

Differences between momentum, heat, & mass transfer

#1

Mass -

More complicated than viscous flow or heat conduction
because for the first time have to deal w/ mixtures

momentum - Newton's Law $\tau_{yx} = -\mu \frac{dv_x}{dy}$

heat - Fourier's Law $q = -k \frac{dT}{dy}$

mass - Fick's Law $N_A = -D_{AB} \frac{dc}{dy}$

heat vs. mass

mass deals w/ mixtures while heat transfer deals w/ single component - energy
mass has $\Delta B.C.$

heat vs. momentum

τ is a tensor + has $\Delta B.C.$ than h.t.

mass vs. momentum

same problems / Δ 's as above

Transport Questions:

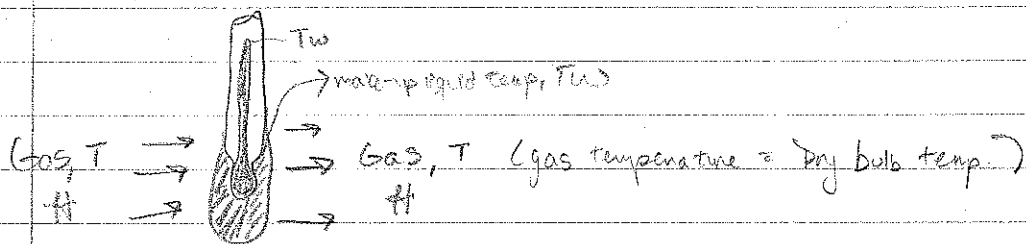
2a

2a. Wet bulb thermometer: (MSH - Pg 748)

Defn (MSH): S.S. non-equilibrium temperature reached by a small mass of liquid immersed under adiabatic conditions in a continuous stream of gas. function (T_{gas} + Humidity gas)

also:

T_w is the S.S. liquid temperature at which the heat needed to evaporate the liquid + heat the vapor to gas temperature = the sensible heat flowing from the gas to the liquid



Theory of wet bulb Temp:

1. neglecting radiation (large enough gas velocity so that sensible heat transfer \gg radiation Q)

q , rate of sensible heat transfer to liquid = rate of vaporization ($\lambda_w + c_p \Delta T$)

$$(1) \quad q = M_w N_A [\lambda_w + c_p (T - T_w)]$$

↳ molar rate of vaporization

latent heat of liq. at T_w sensible part of vapor

To find q :

$$(2) \quad q = h A (T - T_i)$$

↳ h.t.c. between gas + liquid surface.

$T \quad T_i = T_w$

To find N_A : Assuming one-way diffusion.

$$N_A = \frac{k_y (y_i - y) A}{(1 - y)_L}$$

↳ one-way diffusion factor: (Log mean of $(1 - y)_n$ + $(1 - y)_i$)

where $(1 - y)_L = \frac{(1 - y)_n - (1 - y)_i}{\ln \frac{(1 - y)_n}{(1 - y)_i}}$

Since A in (2) + (3) the same

$$\therefore \frac{q}{h(T - T_w)} = \frac{N_A (1 - y)_L}{k_y (y_i - y)}$$

in the usual range of temp + humidities ≈ 1 so can be omitted

plugging in (1) for q

$$h(T - T_w) = \frac{K_y (y_i - y)}{N_A (1 - y)_L} \left[m_A N_A [\lambda_w + c_p (T - T_w)] \right]$$

→ mass transfer coefficient (mol/area/unit mole fraction)

Assumptions: $(1 - y)_L \approx 1$

$c_p \Delta T \ll \lambda_w$ ∴ sensible heat neglected

$$\therefore h_y (T - T_w) = K_y (y_i - y) m_A [\lambda_w]$$

y_i = mole fraction of vapor in saturated gas at T_w (no conc. gradient in liquid)

Now: to put in terms of humidity

From Eqn 23.1 (MSH)

$$H = \frac{m_A P_A}{m_B (1 - P_A)} \rightarrow \text{partial pressure of vapor} = \frac{\text{mass of vapor, A}}{\text{mass of dry gas, B}} \quad \text{At 1 atm!! (since } P_A = y_A P^{\text{sat}})$$

$$H = \frac{m_A y_A}{m_B (1 - y_A)}$$

$$H m_B - H y_A = m_A y_A$$

$$y_A = \frac{H m_B}{m_A + H}$$

mole fraction of vapor in the gas phase

But MSH gives $y = \frac{H/m_A}{1/m_B + H/m_A} = \frac{H m_B}{m_A + H}$

$y \approx \frac{H m_B}{m_A}$ since $H/m_A \ll 1/m_B$

$$\text{Now: } h_y (T - T_w) = K_y [\lambda_w] (H_w m_B - H m_B)$$

$$\text{or } \boxed{\frac{H - H_w}{T - T_w} = \frac{h_y}{m_B K_y \lambda_w}} = \frac{c_p}{\lambda_w} \left(\frac{N_{sc}}{N_{Pr}} \right)^m = \frac{c_p}{\lambda}$$

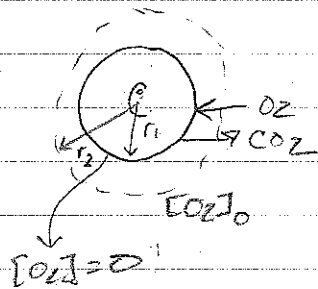
→ $\mu/P_D r$
→ $\frac{c_p \mu}{K}$

$$\therefore \frac{h_y}{m_B K_y} = c_p \left(\frac{N_{sc}}{N_{Pr}} \right)^n$$

if $m = 2/3$, close to experimental value for air in water (but slightly less because it does not account for radiation)

Burning Carbon Particle

2c



$$N_A = -N_B \text{ e.m.c.D}$$

1. Mole balance on O_2

$$I_n - \text{out} + G_{\text{gen}} = \text{Acc.}$$

$$N_{B,r} 4\pi r^2 \Big|_r - N_{B,r} 4\pi r^2 \Big|_{r+\Delta r} + 0 = 0$$

\div by $-4\pi \Delta r$

$$\lim_{\Delta r \rightarrow 0} \frac{(N_{B,r} r^2 \Big|_{r+\Delta r} - N_{B,r} r^2 \Big|_r)}{\Delta r} = 0$$

$$\frac{d(r^2 N_{B,r})}{dr} = 0$$

2. Find expression for $N_{B,r}$

Fick's Law

$$N_{A,r} = x_A (N_A + N_B) - C D_{AB} \frac{dx_A}{dr}$$

$$N_{B,r} = -C D_{BA} \frac{dx_B}{dr}$$

Assume $(D_{AB} \neq f(r))$ i.e. constant temp. (900°C - combustion)

$$C D_{AB} \frac{d(r^2 \frac{dx_B}{dr})}{dr} = 0$$

Integrating once

$$r^2 \frac{dx_B}{dr} = C_1$$

" again:

$$x_B = \int \frac{C_1}{r^2} dr = -\frac{C_1}{r} + C_2$$

2C cont.

