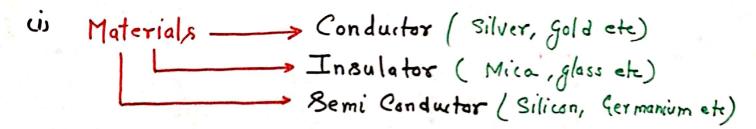
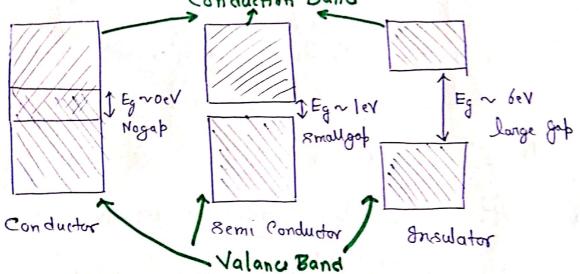
Semi Conductor Basic.



(2) Material classification: on the basis of Band gap.



Duestion. 1 Conductor > large No of tree electrons

Insulator > Small or No free electrons

Semi conductor > few no. of free electrons

Semi-Conductor

(Silicon

Slemental

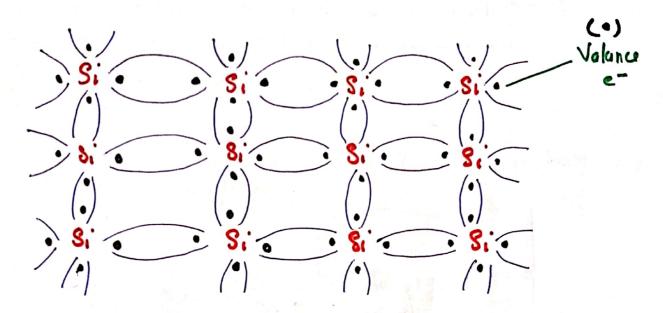
Germanium) Semi Conductor

(GaAr, GaP etc.)

(4) Silicon or Germanium > Both have 4 Valence e- in the aut most Cell ie it require temore to form Stable Status.

1 Si atom makes Cavalent Bond By Sharing its all four valence = to 9 other Si atom.

2-D diagram View



No free electron all valence e- are covalently bonded.

Note: - At OK Semiconductor behaves as Insulator

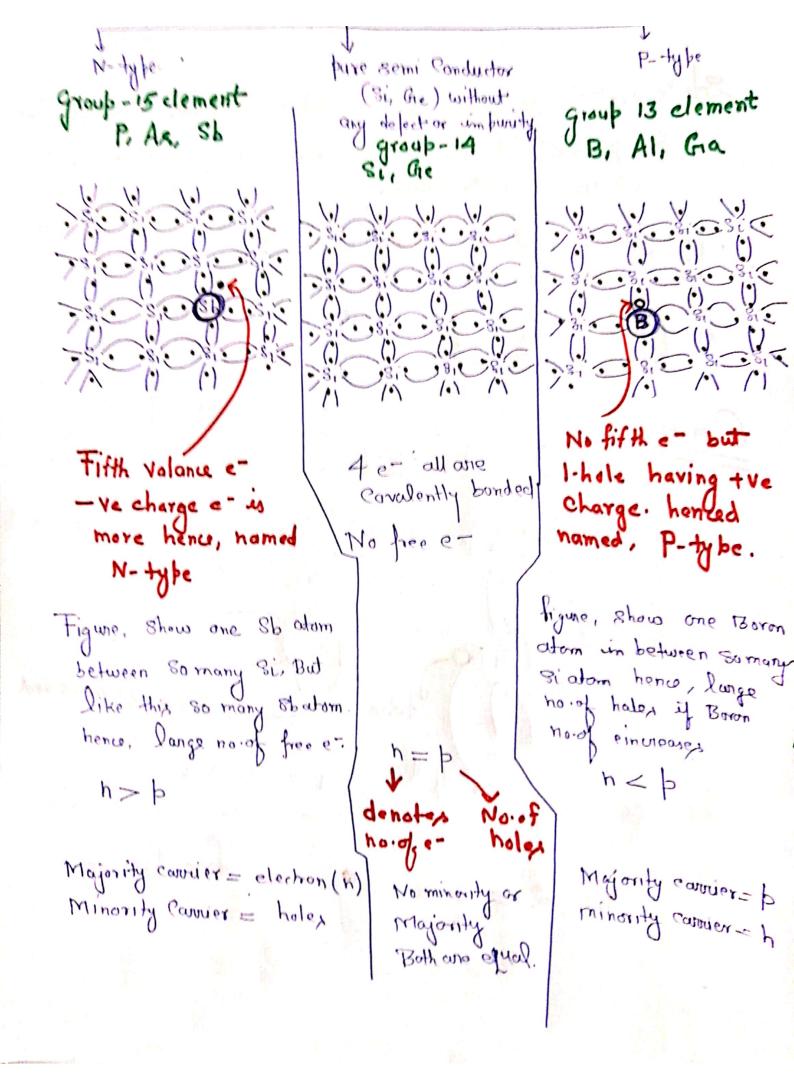
At 300 k or Room temp behaves as little bit Conducting.

Ans: - Because some of the covalent bond absorb k.E from natural cause like light energy and thermal Energy and Covalent bonds breaks and few e-becames free.

Ouestion. 2: Semi Conductor have -ve temberature Coefficient.

Ans: - - ve temp. Coefficient means (increase) in temperature if (reduces) its resistance.

As temp of covalent bonds breaks as a results free e F and Conductivity of hence registance to



MASS Action. LAW! - gives relation ship blu etectron and hale concentration.

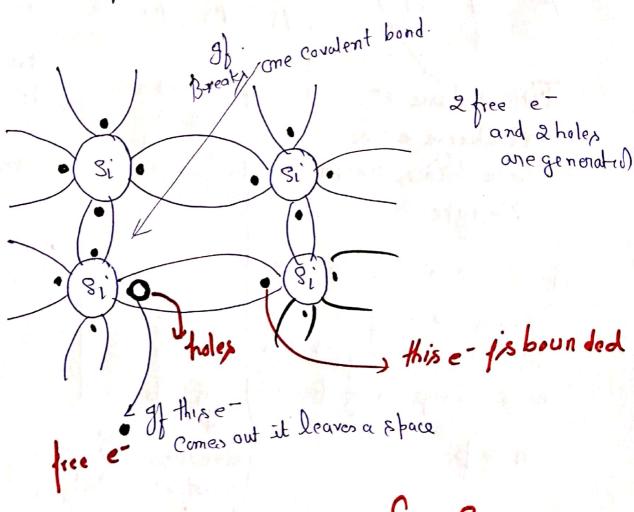
Statement: States that under thermal Equillibrium the product of free -ve and +ve carrier Concentration is a Constant.

 $Nb = ni^2$

hi > intrinsic carrier Concentration

Concentration of free holes in Equillm.

Concept:



hi = 2 = n = b -> for Pure