

# Battery INterFace Ontology (BattINFO) Reference Documentation

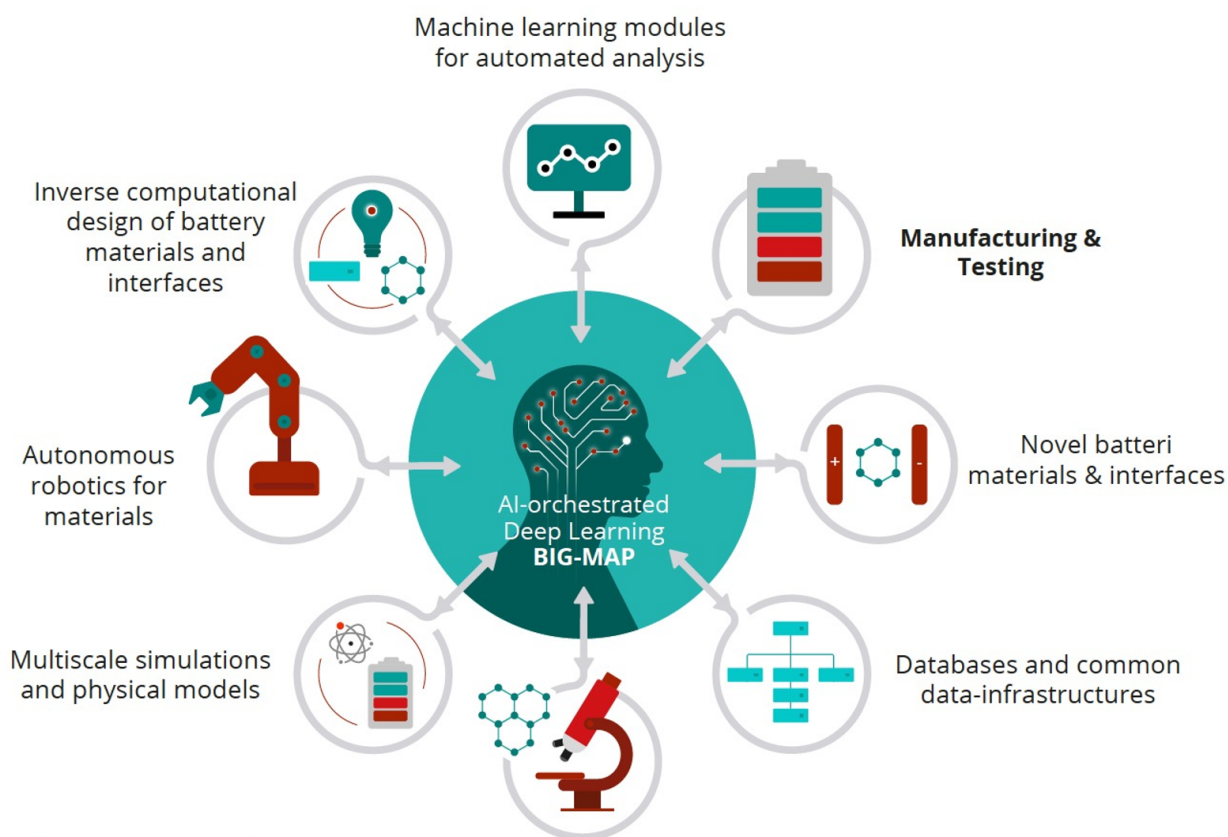
VERSION 0.1

Battery Interface Genome - Materials Acceleration Platform (BIG-MAP)



**BIG MAP**

March 01, 2021



## Abstract

This is a reference documentation for the Battery Interface Ontology (BattINFO).

BattINFO is an ontology of batteries and their interfaces based on the top-level European Materials and Modelling Ontology (EMMO). BattINFO aims to formalize the current state of knowledge on battery interfaces to support the development of computational tools and the deployment of interoperable data in the BIG-MAP project and beyond. The definitions included in BattINFO are based as far as possible on accepted standards defined by the International Union of Pure and Applied Chemistry (IUPAC) or other preeminent textbooks on the subject. BattINFO objects and their relations to each other are designed with three goals in mind: (i) to be scientifically rigorous and accurate, (ii) to reflect current battery orthodoxy and dominant jargon, and (iii) to be flexible to describe a range of battery chemistries, not only Li-ion.

The development of BattINFO is a mammoth undertaking and will continue throughout the project. However, it is important to establish an initial version to support the activities in other BIG-MAP work packages and provide a preliminary platform for collaboration. The objective of this deliverable is to establish the initial version of BattINFO. This report outlines the conceptual foundation for the definitions in the ontology and serves as a guide to help interpret the implementation of BattINFO in the ontology web language (OWL).

**Keywords:** Battery, EMMO, materials science, modelling, characterisation, materials, ontology

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# Chapter 1

## Introduction

Battery development is one of the most important and intensely pursued technical research topics in the world today. From personal electronics to electric mobility to renewable energy storage, batteries are essential to progress. The search for better batteries is supported by a host of databases, methods, models, publications, and presentations. How can we distil this deluge of data into knowledge and translate that knowledge into action?

The answer must rely in some part on artificial intelligence (AI). The breadth of fields necessary to completely describe of battery performance, characterization, and simulation combined with the depth of research being generated in those fields is simply too great for any single person (or even group of people) to manage. However, the challenge is that the wealth of battery data that exists is formatted to be read, understood, and learned by humans, not machines. The field needs a tool to formalize the current state of knowledge about battery interfaces that is both human- and machine-readable.