$$X_B = -\frac{C_1}{r} + C_2$$

→ decrease w/ time. Since Carbon is consumed

B.C. At $r = r_1(t)$ $X_B = 0$

$r = r_2$ $X_B = X_{B,0}$

$$0 = -\frac{C_1}{r_1} + C_2$$

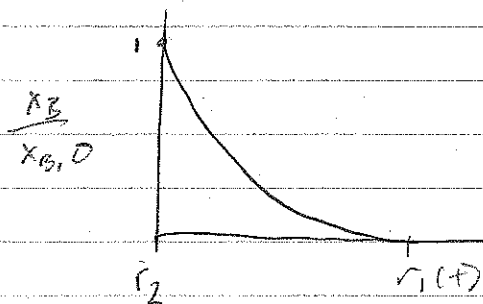
$$C_1 = C_2(r_1)$$

$$X_{B,0} = -\frac{C_2(r_1)}{r_2} + C_2 = C_2 \left(1 - \frac{r_1}{r_2}\right)$$

$$C_2 = \frac{X_{B,0}}{(1 - r_1/r_2)} \quad \text{or} \quad C_1 = \frac{X_{B,0}(r_1)}{(1 - r_1/r_2)}$$

$$\therefore X_B = -\frac{C_2(r_1)}{r} + C_2$$

$$\therefore \frac{X_B}{X_{B,0}} = \frac{1}{(1 - r_1/r_2)} \left(1 - \frac{r_1}{r}\right) = C \left(1 - \frac{r_1(t)}{r}\right)$$



$$\therefore N_{B,r} = -C D_{BA} \frac{dX_B}{dr} = -C D_{BA} \left(\frac{C_1}{r^2}\right) = -\frac{C D_{BA}}{r^2} \left(\frac{X_{B,0}(r_1)}{1 - r_1/r_2}\right)$$

$$N_{B,r} = -C_{B,0} D_{BA} \left(\frac{r_1(t)}{1 - r_1(t)/r_2}\right) \frac{1}{r^2}$$

Part 3 - Mole Balance on Carbon:

2c - contin.

$$I_n = \text{out} + \text{Gen.} = \text{Acc.}$$

$$0 = 0 + r_c 4\pi r_i(t)^2 = \frac{d}{dt} \left(\frac{4\pi r_i^3}{3} \rho_c \right)$$

$$\frac{1}{\rho_c} r_c = \frac{dr_i}{dt}$$

(4) Relate r_c to $N_{B,r}$

$$-r_c \text{ (mol/m}^2\text{)} = -N_{B,r} \Big|_{r=r_i} = \left(C_{B,0} D_{BA} \left(\frac{1}{r_i - \frac{r_i^2}{r_2}} \right) \right)$$

$$\therefore -\frac{dr_i}{dt} = \frac{C_{B,0} D_{BA}}{\rho_c} \left(\frac{1}{r_i - \frac{r_i^2}{r_2}} \right)$$

$$\text{B.C. at } t=0, \quad r_i = r_{i,0}$$

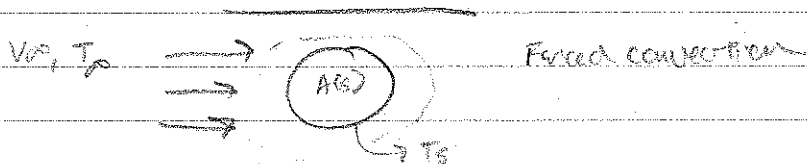
$$t = t_f \quad r_i = 0$$

$$-\int \left(r_i - \frac{r_i^2}{r_2} \right) dr_i = \frac{C_{B,0} D_{BA}}{\rho_c} t$$

$$-\frac{r_i^2}{2} + \frac{r_i^3}{3r_2} + C_1 = \frac{C_{B,0} D_{BA}}{\rho_c} t$$

$$C_1 = -\frac{r_{i,0}^2}{2} + \frac{r_{i,0}^3}{3r_2}$$

$$t_f = \frac{\rho_c}{C_{B,0} D_{BA}} \left(\frac{r_{i,0}^2}{2} - \frac{r_{i,0}^3}{3r_2} \right)$$



Assume: sphere, entirely water under.



Mo balance:

$$I_n - A_r + G_{gen} = A_r$$

$$0 = N_A 4\pi r^2 = \frac{d}{dt} \left(\frac{4}{3} \pi r^3 \rho_c \right) \quad N_A = A k_m (C - C_\infty)$$

$$N_A = \frac{\pi \rho_c}{3} \frac{dr}{dt}$$

Energy balance

$$\dot{q} = h A (T_\infty - T_s) = \lambda_w N_m A$$

energy transfer to melting = energy used in melting iceberg.

For forced convection heat transfer around a single sphere:

$$\frac{h_m D}{k_f} = 2.0 + 0.6 \left(\frac{D V_\infty \rho_f}{\mu} \right)^{1/2} \left(\frac{c_p \mu}{k_f} \right)^{1/3}$$

(Nu) (Re)^{1/2} (Pr)^{1/3}

Solve for $h = f(r)$

$$N_m = \frac{h(r)}{\lambda_w} (T_\infty - T_s) = \frac{\pi \rho_c}{3} \frac{dr}{dt}$$

$$t = \frac{\pi \rho_c \lambda_w}{3 (T_\infty - T_s)} \int_R^0 \frac{1}{h(r)} dr$$

Use Momentum Transport

to get Drag

$$F_k = A f^{1/2} v^2 \quad \text{Skin + body drag}$$

Friction factor:

3

friction factor - measuring constant f

$$F_R = AKf \Rightarrow$$

for conduit



$$F_R = (P_o - P_L + f g (C_o - h_c)) / A$$

$$f = f(\text{Re}, \text{system shape})$$

for sphere



$$F_R =$$

$$= 4/3 \pi R^3 (P_o g - P_{\text{fluid}} g)$$

Experimentally, for flow in conduits:

$$f = \frac{1}{4} \left(\frac{D}{L} \right) \left(\frac{P_o - P_L}{\frac{1}{2} \rho \langle v \rangle^2} \right)$$

For flow in spheres:

$$f = \frac{4}{3} \frac{g D}{v_o^2} \left(\frac{P_o - P}{\rho} \right)$$

Heat transfer will affect friction factor, since it will affect fluid properties, ρ , μ , etc.

$\uparrow Q$, $\uparrow T$, for liquid - $\downarrow \rho$, $\downarrow \mu$, $\uparrow f$?

Differences between diffusion & mass transfer coeff.

4.

