

Mixed Ionic, Polyatomic and Covalent: * note that acids have two correct names

1. Li_2O	lithium oxide	15. Hydrobromic acid	HBr
2. HCl	hydrogen chloride	16. cesium oxide	Cs_2O
3. $\text{Pb}(\text{NO}_3)_2$	lead (II) nitrate	17. silicon tetrafluoride	SiF_4
4. OF_2	oxygen fluoride	18. mercury(II) nitride	Hg_3N_2
5. NH_4ClO_3	ammonium chlorate	19. potassium selenide	K_2Se
6. CuO	copper (II) oxide	20. ammonia	NH_3
7. HI	hydroiodic acid	21. lead(II) nitrate	$\text{Pb}(\text{NO}_3)_2$
8. H_2O	water	22. magnesium carbide	Mg_2C
9. Ag_2S	silver sulfide	23. ammonium hydroxide	NH_4OH
10. PbS	lead (II) sulfide	24. cobalt (III) phosphate	CoPO_4
11. $\text{Sn}(\text{SO}_4)_2$	tin (IV) sulfate	25. carbon tetrachloride	CCl_4
12. N_2S_3	dinitrogen trisulfide	26. lithium chloride	LiCl
13. CuOH	copper (I) hydroxide	27. copper(I) nitrate	CuNO_3
14. K_3N	potassium nitride	28. hydrogen peroxide	H_2O_2