INDIAN INSTITUTE OF TECHNOLOGY, KANPUR LABORATORY REPORT SHEET Name Name of Lab Partner: DHEERAJ KUMAR Subject : PHY461 Date of Experiment Roll No : 210386 Date of Submission 22.10.24 Instructor : Prof. Jayita Nayak 29.10.24 Experiments : Phase Transition in Barium Titanate Remarks by the Instructor Phase Transition in Barium Titanate (BaTiO3) Aim: 1) To study the temperature dependence of a ceramic disc 2) To study the temperature dependence of dielectric constant at 100 Hz and to determine phase transition of BaTiO3. Apparatus: LCR Bridge, Fwinace, Sample, thermocouple and Theory: Let us take a capacitor of any arbitrary shape and dimensions with vacuum in the space between the electrodes (vacuum capacitor) and denotes its capacitance by co. If now inter-electrode space is filled by a material having permittivity  $\varepsilon$ , the new capacitance c is given by  $c = \varepsilon c_0$ In the case of a parallel plate capacitor eq.(1) becomes  $C = \underbrace{\epsilon \, \epsilon_0 \, A}_{d}, \quad \epsilon_0 = 8.854 \times 10^{-12} \, \text{F/m} \quad \text{is the permittivity of}_{d}$ free space. A is the area of plates, d is the separation between the plates. E is the dimensionless parameter which characterizes the dielectric and is called the relative permittivity or dielectric constant. Dielectrics are sub divided into two classes

a) Polar dielectrics - composed of molecules having permanent dipole moments.

b) Non-polar dielectrics - do not have molecules containing permanent dipole moments. A further classification is done as linear and non-linear dielectrics. Linear dielectrics are characterized by the fact that their polarization P and displacement D are directly proportional to the intensity of electric field  $\epsilon$ . P = Ke Eo E , where ke Eo is absolute dielectric susceptibility. D = 880 E E = 1+Ke

for the case of non-linear dielectrics, there is no proportionality between P. & D. Ferencelectrics (FE) are most typical non linear dielectries. The most essential features are:

a hysterisis D-E loop akin to B-H loop of ferromagnetic naterials

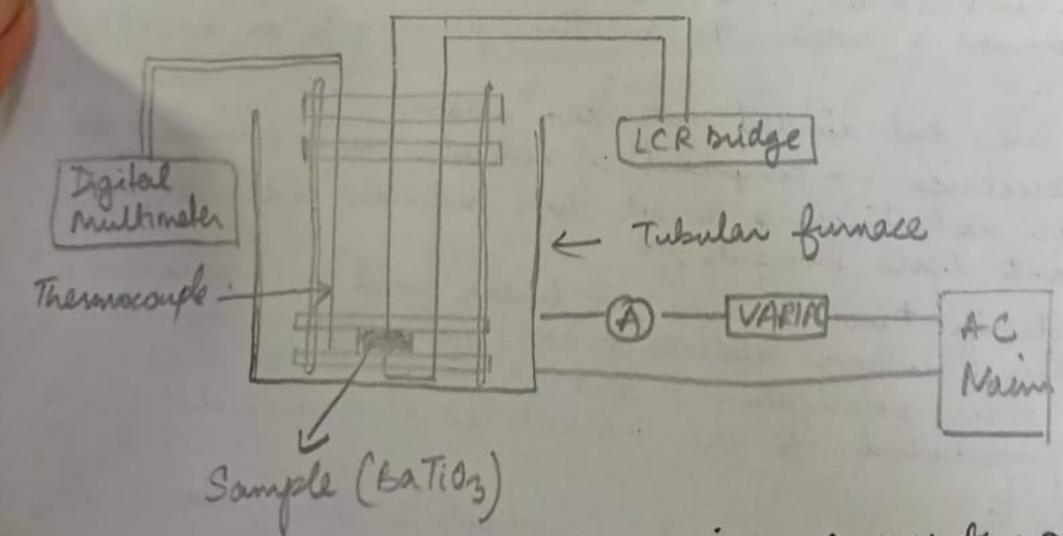
2) Usually a high value of permittivity.
3) Pronounced dependence of dielectric parameters on temperature.
4) Presence of a spontaneous polarization without an external electric field.