The [Battery Interface Ontology \(BattINFO\)](#) is a domain ontology for batteries and their interfaces. It is developed with the goal of creating a formalized description of battery cells to support the interoperability of battery data and support applications of artificial intelligence in battery research.

BattINFO builds upon long-standing and widely accepted principles of electrochemistry as described in preeminent texts such as *Electrochemical Systems* by John Newman and Karen E. Thomas-Alyea [1], *Electrochemical Methods: Fundamentals and Applications* by Allen J. Bard and Larry R. Faulkner [2], and *Handbook of Batteries* by David Linden and Thomas B. Reddy [3], among other seminal sources [4], [5]. The terminology adheres as far as possible to the recommendations and definitions contained in the *Compendium of Chemical Terminology* (also known as the “Gold Book”) from the International Union of Pure and Applied Chemistry (IUPAC) [6] together with IUPAC supplements on electrochemical terminology [7] and recommendations from the Electrochemical Society (ECS) on nomenclature and standards. Places where conflicts exist between sources are noted for further discussion and resolution within the electrochemical community.

BattINFO employs the [European Materials and Modelling Ontology \(EMMO\)](#) as a top-level ontology. EMMO aims at the development of a standard representational ontology framework based on current materials modelling and characterization of knowledge. EMMO starts from the very basic scientific fundamentals and grows to encompass a complex and wide field of knowledge, however it is still functional and clear. This makes it ideal to support the development of BattINFO as an EMMO domain ontology.

The purpose of this report is to lay the groundwork for the development of BattINFO in the [BIG-MAP](#) project.

## Availability and license

The Battery Interface Domain Ontology is available from the github repository <https://github.com/BIG-MAP/BattINFO>.

It is released under the [Creative Commons Attribution 4.0 International license \(CC BY 4.0\)](#).

## References

1. J. Newman and K. E. Thmoas-Alyea, *Electrochemical Systems*, 3rd ed. Hoboken, New Jersey: John Wiley & Sons, 2004.
2. A. J. Bard and L. R. Faulkner, *ELECTROCHEMICAL METHODS: Fundamentals and applications*. 2001.
3. D. Linden and T. Reddy, *Handbook of Batteries*. 2002.
4. P. Atkins and J. De Paula, *Atkins' Physical Chemistry*, 8th Ed. New York: W.H. Freeman and Company, 2006.
5. M. Pourbaix, *Atlas of Electrochemical Equilibria in Aqueous Solutions*, Second. Houston, Texas: National Association of Corrosion Engineers, 1974.
6. IUPAC, *Compendium of Chemical Terminology*, 2nd (the “). Oxford: Blackwell Scientific Publications, 2014.
7. J. M. Pingarrón et al., Terminology of electrochemical methods of analysis (IUPAC Recommendations 2019), *Pure Appl. Chem.*, vol. 92, no. 4, pp. 641-694, 2020.

## Chapter 2

# Generic concepts

These classes are intended to be merged back into EMMO.

## Process subclasses

### FunctionalProcess

**IRI:** [http://emmo.info/emmo#EMMO\\_f7dbce66\\_2822\\_4855\\_9f42\\_1da71aa9e923](http://emmo.info/emmo#EMMO_f7dbce66_2822_4855_9f42_1da71aa9e923)

**elucidation:** The process that makes a product work as intended when in use.

**example:** - The light-emitting process of a diode. - The car crash process for a crash box in a car. - The discharging process of a battery.

**prefLabel:** FunctionalProcess

**Relations:**

- is\_a **Process**

### ChemicalPhenomenon

**IRI:** [http://emmo.info/emmo#EMMO\\_50e36d79\\_b2dd\\_422d\\_81eb\\_a665028a1ead](http://emmo.info/emmo#EMMO_50e36d79_b2dd_422d_81eb_a665028a1ead)

**elucidation:** A ‘process’ that is recognized by chemical sciences and is categorized accordingly.

**prefLabel:** ChemicalPhenomenon

**Relations:**

- is\_a **Process**

### ChemicalReaction

**IRI:** [http://emmo.info/emmo#EMMO\\_ecb0395f\\_ee1e\\_4e9a\\_bf5c\\_d8e56eee2d18](http://emmo.info/emmo#EMMO_ecb0395f_ee1e_4e9a_bf5c_d8e56eee2d18)

**elucidation:** A process that results in the interconversion of chemical species. Chemical reactions may be elementary reactions or stepwise reactions. (It should be noted that this definition includes experimentally observable interconversions of conformers.) Detectable chemical reactions normally involve sets of molecular entities as indicated by this definition, but it is often conceptually convenient to use the term also for changes involving single molecular entities (i.e. ‘microscopic chemical events’).

