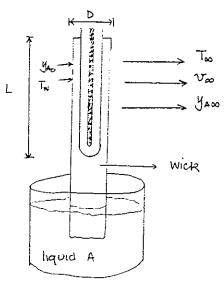
1. WHAT IS THE DIFFERENCE BETWEEN:

- a) HEAT AND MASS TRANSFER ?
 - · Heat transfer is the transport of heat energy where the driving force is the temperature gradient .. Main mechanisms include convection, conduction, and radiation.
 - · Moss transfer is the transport of material species where the driving force is the concentration gradient. Mass transfer also occurs by bulk flow.
- b) HEAT AND MOMENTUM TRANSFER ?
 - · Momentum transfer is the transfer of momentum where the driving force (:
 is the velocity gradient. Main mechanisms include bulk flow, shear
 stress, and various forces acting on the system such as pressure, work,
 and body forces
- C) MASS AND MOMENTUM TRANSFER?
 - · see above

a) A WET BOLB THERMOMETER



· energy balance:

$$N_A \pi D L \Delta h^{Vap} = h (T_{\infty} - T_{N}) \pi D L$$

· moss balance

$$N_{A} = k_{G} (y_{Ao} - y_{Aoo}) + y_{Ao} N_{A}$$

$$N_{A} = \frac{k_{G} (y_{Ao} - y_{Aoo})}{(1 - y_{Ao})}$$
(2)

$$\frac{k_4 (y_{70} - y_{4\infty})}{(1 - y_{40})} \Delta h^{4p} = h (T_{\infty} - T_{w})$$

· equilibrium

$$y_{Ao} = \frac{P_A^S(T_W)}{P}$$

· Chilton Cciburn analogy to get kg

$$j_h = j_m$$

$$St Pr^{\frac{1}{3}} = St_m Sc^{\frac{2}{3}}$$

$$Pr^{\frac{4}{3}} \frac{Nu}{Re Pr} = \frac{Sh}{Re Sc} Sc^{\frac{2}{3}}$$

$$N_u = \frac{hL}{k_g} = \frac{hL}{k_{air}}$$

$$Re = \frac{D v_{\infty} \rho_{air}}{\mu_{air}}$$

$$P_r = \frac{v}{\alpha} = \frac{c_{\rho} M_{air}}{k_{air}}$$

$$Sh = \frac{k_{air} L}{\omega_{A-air}}$$

C) BURNING OF CARBON PARTICLE

$$C + O_2 \rightarrow CO_2$$

assuming complete combustion

· assume transport limitation of 02 to particul

· Shell balance of mass

SS
$$- D \frac{dC_{o_2}}{dr} (4\pi r^2) \Big|_{r \neq a_r} + D \frac{dC_{o_2}}{dr} (4\pi r^2) \Big|_r + r_{o_2} 4\pi r^2 a_r = 0$$

C) BURNING OF CARBON PARTICLE

$$c + o_2 \rightarrow co_2$$

assuming complete combustion

· assume transport limitation of 02 to particle

- Fick's Law:
$$No_2 = -D \frac{dCo_3}{dr}$$

- Fick's Law: Noz = -D dCoz equimolor counter current

· Shell balance of mass

SS
$$-D \frac{dC_{0}}{dr} (4\pi r^{2}) \Big|_{r + \Delta r} + D \frac{dC_{0}}{dr} (4\pi r^{2}) \Big|_{r} + r_{0} + 4\pi r^{2} \Delta r = 0$$

Dillusing - dillusion only mass transler belt - arcompasses contect on alleusion

forster collowed has a line ofference loss defended

between mass (lux and a concentration gradient No. Doc de mass transfer coefficient is defined as a papartionality between mass forx at a boundary of sex point and coefficient bulk $N_A = k_c \left(C_0 - C_\infty \right)$

W; periodic table elections cren't believe so tighting -> News greater dillusivity a) Dallmw L= 3 J M/2 31, b) News (laked 11 up) why? apport equal bic both morebonic gas c) monnas higher K, higher Le? k= 5 cm BSL Pings No. 12 No. 12 What shuts a boltzman "K" Sn't it.

Cv = 5 MR.

C) TEMPERATURE PROFILE

• Find $T_m(z)$, although T = f(z,r)

- energy balance.
$$- \dot{m} c_p T_m \big|_{\mathcal{Z}} + \dot{m} c_p T_m \big|_{\mathcal{Z} + 2} = 2\pi R \, 4Z \, q'' conv$$

$$q''_{conv} = \frac{\dot{m} c_p}{2\pi R} \, \frac{dT_m}{d\mathcal{Z}} = h \, (T_m - T_s)$$

$$\frac{dT_m}{d\mathcal{Z}} = \frac{2\pi R \, h}{\dot{m} c_p} \, (T_m - T_s) = \alpha \, (T_m - T_s)$$

assuming const
$$T_s$$
:
$$T = T_m - T_s$$

$$\frac{dT}{dz} = \alpha T$$

$$T = T_i - T_s \quad \text{at } z = 0$$

$$2 = Stagnand pt. \rightarrow U_2 = 0$$

$$V_1 = V_3$$

* Bernoullis:
$$\frac{1}{3}v_{1}^{2} + \frac{P_{1}}{P} + gh_{1} = \frac{1}{2}v_{2}^{2} + \frac{P_{2}}{P} + gh_{2}$$

$$P_{2} = P_{1} + \frac{1}{2}pv_{1}^{2} + pg(h_{1} - h_{2})$$

$$\frac{1}{2}v_{1}^{2} + \frac{P_{1}}{P} + gh_{1} = \frac{1}{2}v_{3}^{2} + \frac{P_{3}}{P} + gh_{3}$$