Barium Titanate (BaTiO3) is one of the important FE materials and it displays FE properties also in poly crystalline formations obtained as products of required shape and dimensions by ceramic - process nethods. The tempelature dependence of permittivity and some other physical properties of FE are characterized by the presence of transition points called "Curie points" at which the magnitude of permittivity is maximum, and substance acquires or loses FE properties when temperature passes through the curie point. The phase above the curie point is called "Paraclectric (PE) phase " and the phase below it is called the FE phase. Therefore PE & FE phase transition (PT) occurs at the curie point. The permittivity near the PT and above the curie point is governed by Curie - Weiss Law:

E = C, where C is the Curie - Weiss constant and To is the (T-To) curie - Weiss temperature close to the Eurie point.

The capacitance can be measured directly with the help of LCR bridge. The total ownert I through the capacitor is resolved into the two components active current Ia and reactive current  $I_{S}$ .  $I_{A}$  is in phase with V and  $I_{S}$  leads by  $\pi/2$ . The angle S is called the dielectric loss angle .  $S = \frac{1}{\tan S}$  (Quality factor of the dielectric)

Systematic experimental setup:



Dinensions of Pellet: 2r = (10.7 ±0.4) mm d = 3.1 to.1 mm

Observa for frequ

Tempera (oc)

> 29° 35

50 60

60

For Batio3

Frequency = 100 Hz Voffset = 1.1220 - 0.00 = 1.1220

Room temperature = 28°C

Temporature	corresponding			10.00	menula T
	Voltage (Vie - Voffset)	(NF)		Heat cool	
28	(mo)	Heat	cool	Heat Co	
35	0	1.360	-	0.061	=
40	0.29	1.378	0.0658	0.095	1.848
	0.49	1.382	0.0617	0.102	1.629
45 50	0.70	1.394	0.0582	0.102	1.750
	0.90	1.404	0.0612	0.1.05	2.216
55	1.11			0.107	2.730
60	1.31		.6625	0.111	2.778
65	1.52	1.438	0.086	0.112	2.709
70	1.73	1.460	0.109	0.118	2.550
75	1.94	1.489	0.145	1	
80	2.14	0.0650	0.202	2.515	2.233
85	2.35	0.0824	0.297		1.948
90	2.56	0.1237	0.399		1.670
95	2.77	0.1979			7 0. 892
100	2.97	0.3659	1.229	1.961	6.691
105	3.18	0.4928	1.433	1.552	6.590
110	3.39	0.7269	1.634	1.245	
115	3.59	0.9934		1.000	
120	3.80	1.334	2.251	0.85	1 0.407
125	4.00	1.690	2.756	0.774	0.392
	4.21	2.239	3.314	0.683	6.340
130		The state of the s	Part Sale	0.517	0.265
135	4.41	2.704	3-131	0.387	0.228
140	4.61	2.706	2.890		0.206
145	4.81	2.601	2.699	0.303	A 100
150	5.02	2.457	2,534	0.247	- 120
155	5. 22	2.322	2.370	0.211	
160	5.42	2.193	2.178	0.187	0.165

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## LABORATORY REPORT SHEET

Name

Name of Lab Partner

Section

Date of Experiment

Date of Submission:

Experiments

Subject Roll No

Instructor

Remarks by the Instructor

Results and Analysis

\* For the case of the commercial capacitor, we can see that the capacitance decreases with temperature. We have fitted a covere with the data observed and attached it along with the report.

For BaTiO3, if we consider the heating sure part, phase transition happens at T=80°C. While for the Cooling part phase transition happens at T=90°C.

Error Analysis We know that capacitance  $C = \frac{AE}{d}$ . =  $\frac{2\pi Y^2}{d}E$ 

Given,  $2\gamma = (10.7 \pm 0.4)$  mm ,  $d = 3.1 \pm 0.1$  mm

At room temperature, for BaTiO3, C= 1.360 nF  $\Rightarrow \varepsilon = 1.360 \times 10^{-9} \times 3.1 \times 10^{-3} = 5.86 \times 10^{-9} \text{ F/m}$ 2 x x x (10.7)2 x 156

 $\varepsilon_r = \frac{\varepsilon_0}{\varepsilon} = \frac{8.85 \times 10^{-12}}{5} = 1.51 \times 10^{-3}$ 5.86 × 10-9