Diffusivity: D_{AB} (cm^2/sec) $\uparrow T, \downarrow P$

P_2 i.e. T, x_A

Analogous to $\sigma = \mu/P$ For gases, $\neq f(x_A)$

$$\sigma x = \frac{k}{P_{EP}}$$

Diffusivity is a molecular property \rightarrow describes flux for local (infinitesimal) concentration gradients: $D_{AB} = f(T, P, x_A)$
 \rightarrow between phase. \uparrow interface

Mass Transfer coefficients describe fluxes over

measurable concentration gradients & are directly associated with a boundary. Accounts for mass transfer by diffusion or convection.

$$K = f(C_R, Sc)$$

K_c - convective mass transfer coefficient

why is $Pr_{liq} > Pr_{gas}$

5

$$Pr = \frac{C_p \mu}{k}$$

$H_2O > air$

$$\mu_{liq} > \mu_{gas}$$

$$1.794 \text{ cp} > .018$$

$$C_{p,liq} > C_{p,gas}$$

$$1.50 \text{ Btu/lb}^\circ\text{F} > 0.2$$

$k_{liq} > k_{gas}$ but not much

$$0.330 > 0.017 \text{ Btu/hr ft}^\circ\text{F}$$

for H_2O for air

6. Sherwood Number = dimensionless # for mass transfer at slow mass transfer rates

6

$$\frac{K_r D}{C D_{AB}} \quad \text{in BSL} \quad \frac{(\text{mol/L}^2 \text{ s}) L}{(\text{mol/L}^2) (L^2/\text{s})}$$

$$N_{Sh} = \frac{K_r D}{D_{AB}} \quad (\text{m MESH}) \quad \frac{(\text{cm/s})(\text{cm})}{\text{cm}^2/\text{sec}}$$

\uparrow vol. diff. driving

$$N_{Sh} = \frac{K_r D}{D_{AB}} = \frac{\text{convective transport}}{\text{diffusive transport}}$$

$$\text{MESH} \frac{\text{lb}}{\text{ft}^2 \text{ sec}}$$

used to find mass transfer coefficients from making analogies between N_{Nu} + N_{Sh} in order to use heat transfer correlations such as $N_{Nu} = 2.0 + 0.6 Pe^{1/2} Pr^{1/3}$

$$Pe = \frac{C_p u L}{k}$$

$$N_{Sh} = 2.0 + 0.6 Pe^{1/2} Sc^{1/3}$$

$$Sc = \frac{\mu}{\rho D_{AB}}$$

a the Chilton - Colburn analogy

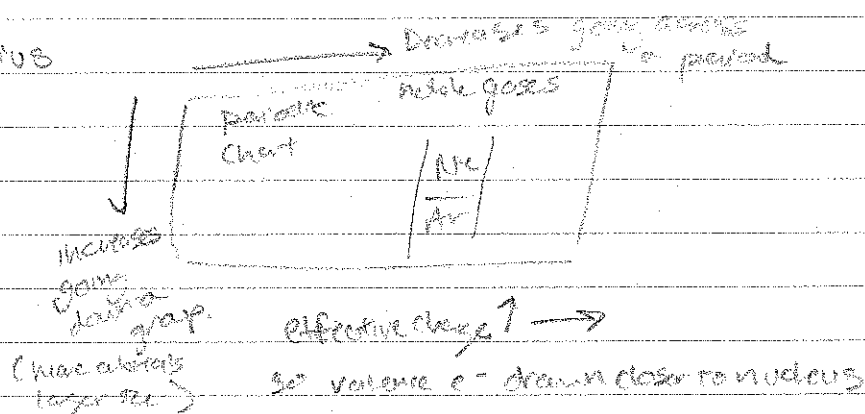
$$j_H = j_D$$

$$Nu_{1/2} = 1/2 = Sh / D_{AB} \text{ s. } 1/2$$

For Ne + Ar, which has the larger:

7.

(a) Atomic radius



∴ Ar > Ne's atomic radius

(b) Diffusivity = f(Temp, pressure, composition)

D_{AB} for smaller Ne > D_{AB} for Ar

Since $D_{AB} \propto 1/\text{MW}$.

The Chapman-Enskog formula $D_{AB} \propto (1/\text{MW})$

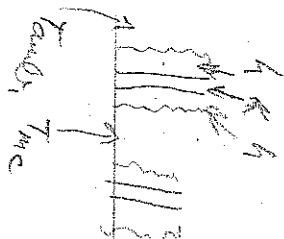
(b) viscosity $\propto \text{MW}$

$\mu_{Ar} > \mu_{Ne}$

(c) heat capacity are the same for monatomic ideal gases

based on structure - $C_p = 20.786 \text{ J/mol K}$
for He, Ne, Ar, Kr,

(d) $N_{pr} = \frac{C_p \mu}{K}$ since $\mu_{Ar} > \mu_{Ne}$
is assume $N_{pr}(Ar) > N_{pr}(Ne)$



cautaten = $P_{\text{fluid}} = P_{\text{req}}$
 $P \downarrow \text{until } = P_{\text{req}}$ causing bubble formation (boiling)

Consider the problem of pumping oil down AK pipeline
 Given D, L & oil properties How would you calculate

(a) pump size

Pumps (msh)

Applying Bernoulli's Egn:

$$\frac{1}{\rho} \Delta P = \left(\frac{P_b}{\rho} + \frac{g z_b}{g_c} + \frac{V_b^2}{2g_c} \right) - \left(\frac{P_a}{\rho} + \frac{g z_a}{g_c} + \frac{V_a^2}{2g_c} \right)$$

From notes:

Need to know flow rate Q

$Q \rightarrow Re \rightarrow f \rightarrow \Delta P \rightarrow \text{Chart}$

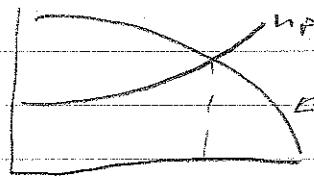
ΔP vs $Q \rightarrow \text{size of pump}$
 need loss

$$W_p = g h_p$$

$$h_p = \frac{\Delta P}{\rho g} + \frac{\Delta V^2}{2g} + \Delta z + \sum K_L \frac{V^2}{2g} + \sum \frac{f L}{D} \frac{V^2}{2g}$$

system curve

head, m



$Q \text{ m}^3/\text{s (Pizcheg)}$

loss coefficients

"pump curve"

2800 rpm centrifugal pump

For given

Q & Re h_p

find pump.

f = friction factor

(Bernoulli's)

$$F_R = \Delta K f = A f \int V^2 = (\Delta P + \rho g \Delta h) A \text{ for tube}$$

$$= \frac{4}{3} \pi R^3 (\rho g - \rho g)$$

mass - buoyancy

Laminar flow = $f = 16/Re$ in tube

Turbulent \Rightarrow calculate f/D & use chart of f vs Re

For spheres:

