20. Ethanol can also be produced by reacting ethene with steam. The equation for this reaction, which is known as hydration, is shown below.

$$CH_2 = CH_2(g) + H_2O(g) \rightarrow CH_3CH_2OH(g) \qquad \Delta H = -45 \text{ kJ mol}^{-1}$$

Which statement about this reaction is **incorrect**?

- (a) The hydration reaction requires higher pressures than fermentation to achieve an economically-viable rate.
- (b) The hydration of ethene is an addition reaction.
- (c) Enzymes decrease the activation energies of both the hydration and fermentation reactions.
- (d) The hydration reaction requires higher temperatures than fermentation to achieve an economically-viable rate.