 $\Rightarrow \Delta E = \left(2 \times \frac{0.4}{10.7} + \frac{0.1}{3.1}\right) \times 5.86 \times 10^{-9} = 6.27 \times 10^{-10} \text{ F/m}$ 80 € = (5.86 € ± 0.63) × 109 F/m

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emperature	conesporolung		
(°c)	Thermo-couple	@ C	)
29°	Voltage (V-Voffset)	Hent 1	and Heat
	0	6.094	0.013
35	0.24	6.143	0.013
40	0.45	6.246	0.014
45	0.65	6.131	0.013
50	0.86	5.895	0.012
56	1.07	5.597	0.012
60	1.27	5.298	0.011
65	1.48	4.948	0.011
70	1.69	4.589	0.011
75	1.90	4.219	0.010
80	2001 2.104	3.883	0.010
85	2052 2.320	3.556	0.010
90 .	2.52	3.257	0.011
95	2,73 2000	2.951	0.012
[00]	2.73 3293	2.639	0.013
105		2.486	6.015
110	3.14 32.05	2.270	0.019
100 115	3.55	2.080	0.024
120	3.76	1.920	0.030
100125	3.96	1.785	0.038
130	4.17	1.650	0.050
135	4.37	1.551	0.063
140	4.57	1.461	0.079
145	4.77	1.429	
150	4.98	1.383	3 4000 0.099

\* We could not get the data for cooling as the heating part took a lot of time and not enough time was left to take the cooling data on the same day.

.1 mm

Solutions to Lab Manual Questions

1. Dielectricis -> They are insulating materials that do not conduct electricity but can be polarized in an electric dipoles. They store electrical energy in the form of electric Ferro electrics - They are materials that exhibit spontaneous polarizations. They have permanent electric dipole moment even in the absence of an electric field. Example - BaTiD3 Paraelectrics -> They are materials that exhibit a weak polarization that can be induced by an external electric Ex. Sr TiO3 (Strontium Titanate)

2. The DC walne of dielectric constant is measured under a steady electric field. It's polarization in this case is mainly due to the displacement of charge carriere and the alignment of dipoles. The AC value of dielectric constant is measured under an alternating electric field. As the frequency increases, the

dielectric material may not have enough time to respond to the rapid changes in the electric field. affecting the

As frequency increases, the dielectric constant typically decreases leading to decrease in capacitance.

3. 1st order phase transition - characterized by a discontinuity in the first - derivative of the free energy with respect to temperature and pressure. Eg. Melting of ice

Il order phase transition -> No discontinuity in first derivative.
However higher derivatives show discontinuity.

BaTiO3 is a first order transition

Polar Dielectries - Have permanent dipoles spontaneous polarization. Ex -> BaTiO3 and exhibit Non Polan Dialectrice -> Lack Permanent dipoles and become polarized only in an external field.

- \* Polan dielectrices can maintain polarization without an external field while non-polar dielectrics require an electric field to exhibit polarization.
- 5. In this Equation E: Real part of the permittivity representing stored energy. E": imaginary part of the permittivity representing energy lost (dielectric loss)

At low frequency, polymer can exhibit high dielectric constant due to dipolar polarization. At high frequency two dipoles may not have enough time to align with field leading to decrease in E', and an increase in E', and an increase in E".

