

- 1. Draw the Lewis structure for XeF<sub>4</sub>.
- 2. Predict the VSEPR geometry at xenon.
- 3. Determine a hybridization scheme to rationalize the geometry.



Draw the MO diagram for  $He_2$ . Calculate the bond order and show the electronic configuration of  $He_2$  molecule.

Draw the MO diagram for  $H_2^-$ . Calculate the bond order and show the electronic configuration of  $H_2^-$  molecule.

Draw the MO energy diagram for  $O_2^+$ . What is the bond order? Is it paramagnetic or diamagnetic? Compare the bond length of  $O_2^-$ ,  $O_2$  and  $O_2^+$ 

Draw the MO energy diagram for BC. What is the bond order? What is the valence electronic configuration?

Fill in the MO energy diagram for CO. What is the bond order?