$$P_{3} = P_{1} + P_{2}(h_{1} - h_{3})$$

$$v_1 = \left\{ \frac{2[(P_2 - P_3) + P_9(z_2 - z_3)]}{P} \right\}^{1/2}$$

$$\frac{10}{p} = \frac{y^2}{a} + g = \frac{1}{2} = \frac{1}{2}$$

restransi i) skedy flow

a) inamplessible flow

Muscid Pow

4) (low along a sixe-mine

$$V_{x} = \frac{(\rho_3 - \rho_L) D^2}{30 \mu L}$$

- relates pressure drap and average velocity for laminar four in a house

also written as Q= TR+ 1P,-P.) 8 LL

Viscous bick on a small sphere when flow is laminar velie - Reco.1 For = GTIFAV

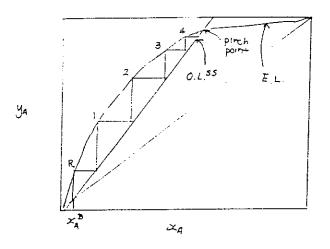
or
$$\frac{\partial F}{\partial D} = -b(\Delta - \Lambda)$$

$$\frac{\partial F}{\partial A} + \Lambda^{2} \frac{\partial X}{\partial D} + \Lambda^{2} \frac{\partial F}{\partial D} + \Lambda^{2} \frac{\partial F}{\partial D} = -b(\Delta - \Lambda)$$
(q)

e) Newtonian figures, pand Lare constant V.V=0 10 DV : - VP+pg+ LV2V

 $b\left(\frac{2F}{3r^{2}} + \frac{2L}{3r^{2}} + \frac{99}{3r^{2}} + \frac{L}{3r^{2}} + \frac{99}{3r^{2}} + \frac{2L}{3r^{2}} + \frac{99}{3r^{2}} + \frac{2}{3r^{2}} +$ $\sqrt{\frac{9^{2}}{9^{4}}} + \sqrt{\frac{9^{2}}{9^{4}}} + \sqrt{\frac{9^$ $\frac{\partial f}{\partial \sqrt{3}} \cdot \frac{\partial c}{\partial \sqrt{3}} \cdot \frac{1}{\sqrt{3}} \frac{\partial c}{\partial \sqrt{3}} + \frac{\partial c}{\partial \sqrt{3}} + \frac{\partial c}{\partial \sqrt{3}} = -\frac{\partial c}{\partial c} \cdot \sqrt{c} \left(\frac{\partial c}{\partial \sqrt{3}} + \frac{\partial c$

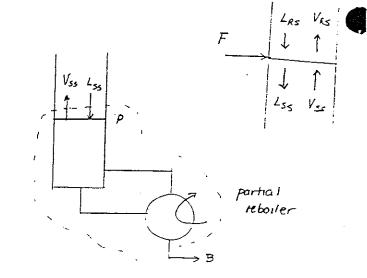
· Switch to new mass balance -> the stripping section



$$L_{SS} = L_{RS} + L_{F}$$

$$V_{SS} = V_{RS} - V_{F}$$

$$V_{SS} \cdot y_{A}^{P} = L_{SS} \cdot x_{A}^{P-1} - B \cdot x_{A}^{B}$$

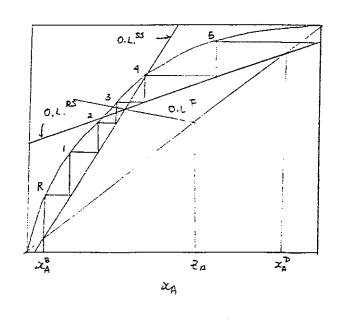


O.L. ss = Stripping section operating line
$$y_A^{\dagger} = \frac{L_{SS}}{V_{SS}} > c_A^{FI} - \frac{B \times A^{S}}{V_{SS}}$$

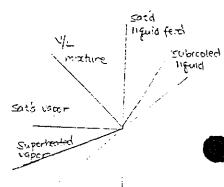
$$\uparrow \qquad \qquad \uparrow$$
slope intercept

· McCabe- Thiele, put them together

Щ



- -We need then, reboiler + little less
 than 5 stages. To get exact R+5
 stages, adjust reflux, r.
- feed stage at 3, where we switch from OL. 55 to O.L. RS
- feed o.L.



a. gas Da I'.75 · Dknosen a VT

light Da T rindepolip by mide range of P

see Alliens

b. ges Lavī hi as Pi liquis La exp: (#)

map of pressure for wide lange of P gas kart liquid k=a:bT * ~!

d) fit to plynomial $C_p = A + Bt + CT^2 + \frac{D}{T^2}$ eiron (or Dis. 20)

Nu: f(Pr, Re, L) force convection C) Nu = a(Pr,GI) fice convection WHILL NU. HD

h is a complex (LT) where k, c, h, sie all changing h is indep of piessure until you get to the knowsen regme 6, 19 geses;

 $\begin{cases} 1 & y = \frac{h}{\rho} \\ 1 & \frac{h}{\rho} \end{cases} = \frac{p m \omega}{RT}.$ gas Da T312 liq p = f(T) -> y x exp(B/T) 15

· Consider s.s flow of fluid of constant p in two systems: a) flow in straight conduit of constant cross section area A. b) flow around submerced object

* examples

a) flow in corduit, circular tubes of radius R & length L, with elevation and $F_{\mu} = (2\pi R L) (\frac{1}{2} p \langle v \rangle^2) f$

- force balance -

$$f = \frac{1}{4} \left(\frac{D}{L} \right) \left(\frac{\mathcal{P}_0 - \mathcal{P}_L}{\vec{s} \cdot \rho \cdot v v^2} \right)$$

aka: fanning fretten, factor

- $f = f(Re, \frac{1}{5})$ f = f(Re) for fully developed $f(ow \ or \frac{1}{5})$

$$f = \frac{16}{Re} \quad \text{for } Re < 2.1 \times 10^{3} \quad \text{laminar}$$

$$f = \frac{0.0791}{Re^{1/4}} \quad \text{for } 2.1 \times 10^{3} < Re \times 10^{5} \quad \text{(Blasius formula)}$$

f = f(Re, k) for turbulent flow, k = height of protuberances due to roughness" of pipe

(6) Lamnar, Far pare boundary agriculturer source 2-d, sleady, incompressible flow by O parsone gradient ωδι + νδι = νδ²; (1) with boundary conditions at 4=0 4.0 y= 00 u= U du= 0 soln of 6in 4 = g(7) where 7 x y based un Stokes for Tux 2.915