- IUPAC Gold Book

**prefLabel:** ChemicalReaction

**Relations:**

- is\_a **ChemicalPhenomenon**

## Participant subclasses

### ActiveParticipant

**IRI:** [http://emmo.info/emmo#EMMO\\_038e37a3\\_1684\\_4980\\_b5e4\\_67ab34cd5bdb](http://emmo.info/emmo#EMMO_038e37a3_1684_4980_b5e4_67ab34cd5bdb)

**elucidation:** A ‘physical’ that stands for a real world object that takes active part of a functional process.

**prefLabel:** ActiveParticipant

**Relations:**

- is\_a Participant
- Inverse(hasProperParticipant) some FunctionalProcess

### FunctionalMaterial

**IRI:** [http://emmo.info/emmo#EMMO\\_d95e6e0d-e8eb-411a-b407-0d1a517e8767](http://emmo.info/emmo#EMMO_d95e6e0d-e8eb-411a-b407-0d1a517e8767)

**elucidation:** Materials that have one or more properties that can be significantly changed in a controlled fashion by external stimuli (temperature, electric/magnetic field, etc.) and are therefore applied in a broad range of technological devices as for example in memories, displays and telecommunication. - NTNU FY3114 - Functional Materials

**prefLabel:** FunctionalMaterial

**Relations:**

- is\_a ActiveParticipant
- is\_a Material

## Physicalistic subclasses

### Pore

**IRI:** [http://emmo.info/emmo#EMMO\\_69b9aead-bb43-4bd5-9168-728cea2116b1](http://emmo.info/emmo#EMMO_69b9aead-bb43-4bd5-9168-728cea2116b1)

**elucidation:** A space within a solid host domain that is filled by a liquid, gas, or vacuum. The characteristic length of the pore is much less than the characteristic length of the host domain. An exception is possible for 1 dimension (e.g. long pores).

**prefLabel:** Pore

**Relations:**

- is\_a Physicalistic
- is\_a Gas or Vacuum or Liquid
- hasContactWith some Solid

## Physical quantities

### VolumetricThermalExpansionCoefficient

**IRI:** [http://emmo.info/emmo#EMMO\\_1c1ec02e\\_4def\\_4979\\_aff9\\_572c06a95391](http://emmo.info/emmo#EMMO_1c1ec02e_4def_4979_aff9_572c06a95391)

**physicalDimension:** T0 L0 M0 I0 Θ-1 N0 J0

**prefLabel:** VolumetricThermalExpansionCoefficient

**Relations:**

- is\_a ThermalExpansionCoefficient

## SingleComponentDiffusivity

**IRI:** [http://emmo.info/emmo#EMMO\\_498d80ae\\_9339\\_49c7\\_8c74\\_44aa704e0395](http://emmo.info/emmo#EMMO_498d80ae_9339_49c7_8c74_44aa704e0395)

**elucidation:** Transport of particles belonging to one component of a material due to a concentration gradient.

**physicalDimension:** T-1 L+2 M0 I0 Θ0 N-1 J0

**prefLabel:** SingleComponentDiffusivity

**Relations:**

- is\_a **ISQDerivedQuantity**
- is\_a **PhysicoChemical**

## SingleComponentDiffusivity

**IRI:** [http://emmo.info/emmo#EMMO\\_498d80ae\\_9339\\_49c7\\_8c74\\_44aa704e0395](http://emmo.info/emmo#EMMO_498d80ae_9339_49c7_8c74_44aa704e0395)

**elucidation:** Transport of particles belonging to one component of a material due to a concentration gradient.

**physicalDimension:** T-1 L+2 M0 I0 Θ0 N-1 J0

**prefLabel:** SingleComponentDiffusivity

**Relations:**

- is\_a **ISQDerivedQuantity**
- is\_a **PhysicoChemical**

## SingleComponentMaximalDiffusivity

**IRI:** [http://emmo.info/emmo#EMMO\\_3bd39834\\_7eb9\\_4c97\\_bb25\\_db88c3df6bab](http://emmo.info/emmo#EMMO_3bd39834_7eb9_4c97_bb25_db88c3df6bab)

**etymology:** Pre-factor in the Arrhenius expression for diffusion.

**physicalDimension:** T-1 L+2 M0 I0 Θ0 N-1 J0

**prefLabel:** SingleComponentMaximalDiffusivity

**Relations:**

- is\_a **PhysicoChemical**
- is\_a **ISQDerivedQuantity**

## SingleComponentActivationEnergyOfDiffusion

**IRI:** [http://emmo.info/emmo#EMMO\\_2f761aff\\_88d1\\_4e79\\_a85e\\_09d6f400de56](http://emmo.info/emmo#EMMO_2f761aff_88d1_4e79_a85e_09d6f400de56)

**elucidation:** The energy barrier for diffusion of a given component.

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** SingleComponentActivationEnergyOfDiffusion

**Relations:**

- is\_a **Energy**
- is\_a **PhysicoChemical**

## MolarHeatCapacity

**IRI:** [http://emmo.info/emmo#EMMO\\_50c5d440\\_683c\\_400f\\_909e\\_b03c0327de9c](http://emmo.info/emmo#EMMO_50c5d440_683c_400f_909e_b03c0327de9c)

**elucidation:** The molar heat capacity of a substance is the heat capacity of one mole of material.

**physicalDimension:** T-2 L+2 M+1 I0 Θ-1 N-1 J0

**prefLabel:** MolarHeatCapacity



**Relations:**

- is\_a **PhysicoChemical**
- is\_a **ISQDerivedQuantity**

**EnergyDensity**

**IRI:** [http://emmo.info/emmo#EMMO\\_686308bd\\_8ed6\\_49d0\\_a204\\_6487dbe56511](http://emmo.info/emmo#EMMO_686308bd_8ed6_49d0_a204_6487dbe56511)

**elucidation:** Energy per unit volume.

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** EnergyDensity

**Relations:**

- is\_a **ISQDerivedQuantity**

**ThermalExpansionCoefficient**

**IRI:** [http://emmo.info/emmo#EMMO\\_7684ddff\\_d99b\\_405d\\_aad2\\_90e830b8403c](http://emmo.info/emmo#EMMO_7684ddff_d99b_405d_aad2_90e830b8403c)

**elucidation:** The coefficient of thermal expansion describes how the fractional change in size of an object changes with a change in temperature.

