

Acids and Bases

Oxyacids

Acid	Name
$\text{HNO}_{3(\text{aq})}$	nitric acid
$\text{HNO}_{2(\text{aq})}$	nitrous acid
$\text{H}_2\text{SO}_{4(\text{aq})}$	sulfuric acid
$\text{H}_2\text{SO}_{3(\text{aq})}$	sulfurous acid
$\text{H}_3\text{PO}_{4(\text{aq})}$	phosphoric acid
$\text{HC}_2\text{H}_3\text{O}_{2(\text{aq})}$	acetic acid
$\text{HClO}_{4(\text{aq})}$	perchloric acid
$\text{HBrO}_{4(\text{aq})}$	perbromic acid
$\text{HIO}_{4(\text{aq})}$	periodic acid
$\text{HClO}_{3(\text{aq})}$	chloric acid
$\text{HBrO}_{3(\text{aq})}$	bromic acid
$\text{HIO}_{3(\text{aq})}$	iodic acid
$\text{HClO}_{2(\text{aq})}$	chlorous acid
$\text{HClO}_{(\text{aq})}$	hypochlorous acid
$\text{HBrO}_{(\text{aq})}$	hypobromous acid
$\text{HIO}_{(\text{aq})}$	hypoiodous acid
$\text{HFO}_{(\text{aq})}$	hypofluorous acid

Concentrated Reagents*

Reagent	Formula	Molar mass (g/mol)	Concentration (mol/L)	Concentration (mass %)
acetic acid	$\text{HC}_2\text{H}_3\text{O}_{2(\text{aq})}$	60.05	17.45	99.8
carbonic acid	$\text{H}_2\text{CO}_{3(\text{aq})}$	62.03	0.039	0.17
formic acid	$\text{HCOOH}_{(\text{aq})}$	46.03	23.6	90.5
hydrobromic acid	$\text{HBr}_{(\text{aq})}$	80.91	8.84	48.0
hydrochloric acid	$\text{HCl}_{(\text{aq})}$	36.46	12.1	37.2
hydrofluoric acid	$\text{HF}_{(\text{aq})}$	20.01	28.9	49.0
nitric acid	$\text{HNO}_{3(\text{aq})}$	63.02	15.9	70.4
perchloric acid	$\text{HClO}_{4(\text{aq})}$	100.46	11.7	70.5
phosphoric acid	$\text{H}_3\text{PO}_{4(\text{aq})}$	98.00	14.8	85.5
sulfurous acid	$\text{H}_2\text{SO}_{3(\text{aq})}$	82.08	0.73	6.0
sulfuric acid	$\text{H}_2\text{SO}_{4(\text{aq})}$	98.08	18.0	96.0
ammonia	$\text{NH}_{3(\text{aq})}$	17.04	14.8	28.0
potassium hydroxide	$\text{KOH}_{(\text{aq})}$	56.11	11.7	45.0
sodium hydroxide	$\text{NaOH}_{(\text{aq})}$	40.00	19.4	50.5

*Typical concentrations of commercial concentrated reagents

Acid–Base Indicators

Common Name	Colour of $\text{HIn}_{(\text{aq})}$	pH range	Colour of $\text{In}^{-}_{(\text{aq})}$	Common name	Colour of $\text{HIn}_{(\text{aq})}$	pH range	Colour of $\text{In}^{-}_{(\text{aq})}$
methyl violet	yellow	0.0 – 1.6	blue	<i>p</i> -nitrophenol	colourless	5.3 – 7.6	yellow
cresol red (acid range)	red	0.2 – 1.8	yellow	litmus	red	6.0 – 8.0	blue
cresol purple (acid range)	red	1.2 – 2.8	yellow	bromothymol blue	yellow	6.2 – 7.6	blue
thymol blue (acid range)	red	1.2 – 2.8	yellow	neutral red	red	6.8 – 8.0	yellow
tropeolin oo	red	1.3 – 3.2	yellow	phenol red	yellow	6.4 – 8.0	red
orange iv	red	1.4 – 2.8	yellow	<i>m</i> -nitrophenol	colourless	6.4 – 8.8	yellow
benzopurpurine-48	violet	2.2 – 4.2	red	cresol red	yellow	7.2 – 8.8	red
2,6-dinitrophenol	colourless	2.4 – 4.0	yellow	<i>m</i> -cresol purple	yellow	7.6 – 9.2	purple
2,4-dinitrophenol	colourless	2.5 – 4.3	yellow	thymol blue	yellow	8.0 – 9.6	blue
methyl yellow	red	2.9 – 4.0	yellow	phenolphthalein	colourless	8.0 – 10.0	red
congo red	blue	3.0 – 5.0	red	α -naphtholbenzein	yellow	9.0 – 11.0	blue
methyl orange	red	3.1 – 4.4	orange	thymolphthalein	colourless	9.4 – 10.6	blue
bromophenol blue	yellow	3.0 – 4.6	blue-violet	alizarin yellow r	yellow	10.0 – 12.0	violet
bromocresol green	yellow	4.0 – 5.6	blue	tropeolin o	yellow	11.0 – 13.0	orange-brown
methyl red	red	4.4 – 6.2	yellow	nitramine	colourless	10.8 – 13.0	orange-brown
chlorophenol red	yellow	5.4 – 6.8	red	indigo carmine	blue	11.4 – 13.0	yellow
bromocresol purple	yellow	5.2 – 6.8	purple	1,3,5-trinitrobenzene	colourless	12.0 – 14.0	orange
bromophenol red	yellow	5.2 – 6.8	red				