$10^{-3} < Re < 1 \Rightarrow$ Stokes Law = $f = 24/Re$

$1 < Re < 1000 \Rightarrow$ intermediate Law = $f = 16/Re$

(8b) Heat loss from a pipe

$$\frac{q}{A_o} = \frac{T_h - T_c}{\frac{1}{h_i} \left(\frac{D_o}{D_i} \right) + \frac{x_w}{k_m} \left(\frac{D_o}{D_i} \right) + \frac{1}{h_o}} = U_o (T_h - T_c)$$

when $A_o \approx A_i$ (thin walled) $U_o \approx U_i$

$$\text{or } U_o \approx U_i = \frac{1}{1/h_o + x_w/k + 1/h_i}$$

Calculate Re to determine laminar/turbulent

$$Re = \frac{\rho D v}{\mu}$$

For laminar flow:

$$N_{Nu} = \frac{h_i D}{k} = \frac{N_{Nu}}{T_i} \ln \left(\frac{T_w - T_i}{T_w - T_o} \right)$$

ASSUME
const. wall
temperature
+ log mean avg.
ext. temp.

For turbulent flow: $Re > 2100$

Dittus-Boelter Egn.

$$\frac{h_i D}{k} = 0.023 \left(\frac{D v}{\mu} \right)^{0.8} \left(\frac{c_p \mu}{k} \right)^{1/3}$$

For long tubes
w/ sharp-edged
entrances,

$$N_{Nu} = 0.023 (Re)^{0.8} (Pr)^{1/3}$$

if $\mu = \mu_w$ throughout use Sieder-Tate

$$\frac{h_i D}{k} = 0.023 Re^{0.8} Pr^{1/3} \left(\frac{\mu}{\mu_w} \right)^{0.14}$$

To find T_w : iterative calculation based on

$$\frac{DT_{avg}}{U} = \frac{\Delta T_i}{D_o/D_i h_i} = \frac{\Delta T_w}{(x_w/k_m) D_o/D_i} = \frac{\Delta T_o}{1/h_o}$$

$$\text{Method: } \Delta T_i = \frac{D_o/D_i h_i}{1/h_o} \Delta T = \frac{1/h_i}{1/h_o + 1/h_i + x_w/k_m} \Delta T$$

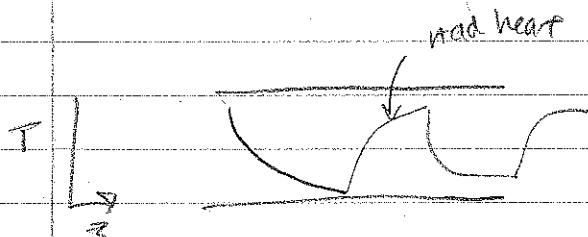
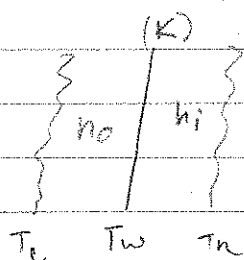
Free convection:

$$Nu = f(Pr, Gr) \approx 0.53 (Nu_{Gr} Nu_{Pr})^{0.25} \quad \text{for } Pr > 0.7$$

$$Gr = \frac{D_o^3 \rho^2 \beta g \Delta T}{\mu^2}$$

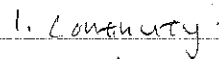
$$\beta = \frac{(\Delta V / \Delta T) \rho}{V} \quad I.G. \approx 1 \quad \text{fluid} = \frac{\Delta Q / \Delta T}{T} = \frac{P_1 - P_2}{F_{avg} (T_2 - T_1)}$$

Temperature Profile



9.

Find Flaw, Q



$$A V_1 = A_0 V_2$$

2. Bernoulli's Egn.

Denn's Bernoulli Equation:

$$\text{Isothermal} \quad \text{ideal gas:} \quad \Delta G = \frac{RT}{m_w} \ln \frac{P_2}{P_1}$$

$$\text{Incompressible} = \frac{P_2 - P_1}{\rho}$$

$$\textcircled{A} \quad \frac{\alpha^2 \langle v \rangle^2}{2} + gh = \frac{\alpha \langle v \rangle^2}{2} + gh - \int_0^R \frac{dP}{P} + \delta W_s - \frac{1}{2} v^2$$

$\frac{M^2}{S^2} = \frac{1}{K_g}$

 $\frac{10 \times 10^2}{100 \times 10^2}$

Let terms of heads: $(\frac{1}{2} A \text{ by } g.)$

$$\frac{\alpha_z \langle V \rangle_z^2}{2g} + h_z + \frac{P_z}{\rho_z g} = \frac{\alpha_1 \langle V_1 \rangle_1^2}{2g} + h_1 + \frac{P_1}{\rho_1 g} + \delta w_s - \frac{dv}{g}$$

$$L_v = \underbrace{\bar{L}}_{\text{Pipe length}} \underbrace{\langle v \rangle^2}_{D} L_f + \sum \frac{1}{2} v^2 K$$

↑
Atmgs
↑

of velocity heads

Thus the B.E. is:

For incompressible fluid

$$L_v = \frac{1}{2} V_z^2 \left(1 - \left(\frac{A_0}{A} \right)^2 \right) + \frac{\rho P}{\rho} \approx \frac{1}{2} K V^2$$

$$K = 1.6 \left[1 - \left(\frac{A_0}{A} \right)^2 \right]$$

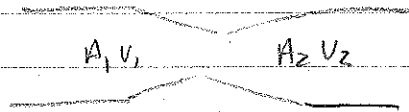
$$V_2^2 = 2 \left(\frac{P_1 - P_2}{\rho \left(1 - \left(\frac{u_0}{u_1} \right)^2 + 1.6 \pm 1.6 \left(\frac{u_0}{u_1} \right)^2 \right)} \right)$$

$$V_2 = \sqrt{\frac{2(P_1 - P_2)}{\rho(1 + K - (A_0/A)^2)}} + Q_c A_0 V_2$$

Venturi Meter

96

like orifice but less head loss



From
FLUIDS
TABLE:

$Q = Q_{\text{orifice meter}} = \overset{\substack{\text{0.62 for orifice}}}{K} A_2 \sqrt{2g \Delta h}$ $\Delta h = \Delta P / \rho g$

$$\frac{1}{2} V_1^2 + g h_1 = \frac{1}{2} V_2^2 + g h_2 + \int \frac{dP}{\rho} + \cancel{8 \frac{L}{D} V_2^2} + \frac{1}{2} K V_2^2$$

$$\frac{1}{2} V_2^2 (1 - (A_2/A_1)^2 + K) = \frac{P_1 - P_2}{\rho}$$

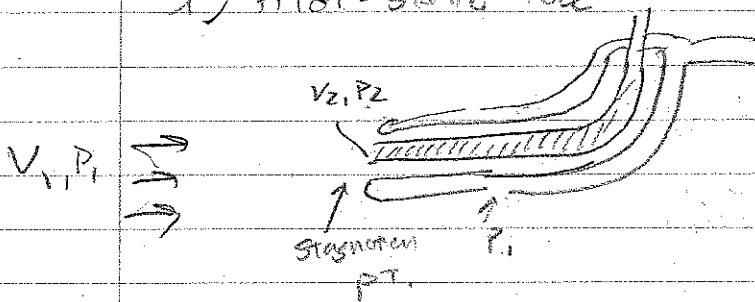
$$V_2 = \sqrt{\frac{2(P_1 - P_2)}{\rho (1 - (A_2/A_1)^2 + K)}}$$

$Q = A_2 V_2 =$ \uparrow look up for venturi meters.

