

Constants and conversions

1 atm	= 101.325 kPa = 1.013 25 bar = 14.696 psi
N_A	$6.022 \times 10^{23} \text{ mol}^{-1}$
e	$1.602 \times 10^{-19} \text{ C}$
1 eV	$1.602 \times 10^{-19} \text{ J}$
ϵ_0	$8.854 \times 10^{-12} \text{ F m}^{-1}$
R	$8.314 \text{ J mol}^{-1} \text{ K}^{-1}$
	$0.082 067 \text{ L atm mol}^{-1} \text{ K}^{-1}$
0 °C	273.15 K
k	$8.62 \times 10^{-5} \text{ eV atom}^{-1} \text{ K}^{-1}$
	$1.38 \times 10^{-23} \text{ J atom}^{-1} \text{ K}^{-1}$
F	$96 486 \text{ C mol}^{-1}$
h	$6.626 \times 10^{-34} \text{ J s}$
	$4.136 \times 10^{-15} \text{ eV s}$
c	$2.99 \times 10^8 \text{ m s}^{-1}$
g	9.81 m s^{-2}

Microstructure

$LD = \frac{\#}{\text{Length}}$	$LPF = \frac{\text{length of atoms}}{\text{length of vector}}$
$PD = \frac{\#}{\text{Area}}$	$PPF = \frac{\text{area of atoms}}{\text{area of plane}}$
$V = \frac{4}{3}\pi r^3$	$A = \pi r^2$
$A_{\text{triangle}} = \frac{1}{2}bh$	$\rho = \frac{nA}{V_C N_A}$
$\rho = \frac{n_A A_A + n_C A_C}{V_C N_A}$	$APF = \frac{V_s}{V_C}$
$N = \frac{NA\rho}{A}$	$N_V = N \exp(-\frac{Q_V}{kT})$
$a = 2\sqrt{2}R$	$a = \frac{4}{\sqrt{3}}R$
$d_{\text{hkl}} = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$	$n\lambda = 2d_{\text{hkl}} \sin \theta$
$n_n = \frac{M_n}{m}$	$n_w = \frac{M_w}{m}$

Mechanical Behaviour

$\sigma = \frac{F}{A_0}$	$\epsilon = \frac{\Delta l}{l_0}$
$\sigma = E\epsilon$	$\sigma_{3\text{-point}} = \frac{3FL}{2wh^2}$
$\sigma_T = \sigma(1 + \epsilon)$	$\epsilon_T = \ln(1 + \epsilon)$
$\sigma_T = \frac{F}{A_i}$	$\sigma_T = K\epsilon_T^n$
$E = 2G(1 + \nu)$	$\nu = -\frac{\epsilon_x}{\epsilon_z} = -\frac{\epsilon_y}{\epsilon_z}$

Magnetic Behaviour

$H = \frac{NI}{L}$	$B_0 = \mu_0 H$
$M = \chi_m H$	$B = \mu_0 H + \mu_0 M$
$B = (1 + \chi_m)\mu_0 H$	$\mu_B = \frac{e\hbar}{2m_e} = \beta$
$\beta = 9.27 \times 10^{-24} \text{ Am}^2$	

Electrical Behaviour

$$\sigma = n|e|\mu_e + p|e|\mu_h \quad \sigma = n|e|\mu_e$$
$$\sigma = p|e|\mu_h$$

Electrochemistry

$$E = E^\circ - \frac{RT}{nF} \ln Q \quad I = \frac{nC}{t}$$
$$E_{\text{at } 25^\circ\text{C}} = E^\circ - \frac{0.0592}{n} \ln Q$$
$$w = nFE^\circ$$