**physicalDimension:** T0 L0 M0 I0 Θ-1 N0 J0

**prefLabel:** ThermalExpansionCoefficient

**Relations:**

- is\_a **ISQDerivedQuantity**
- is\_a **PhysicoChemical**

**HeatCapacity**

**IRI:** [http://emmo.info/emmo#EMMO\\_802c167d\\_b792\\_4cb8\\_a315\\_35797345c0e3](http://emmo.info/emmo#EMMO_802c167d_b792_4cb8_a315_35797345c0e3)

**elucidation:** The amount of heat to be applied to a given mass of material to produce a unit change in its temperature.

**physicalDimension:** T-2 L+2 M+1 I0 Θ-1 N0 J0

**prefLabel:** HeatCapacity

**Relations:**

- is\_a **PhysicoChemical**
- is\_a **ISQDerivedQuantity**

**ThermalConductivity**

**IRI:** [http://emmo.info/emmo#EMMO\\_8dd40ec6\\_2c5a\\_43f3\\_bf64\\_cadcd447a1c1](http://emmo.info/emmo#EMMO_8dd40ec6_2c5a_43f3_bf64_cadcd447a1c1)

**elucidation:** The ability of a material to conduct heat.

**physicalDimension:** T-3 L+1 M+1 I0 Θ-1 N0 J0

**prefLabel:** ThermalConductivity

**Relations:**

- is\_a **PhysicoChemical**
- is\_a **ISQDerivedQuantity**

## SpecificHeatCapacity

**IRI:** [http://emmo.info/emmo#EMMO\\_b4f4ed28\\_d24c\\_4a00\\_9583\\_62ab839abeca](http://emmo.info/emmo#EMMO_b4f4ed28_d24c_4a00_9583_62ab839abeca)

**elucidation:** The specific heat capacity (symbol cp) of a substance is the heat capacity of a sample of the substance divided by the mass of the sample.

**physicalDimension:** T-2 L+2 M0 I0 Θ-1 N0 J0

**prefLabel:** SpecificHeatCapacity

**Relations:**

- is\_a **PhysicoChemical**
- is\_a **ISQDerivedQuantity**

## Physical dimensions

### PerTemperatureDimension

**IRI:** [http://emmo.info/emmo#EMMO\\_6e9aef15\\_272b\\_4eea\\_aaa9\\_2f38b8ae951f](http://emmo.info/emmo#EMMO_6e9aef15_272b_4eea_aaa9_2f38b8ae951f)

**prefLabel:** PerTemperatureDimension

**Relations:**

- is\_a **PhysicalDimension**
- equivalent\_to **hasSymbolData** value “T0 L0 M0 I0 Θ-1 N0 J0”

## Chapter 3

# Electrochemical and battery-specific concepts

All classes under here are defined with the <http://emmo.info/BattINFO#> namespace.

### Active Participant branch

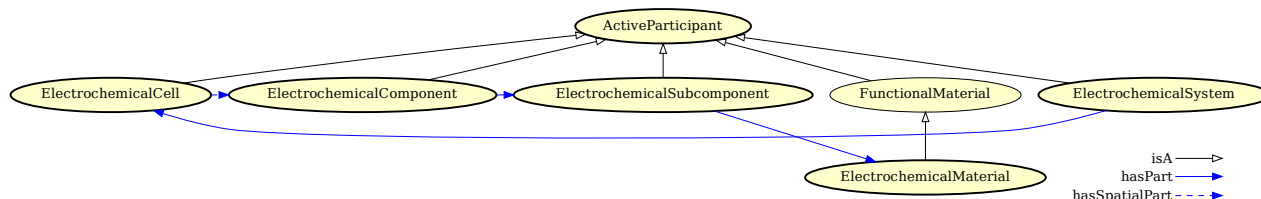


Figure 3.1: Active Participant branch.

### FunctionalMaterial

**IRI:** [http://emmo.info/emmo#EMMO\\_d95e6e0d-e8eb-411a-b407-0d1a517e8767](http://emmo.info/emmo#EMMO_d95e6e0d-e8eb-411a-b407-0d1a517e8767)

**elucidation:** Materials that have one or more properties that can be significantly changed in a controlled fashion by external stimuli (temperature, electric/magnetic field, etc.) and are therefore applied in a broad range of technological devices as for example in memories, displays and telecommunication. - NTNU FY3114 - Functional Materials

**prefLabel:** FunctionalMaterial

**Relations:**

- is\_a **ActiveParticipant**
- is\_a **Material**

### ActiveParticipant

**IRI:** [http://emmo.info/emmo#EMMO\\_038e37a3\\_1684\\_4980\\_b5e4\\_67ab34cd5bdb](http://emmo.info/emmo#EMMO_038e37a3_1684_4980_b5e4_67ab34cd5bdb)

**elucidation:** A ‘physical’ that stands for a real world object that takes active part of a functional process.

**prefLabel:** ActiveParticipant

**Relations:**

- is\_a **Participant**
- Inverse(**hasProperParticipant**) some **FunctionalProcess**

## Electrochemical System branch

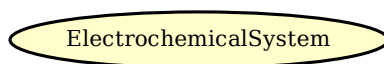


Figure 3.2: Electrochemical System branch.