Bizzers

intiscies sticom (xn & where n: JA 10-JA

define dimensionless stream fxn

$$f(7) = \frac{\psi}{V \times U}$$
 where $f(7) = dependent variable$
 $\eta = independent variable$

$$U = \frac{\partial \psi}{\partial y} = \frac{\partial \psi}{\partial \eta} \frac{\partial \eta}{\partial y} = \sqrt{v \times U} \frac{\partial f}{\partial \eta} \left[\frac{U}{v \times V} = U \frac{\partial f}{\partial \eta} \right]$$

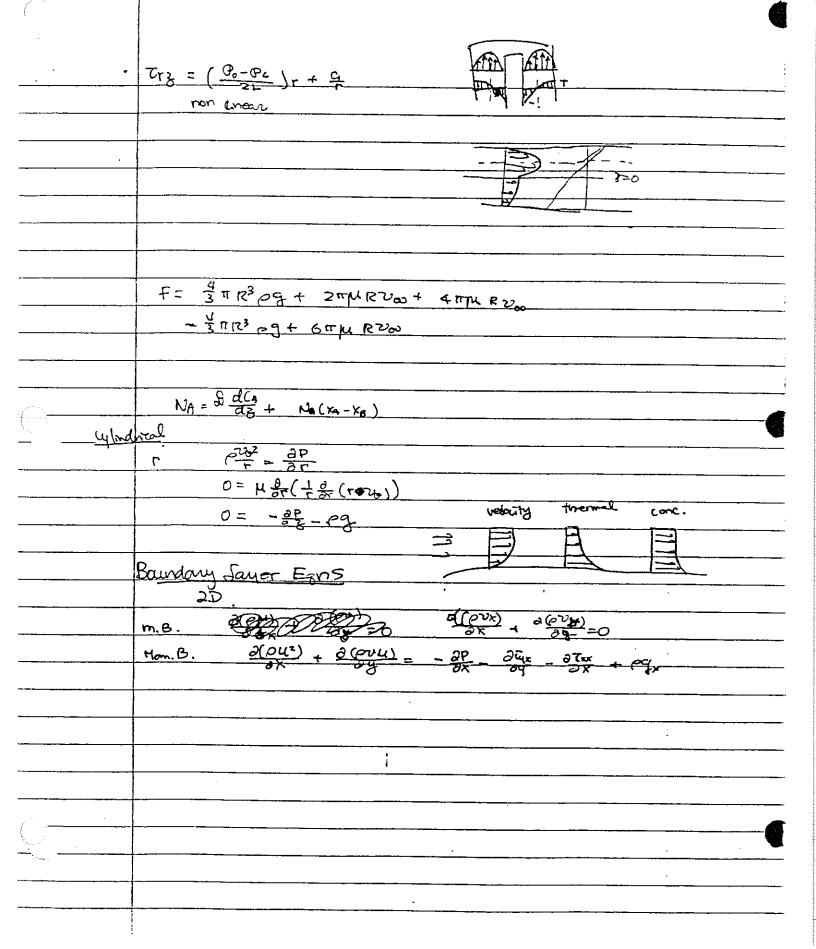
$$U = \frac{\partial \psi}{\partial x} = -\left[\sqrt{v \times U} \frac{\partial f}{\partial x} + \frac{1}{2} \sqrt{\frac{v U}{x}} f \right]$$

$$= -\left[\sqrt{v \times v} \frac{df}{d\eta} \left(-\frac{1}{2} \eta \frac{1}{\chi}\right) + \frac{1}{2} \sqrt{\frac{yv}{\chi}} f\right]$$

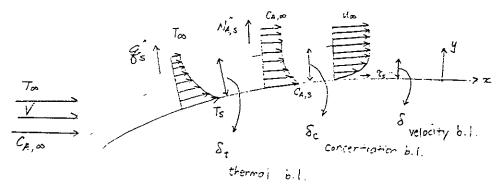
$$V = \frac{1}{2} \sqrt{\frac{yU}{x}} \left[\frac{\eta df}{d\eta} - f \right]$$

$$\frac{\partial u}{\partial x} = -\frac{U}{2x} \eta \frac{d^2 f}{d \eta^2}$$

2 h = U2 d3f



16 JERIVE THE BOUNDARY LAYER EQUATIONS



· mass belonee (continuty)

$$mass = -mass she = 0$$

$$2 + dx$$

$$(\rho u) dy + (\rho v) dx - \left[(\rho u) + \frac{\partial (\rho u)}{\partial x} dx \right] dy - \left[(\rho v) + \frac{\partial (\rho u)}{\partial y} dx \right] dx = \frac{\partial (\rho u)}{\partial x} + \frac{\partial (\rho v)}{\partial y} = c$$

· momentum balance (x-direction) -> velocity boundary layer

$$\frac{\partial(\rho u \cdot u)}{\partial x} + \frac{\partial(\rho v \cdot u)}{\partial y} = -\frac{\partial p}{\partial x} - \frac{\partial v_{ux}}{\partial y} - \frac{\partial v_{xx}}{\partial x} + \frac{\partial v_{xx}}{\partial x}$$

- expord out left side and use continuity and thuse definitions

$$\sigma_{2x} = -2\mu \frac{\partial u}{\partial x} + \frac{2}{3}\mu \left(\frac{\partial x}{\partial x} + \frac{\partial x}{\partial y} \right)$$

$$\nabla_{yx} = -\mu \left(\frac{\partial u}{\partial y} + \frac{\partial x}{\partial x} \right)$$

$$\varphi\left(u\frac{\partial u}{\partial x}+v\frac{\partial u}{\partial y}\right)=-\frac{\partial F}{\partial x}+\frac{\partial}{\partial x}\left[2\mu\frac{\partial v}{\partial x}-\frac{2}{3}\mu\left(\frac{\partial v}{\partial x}+\frac{\partial v}{\partial y}\right)\right]+\frac{\partial}{\partial y}\left[\mu\left(\frac{\partial v}{\partial y}-\frac{\partial v}{\partial y}\right)\right]$$