9c - Pitot tube - measures velocity by Δp

9c

i) Pitot-static tube



$$\frac{1}{2} V_1^2 + g z_1 = \frac{1}{2} V_2^2 + g z_2 + \int \frac{dP}{\rho}$$

$$\frac{1}{2} V_1^2 = \frac{P_2 - P_1}{\rho}$$

$$V_1 = \sqrt{\frac{2(P_2 - P_1)}{\rho}}$$

from manometer.

can neglect Δv except at low Re

From Denn's book

(10) Bernoulli's equation

$$\frac{\alpha}{2} V_1^2 + g z_1 = \frac{\alpha}{2} V_2^2 + g z_2 + \int_{P_1}^{P_2} \frac{dp}{\rho} + \sum W_s + l_v \quad \left\{ \right.$$

$$\int \frac{dp}{\rho} = \frac{P_2 - P_1}{\rho} \quad \text{incomp. fluids}$$

Assuming
turbulent flow

$\alpha = 1$ uniform
velocity over cross-section

$$l_v = \frac{1}{2} K V^2 \quad \text{viscous losses}$$

$$\frac{1}{2} V_1^2 + g z_1 + \frac{P_1}{\rho} = \frac{1}{2} V_2^2 + g z_2 + \frac{P_2}{\rho} + l_v$$

(b) Hagen-Poiseuille Egn.

For laminar flow in a pipe, $Re < 2100$

$$Q = \frac{\pi}{128} \frac{|\Delta P| D^4}{L \eta} \quad \left(\text{from } f = \frac{16}{Re} \right)$$

(c) Stokes Law

For $Re < 1$, Drag Force on a sphere is $F_D = 3\pi\eta V D$ (Denn's book)

\Uparrow same

$$F_R = 6\pi\mu R V_{\infty} \quad (\text{BSL})$$

(d) continuity equation (Denn) Pg 85

$$\frac{d}{dt} \int_{z_1}^{z_2} \langle \rho \rangle A dz = \langle \rho V \rangle_1 A_1 - \langle \rho V \rangle_2 A_2$$

Single fluid

$$\frac{d}{dt} \int_{z_1}^{z_2} \rho A dz = \rho V_1 A_1 - \rho V_2 A_2$$

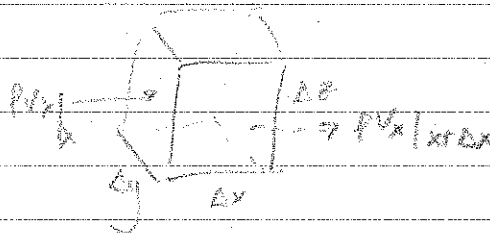
if incompressible $\frac{d}{dt} \int_{z_1}^{z_2} A dz = A_1 V_1 - A_2 V_2$

Give the following:

(10)

(b) Equation of Continuity

$$\Delta \rho = \rho_{in} - \rho_{out}$$



$$\Delta x \Delta y \Delta z \frac{\partial \rho}{\partial t} = \Delta y \Delta z (\rho v_x|_x - \rho v_x|_{x+\Delta x}) + \Delta x \Delta z (\rho v_y|_y - \rho v_y|_{y+\Delta y}) + \Delta x \Delta y (\rho v_z|_z - \rho v_z|_{z+\Delta z})$$

by $\Delta x \Delta y \Delta z$, & take limits as $\Delta x, \Delta y, \Delta z \rightarrow 0$

$$\frac{\partial \rho}{\partial t} = - \left(\frac{\partial (\rho v_x)}{\partial x} + \frac{\partial (\rho v_y)}{\partial y} + \frac{\partial (\rho v_z)}{\partial z} \right) \quad \text{rate of } \Delta \text{ decreasing as a fixed point.}$$

$$\text{or } \frac{\partial \rho}{\partial t} = - \nabla \cdot \rho \mathbf{v} \quad \text{"divergence of } \rho \mathbf{v} \text{"}$$

or, for the Δ in ρ as seen from a path following the fluid motion:

$$\frac{D\rho}{Dt} = - \rho (\nabla \cdot \mathbf{v})$$

Simplifications:

$\rho = \text{const.}$ (incompressible) then

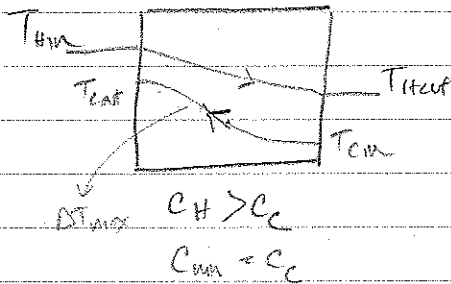
$$\boxed{\nabla \cdot \mathbf{v} = 0}$$

- when $\rho = \text{const.}$

11. NTU? How to calculate.

11.

(WWW - Pg 414)



For counterflow

ΔT_{max} occurs in fluid with

Smallest C_p

$$q = m C_p (\Delta T)$$

↓ ↑

effectiveness factor $\epsilon = \frac{\text{Actual h.t.}}{\text{max. h.t. possible}} = \frac{C_{max} \left(\begin{matrix} \Delta T_H \Rightarrow C_{min} = C_C \\ \text{OR} \\ \Delta T_C \Rightarrow C_{min} = C_H \end{matrix} \right)}{C_{min} (T_{Hin} - T_{Cin})}$

OR
$$\epsilon = \frac{q}{C_{min} (T_{Hin} - T_{Cin})}$$

↓ occurs in fluid w/ max. ΔT

To find ϵ

$$q = C_c (T_{C2} - T_{C1}) = UA \Delta T_{lm}$$

$$NTU = \frac{AU}{C_{min}}$$

1. To find q

1. Calculate $C_{min} + C_{max}$

2. Calculate $NTU = UA/C_{min}$

3. Use Figure to calculate ϵ (or derive eqn)

4. then calc. $q = \epsilon C_{min} (T_{Hin} - T_{Cin})$

then use q to calculate

$T_{Hout} + T_{Cout}$

McCabe Thiele Diagram. - compare # of
ideal plates needed to

12

2 operating lines:

achieve certain concn. Δ

1. Rectifying Section (Above feed plate)

$$y_{n+1} = \frac{L_n}{L_n + D} x_n + \frac{D x_D}{L_n + D}$$

$$\text{For } R_D = \frac{L}{D}$$

Assumes constant
molar overflow
(similar ΔH_{vap} 's)

