CLL371 | Chemical Process Technology and Economics

NH₃ PRODUCTION FROM H₂ AND N₂

Presentation by

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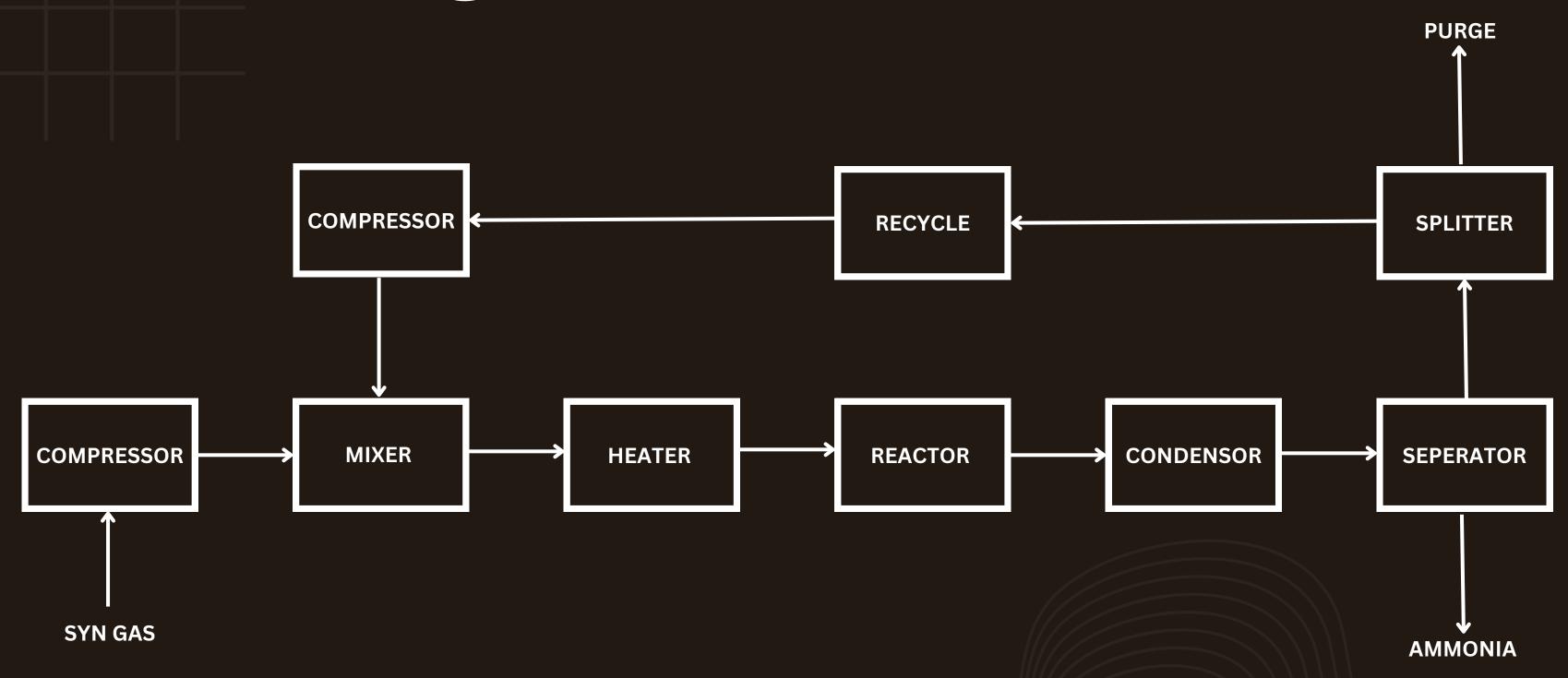
Avik Ghosh 2020CH10081