Thermodynamics

$$PV = nRT \quad \Delta U = q + w$$
$$\Delta U = q - P_{\text{ext}}\Delta V \quad H \equiv U + PV$$
$$G \equiv H - TS \quad \Delta S = \frac{q_{\text{rev}}}{T}$$
$$\text{constant T: } \Delta G = \Delta H - T\Delta S$$
$$q = mc\Delta T \quad q = nC_P\Delta T$$
$$\text{For } aA + bB \rightarrow cC + dD, Q = \frac{a_C^c a_D^d}{a_A^a a_B^b}$$
$$\Delta_r G = \Delta G^\circ + RT \ln Q$$
$$\Delta_r H^\circ = (\sum v_i \Delta_{f,i} H^\circ)_{\text{prod.}} - (\sum v_i \Delta_{f,i} H^\circ)_{\text{react.}}$$
$$\Delta_r S^\circ = (\sum v_i \Delta_{f,i} S^\circ)_{\text{prod.}} - (\sum v_i \Delta_{f,i} S^\circ)_{\text{react.}}$$
$$W_{\text{phase}} = \frac{\text{length of opp. side of lever}}{\text{total length of lever}}$$
$$E = h\nu = \frac{hc}{\lambda}$$

Specific heats and heat capacities

Substance	$c \left(\frac{\text{J}}{\text{g}\cdot\text{K}} \right)$	$C_P \left(\frac{\text{J}}{\text{mol}\cdot\text{K}} \right)$
Air(g)	1.0	-
CO ₂ (g)	0.843	37.1
H ₂ (g)	14.304	28.836
H ₂ O(g)	2.03	36.4
H ₂ O(l)	4.184	75.3
H ₂ O(s)	2.09	37.7
NaCl	0.853	50.5
O ₂ (g)	0.918	29.378

Temperatures and enthalpies of phase changes

Substance	M.P. (°C)	$\Delta_{fus}H$ $\frac{\text{kJ}}{\text{mol}}$	B.P. (°C)	$\Delta_{vap}H$ $\frac{\text{kJ}}{\text{mol}}$
Al	658	10.6	2467	284
Ca	851	9.33	1487	162
CH ₄	-182	0.92	-164	8.18
H ₂ O	0	6.01	100	40.7
Fe	1530	14.9	2735	354

Standard formation enthalpy, standard entropy and standard formation Gibbs energy at 298.15 K

Species	$\Delta_f H^\circ$ $\left(\frac{\text{kJ}}{\text{mol}} \right)$	S° $\left(\frac{\text{J}}{\text{mol}\cdot\text{K}} \right)$	$\Delta_f G^\circ$ $\left(\frac{\text{kJ}}{\text{mol}} \right)$
C	0	5.74	0
CH ₄ (g)	-74.81	186.2	-50.75
C ₂ H ₂ (g)	-83.9	200.93	-
C ₃ H ₈ (g)	-103.8	269.9	-23.49
CaC ₂ (s)	-59.8	70.3	-
CaO(s)	-635	38.1	-
CaF ₂ (s)	-1225	68.87	-1162
CaF ₂ (l)	-1186	92.6	-
Ca(OH) ₂ (s)	-987.0	83.0	-
CO ₂ (g)	-393.5	213.6	-394.4
Cu ₂ O(s)	-168.6	93.1	-
Cu ₂ O(l)	-154.79	-	-
Cu(s)	-	33.2	-
Fe(s)	0	27.3	0
Fe ₂ O ₃ (s)	-824.2	87.4	-
H ₂ (g)	-	130.68	-
H ₂ O(g)	-241.8	188.7	-228.6
H ₂ O(l)	-285.8	69	-
O ₂ (g)	0	205.0	0

Miscellaneous enthalpies

Substance	Reaction	$\Delta H \left(\frac{\text{kJ}}{\text{mol}} \right)$
F ₂	$F_2 \rightarrow F(g)$	157
F	$F(g) \rightarrow F^-(g)$	-328
Ca	$Ca(g) \rightarrow Ca^{2+}(g)$	1734
NaCl	$NaCl(s) \rightarrow Na^+(aq) + Cl^-(aq)$	3.9