GAS

$$P_i = 10^2$$

13

(BSL)

	<u>T</u>	<u>P</u>
P_{AB}	\uparrow	\downarrow
μ	\uparrow (liq \downarrow)	\uparrow (liq \leftrightarrow)
K	\uparrow (liq \downarrow)	\uparrow
C_p	\uparrow	\leftrightarrow $C_p = \left(\frac{\partial H}{\partial T}\right)_P$
h	\uparrow as \uparrow	\uparrow as \leftrightarrow
ν	\uparrow	\uparrow

coeff (pressure) whenever the substance is independent of pressure.
(valid for i.g. liquids, solids, + low pressure gases)

Reynold's Analogy \Rightarrow

transport energy distribution
the solid-fluid interface

similar ratio for momentum transport.

Assume

$$1. \frac{z}{\alpha} \equiv N_{Pr} = \frac{\epsilon_m}{\epsilon_{it}}$$

2. q/A varies linearly w/r

3. the point where $T_{loc} = T_{bulk}$ is the same as where $u = \bar{u}$

Then

$$\frac{Nu}{Re Pr} = \frac{f}{2} \quad \text{for } Pr = 1$$

$$\frac{h D}{k} = \frac{K}{k_{eff} C_p \rho}$$

$$\frac{h}{C_p \rho \nu} = N_{SE} = \frac{f}{2}$$

flow at high Re in straight round tubes

Colburn Analogy: better than Re analogy ($Pr=1$)

146

For $Re > 10,000$
long smooth tube \rightarrow

since applies to $0.6 < Pr < 100$

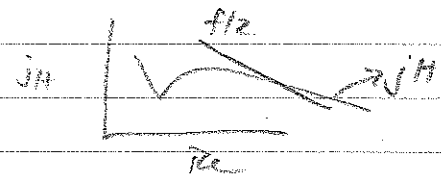
Colburn j factor:

$$j_H = \frac{h}{C_{p\rho V}} N_{Pr}^{2/3} \left(\frac{\mu_w}{\mu} \right)^{0.14} = \frac{f}{2} \quad \text{to find } f$$

$$j_H = 0.023 N_{Re}^{-0.2} \quad (\text{for heating } h)$$

j_H vs. Re at $Re > 10,000$

the same as $f/2$ vs Re



$$\frac{f}{2} = \frac{h}{C_{p\rho V}} A^{2/3}$$

$$\text{Power} = \dot{P} = Q|\Delta P|$$

friction factor?

15

$$\boxed{F_k = fKA} = fK \frac{1}{2} V^2 \quad (\text{BSC})$$

definition of f

$$K = \text{kinetic energy/volume} = \frac{1}{2} \rho V^2$$

A = wetted surface

Also from a force balance

on a tube:

$$F_k = \Delta P A + \rho g \Delta h (A) = (\Delta P + \rho g \Delta h) \pi r^2$$

$$f = \frac{(\Delta P + \rho g \Delta z) \pi r^2}{\frac{1}{2} \rho V^2 \cdot 2\pi r L} = \frac{(\Delta P + \rho g \Delta z)}{\frac{1}{2} \rho V^2}$$

Boundary Layer Equations

16.

Denn (202, 220, 286, 325)

Prandtl B.L. equations (BSL - 146, 331, 367, 609)

Describe the friction factor vs Re relation for a

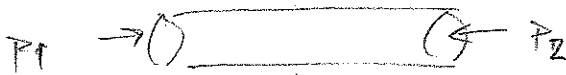
18

↓
Solve for f
Solve for V (Hagen-Poiseuille)
Relate f to V → Re

a) pipe

$$F_k = f K A = f \frac{1}{2} \rho V^2 (\pi R L)$$

$$k = \frac{1}{2} \rho V^2$$



$$F_k = (P_1 - P_2) A + \rho g (z_1 - z_2) A$$

$$f \frac{1}{2} V^2 (\pi R L) = \pi R^2 ((P_1 - P_2) + \rho g \Delta z)$$

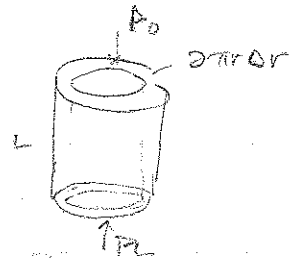
$$f = \frac{R (\Delta P + \rho g \Delta z)}{(\pi L) \frac{1}{2} \rho V^2} = \frac{D (\Delta P + \rho g \Delta z)}{4 L (\frac{1}{2} \rho V^2)}$$

$$Re = \frac{\rho D V}{\mu}$$

Need to find V_z

Assume s.s.

$$0 = \dot{m} - \dot{m}_{in} + \dot{m}_{out}$$



$$\frac{d}{dr} (r \tau_{rz}) = \left[\frac{(P_0 - P_L)}{L} + \rho g \right] r$$

For Newtonian Fluids: $\tau_{rz} = -\mu \frac{dv_z}{dr}$

Integrate

$$r \tau_{rz} = \left(\frac{P_0 - P_L}{L} + \rho g \right) \frac{r^2}{2} + C_1$$

$$\tau_{rz} = \left(\frac{P_0 - P_L}{L} + \rho g \right) \frac{r}{2} + \frac{C_1}{r}$$

$$\frac{dv_z}{dr} = -\frac{1}{\mu} \left(\frac{P_0 - P_L}{L} + \rho g \right) \frac{r}{2}$$

$$v_z = -\frac{1}{\mu} \left(\frac{P_0 - P_L}{L} + \rho g \right) \frac{r^2}{4} + C_2$$

$$r = R, v_z = 0$$

$$V_z = \frac{1}{4\mu} \left(\frac{\Delta P}{L} + \rho g \right) (R^2 - r^2)$$

18-cont.

$$\langle v_z \rangle = \frac{\int_0^{2\pi} \int_0^R v_z r dr d\theta}{\int_0^{2\pi} \int_0^R r dr d\theta} = \left(\frac{\Delta p + \rho g}{8\mu} \right) R^2$$

$$F = \frac{D}{4} \frac{(\Delta p/L + \rho g)}{\frac{1}{2} \rho V (\Delta p/L + \rho g) R^2} \frac{8\mu}{\rho V R^2} = \frac{4\mu}{\rho V R^2} = \frac{16\mu}{\rho V D^2}$$

f. = $\boxed{\frac{16}{Re}}$

(b) Sphere

$$F_K = \frac{1}{2} \rho V^2 f A = \pi r^2 \left(\frac{1}{2} \rho V_\infty^2 \right) f$$

using Stokes Law: $6\pi \mu R V_\infty = F_K$

$$f = \frac{6\pi \mu R V_\infty}{\pi r^2 \left(\frac{1}{2} \rho V_\infty^2 \right)} = \frac{24}{Re} \quad \text{for } Re < 0.1$$