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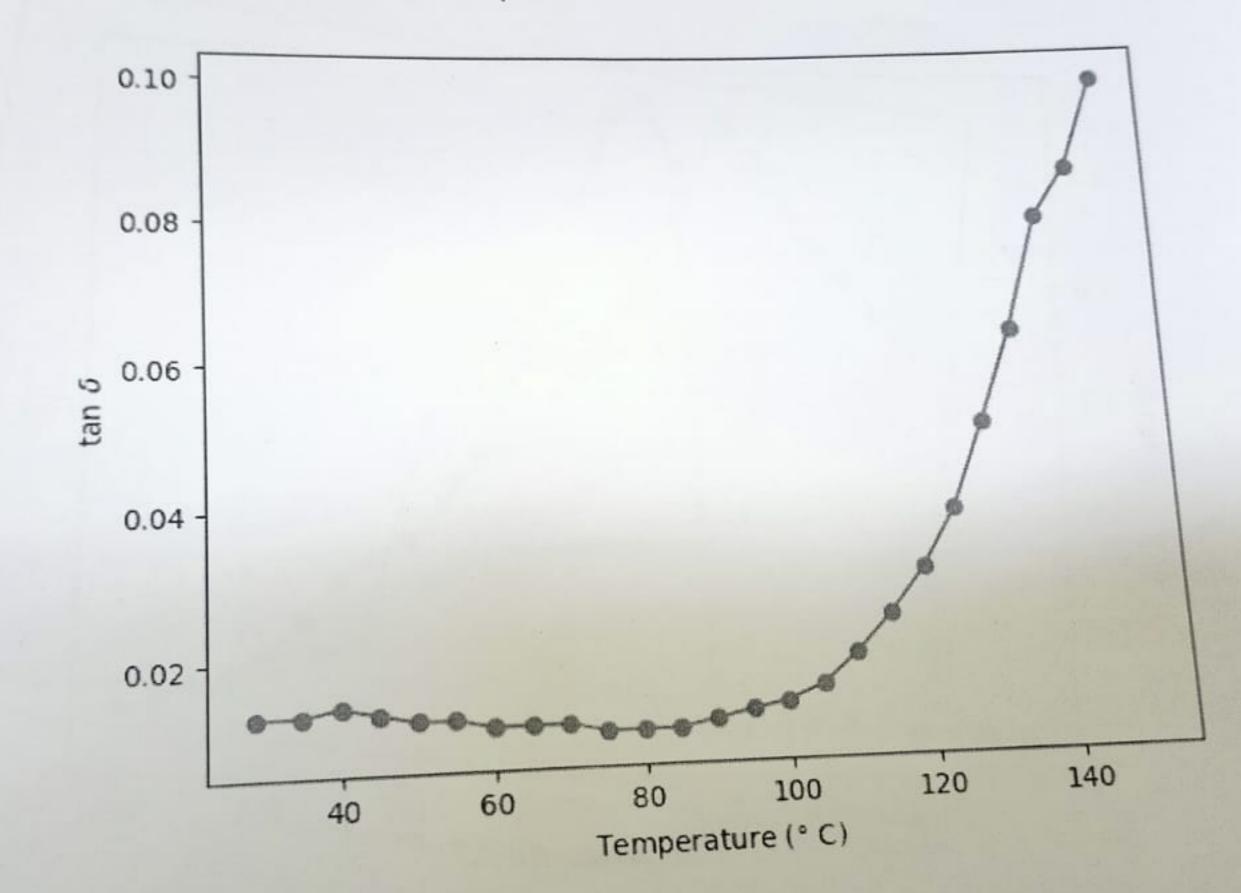
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Curie's constant =  $\mathcal{E}(T-T_0)$   $T_0 = 80^{\circ} \text{ C}$  (for heating part)  $\Rightarrow C = 5.86 \times 10^{-9} (T-80)$ 

