

When a stationary neutron decays into a proton, an electron and an electron anti-neutrino are also produced. Total energy is conserved during the decay process. The reaction is described by the following equation:



- (a)
- In the reaction, the quark composition of a neutron changes from udd to uud. Show how the reaction conserves both baryon number and lepton number by filling in the table below.
- (6 marks)

	n^0	\rightarrow	p^+	+	e^-	+	$\bar{\nu}_e$
Baryon number		\rightarrow		+		+	
Lepton number		\rightarrow		+		+	

- (b)
- The mass of a stationary neutron is 1.675×10^{-27} kg. The mass of a proton is 1.673×10^{-27} kg. The mass of an electron is 9.109×10^{-31} kg. If we assume the total energy of the anti-neutrino is 0 J, calculate the total kinetic energy of the particles emitted in keV.
- (5 marks)

- (c) If the electron accounts for 90.0% of the kinetic energy produced, calculate the velocity of the emitted proton in terms of c . If you could not determine an answer for part (b), use 581 keV (9.30×10^{-14} J). (4 marks)

Answer: _____ c