

Balancing a Chemical Equation



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Abstract—This manual shows how to balance chemical equations using matrices.

Download python codes using

svn co https://github.com/gadepall/school/trunk/ training

1 CHEMISTRY

1. Express the problem of balancing the following chemical equation as a matrix equation.

$$Fe + H_2O \rightarrow Fe_3O_4 + H_2$$
 (1.1.1)

Solution: Let the balanced version of (1.1.1) be

$$x_1Fe + x_2H_2O \rightarrow x_3Fe_3O_4 + x_4H_2$$
 (1.1.2)

which results in the following equations

$$(x_1 - 3x_3) Fe = 0$$

$$(2x_2 - 2x_4) H = 0$$

$$(x_2 - 4x_3) O = 0$$
(1.1.3)

which can be expressed as

$$x_1 + 0.x_2 - 3x_3 + 0.x_4 = 0$$

$$0.x_1 + 2x_2 + 0.x_3 - 2x_4 = 0$$

$$0.x_1 + x_2 - 4x_3 + 0.x_4 = 0$$
(1.1.4)

resulting in the matrix equation

$$\begin{pmatrix} 1 & 0 & -3 & 0 \\ 0 & 2 & 0 & -2 \\ 0 & 1 & -4 & 0 \end{pmatrix} \mathbf{x} = \mathbf{0}$$
 (1.1.5)

where

$$\mathbf{x} = \begin{pmatrix} x_1 \\ x_2 \\ x_3 \\ x_4 \end{pmatrix} \tag{1.1.6}$$

2. Solve (1.1.2) by row reducing the matrix in (1.1.5).

Solution: (1.1.5) can be row reduced as follows

$$\begin{pmatrix}
1 & 0 & -3 & 0 \\
0 & 2 & 0 & -2 \\
0 & 1 & -4 & 0
\end{pmatrix}
\xrightarrow{R_2 \leftarrow \frac{R_2}{2}}
\begin{pmatrix}
1 & 0 & -3 & 0 \\
0 & 1 & 0 & -1 \\
0 & 1 & -4 & 0
\end{pmatrix}$$

$$\xrightarrow{R_3 \leftarrow R_3 - R_2}
\begin{pmatrix}
1 & 0 & -3 & 0 \\
0 & 1 & 0 & -1 \\
0 & 0 & -4 & 1
\end{pmatrix}$$

$$\xrightarrow{R_1 \leftarrow 4R_1 - 3R_3}
\begin{pmatrix}
4 & 0 & 0 & -3 \\
0 & 1 & 0 & -1 \\
0 & 0 & -4 & 1
\end{pmatrix}$$

$$\xrightarrow{R_1 \leftarrow \frac{1}{4}}
\xrightarrow{R_3 \leftarrow -\frac{1}{4}R_3}
\begin{pmatrix}
1 & 0 & 0 & -\frac{3}{4} \\
0 & 1 & 0 & -1 \\
0 & 0 & 1 & -\frac{1}{4}
\end{pmatrix}$$

$$\xrightarrow{R_1 \leftarrow \frac{1}{4}}$$

$$\xrightarrow{R_3 \leftarrow -\frac{1}{4}R_3}
\begin{pmatrix}
1 & 0 & 0 & -\frac{3}{4} \\
0 & 1 & 0 & -1 \\
0 & 0 & 1 & -\frac{1}{4}
\end{pmatrix}$$

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Thus,

$$x_1 = \frac{3}{4}x_4, x_2 = x_4, x_3 = \frac{1}{4}x_4$$
 (1.2.5) (1.2.6)

$$\implies \mathbf{x} = x_4 \begin{pmatrix} \frac{3}{4} \\ 1 \\ \frac{1}{4} \\ 1 \end{pmatrix} = \begin{pmatrix} 3 \\ 4 \\ 1 \\ 4 \end{pmatrix} \tag{1.2.7}$$

upon substituting $x_4 = 4$. (1.1.2) then becomes

$$3Fe + 4H_2O \rightarrow Fe_3O_4 + 4H_2$$
 (1.2.8)

3. Verify your answer through a python code. **Solution:** Execute

codes/chembal.py

2 MATHEMATICS

1. Find the equation of the plane P that contains the point $\begin{pmatrix} 1 \\ -1 \\ 2 \end{pmatrix}$ and is perpendicular to each of the planes

$$P_1: (2 \ 3 \ -2)\mathbf{x} = 5$$
 (2.1.1)

$$P_2: (1 \ 2 \ -3)\mathbf{x} = 8$$
 (2.1.2)

From (2.1.1), the normals to P_1, P_2 are

$$\mathbf{n}_1 = \begin{pmatrix} 2 & 3 & -2 \end{pmatrix}$$

$$\mathbf{n}_2 = \begin{pmatrix} 1 & 2 & -3 \end{pmatrix}$$
(2.1.3)

 $P \perp P_1, P \perp P_2$, if **n** be the normal to P, **n** \perp **n**₁, **n** \perp **n**₂, which can be expressed using (2.1.3) as

$$\begin{pmatrix} 2 & 3 & -2 \\ 1 & 2 & -3 \end{pmatrix} \mathbf{n} = 0 \tag{2.1.4}$$

Obtain **n** using row reduction.

- 2. Verify your answer through a python code.
- 3. Verify that $\mathbf{n} = \mathbf{n}_1 \times \mathbf{n}_2$.

3 Physics

- 1. A force of $\mathbf{F} = 7\hat{i} + 3\hat{j} 5\hat{k}$ acts on a particle whose position vector is $\mathbf{r} = \hat{i} \hat{j} + \hat{k}$. Find the torque about the origin given by $\mathbf{F} \times \mathbf{r}$ using a matrix equation.
- 2. Verify your answer using python.