- for constant µ and incompressible fluid, steady-state.

$$\rho\left(u\frac{\partial u}{\partial x} + v\frac{\partial u}{\partial y}\right) = -\frac{\partial p}{\partial x} + \mu\left(\frac{\partial^{2} u}{\partial z^{2}} + \frac{\partial^{2} u}{\partial y^{2}}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) - \frac{\partial p}{\partial y} + \mu\left(\frac{\partial^{2} v}{\partial z^{2}} + \frac{\partial^{2} v}{\partial y^{2}}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial x} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_{g}^{g} \times \rho\left(u\frac{\partial v}{\partial y} + v\frac{\partial v}{\partial y}\right) + \rho_$$

x-direction

y - direction -

or.

$$\frac{\partial}{\partial t} (\rho \vec{v}) = -(\nabla \cdot \rho \vec{v} \vec{v}) - \vec{\nabla} \cdot \rho - \vec{\nabla} \cdot \vec{z} + \rho \vec{g}$$

· energy balance - thermal boundary layer

$$\rho c_{p} \left(u \frac{\partial T}{\partial x} + v \frac{\partial T}{\partial y} \right) = k \left(\frac{\partial^{2} T}{\partial x^{2}} + \frac{\partial^{2} T}{\partial y^{2}} \right) + \mu \overline{\Phi} + \dot{q}$$

Òſ

$$\frac{\rho c_{p} \left(\frac{\partial T}{\partial t} + \vec{v} \cdot \vec{\nabla} \cdot T \right) = k \vec{\nabla}^{2} T + \mu \vec{q} - T \left(\frac{\partial p}{\partial T} \right)_{p} (\vec{\nabla} \cdot \vec{v})}{\left(\vec{\nabla} \cdot \vec{v} \right)}$$

· species mass balance -> concentration boundary layer

$$u\frac{\partial C_A}{\partial x} + v\frac{\partial C_A}{\partial y} = \frac{\partial}{\partial z} \left(D_{AB} \frac{\partial C_A}{\partial z} \right) + \frac{\partial}{\partial y} \left(D_{EB} \frac{\partial C_B}{\partial y} \right) + \dot{N}_A$$

or.
$$\frac{\partial C_A}{\partial t} + \vec{v} \cdot \vec{\nabla} C_A = \mathcal{D}_{AS} \nabla^2 C_A + R_A$$

17. SKETCH GOVERNING DIAGRAMS FOR A STRIPPER HID FREDERZER

· Absorber - Coursic rice + Huttistage Operation (absorption of A vita liquid)

$$\begin{array}{ccc}
\downarrow & \uparrow \\
L & G_7 \\
x_2 & y_2 \\
X_2 & Y_2
\end{array}$$

$$\begin{array}{ccc}
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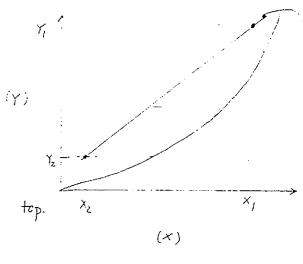
$$Y_{i} = \frac{g_{i}}{1 - g_{i}} = \frac{molst \ F \ ir \ gos}{noise \ restricted gos}$$

$$X_{i} = \frac{x_{i}}{1 - x_{i}} = \frac{molst \ of \ F \ ir \ irg.}{noise \ of \ Solveint}$$

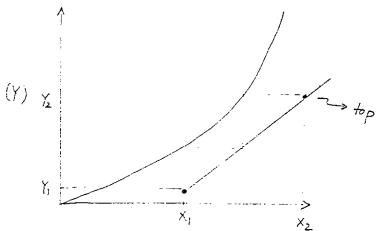
more balance.

bottom

$$Y = \frac{L}{G} \times + \frac{GY_1 - LX_1}{G}$$



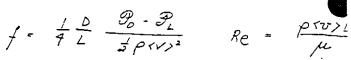
abscroer



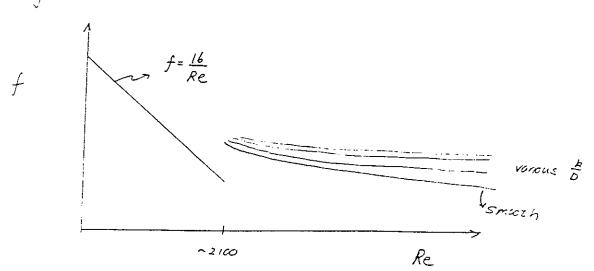
(X)

Stripper }

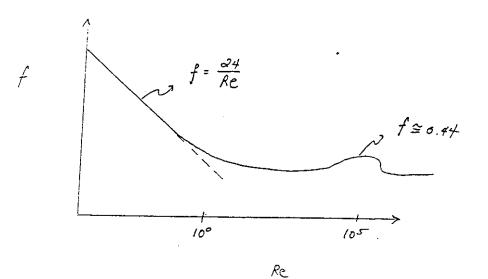
Moody Chare for $f = \frac{1}{4} \frac{D}{L} \frac{g_0 - g_1}{f \rho \langle v \rangle^2}$



on log-log



$$f = \frac{4}{3} \frac{2D}{v_o^2} \left(\frac{\rho_{spi} \cdot \rho}{\rho} \right)$$
. Re = $\frac{\rho v_{so} D}{\mu}$



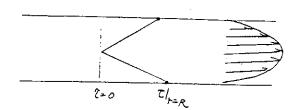
C) FLAT PLATZ

$$f = 1.328 Re^{-1/2}$$

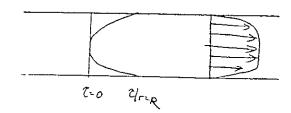
SKETCH THE SHEAR STRESS (2) PROFILE FOR A PIPE. 19.

laminar flow

$$V_Z = V_{max} \left(1 - \frac{r^2}{R^2} \right)$$



turbulent flow



using Hagen Paiseuk T = (Po-Pr) r TIGA MAX @ the ppe wells

monunium flux profile pistile

· Momentum transfer

Re =
$$\frac{\rho VL}{\mu}$$
 = $\frac{Inertial forces}{Viscons forces}$

Gr = $\frac{8\beta(T_5 - T_\infty)L^3}{\nu}$ = $\frac{bucyancy forces}{\nu is cons forces}$

