Assign. #7 MAT E 201 Dr. S. Djokić Q1 I= 20A, d=3mm, l=1000 m Power Loss: P=VI= I2R, R= 1/46, A= 02/1 $P = I^{2} \frac{\ell}{6} \frac{4}{d^{2} \pi} = (204)^{2} \frac{1000 \text{ m}}{6} \frac{4}{(2.10^{3} \text{ m})^{2} \pi}$ P= 5.6588.10" - [A/m] = 5.6588.10. - [A/c] a) Aluminum = 3.77.10 52 cm-1 P= 1501 W 6) Nickel 5=1.46.105 2cm-1 P=3876W c) SiC = 10-10 2cm-1 P= 5.6588.1018 W

Q2 AlAs fibre, d=0.5mm = 0.05cm, l=0.7cm GARAS = 0. 1 scm', V=110 V a) $V = IR = I \frac{\ell}{AG}$, $A = \frac{d^2 \pi}{4} = 7 I = \frac{V d^2 \pi}{4\ell}$ I = 110V. (0.55cm) T. 2.15cm = 0.03085A 6) J= I = ng v $\frac{h}{t} = \frac{4I}{d^2 \pi g \ell}$ $\frac{1}{A} = ng\frac{\ell}{\ell} \Longrightarrow$ $\left(\frac{n}{t}\right)$ - number of electrons per second $\frac{n}{t} = \frac{4 \cdot 0.03085 \text{ A}}{(0.05 \text{ cm})^2 \text{ T.} \cdot 1.6 \cdot 10^{-19} \text{ c.} \cdot 0.7 \text{ cm}} = 1.4028 \cdot 10^{20} \frac{1}{\text{cm}^3 \text{ s}}$ Q3 d=0.3 mm Cu wire, P < 300W, I=5A $A = \frac{d^2 \pi}{4}$, $\ell = ?$ $G_{cu} = 5.598.10^5 \pi cm^{-1}$ $R = \frac{\ell}{6A}$, $V = IR = I = \frac{I}{6d^2\pi}$, $P = V = I = \frac{I^2 4\ell}{6d^2\pi}$ $l = \frac{P6d^{2}\pi}{4\bar{I}^{2}} = \frac{300.5.598.105.6m^{2}.(6.03cm)^{2}\pi}{4.(5A)^{2}}$ l=4748.39 cm = 47.48 m Q4 V=5000V, Cu wire, 6c=5.598.10 Sani 1=1000m R=502 $R = \frac{1}{6} \cdot \frac{\ell}{A} = \frac{1}{6} \cdot \frac{4\ell}{d^2 J_l} \implies d = \sqrt{\frac{4\ell}{R \ll J_l}}$ $d = \sqrt{\frac{4.1000.100 \text{ cm}}{5\Omega \cdot 5.598.10^5 \Omega^{2} \text{cm}^{1} \pi}} = 0.213 \text{ cm}$ V= IR = I. & = J. & . 1 J= V6 - 5000.5.98.105 = 29900 A/cm²
1000.100

Q5 Ag, M=85 cm²/Vs From Appendix A, Ag => ao= 4.0862.10 cm FCC structure => 4 at/u.c. Total number of electrons per cm3 n_ = \frac{4at/\(\text{u.c.}\) \left[\text{electron/at} \] - 5.8627.10^2 \frac{\text{electr.}}{\text{cm}^3} Since $\mu = \frac{6}{hg}$, $6 = 6.80 \cdot 10^5 \Omega cm^{-1}$ Number of charge carriers: $n = \frac{6.80 \cdot 10^{3} \text{ cm}^{-1}}{\text{Mg}} = \frac{6.80 \cdot 10^{3} \text{ cm}^{-1}}{85 \text{ cm}^{2}/\text{Vs} \cdot 1.6 \cdot 10^{-19} \text{ c}}$ 5-10 -electr. Fraction of electrons that carry charge (F) $F = \frac{n}{n_{T}} = \frac{5.10^{22}}{5.8627.10^{22}} = 0.8525$