

## WORK INSTRUCTION

## Production Process Control Testing : Chlorine Test (Titration)

Purposes/Function/Objective

- To determine percentage of chlorine in chlorine tank

Materials/ Chemicals/Tools/Equipment

- 1) Potassium Iodide
- 2) 0.1 M Sodium Thiosulphate
- 3) 25 ml Pipette
- 4) 100 ml conical flask

Specification

calculation:-

$$\% \text{ Chlorine ppm} = \frac{V}{V \times 0.0142 \times 10000}$$

Where is:

V=Volume of  $\text{Na}_2\text{SO}_3$  solution required for titration ml

Form

- LA/F02

References :

- nil

Procedures

Apparatus use for Chlorine test



Weigh 0.5g potassium iodide into conical flask.



Add (pipette) 25ml chlorine sample into the conical flask



Titrate with 0.1M of Sodium Thiosulphate ( $\text{Na}_2\text{SO}_3$ )



End Point: Brown to colorless

Calculation :

$$\% \text{ Chlorine ppm} = \frac{V}{V \times 0.0142 \times 10000}$$

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