Fr = $\frac{\rho VL}{\rho gL}$ = $\frac{Inertial forces}{glavity forces}$

· Heat transfer

$$Nu = \frac{hL}{k} = \frac{\text{convective heat transfer}}{\text{ksat transfer by conduction}}$$

$$Pr = \frac{y}{\alpha} = \frac{\text{Visacus momentum transport}}{\text{ciffusional heat transport}}$$

$$B_i = \frac{hL}{k_s} = \frac{\text{resistance to conduction}}{\text{resistance to convection}}$$

$$Fo = \frac{\alpha t}{L^2} = \frac{\text{heat conduction rate}}{\text{rate of thermal energy shorage}}$$

· Mass transfer

$$Sh = \frac{h_m L}{D} = \frac{convective \ mass \ transfer}{cliffvsive \ mass \ transfer}$$

$$Sc = \frac{D}{D_{AB}} = \frac{viscons \ momentum \ transport}{cliffvsive \ mass \ transfer}$$

$$Bim = \frac{h_m L}{D_{AB}} = \frac{resistance \ te \ mass \ transport \ by \ diffvsion}{resistance \ to \ mass \ transport \ by \ convection}$$

BI GIVEN A PCOL OF ORGANIC LIQUID (EQ FROM A SPILL), ESTIMATE ITS

· lancre rest transfer, isotherms

$$N_{A_{\overline{e}}} = -\widetilde{\omega}_{AB} \frac{\partial C_{A}}{\partial z} + \frac{1}{2} \left(1/_{F_{\overline{e}}} + \frac{N_{B_{\overline{e}}}}{N_{B_{\overline{e}}}} \right)$$

OSSUME IZ & O

rate =
$$N_{A_{\overline{A}}} / A_{pool} = - \mathcal{J}_{A_{\overline{A}}} \cdot A \frac{c' \gamma_z}{c' \overline{z}} /_{\overline{z}=0}$$

· governing egn

$$\frac{\partial C_{A}}{\partial t} = \mathcal{L}_{R3} \frac{\partial^{2} C_{R}}{\partial z^{2}}$$

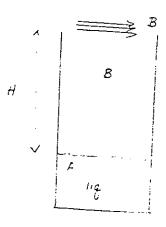
$$C_{R} = 0 \quad \vec{x} \rightarrow \mathcal{C}$$

$$C_{R} = \frac{\mathcal{D}_{R}^{20}(T_{2/R})}{RT} \quad \vec{x} = 0$$

then get
$$\frac{\partial C_R}{\partial z}\Big|_{z=0}$$
 -> $rot_z = -2c_{rz}A + \frac{\partial C_R}{\partial z}\Big|_{z=0}$

· Diffusivity

- can use kinetic theory and corresponding states to obtain Dag for goses at low P.



at s.s
$$N_E = 0$$

$$N_A = -c \partial_{AB} \frac{dC_E}{d\hat{z}} + x_A N_A$$

$$N_A = \frac{-c \partial_{AB} \frac{dC_F}{d\hat{z}}}{1 - x_A}$$

$$\frac{dN_A}{d\hat{z}} = 0 \qquad \text{Soive} \quad x_A = f(\hat{z})$$

$$\hat{z} \cdot C \cdot z = 0 \quad C_A = \frac{\hat{r}_A}{RT} = C_A$$

$$\hat{z} = H \quad C_A = 0$$

B.S.L. Egn 17.2-10

$$\frac{1-\chi_{A}}{1-\chi_{A_{0}}} = \left(\frac{1}{1-\chi_{A_{0}}}\right)^{\frac{Z}{H}}$$

$$N_{A} \Big|_{Z=C} = \frac{C \hat{\mathcal{Q}}_{AB}}{H \cdot \chi_{B_{IN}}} \chi_{A_{0}}$$

$$= \frac{P \hat{\mathcal{Q}}_{AB} / RT}{H \cdot (P_{B})_{IN}} P_{A_{1}}$$

$$\hat{\mathcal{Q}}_{AB} = \frac{N_{A} |_{Z=0} \cdot H \cdot (P_{B})_{IN}}{P / R_{7} \cdot P_{A_{1}}}$$

· VISCOSHY

$$\mu = \frac{Gh}{2\pi R^3 I \Omega}$$
 G= total torque.

experimental determination of defosivily - erapsiate a pure liquid in a narrow tube with a gas passed over the top. - ar measure for fall in liquid live withtime Li Palzi - 2,2) RT pam

ama Das P(Da, -Paz) 7, P.

- rate of evaporation of sphere

For a gaseous mixture: a bulb method

- · capillary of area A and length L · Vcp. << V, orv,

 - · Open valve · Wait · Close valive
 - · semple champer contents

rate of diffusion of A going to V2 = 12 to of accumulation in V

 $A \int_{A}^{2} = -D_{AG} \left(C_{2} - C_{i} \right) A = V_{i} \frac{dC_{2}}{dt}$ (1)

· calc car, average value to easin from starting compositions c' and Co at t=0

(V, + V_) Cav = 1, C, + 1, C2

· balance at time t gives

(V, +V2) Cay = V, C, + V2 C2

$$\frac{C_{av}-C_{2}}{C_{av}-C_{2}} = \exp\left(\frac{-D_{AB}(V_{1}+V_{2})}{L_{IA}(V_{2}V_{1})}E\right)$$

VISCOSIFY

· brooklie a Visio. is ten

liquids

dT = ref :

1= mdv => 2 = mwr.

small gap

dA: 1:100

integrate T = DITTI LAWTE

gaseous mixture?

1st estimate - mean visiosity?

ving mare kiactons

Com Alkins:

amain tecniques

1. rate of damping of the torsional usuallations a disk hanging in the gas

> use Biseville dr = (p, 2-p, 2) Tr 4 length L

> > V= Volume Flowing

BSL

Explosions are fost processes, so diffusion not important Exponsion of gases due to bulk flow. Also important is the void space for gos to expand.