### ElectrochemicalSystem

**IRI:** [http://emmo.info/BattINFO#EMMO\\_4e4d7f4b-680b-469e-bdd4-728dd3e465bf](http://emmo.info/BattINFO#EMMO_4e4d7f4b-680b-469e-bdd4-728dd3e465bf)

**elucidation:** A system comprising at least one electrochemical cell and the components necessary to support it.

**prefLabel:** ElectrochemicalSystem

**Relations:**

- is\_a **ActiveParticipant**
- hasPart some **ElectrochemicalCell**

## Electrochemical Cell branch

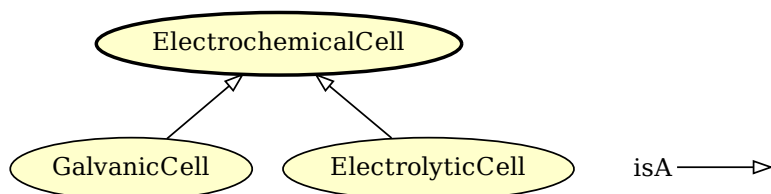


Figure 3.3: Electrochemical Cell branch.

### ElectrochemicalCell

**IRI:** [http://emmo.info/BattINFO#EMMO\\_6f2c88c9\\_5c04\\_4953\\_a298\\_032cc3ab9b77](http://emmo.info/BattINFO#EMMO_6f2c88c9_5c04_4953_a298_032cc3ab9b77)

**elucidation:** A system capable of either generating electrical energy from chemical reactions or using electrical energy to cause chemical reactions. The key feature of an electrochemical cell is that it contains two (or more) electrodes that allow transport of electrons, separated by a salt bridge that allows the movement of ions but blocks movement of electrons.

– Adapted from J. Newman, *Electrochemical Systems* (p. 3) and other sources

**prefLabel:** ElectrochemicalCell

**Relations:**

- is\_a **Matter**

- is\_a **ActiveParticipant**
- is\_a **Object**
- hasConventionalQuantity some **SpecificHeatCapacity**
- hasConventionalQuantity some **ThermalExpansionCoefficient**
- hasConventionalQuantity some **Volume**
- hasConventionalQuantity some **OpenCircuitVoltage**
- hasConventionalQuantity some **InternalResistance**
- hasSpatialPart some **ElectrochemicalComponent**
- hasConventionalQuantity some **ThermalConductivity**
- hasConventionalQuantity some **HeatCapacity**
- hasConventionalQuantity some **ElectricPotential**
- hasConventionalQuantity some **Density**
- hasConventionalQuantity some **InternalConductance**
- hasConventionalQuantity some **SpecificChargeCapacity**
- hasConventionalQuantity some **SpecificEnergy**
- hasConventionalQuantity some **ElectricImpedance**
- hasConventionalQuantity some **Mass**
- hasConventionalQuantity some **StoredEnergy**
- hasConventionalQuantity some **ChargeCapacity**
- hasConventionalQuantity some **ThermodynamicTemperature**
- hasConventionalQuantity some **EnergyDensity**

## GalvanicCell

**IRI:** [http://emmo.info/BattINFO#EMMO\\_e248373f\\_294f\\_4ca4\\_9edf\\_0ad6653bb64f](http://emmo.info/BattINFO#EMMO_e248373f_294f_4ca4_9edf_0ad6653bb64f)

**elucidation:** An electrochemical cell that spontaneously produces work.

– J. Newman, Electrochemical Systems (p. 6)

**prefLabel:** GalvanicCell

**Relations:**

- is\_a **ElectrochemicalCell**

## ElectrolyticCell

**IRI:** [http://emmo.info/BattINFO#EMMO\\_e931087f\\_7681\\_4096\\_b200\\_5223bcc47eb4](http://emmo.info/BattINFO#EMMO_e931087f_7681_4096_b200_5223bcc47eb4)

**elucidation:** An electrochemical cell that requires input of work to drive the reaction.

– J. Newman, Electrochemical Systems (p. 6)

**prefLabel:** ElectrolyticCell

**Relations:**

- is\_a **ElectrochemicalCell**

## Electrochemical Component branch

### ElectrochemicalComponent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_3597a1e0\\_09ef\\_48ad\\_b913\\_b3e71ea21c94](http://emmo.info/BattINFO#EMMO_3597a1e0_09ef_48ad_b913_b3e71ea21c94)

**elucidation:** A component that is essential to the function of an electrochemical cell.

**prefLabel:** ElectrochemicalComponent

**Relations:**

- is\_a **ActiveParticipant**
- hasPart some **ElectrochemicalSubcomponent**

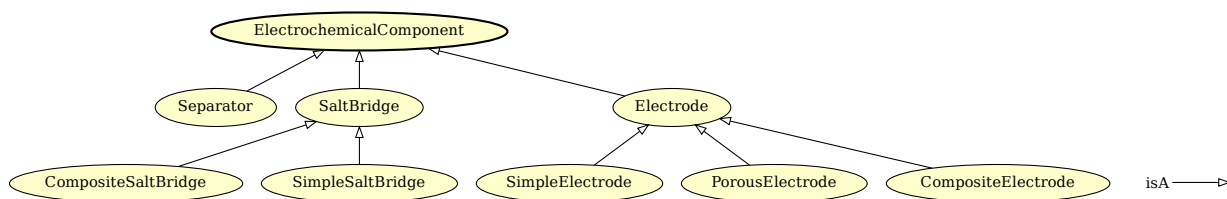


Figure 3.4: Electrochemical Component branch.

