



With the definitions of

$$K_{S_1} = \frac{[S][ES]}{[ES_2]}, \quad K'_m = \frac{[S][E]}{[ES]} \quad (3.33)$$

the assumption of rapid equilibrium yields

$$v = \frac{V_m[S]}{K'_m + [S] + \frac{[S]^2}{K_{S_1}}} \quad (3.34)$$

A double-reciprocal plot describing substrate inhibition is given in Fig. 3.10.

At low substrate concentrations,  $[S]^2/K_{S_1} \ll 1$ , and inhibition effect is not observed. The rate is

$$v = \frac{V_m}{\left[1 + \frac{K'_m}{[S]}\right]} \quad (3.35)$$

or

$$\frac{1}{v} = \frac{1}{V_m} + \frac{K'_m}{V_m} \frac{1}{[S]} \quad (3.36)$$

A plot of  $1/v$  versus  $1/[S]$  results in a line of slope  $K'_m/V_m$  and intercept of  $1/V_m$ .

At high substrate concentrations,  $K'_m/[S] \ll 1$ , and inhibition is dominant. The rate in this case is

$$v = \frac{V_m}{\left(1 + \frac{[S]}{K_{S_1}}\right)} \quad (3.37)$$

or

$$\frac{1}{v} = \frac{1}{V_m} + \frac{[S]}{K_{S_1} V_m} \quad (3.38)$$

A plot of  $1/v$  versus  $[S]$  results in a line of slope  $1/K_{S_1} \cdot V_m$  and intercept of  $1/V_m$ .