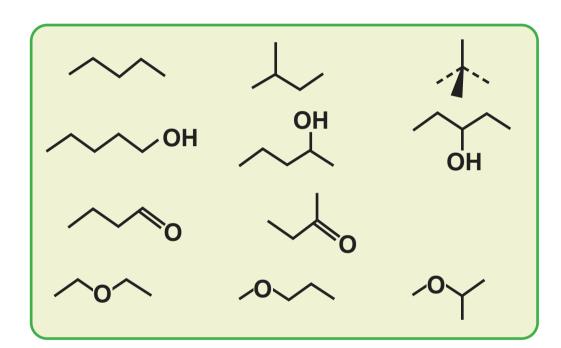


CONSTITUTIONAL ISOMERISM

CONSTITUTIONAL ISOMERISM [STRUCTURAL]



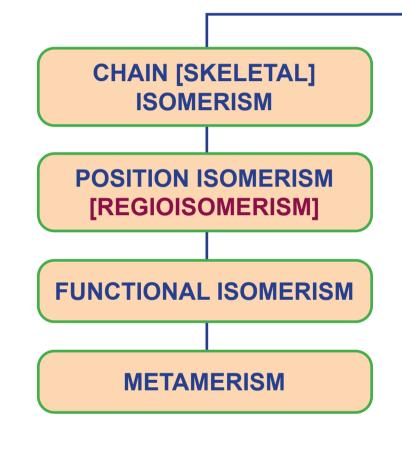
Pair of functional isomers:

- Alcohols and ethers (C_nH_{2n+2}O)
- Aldehydes and ketones (C_nH_{2n}O)
- Carboxylic acids and esters (C_nH_{2n}O₂)
- 1°, 2° & 3° amines (C_nH_{2n+3} N)

DBE =
$$\frac{\sum n (v - 2)}{2}$$
 + 1 = sum of no. of π bonds + rings

in the molecule

(n is no of atoms of particular element & v is corresponding valency in given molecule).



PROTOTROPIC TAUTOMERISM [PROTOTROPY] H-X-Y=Z, X=Y-Z-H

$$H_3C-N$$
 $H_2C=N$

O-H

Nitroso form

oxime

- **Q.** The number of structural isomers possible from the molecular formula is
- (a) 4 (b) 5 (c) 2

(d) 3

- **Q.** Identify the compound that exhibits tautomerism
 - (a) 2-Pentanone
- (b) Phenol
- (c) 2-Butene
- (d) Lactic acid
- **Q.** The number of structural isomers for C_6H_{14} is

(c) 5

- (a) 3
- (b) 4
- (d) 6

$$R_2CH-N$$
 O
 R_2CH-N
 O
 O
 O
 O
Aci form