Intermetallic compounds are formed from two or more metals. They represent a new phase with its own composition, crystal structure and properties.

TiAl is an intermetallic compound composed of two metals: Ti and Al.

Al203 is a ceramic compound composed of a metal (Al) and non-metal (O).

Q2.

Eutectic: A three phase invariant reaction in which one liquid phase solidifies to produce two solid phases

Eutectoid: A solid phase transforms into two different solid phases

Peritectic: A liquid and a solid phase combine to produce a second solid phase

Perifectoid: Two solid phases combine to produce a new solid phase

Monotectic: One liquid phase transforms into a solid phase and a second liquid phase

Q3 Refer to Fig. 1

a) Pouring temperature 1150°C

6) Superheat = 1150-1000 = 150°C

c) Liquidus temperature = 1000°C

d) Eutectic temperature = 577°C

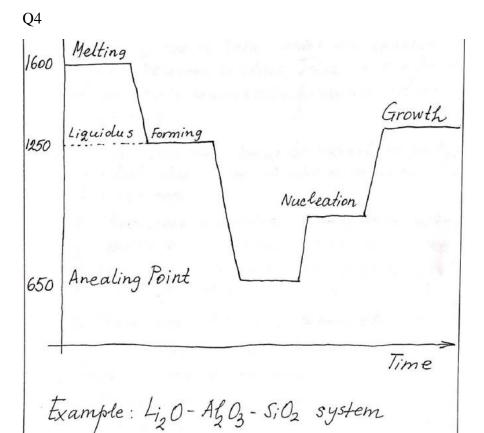
e) Freezing Range = 1000-577=423°C

f) Solidification Time = 11.8-1= 10.8 min

9) Total Solidification Time = 11. 8 min

L) Composition: Refer to Figure 11-19 (Book)

=> Approximately 45%



Glass-ceramics: Materials formed in the glassy-state and allowed to crystallize during a heat treatment

Q5

Most widely used ceramic materials: Als O3 (alumina), Diamond (C), Silica (SiO2) Silicon carbide (SiC), Silicon Nitride (Si3 N4) TiO2, ZrO2

**Q**6

The temperature below which an undercooled liquid becomes a glass. This is not a fixed temperature and is also known as glass - transition.

Q7

- a) Thermoplastie: Linear or branched polymer in which chains are not inter-connected to one another.
- b) Thermosetting plastics: Linear or branched polymers in which chains are cross-linked to one another to form three-dimensional network structures.

c) Elastomers - Polymers (thermoplastics or lightly cross-linked thermosets) that have an elastic deformation > 200% d) Thermoplastic Elastomers: Polymers that are heavily cross-linked to produce strong three-dimensional structures.

 $Q = \frac{R_{olyvinyl} \ chloride, From Table 16-3 \ Repead unit}{Cl \ Cl \ Cl} \ M_r (C_2 H_3 Cl) = 2.12+3.1+35.5 = 62.5 g/mot ce cl \ M_r = 150000 g/mot \ a) Degree of Polymerization (DP) <math>DP = \frac{M_r}{M_r'}$   $DP = \frac{150000}{62.5} = 2400$ 6) Number of chains in 5g of polymer:  $Imol(150000g) = 6.023.10^{23} chains$   $Imol(150000g) = 6.023.10^{23} chains$ 

$$m = \frac{600000 \cdot 5 \cdot 10^{21}}{6.023 \cdot 10^{23}} = 4150.76 g$$