15. The reaction of aniline (C₆H₅NH₂) with water is an equilibrium process:

$$C_6H_5NH_2(\ell) + H_2O(\ell) \rightleftharpoons C_6H_5NH^-(aq) + H_3O^+(aq)$$

A conjugate acid-base pair in this process is

- C₆H₅NH⁻ and H₂O (a)
- $C_6^6H_5^5NH_2$ and $C_6^2H_5NH_3^ C_6^6H_5^5NH_3^-$ and $H_3^3O_3^+$ $H_3^3O_3^+$ and $H_3^3O_3^+$ (b)
- (c)
- (d)