8. Consider the hydrolysis equation below.

$$CH_3NH_2(aq) + H_2O(\ell) \rightleftharpoons CH_3NH_3^+(aq) + OH^-(aq)$$

Which of the following are conjugate acid-base pairs?

- (i) CH_3NH_2 and H_2O
- (ii) CH₃NH₂ and CH₃NH₃⁺
- (iii) H₂O and OH-
- (iv) CH₃NH₃+ and OH-
- (a) ii only
- (b) ii and iii only
- (c) i and iv only
- (d) i, ii, iii and iv