DOES DROP T CHANGE AS IT FALLS.

- forced convection heat transfer as drop falls
- evaporation occurs as drop falls
- rate of evaporation = rate at which heat is transferred to drop

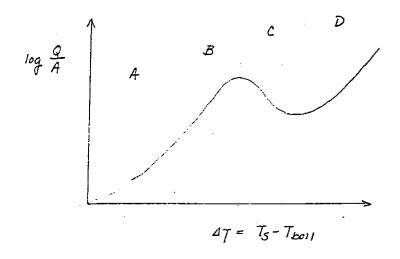
Trg = (Po-Pa)r + G F= 3 11 12 pg + 211/4 1200 + 411/4 1200 - 3 1123 Pg + 6114 R200 NA = & d(A + NA (XA-XB) Property of the second of the 0= H & (+ & (recol) 0= -38-19 $\frac{\partial(\partial u^2)}{\partial x} + \frac{\partial(\partial vu)}{\partial y} = -\frac{\partial p}{\partial x} - \frac{\partial \overline{u}_{xx}}{\partial y} - \frac{\partial \overline{v}_{xx}}{\partial x} + \frac{\partial q}{\partial x}$

A) nig balance $\rho(\rho V_{cl}T) = hA (T - T_{lower})$ $hk = Nu = 2 + 0.6 Re^{2} Pr^{2}$ $\rho assume dop (2 a uniform T)$ $\rho assume laminal$ $h: D^{2} (P_{sprine} - P_{air}) from Stokes law$ 18 h

DRAW THE BOILING CURVE AND DESCRIBE THE PHYSICAL 27 RESPONSIBLE FOR THE OBSERVED BEHAVIOR DEAN AND EXPLAIN THE SIMILAR CURVE FOR CONDENSATION.

· Heat transfer to boiling liquid

- liquid at Thou and P of equipment
- heat transfer from Surface W/ 5 > Tboll
- bubbles of vapor generated at surface



Natural convection : few bubbles, mostly natural convection (AT≤5°C A:

Nucleate boiling more bubbles (5 \ 47 \ 30°C) \mathcal{B}

many bubbles form so quickly that transition boiling: coalesce and form layer of insulating vapor on surface. As 71, = q" L

(30 ≤ 4T ≤ 120°C)

Film boiling: bubbles detach faster than they coalesce **)**. radiation thru vapor layer next to surface becom significant

· Heat transfer from condensing vapor

- Ts < Toondensate

Film condensation: film of condensate forms or er surface and causes the main resistance to heat transfer

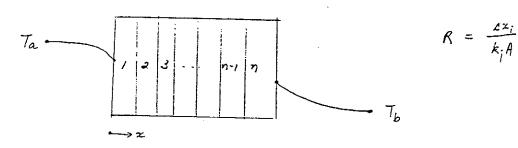
Iropwise Condensation. Only drops of liquid are formed on surface more clean surface for heat transfer.

.

.

- 30. HOW IS THE OVERALL HEAT TRANSFER COEFFICIENT FOR A HEAT EXCHANGER
 FOUND?
 - · for flot plote geometry:

resistances:



- all fluxes must equal:
$$g = Aq'' = Aq'' = Aq'' = Aq'' = Aq'' = Aq''$$

$$f_0'' = -k_1 \frac{dT_1}{dx} \qquad integrate \qquad -\frac{f_0''}{k_1} \Delta x_1 = T_5 - T_1$$

$$f_0'' = -k_2 \frac{dT_2}{dx} \qquad \Rightarrow \qquad -\frac{f_0''}{k_2} \Delta x_2 = T_3 - T_2$$

$$\xi_0'' = -k_n \frac{dT_n}{dx}$$

$$- \frac{\xi_0''}{k_n} z = T_{n+j} - T_n$$

- for faces:

$$g_o'' = h_a (T_a - T_i)$$

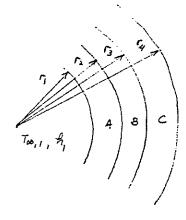
$$g_o'' = h_b (T_n - T_b)$$

$$- \operatorname{add} \operatorname{all} \Rightarrow \begin{cases} g'' \left[\frac{\Delta x_1}{k_1} + \frac{\Delta x_2}{k_2} + \dots + \frac{\Delta x_n}{k_n} + \frac{1}{k_a} + \frac{1}{k_b} \right] - T_a - T_b \end{cases}$$

$$f_{a}'' = U(T_a - T_b)$$

$$\overline{U} = \frac{1}{\frac{1}{h_a} + \frac{dx_1}{k_1} + \frac{dx_2}{k_2} + \dots + \frac{dx_n}{k_n} + \frac{1}{h_b}}$$

· for cylmolocal heat exchanger



To, 4, 24

- energy balance

$$g'' \cdot 2\pi r L \Big|_{r+2r} - g'' \cdot 2\pi r L \Big|_{r} = 0$$

$$\frac{d(rg'')}{dr} = 0 \qquad g'' = -k \frac{dT}{dr}$$

$$T(r) = \frac{T_{5,1} - T_{5,2}}{\ln(\frac{r}{r_2})} \ln(\frac{r}{r_2}) + T_{5,2}$$

$$g'' = \frac{2\pi L k}{\ln(\frac{r_{5,1} - T_{5,2}}{r_{5,2}})}$$

Oddition of resistances:

$$R_{cond} = \frac{\ln\left(\frac{r_{2}}{r_{1}}\right)}{2\pi Lk} \cdot \frac{1}{r} \frac{d}{dr} \left(\frac{r}{dr}\right) = 0$$

$$g = 2\pi r_1 L \cdot U_1 (T_{\infty,1} - T_{\infty,4}) \rightarrow$$

Ui based on A,.