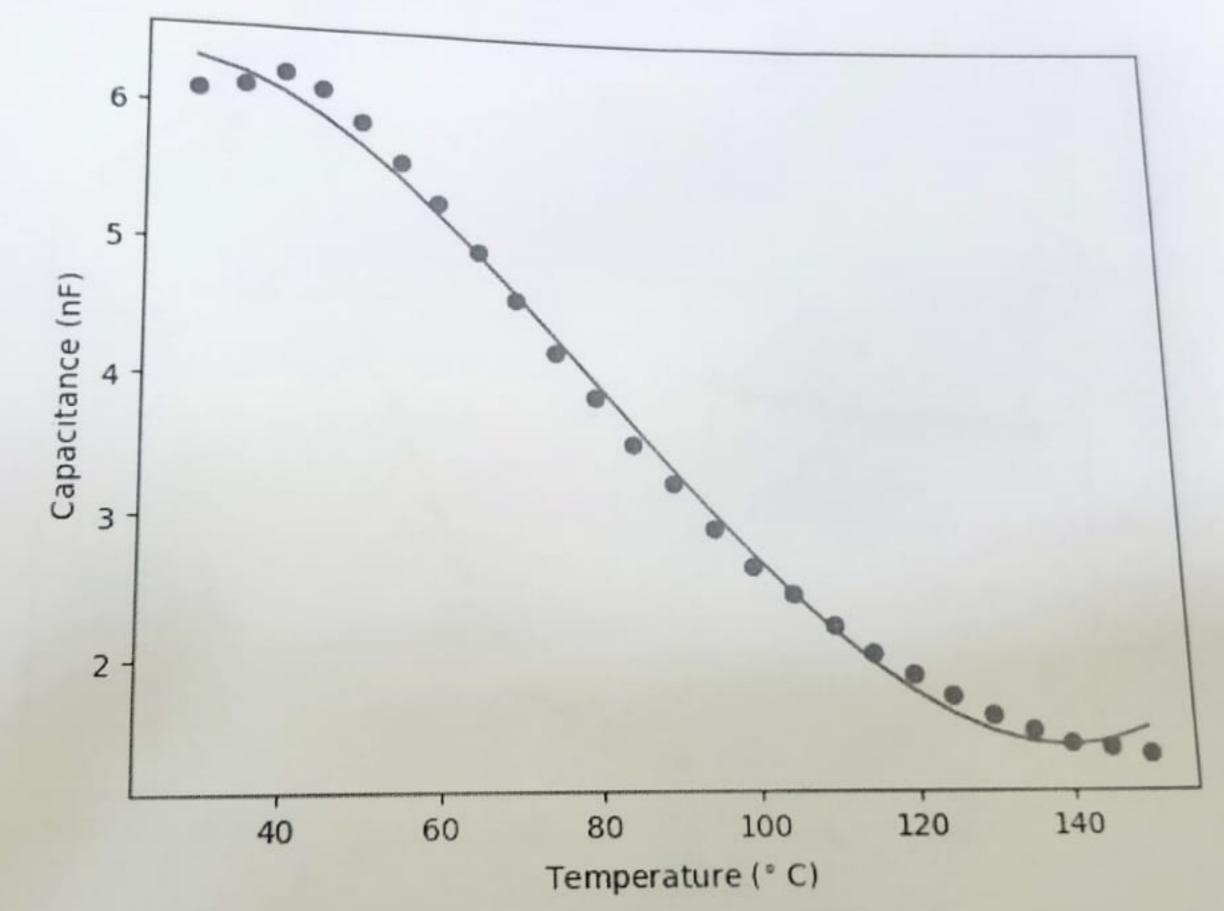
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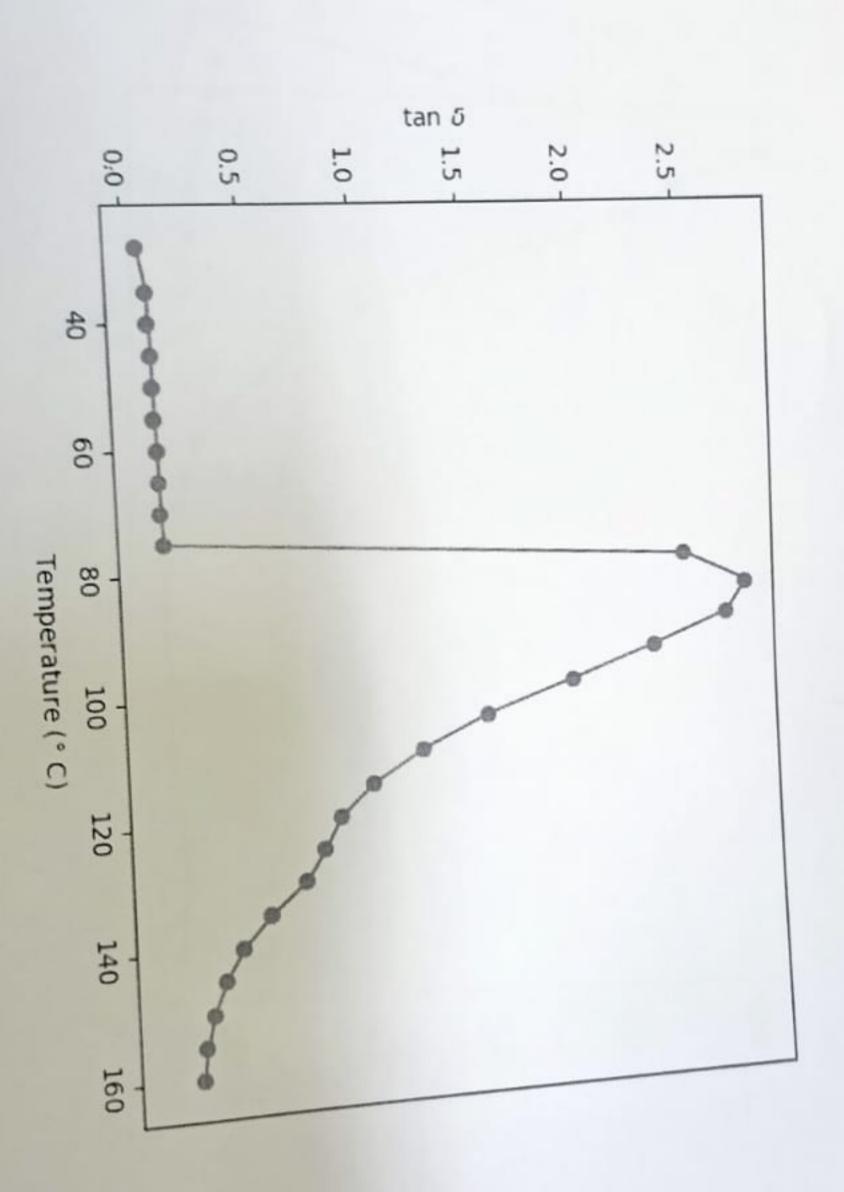
. To the two comp. Thus the angle loss