## CompositeSaltBridge

**IRI:** [http://emmo.info/BattINFO#EMMO\\_6cae5943-737a-4f88-9903-9de4cffe4d11](http://emmo.info/BattINFO#EMMO_6cae5943-737a-4f88-9903-9de4cffe4d11)

**elucidation:** A salt bridge consisting of at least two subcomponents, one of which is an IonicSubcomponent.

**prefLabel:** CompositeSaltBridge

**Relations:**

- is\_a **State**
- is\_a **SaltBridge**
- hasSpatialDirectPart some **IonicSubcomponent**
- hasSpatialDirectPart min 2 **ElectrochemicalSubcomponent**

## SimpleSaltBridge

**IRI:** [http://emmo.info/BattINFO#EMMO\\_6e4f4681-f327-4300-96e4-5905fcea36e3](http://emmo.info/BattINFO#EMMO_6e4f4681-f327-4300-96e4-5905fcea36e3)

**elucidation:** A salt bridge consisting of exactly 1 subcomponent that is an IonicSubcomponent.

**prefLabel:** SimpleSaltBridge

**Relations:**

- is\_a **State**
- is\_a **SaltBridge**
- hasSpatialDirectPart exactly 1 **IonicSubcomponent**

## Separator

**IRI:** [http://emmo.info/BattINFO#EMMO\\_331e6cca\\_f260\\_4bf8\\_af55\\_35304fe1bbe0](http://emmo.info/BattINFO#EMMO_331e6cca_f260_4bf8_af55_35304fe1bbe0)

**definition:** “A permeable membrane placed between the positive and negative electrodes to keep them physically separated and prevent an internal short circuit.”

**prefLabel:** Separator

**Relations:**

- is\_a **ElectrochemicalComponent**

## SimpleElectrode

**IRI:** [http://emmo.info/BattINFO#EMMO\\_029f0b45-70a7-481f-8154-bf982a77e08c](http://emmo.info/BattINFO#EMMO_029f0b45-70a7-481f-8154-bf982a77e08c)

**elucidation:** An electrode consisting of a single ElectrochemicalSubComponent

**example:** Metal foil.

**prefLabel:** SimpleElectrode

**Relations:**

- is\_a **Electrode**

## PorousElectrode

**IRI:** [http://emmo.info/BattINFO#EMMO\\_3663991d-9319-4f7a-922b-f0e428b58801](http://emmo.info/BattINFO#EMMO_3663991d-9319-4f7a-922b-f0e428b58801)

**elucidation:** Porous electrodes consist of porous matrices of a single reactive electronic conductor or a mixture of solids that include essentially non-conducting, reactive materials in addition to electronic conductors. An electrolytic solution fills the void spaces of the porous matrix. At a given time, there may be a large range of reaction rates within the pores. The distribution of these rates will depend on physical structure, conductivity of the matrix and of the electrolyte, and on parameters characterizing the electrode processes themselves. - Newman and Thomas-Alyea, Electrochemical Systems.

**prefLabel:** PorousElectrode

**Relations:**

- is\_a **Electrode**
- hasSpatialPart some **ElectrodePore**

## CompositeElectrode

**IRI:** [http://emmo.info/BattINFO#EMMO\\_7aa79b12-6b34-4724-9728-f31b5f7ed83d](http://emmo.info/BattINFO#EMMO_7aa79b12-6b34-4724-9728-f31b5f7ed83d)

**elucidation:** An electrode consisting of multiple ElectrochemicalSubComponent

**prefLabel:** CompositeElectrode

**Relations:**

- is\_a **Electrode**

## SaltBridge

**IRI:** [http://emmo.info/BattINFO#EMMO\\_637c576e\\_a50e\\_47ae\\_8c74\\_2024ce4c6d0f](http://emmo.info/BattINFO#EMMO_637c576e_a50e_47ae_8c74_2024ce4c6d0f)

**elucidation:** “Means of making electrolytic connection between two half cells without introducing a significant liquid junction potential. Note: A typical construction is a tube of an inert material (e.g. agar agar) filled with a solution containing an electrolyte with approximately equal ion mobilities of the cation and the anion (e.g., KNO<sub>3</sub>, KCl), with the ends of the tube immersed in the electrolyte solution of the half cells.” Pingarron et al., Terminology of electrochemical methods of analysis

**prefLabel:** SaltBridge

**Relations:**

- is\_a **ElectrochemicalComponent**

## Electrode

**IRI:** [http://emmo.info/BattINFO#EMMO\\_0f007072-a8dd-4798-b865-1bf9363be627](http://emmo.info/BattINFO#EMMO_0f007072-a8dd-4798-b865-1bf9363be627)

**elucidation:** Electron conductor in an electrochemical cell connected to the external circuit. - Terminology of electrochemical methods of analysis (IUPAC Recommendations 2019)

**prefLabel:** Electrode

**Relations:**

- is\_a **ElectrochemicalComponent**
- is\_a **Object**
- hasContactWith some **Electrolyte**
- hasConventionalQuantity some **ChargeCapacity**

## Electrochemical Subcomponent branch

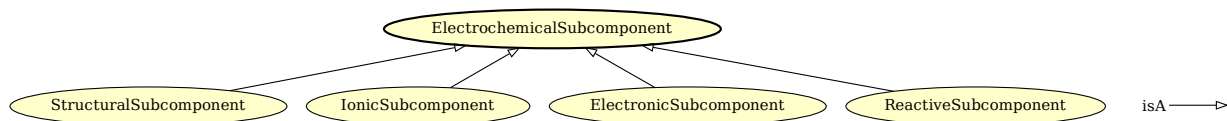


Figure 3.5: Electrochemical Subcomponent branch.