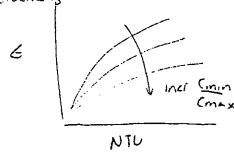
• sphere

$$R_{cond} = \frac{\frac{1}{r_1} - \frac{1}{r_2}}{4\pi k}, \quad \frac{1}{r^2} \frac{d}{dr} \left(r^2 \frac{d\tau}{dr}\right) = 0$$

NTU method

C = mEp

uce wielations

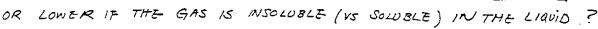


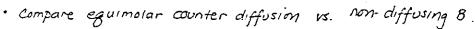
combine with 9= mcp DT in cgn: and solve

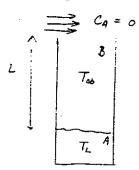
2 cans 2 unknowns

$$\mathcal{E} = \frac{T_{h,i} - T_{h,o}}{T_{h,i} - T_{c,i}}$$
 effectiveness factor

 $NTU = \frac{UA}{C_{min}}$







$$g'' = h(T_L - T_\infty)$$

$$N_A = k(C_{A_S} - C_A^\infty) = kC_{A_S}$$

consider diffusion to fird k, then use chilton-Colburn to find &.

- Non diffusing B
$$NA = -c\partial_{AB} \frac{dx_{A}}{dz} + x_{A} \left(N_{A} + N_{B}\right) \rightarrow N_{A} = \frac{-c\partial_{AB} \frac{dx_{A}}{dz}}{1-x_{A}}$$

$$\frac{dN_A}{d\vec{z}} = 0 \implies N_A = C_1 = -\frac{c\partial_{AS}}{1 - z_A} \frac{dz_A}{dz}$$

8.C.

$$Z=L$$
, $Z_{\mu}=0$
 $Z=0$, $Z_{\mu}=\frac{P_{\mu}s}{D}=Z_{\mu}$

$$C_{1} = -\frac{c\partial_{AB} \ln (1-x_{A_{0}})}{L}$$

$$C_{2} = c\partial_{AB} \ln (1-x_{A_{0}})$$

$$z_{A} = 1 - \exp\left[\frac{1}{c\partial_{AB}}\left(C_{1}z+C_{2}\right)\right]$$

$$\frac{dz_{A}}{dz} = -\frac{1}{c\partial_{AB}}C_{1}\exp\left[\frac{1}{c\partial_{AB}}\left(C_{1}z+C_{2}\right)\right]$$

$$N_A = -c \partial_{AB} \frac{dx_A}{dz} \rightarrow$$

$$x_{A} = -\frac{1}{C\Omega_{AB}} \left(C_{1} + C_{2} \right)$$

$$C_{2} = -C\Omega_{AB} \times A_{BB}$$

$$C_{3} = -C\Omega_{AB} \times A_{BB}$$

$$C_{1} = \frac{\omega_{18} \times_{18}}{2}$$

$$C_{2} = \frac{\omega_{18} \times_{18}}{2}$$

$$J_{H} = \frac{1}{c_{\rho}^{2}} \frac{(N_{\rho r})^{-1}}{k}$$

$$J_{H} = \frac{h}{c_{\rho}^{2}} \frac{(C_{\rho} h)^{-1} J_{\Delta}}{k}$$

$$J_{H} = \frac{h}{k} \frac{(C_{\rho} h)^{-1} J_{\Delta}}{k} \frac{h}{\rho VD}$$

$$J_{H} = N_{U} (Re)^{-1} (\rho_{I})^{-1} J_{\Delta}$$

$$j_{H} = St \cdot Pr^{2/3} = \frac{Nu}{Re Pr} Pr^{2/3} = \frac{Nu}{Re Pr^{1/3}}$$

$$j_{D} = St_{M} Sc^{2/3} = \frac{Sh}{Re Sc} Sc^{2/3} = \frac{Sh}{Re Sc^{1/3}}$$

$$Nu = \frac{hL}{R}$$

$$Sh = \frac{h_{M}L}{D_{AB}}$$

$$Re = \frac{PVL}{M}$$

$$Pr = \frac{Q}{R} = \frac{M}{P} \cdot \frac{PCp}{R} = \frac{MCp}{R}$$

36 HOW WOULD YOU SEPARATE OZ FROM SALT WATER? WHAT VARIABLES

AFFECT SOLUBILITY? WHERE IS THE MASS TRANSFER RESISTANCE?

$$\begin{aligned}
\beta_{2} & \phi_{2} P = H_{3,1} \times_{2} \\
& H_{3,1}(P) = H_{3,1}(P_{1}) \left[\exp \int_{P_{1}}^{P} \frac{\overline{V_{3}}^{\infty}}{RT} dP \right] \\
& \times_{2} = \frac{V_{2} \phi_{2} P}{H_{2,1}(P_{1}) \exp \left[\frac{\overline{V_{2}}^{\infty}}{RT}(P-P_{1}) \right]}
\end{aligned}$$

decreasing P 1 x2

- -> throttle sea water through valve to tank under vaccuum. Oz buchles will form.
- most of the moss transfer resistance. Is on the liquid phase side because it is in this phase that Oz is scarce

pcp VdT = WcpTin - WcpT.

·put in deration vers

·take Lopicse

pcp V sT(a) = W (p Tin(a) - WcpT(a)

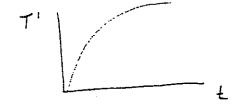
T'(a) = Tin(a)

24+1

1e 1st order system

 $T''(L) = \frac{m}{5}$ m is magnitude of step change T'(L) = m(1-c-t/c)

balance



would level off to a new temp

39

WHY ARE ANALOGIES BETWEEN HEAT & MASS TRANSFER MUCH

MORE STRAIGHT FORWARD TO USE THAN ANALOGIES BETWEEN

MASS & MOMENTUM TRANSFER?

Fluz of heat and mass are vectors but momentum is a tensor.