Scott Ramsay, December 2024

IUPAC Periodic Table of the Elements

<div>1<div>1 H hydrogen 1.0080 ± 0.0002</div></div>												<div>2<div>2 He helium 4.0026 ± 0.0001</div></div>																																																																																																																																																																																																											
<div>3<div>3 Li lithium 6.94 ± 0.06</div></div>												<div>4<div>4 Be beryllium 9.0122 ± 0.0001</div></div>												<div>Key:<div>atomic number Symbol name abridged standard atomic weight</div></div>												<div>13<div>5 B boron 10.81 ± 0.02</div></div>												<div>14<div>6 C carbon 12.011 ± 0.002</div></div>												<div>15<div>7 N nitrogen 14.007 ± 0.001</div></div>												<div>16<div>8 O oxygen 15.999 ± 0.001</div></div>												<div>17<div>9 F fluorine 18.998 ± 0.001</div></div>												<div>18<div>10 Ne neon 20.180 ± 0.001</div></div>																																																																																																																							
<div>11<div>11 Na sodium 22.990 ± 0.001</div></div>												<div>12<div>12 Mg magnesium 24.305 ± 0.002</div></div>												<div>3</div>												<div>4</div>												<div>5</div>												<div>6</div>												<div>7</div>												<div>8</div>												<div>9</div>												<div>10</div>												<div>11</div>												<div>12</div>												<div>13<div>13 Al aluminium 26.982 ± 0.001</div></div>												<div>14<div>14 Si silicon 28.085 ± 0.001</div></div>												<div>15<div>15 P phosphorus 30.974 ± 0.001</div></div>												<div>16<div>16 S sulfur 32.06 ± 0.02</div></div>												<div>17<div>17 Cl chlorine 35.45 ± 0.01</div></div>												<div>18<div>18 Ar argon 39.95 ± 0.16</div></div>											
<div>19<div>19 K potassium 39.098 ± 0.001</div></div>												<div>20<div>20 Ca calcium 40.078 ± 0.004</div></div>												<div>21<div>21 Sc scandium 44.956 ± 0.001</div></div>												<div>22<div>22 Ti titanium 47.867 ± 0.001</div></div>												<div>23<div>23 V vanadium 50.942 ± 0.001</div></div>												<div>24<div>24 Cr chromium 51.996 ± 0.001</div></div>												<div>25<div>25 Mn manganese 54.938 ± 0.001</div></div>												<div>26<div>26 Fe iron 55.845 ± 0.002</div></div>												<div>27<div>27 Co cobalt 58.933 ± 0.001</div></div>												<div>28<div>28 Ni nickel 58.693 ± 0.001</div></div>												<div>29<div>29 Cu copper 63.546 ± 0.003</div></div>												<div>30<div>30 Zn zinc 65.38 ± 0.02</div></div>												<div>31<div>31 Ga gallium 69.723 ± 0.001</div></div>												<div>32<div>32 Ge germanium 72.630 ± 0.008</div></div>												<div>33<div>33 As arsenic 74.922 ± 0.001</div></div>												<div>34<div>34 Se selenium 78.971 ± 0.008</div></div>												<div>35<div>35 Br bromine 79.904 ± 0.003</div></div>												<div>36<div>36 Kr krypton 83.798 ± 0.002</div></div>											
<div>37<div>37 Rb rubidium 85.468 ± 0.001</div></div>												<div>38<div>38 Sr strontium 87.62 ± 0.01</div></div>												<div>39<div>39 Y yttrium 88.906 ± 0.001</div></div>												<div>40<div>40 Zr zirconium 91.224 ± 0.002</div></div>												<div>41<div>41 Nb niobium 92.906 ± 0.001</div></div>												<div>42<div>42 Mo molybdenum 95.95 ± 0.01</div></div>												<div>43<div>43 Tc technetium [97]</div></div>												<div>44<div>44 Ru ruthenium 101.07 ± 0.02</div></div>												<div>45<div>45 Rh rhodium 102.91 ± 0.01</div></div>												<div>46<div>46 Pd palladium 106.42 ± 0.01</div></div>												<div>47<div>47 Ag silver 107.87 ± 0.01</div></div>												<div>48<div>48 Cd cadmium 112.41 ± 0.01</div></div>												<div>49<div>49 In indium 114.82 ± 0.01</div></div>												<div>50<div>50 Sn tin 118.71 ± 0.01</div></div>												<div>51<div>51 Sb antimony 121.76 ± 0.01</div></div>												<div>52<div>52 Te tellurium 127.60 ± 0.03</div></div>												<div>53<div>53 I iodine 126.90 ± 0.01</div></div>												<div>54<div>54 Xe xenon 131.29 ± 0.01</div></div>											
<div>55<div>55 Cs caesium 132.91 ± 0.01</div></div>												<div>56<div>56 Ba barium 137.33 ± 0.01</div></div>												<div>57-71<div>lanthanoids</div></div>												<div>72<div>72 Hf hafnium 178.49 ± 0.01</div></div>												<div>73<div>73 Ta tantalum 180.95 ± 0.01</div></div>												<div>74<div>74 W tungsten 183.84 ± 0.01</div></div>												<div>75<div>75 Re rhenium 186.21 ± 0.01</div></div>												<div>76<div>76 Os osmium 190.23 ± 0.03</div></div>												<div>77<div>77 Ir iridium 192.22 ± 0.01</div></div>												<div>78<div>78 Pt platinum 195.08 ± 0.02</div></div>												<div>79<div>79 Au gold 196.97 ± 0.01</div></div>												<div>80<div>80 Hg mercury 200.59 ± 0.01</div></div>												<div>81<div>81 Tl thallium 204.38 ± 0.01</div></div>												<div>82<div>82 Pb lead 207.2 ± 1.1</div></div>												<div>83<div>83 Bi bismuth 208.98 ± 0.01</div></div>												<div>84<div>84 Po polonium [209]</div></div>												<div>85<div>85 At astatine [210]</div></div>												<div>86<div>86 Rn radon [222]</div></div>											
<div>87<div>87 Fr francium [223]</div></div>												<div>88<div>88 Ra radium [226]</div></div>												<div>89-103<div>actinoids</div></div>												<div>104<div>104 Rf rutherfordium [267]</div></div>												<div>105<div>105 Db dubnium [268]</div></div>												<div>106<div>106 Sg seaborgium [269]</div></div>												<div>107<div>107 Bh bohrium [270]</div></div>												<div>108<div>108 Hs hassium [269]</div></div>												<div>109<div>109 Mt meitnerium [277]</div></div>												<div>110<div>110 Ds darmstadtium [281]</div></div>												<div>111<div>111 Rg roentgenium [282]</div></div>												<div>112<div>112 Cn copernicium [285]</div></div>												<div>113<div>113 Nh nihonium [286]</div></div>												<div>114<div>114 Fl flerovium [290]</div></div>												<div>115<div>115 Mc moscovium [290]</div></div>												<div>116<div>116 Lv livermorium [293]</div></div>												<div>117<div>117 Ts tennessine [294]</div></div>												<div>118<div>118 Og oganeson [294]</div></div>											