For commercial capacitor Freq = 100 HZ



For commercial capacitor Freq = 100 HZ





Freq = 100 Hz
Freq = 100 Hz
During Healing

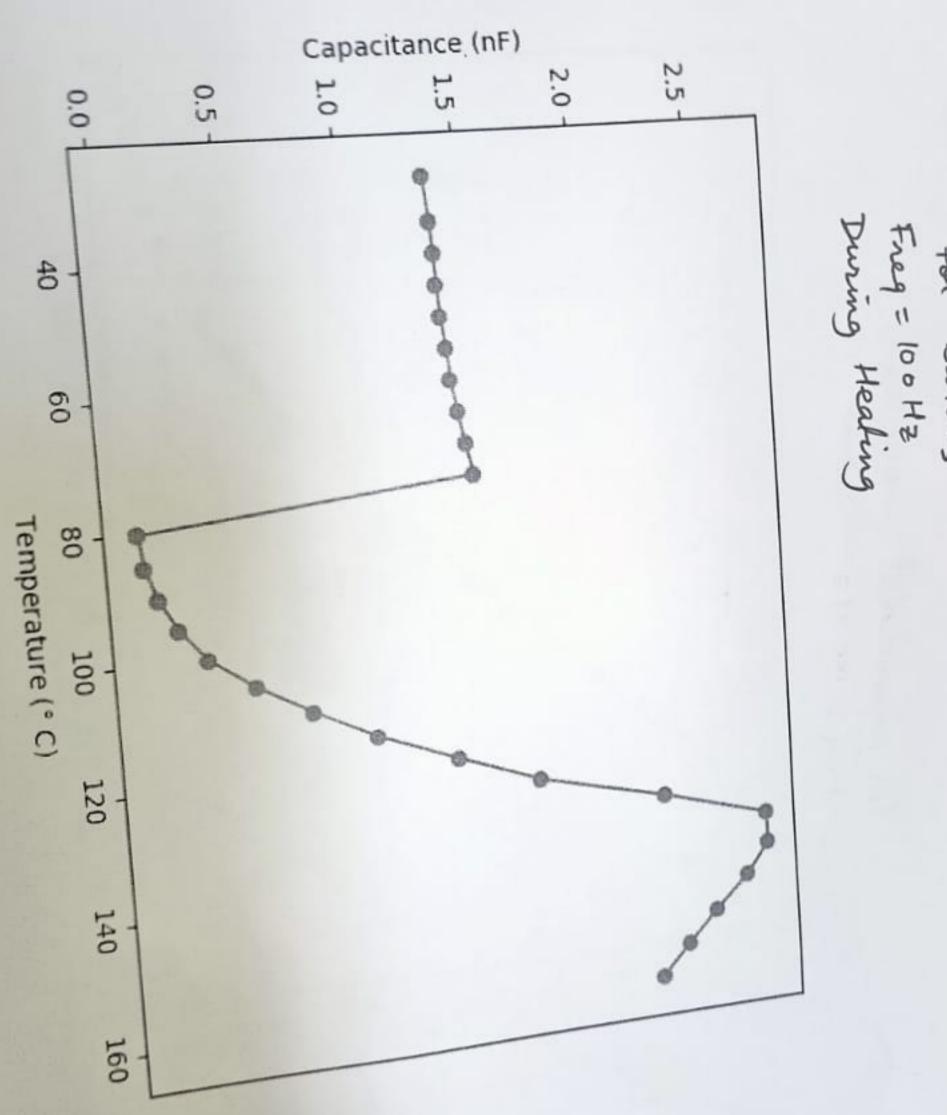