Creeping flows

(c) using (BSL - 4.4-27)

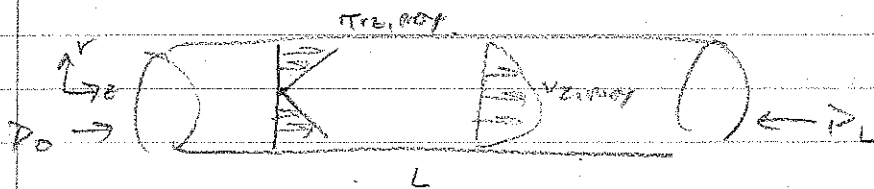
$$F_K = \text{drag force} = 1.328 \sqrt{\rho \mu L \underset{\substack{\uparrow \\ \text{width}}}{W}^2 V_\infty^2}$$

$$F_K = \frac{1}{2} \rho V_\infty^2 f A$$

$$f = \frac{F_K}{\frac{1}{2} \rho V_\infty^2 A}$$

Shear stress profile for a pipe

19.



Momentum balance - shell balance on $dr/2\pi r L$
 Acc \dot{m} - cut, + forces.

$$0 = (2\pi r L) (T_{rz}|_r - T_{rz}|_{r+dr}) + (P_0 - P_L)(2\pi r dr)$$

\div by $2\pi r L dr$ $\lim_{dr \rightarrow 0}$

$$-\frac{d(r T_{rz})}{dr} = (P_L - P_0)r$$

$$\frac{d(r T_{rz})}{dr} = \frac{(P_0 - P_L)r}{L}$$

$$r T_{rz} = \frac{(P_0 - P_L)L}{2} \frac{r^2}{2} + C_1$$

$$T_{rz} = \frac{(P_0 - P_L)r^2}{4L} = \frac{\Delta P}{4L} r^2$$

Dimensionless Parameters

20.

h.t.

$$Re = \frac{\rho D V}{\mu} = \frac{\text{inertial forces}}{\text{viscous forces}}$$

$$Pr = \frac{c_p \mu}{k} = \frac{\text{diffusivity of momentum}}{\text{thermal diffusivity}}$$

$$Nu = \frac{hL}{k}$$

mass

$$Sh = \frac{k^c L}{D_{AB}} = \frac{\text{mass transfer vel}}{\text{diffusion velocity}}$$

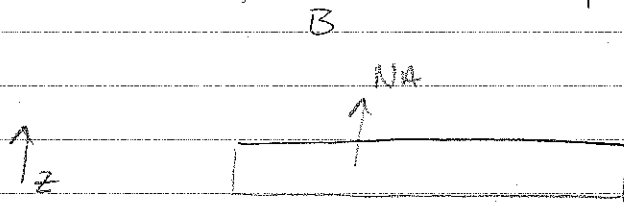
$$Sc = \frac{\mu}{\rho D} = \frac{\text{diff. of momentum}}{\text{diff. of mass}}$$

$$Gr = \frac{D^3 \Delta \rho g}{\rho \nu^2}$$

free convection

Pool of organic liquid - find rate of evaporation.

21



Fick's Law

$$N_A = -C \frac{D_{AB}}{z} \frac{dx_A}{dz} + x_A (N_A + N_B) \quad \text{Assume B insoluble in A}$$

$$N_A = -C \frac{D_{AB}}{(1-x_A)} \frac{dx_A}{dz} \quad \frac{\text{mol}}{\text{m}^2 \cdot \text{s}} = \frac{L^2}{L \cdot \text{s}}$$

Continuity eqn.

$$\frac{\partial C}{\partial t} = - \frac{\partial}{\partial z} (N_{A,z} + N_{B,z}) = 0 \quad \text{Assume S.S.}$$

$$0 = - \frac{\partial}{\partial z} \left(C \frac{D_{AB}}{(1-x_A)} \frac{dx_A}{dz} \right)$$

$$C = \frac{C D_{AB}}{(1-x_A)} \frac{dx_A}{dz}$$

$$C, z = -C D_{AB} \ln(1-x_A) + C_2$$

$$z=0, \quad x=$$

Diffusivity -

525, 526, 552,

Stokes - 514,

502, 507

22

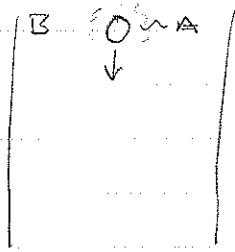
Temp profile in heat exchanger

23

Important phenomena during an underground explosion? 24

radiation =

$$q_{12} = A_{12} \sigma (T_1^4 - T_2^4)$$



25.

Concurrently

$$\frac{\partial C_A}{\partial t} = - \frac{\partial N_A}{\partial z} \quad \frac{\partial C_B}{\partial t} = - \frac{\partial N_B}{\partial z}$$

T_I = tower temp

T_i = initial drop temp.

How does drop temp Δ as it falls

$$\frac{\partial C}{\partial t} = \frac{\partial N_A}{\partial z} - T_i$$

For no fluid motion, All heat transfer by conduction

$$\nabla \cdot k \nabla T = \rho C_p \frac{\partial T}{\partial t}$$

$$k \nabla T = k \frac{\partial T}{\partial r} \vec{e}_r + \frac{k}{r} \frac{\partial T}{\partial \theta} \vec{e}_\theta + \frac{k}{r \sin \theta} \frac{\partial T}{\partial \phi} \vec{e}_\phi$$

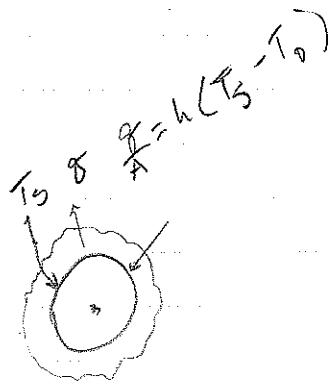
$$\nabla \cdot k \nabla T = \frac{1}{r^2} \frac{\partial}{\partial r} (r^2 k \frac{\partial T}{\partial r}) + \frac{1}{r \sin \theta} \frac{\partial}{\partial \theta} \left(\frac{k}{r} \frac{\partial T}{\partial \theta} \sin \theta \right) + \frac{1}{r \sin \theta} \frac{\partial}{\partial \phi} \left(\frac{k}{r} \frac{\partial T}{\partial \phi} \sin \theta \right)$$

$$\frac{1}{r^2} \frac{\partial}{\partial r} (r^2 k \frac{\partial T}{\partial r}) = \rho C_p \frac{\partial T}{\partial t}$$

I.e. $t=0$, $T = T_i$ for drop.

