

Question 33**(12 marks)**

Organic molecules have a hydrocarbon skeleton and can contain functional groups that are responsible for the molecules' characteristic chemical properties.

Complete the following tables by

- (i) writing the structural formula of each compound listed
- (ii) writing the structural formula of the organic product from the reaction
- (iii) naming the organic product from the reaction.

When writing the structural formula, show the bonds between carbon atoms and within any functional group e.g. $\text{CH}_3\text{—CH}_2\text{—}\overset{\text{O}}{\underset{\text{||}}{\text{C}}}\text{—CH}_3$

| Name of compound | | Structural formula of compound |
|--------------------------------------|---------------------------------------|--------------------------------|
| pent-2-ene | | |
| Reacts with $\text{Br}_2(\text{aq})$ | Structural formula of organic product | |
| | Name of organic product | |

| Name of compound | | Structural formula of compound |
|---|--|--------------------------------|
| ethanal | | |
| Reacts with $\text{KMnO}_4(\text{aq}) / \text{H}^+(\text{aq})$ | Structural formula of organic product | |
| | Name of organic product | |

| Name of compound | | Structural formula of compound |
|--|--|--------------------------------|
| butanoic acid | | |
| Reacts with $\text{Na}_2\text{CO}_3(\text{aq})$ | Structural formula of organic product | |
| | Name of organic product | |