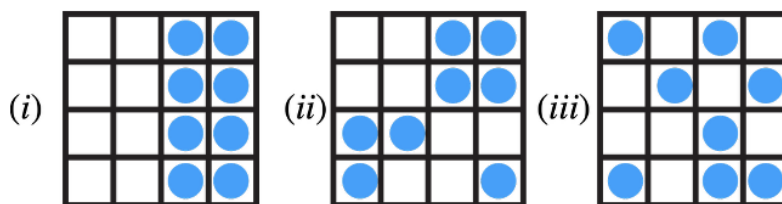


MSEN 640 Fall 2025 – Homework # 5

Instructions: Please answer the problems below either using a computer, tablet, or paper. Save your solutions as a PDF document; remember: if using paper, please **scan** your answers! Remember to **show all of your work** for a full grade.

1.
 - a. (5 points) How many possible microstates exist for a system containing eight atoms of element A and one atom of element B? Draw all possible microstates for such a system on a three-by-three two-dimensional lattice.
 - b. (5 points) How many possible microstates exist for a system containing four atoms of element A, two atoms of element B, and three atoms of element C?
 - c. (5 points) The following lattices show microstates of a 4x4 lattice. Which is the most probable microstate consistent with the macrostate of being 50% occupied. Explain your answer.



2. Consider a small colloidal glass bead of mass m in a container of water held at temperature T . The gravitational potential energy U of the glass particle depends on its height above the bottom of the container z : $U(z) = mgz$, where g is the acceleration due to gravity.

(a) Determine the probability $P(z)$ that the bead is found at a certain height z . Normalize the probability distribution such that $\int_0^\infty P(z) dz = 1$.

(b) Calculate the average height of the bead $\langle z \rangle$. Under what conditions would we expect particles to “sediment” out of a solution?

(c) In the high temperature limit (or $T \rightarrow \infty$), where can we expect to find the bead?

(d) Calculate the variance of the bead height $\langle z^2 \rangle - \langle z \rangle^2$.

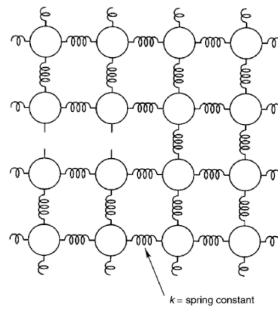
(e) Given what we have learned in part (d), re-interpret your answer to part (c). That is, at any instant in time, where might the bead be found at high temperatures?

(f) Calculate the average energy $\langle U \rangle$

3. One simple model of a crystal consists of a collection of masses and springs. Consider one particular “normal mode” of the crystal, characterized by natural frequency ω . The vibrational energy associated with this normal mode can be quantized such that the n^{th} energy level is given by

$$\epsilon_n = \left(n + \frac{1}{2} \right) \hbar \omega$$

where $n = 0, 1, 2, \dots$ and \hbar is Planck's constant divided by 2π . This is called the “Einstein model” of a crystalline solid.



(a) Calculate the partition function Z . To simplify, take advantage of the formula for a convergent geometric series:

$$\sum_{n=0}^{\infty} x^n = \frac{1}{1-x}$$

(b) Calculate the entropy, internal energy, and Helmholtz free energy of the crystal. You can express your results in terms of the “Einstein temperature” $\vartheta_E \equiv \hbar\omega/k_B$. (Note that for diamonds, $\vartheta_E \approx 1320$ K.)

(c) Calculate the molar heat capacity C_V . Plot C_V/k_B as a function of T/ϑ_E . (This model actually does a good job at capturing the high-temperature behavior of real crystals! The low-temperature requires some corrections, however.)