B.c. $r=0$ finite

$$r=R \quad h_A (T_s - T_r) = -k \frac{dT}{dr}$$

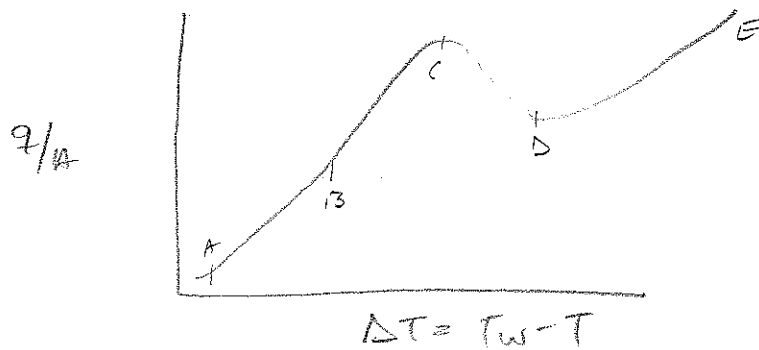


What is the angular dependence of the Nusselt number for a falling drop?

25.

Boiling Curve

27



- AB - natural convection $q = h A \Delta T$
- BC - nucleate boiling
- CD - transition boiling - so many bubbles form they create insulating layer of vapor on heating surface
- DE - film boiling

Given the free stream velocity & particle diameter calculate the B.L. thickness at a 45° Angle. What is the pressure at the forward & backward stagnation points? Why the difference? 28.

(Model as a flat plate?)

Derive the S.S. momentum balance for
fully developed laminar flow in a pipe.

29.

To find U for shell & tube heat exchangers.

1930

$$q = U_o A \Delta T_{lm} \quad (\text{MSH 434})$$

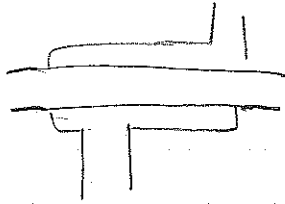
$$U_o = \frac{1}{\frac{1}{h_i} \frac{D_i}{D_o} + \frac{x}{K} \left(\frac{D_o}{D_i} \right) + \frac{1}{h_o}}$$

for thin walled

$$\frac{1}{U_o} = \frac{1}{h_i} + \frac{x}{K} + \frac{1}{h_o}$$

for $h_i \Rightarrow$ Dittus Boelter

for $h_o \Rightarrow$ Colburn Egn.



MSH (313)

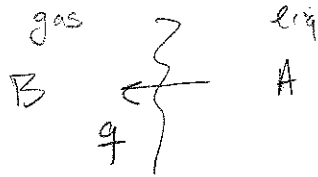
31

$$q = \dot{m}_c C_{p,c} (T_{c,o} - T_{c,i}) = \dot{m}_h C_{p,h} (T_{h,i} - T_{h,o}) = \frac{UA (\Delta T_1 - \Delta T_2)}{\ln \left(\frac{\Delta T_1}{\Delta T_2} \right)}$$

Wet Bulb/Dry Bulb Psychrometer.

32

more heat flux if gas is soluble in liquid



~~EMCD~~

Pick's Law:

$$N_A = -C D_{AB} \frac{dx_A}{dz} + x_A (N_A + N_B)$$

For EMCD $N_A = -N_B$

$$N_A = -C D_{AB} \frac{dx_A}{dz}$$

For stationary comp.

$$N_A = \frac{-C D_{AB} \frac{dx_A}{dz}}{(1-x)} > \text{EMCD}$$

$$\frac{\dot{q}}{A} = -k \Delta T + \sum N_i H_i$$

$$N_A (H_A - H_B) \text{ EMCD}$$

$$N_A H_A + N_B H_B \text{ stationary}$$

$$\dot{q}_s > \dot{q}_{\text{EMCD}}$$

Chilton - Colburn

$$j_H \equiv \frac{Nu}{Re Pr^{1/3}} = j_D \equiv \frac{Sh}{Re Sc} = f/2$$

$$= \frac{\frac{h}{k} \frac{D}{L}}{\frac{\rho V}{\mu} \frac{C_p \mu}{k}} = \frac{\frac{k L / D}{\rho V}}{\frac{\mu}{\rho V} \frac{\mu}{\rho V D}} = \frac{k L / D}{\mu} \frac{\rho V}{\mu}$$

Joule Thompson coefficient =

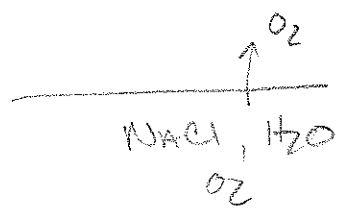
38.

$$= \left(\frac{dT}{dP} \right)$$

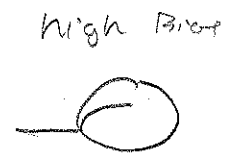
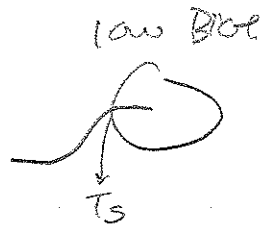
μ_J is negative

$C_p + W_J$ is positive

36.



Assume
 $k > hD$



60

$$Bi = \frac{hD}{k}$$

low Biot #:

$$q = \frac{h 4 \pi r^2 (T_s - T_\infty)}{3} = \rho C_p \frac{4}{3} \pi R^3 \frac{dT_s}{dt}$$

38,

$$\text{Lewis \#} = \frac{K}{\text{pepD}}$$

From Navier-Stokes Eqn.

Le3

$$\rho \frac{Dv}{Dt} = -\nabla P + \mu \nabla^2 v + \rho g$$

$$\rho \left[\frac{dv_z}{dt} + v_r \frac{\partial v_z}{\partial r} + v_z \frac{\partial v_z}{\partial z} \right]$$

$$= -\frac{dP}{dz} + \mu \left(\frac{d^2 v_z}{dz^2} + \frac{1}{r} \frac{d}{dr} \left(r \frac{dv_z}{dr} \right) \right)$$

$$\frac{dP}{dz} = \frac{\mu}{r} \frac{d}{dr} \left(r \frac{dv_z}{dr} \right)$$

$$v^* = v/V \quad r^* = r/L$$

$$\frac{1}{\mu L} \frac{dP}{dz} = \frac{1}{r} \frac{d}{dr} \left(r \frac{dv_z}{dr} \right)$$

$$\frac{1}{\mu} \frac{\Delta P L}{V} = \frac{1}{r^*} \frac{d}{dr^*} \left(r^* \frac{dv^*}{dr^*} \right)$$

Bubble leaving straw



Summing Forces

$$\frac{4}{3}\pi R^3 \rho$$

$$\rho g = \sigma \cdot 2\pi R$$