Key:
atomic number
Symbol
name
abridged standard
atomic weight



INTERNATIONAL UNION OF
PURE AND APPLIED CHEMISTRY

57 La lanthanum 138.91 ± 0.01	58 Ce cerium 140.12 ± 0.01	59 Pr praseodymium 140.91 ± 0.01	60 Nd neodymium 144.24 ± 0.01	61 Pm promethium [145]	62 Sm samarium 150.36 ± 0.02	63 Eu europium 151.96 ± 0.01	64 Gd gadolinium 157.25 ± 0.03	65 Tb terbium 158.93 ± 0.01	66 Dy dysprosium 162.50 ± 0.01	67 Ho holmium 164.93 ± 0.01	68 Er erbium 167.26 ± 0.01	69 Tm thulium 168.93 ± 0.01	70 Yb ytterbium 173.05 ± 0.02	71 Lu lutetium 174.97 ± 0.01
89 Ac actinium [227]	90 Th thorium 232.04 ± 0.01	91 Pa protactinium 231.04 ± 0.01	92 U uranium 238.03 ± 0.01	93 Np neptunium [237]	94 Pu plutonium [244]	95 Am americium [243]	96 Cm curium [247]	97 Bk berkelium [247]	98 Cf californium [251]	99 Es einsteinium [252]	100 Fm fermium [257]	101 Md mendelevium [258]	102 No nobelium [259]	103 Lr lawrencium [262]

For notes and updates to this table, see www.iupac.org. This version is dated 4 May 2022.
Copyright © 2022 IUPAC, the International Union of Pure and Applied Chemistry.