

After a while, the following equilibrium is established inside Chamber 1:

$$2 \text{ NO}_2(g) \rightleftharpoons \text{N}_2\text{O}_4(g) + 58 \text{ kJ mol}^{-1}$$
reddish colourless
brown

25. The valve between the chambers is opened, allowing the gas mixture in Chamber 1 to fill both chambers. Which statement describes the initial and final observations of the gas mixture's colour as a new equilibrium is established?

	Initial observation when tap is opened	Observation as equilibrium is re-established
(a)	paler brown	became darker
(b)	paler brown	became paler
(c)	darker brown	became darker
(d)	darker brown	became paler