



**Figure 11.17.** Movement of adsorption zone through a fixed-bed downflow adsorber and the corresponding breakthrough curve. (With permission, from D. W. Sundstrom and H. E. Klei, *Wastewater Treatment*, Pearson Education, Upper Saddle River, NJ, 1979, p. 281.)

would observe the *breakthrough* curve of the adsorption zone. This process is depicted in Fig. 11.17.

Under these circumstances, eq. 11.47 would describe the concentration profile in the adsorption zone of a height  $H$ . If the column were of total height  $H_t$ , then the time to exhaustion of the column would be equal to the ratio of  $H$  to  $H_t$  multiplied by the velocity of the adsorption zone, plus the time required to form the adsorption zone (usually negligible).

### Example 11.3.

Cephalosporin is separated from fermentation broth by adsorption on an ion-exchange resin in a moving-bed column. The bed is 4 cm in diameter and contains  $0.8 \text{ cm}^3 \text{ resin/cm}^3 \text{ bed}$ . The density of the resin is  $1.3 \text{ g/cm}^3$ , and the feed solution contains  $5 \text{ g/l}$  of the antibiotic. If the superficial velocity of liquid is  $1.5 \text{ m/h}$  and the overall mass transfer coefficient is  $15 \text{ h}^{-1}$ , determine the height of the column for an effluent antibiotic concentration of  $C_L = 0.2 \text{ g/l}$ . Assume an equilibrium relationship of

$$C_S = 25(C_L^*)^{1/2}$$

where  $C_S$  is g solute/l resin and  $C_L$  is g solute/l solution. The ratio of volumetric flow rate of broth to resin is 10.