42. WHAT IS THE THEORETICAL BASES FOR ALL THE "FAMOUS" ANALOGIES

BETWEEN HEAT, MASS, AND MOMENTUM TRANSPORT?

$$\frac{\partial v_{x}}{\partial t} + v_{x} \frac{\partial v_{x}}{\partial x} + v_{y} \frac{\partial v_{x}}{\partial y} + v_{z} \frac{\partial v_{x}}{\partial z} = -\frac{\partial P}{\partial x} \frac{1}{P} + v_{z} \frac{\partial^{2} v_{x}}{\partial x^{2}} + \frac{\partial^{2} v_{x}}{\partial y^{2}} + \frac{\partial^{2} v_{x}}{\partial z^{2}} + \frac$$

write variables in dimensionless forms:

$$T^* = \frac{V}{V}$$

$$T^* = \frac{T - T_0}{T_{\infty} - T_0}$$

$$C^* = \frac{C - C_0}{C_{\infty} - C_0}$$

let is a dimensionless variable, then

all egns collapse down to

$$\frac{\partial \dot{s}}{\partial t} + \vec{s}^* \cdot \nabla \vec{s} = \begin{cases} \frac{1}{Re} & momentum \\ \frac{1}{RePr} & heat \\ \frac{1}{ReSc.} & mass \end{cases}$$

43)

Skin fretion drag - due to tengential to rae of Cruid
on body

form drag = pressure drag - due to normal trae of fluid or
body

- 45. WHY DOES FROST NOT FORM UNDER A TREE LUHEN 17 IS ON THE GROUND ALL AROUND THE TREE ?
 - · Insulating Stegnant air under foiloge keeps convection to aminimal & there fore T is higher.
 - · also consider radiotive insulation. Foilage blocks radiative loss of heat to sky.

a) MASS TRANSFER

$$\frac{\partial \mathcal{C}}{\partial t} + \vec{v} \cdot \vec{\nabla} \mathcal{C} = \partial_{AB} \vec{\nabla}^2 \mathcal{C} + R_A$$

b) MOMENTUM TRANSFER

$$\rho\left(\frac{S\vec{v}}{\partial t} + \vec{v} \cdot \vec{\nabla}\vec{v}\right) = -\vec{\nabla}p + \mu \vec{\nabla}^2 \cdot \vec{v} + \rho \vec{g}$$

C) HEAT TRANSFER

$$p_{\mathcal{G}}\left(\frac{\partial T}{\partial t} + \vec{v} \cdot \vec{\nabla} T\right) = k \vec{\nabla}^2 T - T \left(\frac{\partial P}{\partial T}\right)_{P} (\vec{\nabla} \cdot \vec{v}) + \mu^{\frac{\sigma}{2}} \nabla T + \mu^{\frac{\sigma}$$

Mass
$$\frac{d(u_1C_A)}{d^2} = \Gamma_A \rho_B$$

Eraun egn:

$$F_{p} = AKf \Rightarrow \frac{p_{o} - p_{c}}{5 p_{o}^{2}} = \frac{L}{O_{p}} \cdot 4f$$

$$velocity$$

$$J_{p} = partical$$

$$diameter$$

recall laminar flow
$$\Rightarrow$$
 $Q = \frac{\pi (\Delta P) R^4}{8 \mu L} = \langle v \rangle \pi R^2$
 $\langle v \rangle = \frac{(P_0 - P_L) R^2}{8 \mu L}$
 $\Rightarrow \langle v \rangle = \frac{(P_0 - P_L) R_h^2}{8 \mu L}$

$$Q = a_v (1 - \varepsilon) \qquad D_p = \frac{6}{a_v} \qquad v_o = \langle v \rangle \varepsilon$$

$$v_o = \frac{(\mathcal{P}_o - \mathcal{P}_L)}{L} \frac{D_p^2}{2(36\mu)} \frac{\varepsilon^3}{(1-\varepsilon)^2}$$

need flowing air stream post wick for derivation to be value air humidity must be constant

Stagnant air. b.L. may build up w/ higher humidity.

WHAT IS INSIDE A LIGHTBULB & WHY? 51.

> We want radiation of light, so a vacuum exists to prevent Conduction & Convection

HOW DOES A LAWN SPRINKLER WORK? 54.

Macroscopic momentum balance

$$\frac{dP}{dt} = P(v_x^2), A_1 - P(v_x^2), A_2 + P_1S_1 - P_2S_2 + m_{tot}g_x - F_x$$

1

total

momentum

change

change

change

by flow

force acting

no c.v

Fx

x nom: $D = f(v_i^2)A + P_iA - F_z \rightarrow F_{zc} = f(v_i^2)A + P_iA$

y nom: $0 = -\rho(v_z^2) A - \rho_1 A - F_y \rightarrow F_y = -\left(\rho(v_z^2) A + \rho_2 A\right)$

: lawn sprinkler rotates!

	, ,	
Steam	dy momentum balance on a horizon	ral notzie
· ·	→× 1 2	so that it doesn't include pipe wal
	mizm m	
	V _a > V _a smaller area	
	evaluate operación preserve using r	nechanical orgbalance
	$\frac{1}{\sqrt{1}} + \frac{1}{\sqrt{2}} + \frac{1}{\sqrt{2}} + \frac{1}{\sqrt{2}} + \frac{1}{\sqrt{2}}$	
	A=0 garge pressure	2
	D 15" = 1022 F2	
	$P: (\frac{\sqrt{2}-\sqrt{2}}{2}) \rho = + ve$ large	cramples numberal
······.		6, Q.03154m3
- 21 - 21	Rx = mv2-mv, + p2A2-p,A,	Q=0.028cm
	4ve 40E	P = 1.156×10' N
	in numerical examples that	1= 9.9(m/s ==
	I have seen	V = 491m/
	Fx is -ve = force of nozz	le on livid

12 maintain this tension

FIREMEN MUST PULL

CONSIDER A DEPARTMENT STORE PING PONG BALL "FLOATING" ABOVE A VACUUM CLEANER DISCHARGE WHAT DETERMINES HOW HIGH

THE BALL WILL BE? WHAT KEEPS THE BALL FROM MOVING

LATERALLY OUT OF THE PATH OF THE AIR? WHAT DOES

THE VELOCITY PROFILE LOOK LIKE CLOSE TO, AROUND, AND AROVE

THE BALL? WHAT DETERMINES WHETHER THE BALL WILL FALL

TO THE GROUND IF THE JET IS POINTING AT AN ANGLE RETHER