Identify the oxidant in the following equation. 5.

$$2 \ BrO_{_3}^{\ -}(aq) + 10 \ I^-(aq) + 12 \ H^+(aq) \rightarrow Br_{_2}(aq) + 5 \ I_{_2}(aq) + 6 \ H_{_2}O(\ell)$$

- BrO₃⁻(aq) I⁻(aq) (a)
- (b)
- H⁺(aq) (c)
- Br₂(aq) (d)