

Notes of **Modern Quantum Chemistry, Szabo & Ostlund**

hebrewsnabla

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Contents

0		2
1		2
1.1	2
1.2	2
1.3	2
1.4	N-D Complex Vector Spaces	2
1.4.1	Change of Basis	2
2		3
2.1	The Electronic Problem	3
2.1.1	Atomic Units	3
2.1.2	The B-O Approximation	3
2.1.3	The Antisymmetry or Pauli Exclusion Principle	3
2.2	Orbitals, Slater Determinants, and Basis Functions	3
2.2.1	Spin Orbitals and Spatial Orbitals	3
2.2.2	Hartree Products	3
2.2.3	Slater Determinants	3
2.2.4	The Hartree-Fock Approximation	3
2.2.5	The Minimal Basis H ₂ Model	3
2.2.6	Excited Determinants	3

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spatial mol orb – $\psi - i, j, k, \dots$
 spatial basis fcn – $\phi - \mu, \nu, \lambda, \dots$
 spin orb – χ
 occ mol orb – a, b, c, \dots
 vir mol orb – r, s, t, \dots
 exact many-elec wfn – Φ
 approx many-elec wfn – Ψ
 exact energy – \mathcal{E}
 approx energy – E

1

1.1

1.2

1.3

1.4 N-D Complex Vector Spaces

Suppose

$$\mathcal{O}|a\rangle = |b\rangle \quad (1.1)$$

$$\langle i|\mathcal{O}|j\rangle = O_{ij} \quad (1.2)$$

def the **adjoint** of \mathcal{O} as \mathcal{O}^\dagger

$$\langle a|\mathcal{O}^\dagger = \langle b| \quad (1.3)$$

$$\langle i|\mathcal{O}^\dagger|j\rangle = O_{ji}^* \quad (1.4)$$

1.4.1 Change of Basis

$$|\alpha\rangle = \sum_i |i\rangle \langle i|\alpha\rangle = \sum_i |i\rangle U_{i\alpha} \quad (1.5)$$

$$|i\rangle = \sum_\alpha |\alpha\rangle \langle i|\alpha\rangle = \sum_\alpha |\alpha\rangle U_{i\alpha}^* \quad (1.6)$$

If i, α are all orthonormal, \mathbf{U} must be unitary.

$$\Omega_{\alpha\beta} = \langle \alpha|\mathcal{O}|\beta\rangle = \dots \sum_{ij} U_{\alpha i}^* O_{ij} U_{j\beta} \quad (1.7)$$

or

$$\mathbf{\Omega} = \mathbf{U}^\dagger \mathbf{O} \mathbf{U} \quad (1.8)$$

2

2.1 The Electronic Problem

2.1.1 Atomic Units

2.1.2 The B-O Approximation

2.1.3 The Antisymmetry or Pauli Exclusion Principle

2.2 Orbitals, Slater Determinants, and Basis Functions

2.2.1 Spin Orbitals and Spatial Orbitals

2.2.2 Hartree Products

2.2.3 Slater Determinants

def

$$|\chi_i(\mathbf{x}_1)\chi_j(\mathbf{x}_2)\cdots\chi_k(\mathbf{x}_N)\rangle \equiv \frac{1}{\sqrt{N!}} \begin{vmatrix} \chi_i(\mathbf{x}_1) & \chi_j(\mathbf{x}_1) & \cdots & \chi_k(\mathbf{x}_1) \\ \chi_i(\mathbf{x}_2) & \chi_j(\mathbf{x}_2) & \cdots & \chi_k(\mathbf{x}_2) \\ \vdots & \vdots & & \vdots \\ \chi_i(\mathbf{x}_N) & \chi_j(\mathbf{x}_N) & \cdots & \chi_k(\mathbf{x}_N) \end{vmatrix} \quad (2.1)$$

It can be further shortened to

$$|\chi_i\chi_j\cdots\chi_k\rangle \quad (2.2)$$

2.2.4 The Hartree-Fock Approximation

2.2.5 The Minimal Basis H₂ Model

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2.2.6 Excited Determinants

Suppose the ground state det

$$|psi_0\rangle = |\chi_1 \cdots \chi_a \cdots \chi_b \cdots \chi_N\rangle \quad (2.3)$$

thus, singly excited det

$$|psi_a^r\rangle = |\chi_1 \cdots \chi_r \cdots \chi_b \cdots \chi_N\rangle \quad (2.4)$$

$$|psi_{ab}^{rs}\rangle = |\chi_1 \cdots \chi_r \cdots \chi_s \cdots \chi_N\rangle \quad (2.5)$$

2.3 Operators and Matrix Elements

2.3.1 Minimal Basis H₂ Matrix Elements

2.3.2 Notations for 1- and 2-Electron Integrals

For spin orb

$$\langle ij | kl \rangle = \left\langle \chi_i(1)\chi_j(2) \left| \frac{1}{r_{12}} \right| \chi_k(1)\chi_l(2) \right\rangle \quad (2.6)$$

$$\langle ij || kl \rangle = \langle ij | kl \rangle - \langle ij | lk \rangle \quad (2.7)$$

$$[ij|kl] = \left\langle \chi_i(1)\chi_j(1) \left| \frac{1}{r_{12}} \right| \chi_k(2)\chi_l(2) \right\rangle \quad (2.8)$$

For spatial orb

$$(ij|kl) = \left\langle \psi_i(1)\psi_j(1) \left| \frac{1}{r_{12}} \right| \psi_k(2)\psi_l(2) \right\rangle \quad (2.9)$$