### StructuralSubcomponent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_dd15b4b0-11e7-4900-b379-9702a8caa6bb](http://emmo.info/BattINFO#EMMO_dd15b4b0-11e7-4900-b379-9702a8caa6bb)

**prefLabel:** StructuralSubcomponent

**Relations:**

- is\_a [ElectrochemicalSubcomponent](#)

### IonicSubcomponent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_23b866e8-27c6-4fd8-a1d2-6b58ad4445af](http://emmo.info/BattINFO#EMMO_23b866e8-27c6-4fd8-a1d2-6b58ad4445af)

**prefLabel:** IonicSubcomponent

**Relations:**

- is\_a [ElectrochemicalSubcomponent](#)

### ElectrochemicalSubcomponent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_f89bb8bc-ef9b-43d5-b5df-14e12b0d93b8](http://emmo.info/BattINFO#EMMO_f89bb8bc-ef9b-43d5-b5df-14e12b0d93b8)

**elucidation:** A subcomponent of an ElectrochemicalComponent.

**prefLabel:** ElectrochemicalSubcomponent

**Relations:**

- is\_a [ActiveParticipant](#)
- hasPart some [ElectrochemicalMaterial](#)

### ElectronicSubcomponent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_9c4e61c6-4a7b-41c2-9133-e780e144ddcd](http://emmo.info/BattINFO#EMMO_9c4e61c6-4a7b-41c2-9133-e780e144ddcd)

**elucidation:** An ElectrochemicalSubcomponent whose primary role is electronic

**example:** Current Collector Conducting Additive

**prefLabel:** ElectronicSubcomponent

**Relations:**

- is\_a [ElectrochemicalSubcomponent](#)



## ReactiveSubcomponent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_6ab1ca1a-3809-4e9a-aaf7-374915288f73](http://emmo.info/BattINFO#EMMO_6ab1ca1a-3809-4e9a-aaf7-374915288f73)

**elucidation:** An ElectrochemicalSubcomponent whose primary role is to participate in a reaction.

**prefLabel:** ReactiveSubcomponent

**Relations:**

- is\_a **ElectrochemicalSubcomponent**

## Electrochemical Material branch

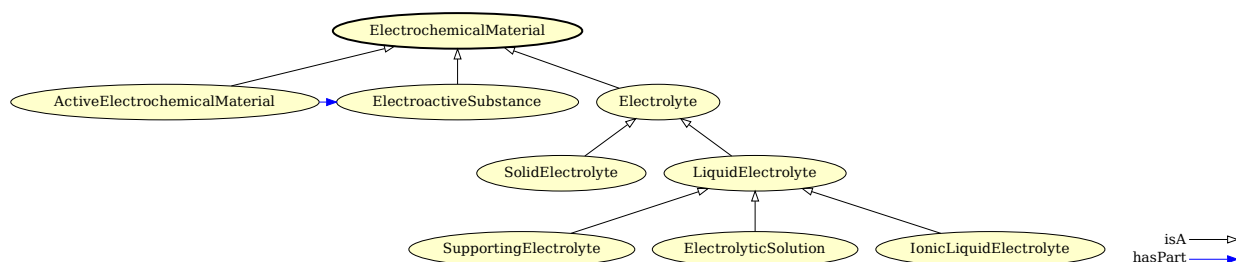


Figure 3.6: Electrochemical Material branch.

## Electrolyte

**IRI:** [http://emmo.info/BattINFO#EMMO\\_fb0d9eef\\_92af\\_4628\\_8814\\_e065ca255d59](http://emmo.info/BattINFO#EMMO_fb0d9eef_92af_4628_8814_e065ca255d59)

**definition:** 1. Conducting medium in which the flow of electric current is accompanied by the movement of ions. Pingarron et al., Terminology of electrochemical methods of analysis

**prefLabel:** Electrolyte

**Relations:**

- is\_a **ElectrochemicalMaterial**

## SupportingElectrolyte

**IRI:** [http://emmo.info/BattINFO#EMMO\\_1fc5642c\\_b7b2\\_43bf\\_ad20\\_f96001db8800](http://emmo.info/BattINFO#EMMO_1fc5642c_b7b2_43bf_ad20_f96001db8800)

**definition:** “Electrolyte solution, the ions of which are electroinactive in the range of applied potential being studied, and whose ionic strength (and, therefore, contribution to the overall conductivity) is usually much greater than the concentration of an electroactive substance to be dissolved in it.” Pingarron et al., Terminology of electrochemical methods of analysis

**prefLabel:** SupportingElectrolyte

**Relations:**

- is\_a **LiquidElectrolyte**

## ElectrolyticSolution

**IRI:** [http://emmo.info/BattINFO#EMMO\\_fa22874b\\_76a9\\_4043\\_8b8f\\_6086c88746de](http://emmo.info/BattINFO#EMMO_fa22874b_76a9_4043_8b8f_6086c88746de)

**definition:** “A liquid electrolyte that consists of solutes dissolved in a solvent.”

**prefLabel:** ElectrolyticSolution

**Relations:**

- is\_a **LiquidElectrolyte**

**SolidElectrolyte**

**IRI:** [http://emmo.info/BattINFO#EMMO\\_0508a114\\_544a\\_4f54\\_a7de\\_9b947fb4b618](http://emmo.info/BattINFO#EMMO_0508a114_544a_4f54_a7de_9b947fb4b618)

