q_3 @ \text{origin}} = + \frac{k|q_3|}{3^2} \hat{j} = 7.99 \times 10^9 \hat{j}$$

$$\therefore \vec{E}_{@ \text{origin}} = -5.62 \times 10^9 \hat{i} + 7.99 \times 10^9 \hat{j}$$

What would be the force on

a -5C charge placed @ the origin?

$$\vec{F} = q \vec{E} = (-5)(-5.62 \times 10^9 \hat{i} + 7.99 \times 10^9 \hat{j})$$
$$= +2.81 \times 10^{10} \hat{i} - 4 \times 10^{10} \hat{j}$$



Consistent