Aryaman Das 2020CH10077

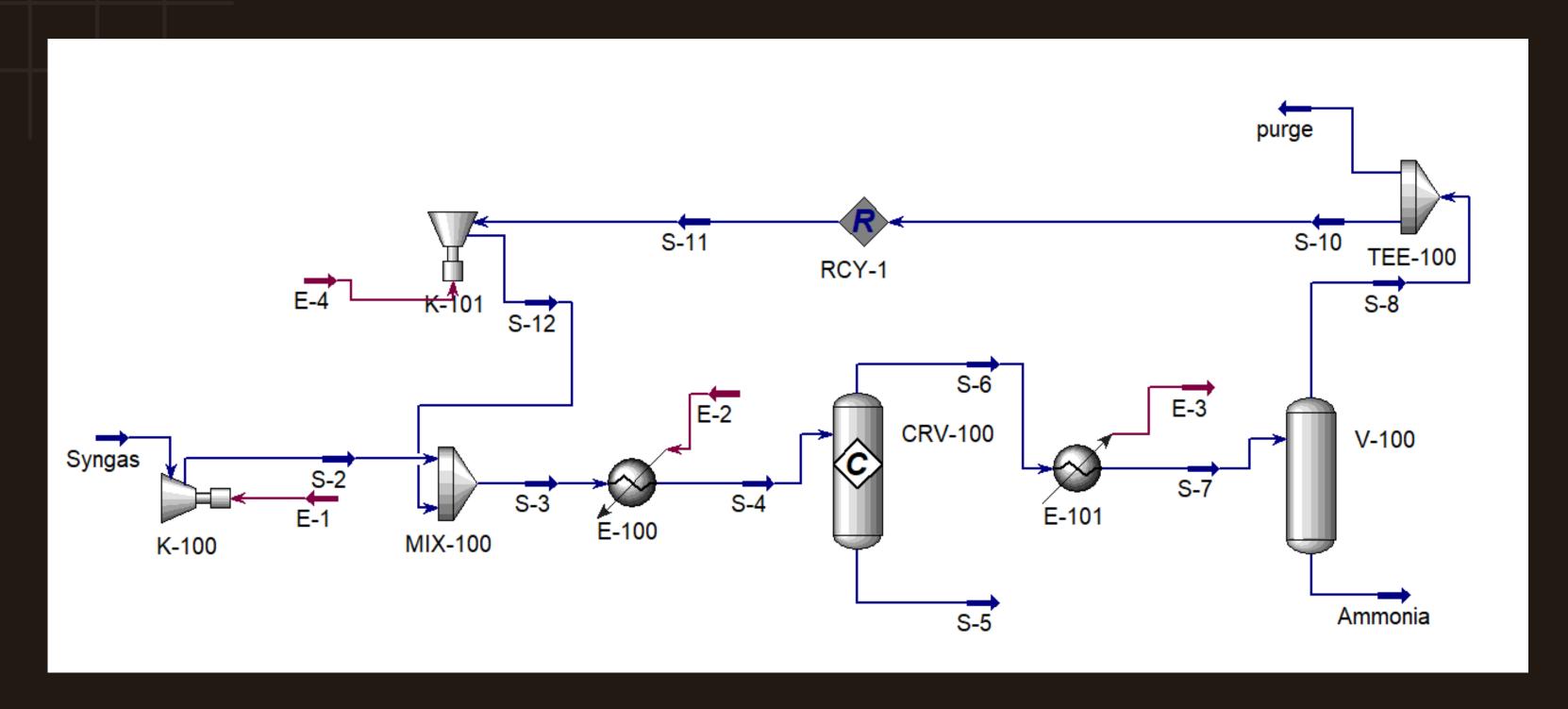
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Block Diagram



Process Flow Diagram



Reaction Involved

Syn-Gas Conditions (Feed Conditions)

- Temperature = 280 C
- Pressure = 25.5 bar
- → Molar Flow Rate = 7000 kgmol/h

MOLAR CONCENTRATIONS

$$+$$
 N2 = 0.2474

$$+$$
 H2 = 0.7372

$$+$$
 CO = 0.0024

$$+$$
 Ar = 0.0027

$$+$$
 CH4 = 0.0103

Equipment Specifications

Compressor 1

Delta P = 249.5 bar Adibatic efficiency = 75%

Heater

Temperature = 500 C Pressure drop = 0.1 bar

Conversion reactor

Conversion = 40% Base component : Nitrogen

Cooler

Temperature = 10 C Pressure drop 100bar

Purge rate

1% of S-8 Steam

Compressor 2

Delta P = 100.1 bar Adiabatic Efficiency = 75%



Stream Conditions

Strea m No.	Stream Description	Contents	Temp.	Pressure (bar)	Mass flow rate (kg/hr)
1	Gas feed to compressor	N2 ,H2, CO, Ar, NH3, CH4	280	25.5	61300
2	Compressed stream to mixer	N2, H2, other	948.7	275	61300
3	Feed to heater	N2, H2, other	948.7	275	61300
4	Heated feed to ammonia reactor	N2, H2, other	500	274.9	61300
5	bottom liquid stream from reactor	N2, H2, NH3	856.6	274.9	0
6	heated feed from Reactor to cooler	N2, H2, NH3, other	856.6	274.9	61300
7	feed to separator	N2, H2, NH3, other	10	174.9	61300
8	Top product stream from separator	N2, H2, other	10	174.9	42640
9	Bottom product ammonia	NH3, other	10	174.9	59540
10	Input feed to Recycler R	N2, H2	10	174.9	42214
11	Recycle stream	N2, H2	10	174.9	42214
12	Outlet compressed stream to mixer	N2, H2	61.91	275	42214



Thank You!