

## L6 – 6.5 – Applications of Logarithms in Physical Sciences

MHF4U

### Part 1: Review of Solving Logarithmic Equations

**Example 1:** Solve for  $x$  in the following equation

$$\log_2(x - 6) = 4 - \log_2 x$$

$$\log_2(x - 6) + \log_2 x = 4$$

$$\log_2[(x - 6)(x)] = 4$$

$$2^4 = (x - 6)(x)$$

$$16 = x^2 - 6x$$

$$0 = x^2 - 6x - 16$$

$$0 = (x - 8)(x + 2)$$

$$x = 8$$

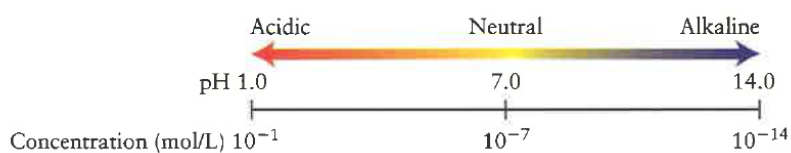
Reject  $x = -2$  as both original logarithmic expressions are undefined for this value

### Part 2: pH Scale

The pH scale is used to measure the acidity or alkalinity of a chemical solution. It is defined as:

$$pH = -\log[H^+]$$

where  $[H^+]$  is the concentration of hydronium ions, measured in moles per liter.



pH = 0	battery acid, strong hydrofluoric acid
pH = 1	hydrochloric acid secreted by stomach lining
pH = 2	lemon juice, gastric acid, vinegar
pH = 3	grapefruit, orange juice, soda
pH = 4	tomato juice, acid rain
pH = 5	soft drinking water, black coffee
pH = 6	urine, saliva
pH = 7	"pure" water
pH = 8	seawater
pH = 9	baking soda
pH = 10	Great Salt Lake, milk of magnesia
pH = 11	ammonia solution
pH = 12	soapy water
pH = 13	bleaches, oven cleaner
pH = 14	liquid drain cleaner

**Example 2:** Answer the following pH scale questions

**a)** Tomato juice has a hydronium ion concentration of approximately 0.0001 mol/L. What is its pH?

$$pH = -\log 0.0001$$

$$pH = -(-4)$$

$$pH = 4$$

**b)** Blood has a hydronium ion concentration of approximately  $4 \times 10^{-7}$  mol/L. Is blood acidic or alkaline?

$$pH = -\log(4 \times 10^{-7})$$

$$pH \cong 6.4$$

Since this is below the neutral value of 7, blood is acidic.

**c)** Orange juice has a pH of approximately 3. What is the concentration of hydronium ions in orange juice?

$$3 = -\log[H^+]$$

$$-3 = \log[H^+]$$

$$10^{-3} = [H^+]$$

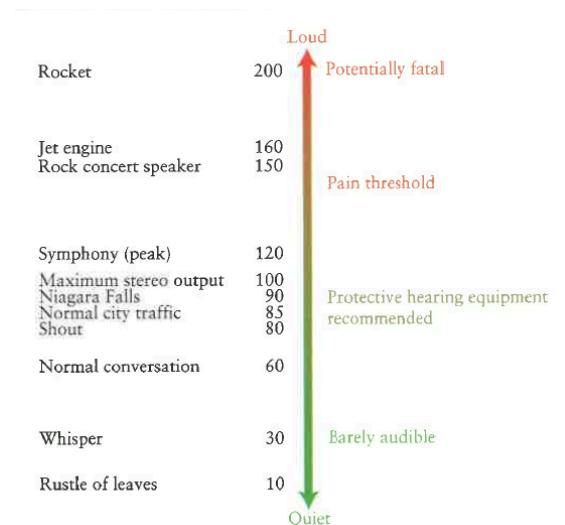
$$[H^+] = 0.001 \text{ mol/L}$$

### Part 3: Decibel Scale

Some common sound levels are indicated on the decibel scale shown. The difference in sound levels, in decibels, can be found using the equation:

$$\beta_2 - \beta_1 = 10 \log \left( \frac{I_2}{I_1} \right)$$

where,  $\beta_2 - \beta_1$  is the difference in sound levels, in decibels, and  $\frac{I_2}{I_1}$  is the ratio of their sound intensities, where  $I$  is measured in watts per square meter ( $W/m^2$ )



**Example 3:** Answer the following questions about decibels

**a)** How many times as intense as a whisper is the sound of a normal conversation

$$60 - 30 = 10 \log\left(\frac{I_2}{I_1}\right)$$

$$30 = 10 \log\left(\frac{I_2}{I_1}\right)$$

$$3 = \log\left(\frac{I_2}{I_1}\right)$$

$$10^3 = \frac{I_2}{I_1}$$

$$\frac{I_2}{I_1} = 1000$$

A conversation sounds 1000 times as intense as a whisper.

**b)** The sound level in normal city traffic is approximately 85 dB. The sound level while riding a snowmobile is about 32 times as intense. What is the sound level while riding a snowmobile, in decibels?

$$\beta_2 - 85 = 10 \log(32)$$

$$\beta_2 = 10 \log(32) + 85$$

$$\beta_2 \cong 100 \text{ dB}$$

#### **Part 4: Richter Scale**

The magnitude,  $M$ , of an earthquake is measured using the Richter scale, which is defined as:

$$M = \log\left(\frac{I}{I_0}\right)$$

where  $I$  is the intensity of the earthquake being measured and  $I_0$  is the intensity of a standard, low-level earthquake.

**Example 4:** Answer the following questions about the Richter Scale

**a)** How many times as intense as a standard earthquake is an earthquake measuring 2.4 on the Richter scale?

$$2.4 = \log\left(\frac{I}{I_0}\right)$$

$$10^{2.4} = \frac{I}{I_0}$$

$$\frac{I}{I_0} \cong 251.19$$

It is about 251 times as intense as a standard earthquake.

**b)** What is the magnitude of an earthquake 1000 times as intense as a standard earthquake?

$$M = \log(1000)$$

$$M = 3$$