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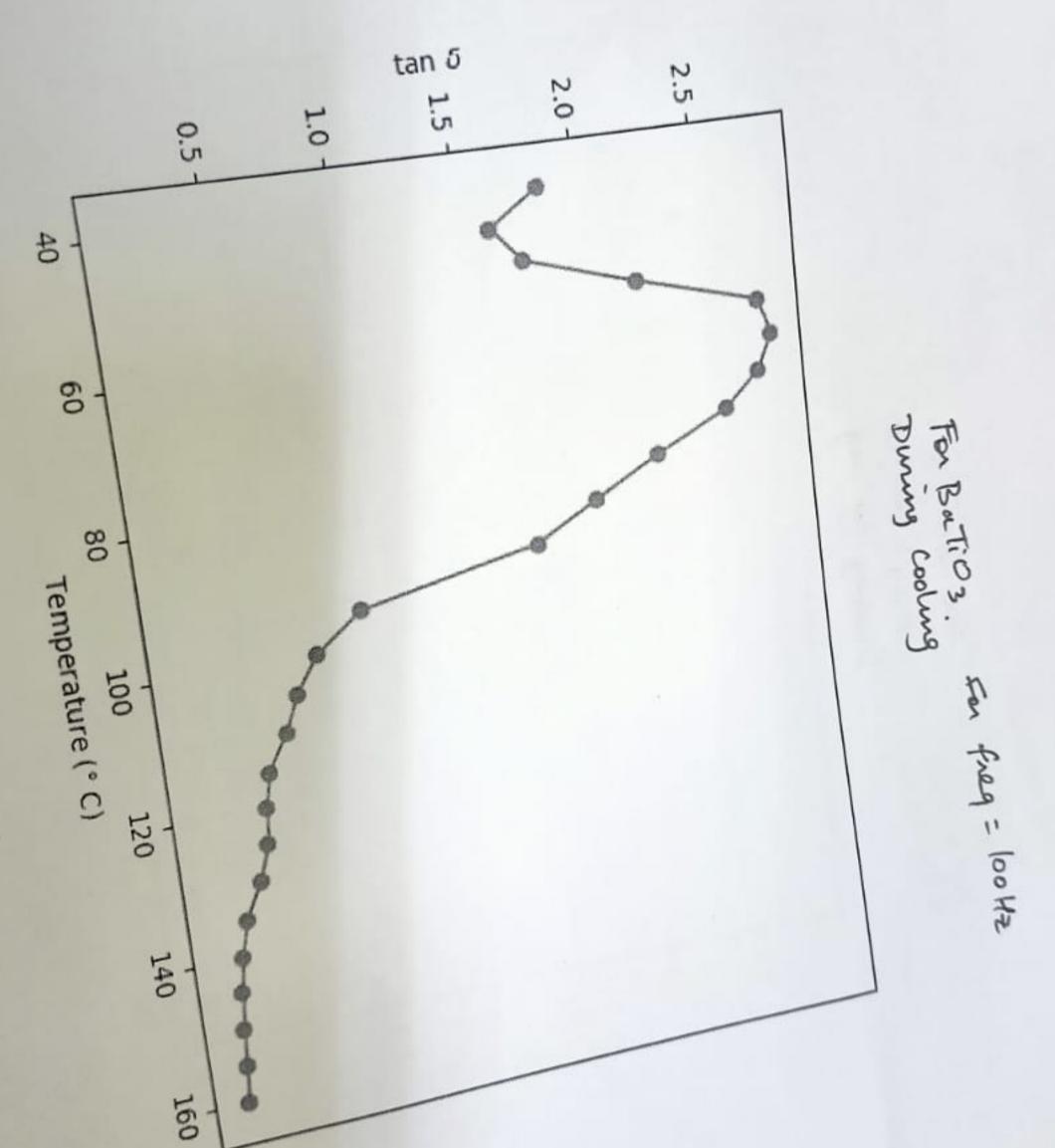
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