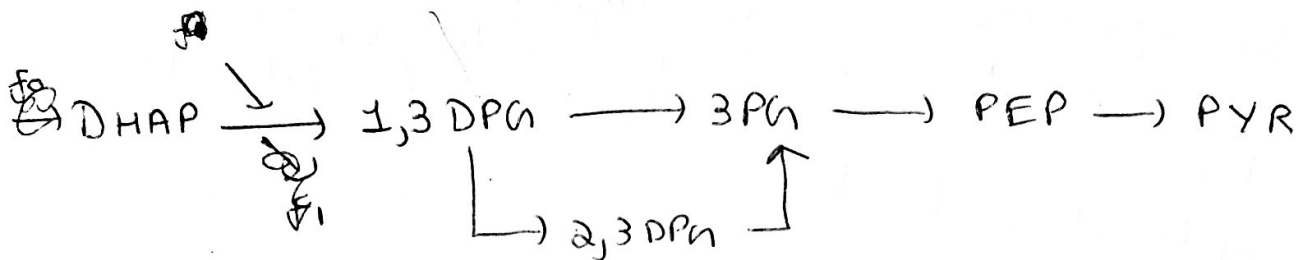


Assignment - 2

Arvind
MT16122



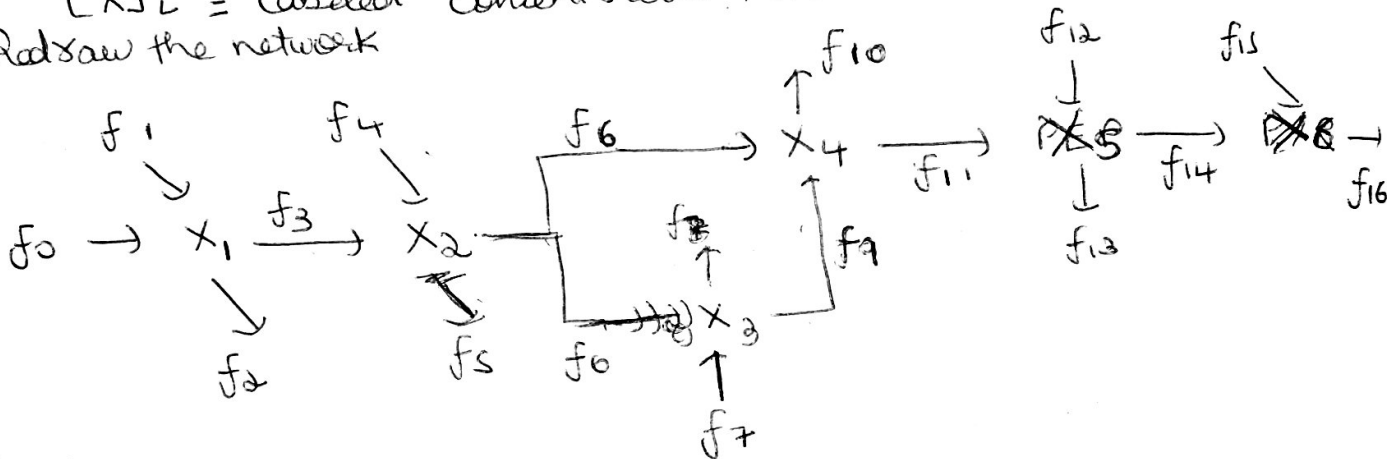
Now
Given

$$[X]_T = \text{Total concentration of X metabolite}$$

$[X]_0$ = Unlabeled concentration of * metabolite

$[X]_L$ = labeled concentration x metabolite

→ Rod saw the network



whose

$$X_1 = [\text{DHAP}] \quad X_3 = [2,3 \text{ DPG}] \quad X_5 = [\text{PEP}]$$

$$X_2 = [1, 3 \text{ DPh}] \quad X_4 = [3 \text{ Ph}] \quad X_6 = [P, Y, R]$$

c) Input flux = Output flux at node [i.e. x_1, x_2, \dots, x_6]
and $[X]_u + [X]_L = [X]^T$ is condition to be in steady state.

$$\begin{array}{l|l|l} \text{b)} & \text{For } x_1 & \text{For } x_2 & \text{For } x_3 \\ f_0 + f_1 = f_2 + f_3 & f_3 + f_4 = f_5 + f_6 & f_6 + f_7 = f_8 + f_9 \\ \\ \text{For } x_4 & \text{For } x_5 & \text{For } x_6 \\ f_6 + f_9 = f_{10} + f_{11} & f_{11} + f_{12} = f_{13} + f_{14} & f_{14} + f_{15} = f_{16} \end{array}$$

c]

$$\frac{dx_1^u}{dt} = f_1 - f_x (x_1^u / x_1^T) \quad \text{where } f_x = f_0 + f_1 = f_2 + f_3$$

The equation gives the decaying of unlabeled isotope metabolite with respect to time. Or it tells the rate of disappearance of unlabeled/unlabeled nutrient.