**definition:** “A solid electrolyte is a solid material where the predominant charge carriers are ions. For example: NASICON (Na Super Ionic Conductor), which has the general formula  $\text{Na}_{1+x}\text{Zr}_2\text{P}_3\text{O}_{12}$ ,  $0 < x < 3$ .” Pingarron et al., Terminology of electrochemical methods of analysis

**prefLabel:** SolidElectrolyte

**Relations:**

- is\_a **Electrolyte**

**ActiveElectrochemicalMaterial**

**IRI:** [http://emmo.info/BattINFO#EMMO\\_79d1b273-58cd-4be6-a250-434817f7c261](http://emmo.info/BattINFO#EMMO_79d1b273-58cd-4be6-a250-434817f7c261)

**prefLabel:** ActiveElectrochemicalMaterial

**Relations:**

- is\_a **ElectrochemicalMaterial**
- hasPart some **ElectroactiveSubstance**

**IonicLiquidElectrolyte**

**IRI:** [http://emmo.info/BattINFO#EMMO\\_c3f4b34a\\_0e2c\\_46f3\\_baab\\_4ebd2682d26f](http://emmo.info/BattINFO#EMMO_c3f4b34a_0e2c_46f3_baab_4ebd2682d26f)

**definition:** “An ionic liquid is an electrolyte composed of a salt that is liquid below 100 °C. Ionic liquids have found uses in electrochemical analysis, because their unconventional properties include a negligible vapor pressure, a high thermal and electrochemical stability, and exceptional dissolution properties for both organic and inorganic chemical species.” Pingarron et al., Terminology of electrochemical methods of analysis

**prefLabel:** IonicLiquidElectrolyte

**Relations:**

- is\_a **LiquidElectrolyte**

**ElectrochemicalMaterial**

**IRI:** [http://emmo.info/BattINFO#EMMO\\_ebdb68e9\\_c4b5\\_4d57\\_a042\\_c0f51d446755](http://emmo.info/BattINFO#EMMO_ebdb68e9_c4b5_4d57_a042_c0f51d446755)

**elucidation:** A material that participates in a functional process in an electrochemical assembly.

**prefLabel:** ElectrochemicalMaterial

**Relations:**

- is\_a **FunctionalMaterial**

**ElectroactiveSubstance**

**IRI:** [http://emmo.info/BattINFO#EMMO\\_92ba4a12-146e-4b1f-86f3-bcc66ac52763](http://emmo.info/BattINFO#EMMO_92ba4a12-146e-4b1f-86f3-bcc66ac52763)

**prefLabel:** ElectroactiveSubstance

**Relations:**

- is\_a **ElectrochemicalMaterial**

## LiquidElectrolyte

**IRI:** [http://emmo.info/BattINFO#EMMO\\_609b340f\\_3450\\_4a10\\_95c2\\_c457e3eb8a89](http://emmo.info/BattINFO#EMMO_609b340f_3450_4a10_95c2_c457e3eb8a89)

**definition:** “An electrolyte in the liquid phase”

**prefLabel:** LiquidElectrolyte

**Relations:**

- is\_a **Electrolyte**

## Electrochemical Quantity branch

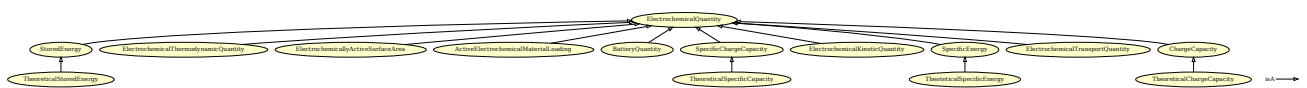


Figure 3.7: Electrochemical Quantity branch.

## ChargeCapacity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_791c1915\\_a791\\_4450\\_acd8\\_7f94764743b5](http://emmo.info/BattINFO#EMMO_791c1915_a791_4450_acd8_7f94764743b5)

**elucidation:** Amount of electric charge that can be stored.

**physicalDimension:** T+1 L0 M0 I+1 Θ0 N0 J0

**prefLabel:** ChargeCapacity

**Relations:**

- is\_a **ElectricCharge**
- is\_a **ElectrochemicalQuantity**

## StoredEnergy

**IRI:** [http://emmo.info/BattINFO#EMMO\\_4f1ed4ee\\_06ba\\_44a4\\_8ece\\_1ee56bf12afe](http://emmo.info/BattINFO#EMMO_4f1ed4ee_06ba_44a4_8ece_1ee56bf12afe)

**elucidation:** Amount of energy stored in a physical object.

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** StoredEnergy

**Relations:**

- is\_a **ElectrochemicalQuantity**
- is\_a **InternalEnergy**

## ElectrochemicalQuantity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_aecc6094\\_c6a5\\_4a36\\_a825\\_8a497a2ae112](http://emmo.info/BattINFO#EMMO_aecc6094_c6a5_4a36_a825_8a497a2ae112)

**elucidation:** Physical quantities defined within the domain of electrochemistry.

**prefLabel:** ElectrochemicalQuantity

**Relations:**

- is\_a **PhysicoChemical**

## TheoreticalSpecificCapacity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_8632dee1\\_0adf\\_4a47\\_8400\\_820b48b86732](http://emmo.info/BattINFO#EMMO_8632dee1_0adf_4a47_8400_820b48b86732)

**elucidation:** TheoreticalCapacity divided by the mass of the cell.

**physicalDimension:** T+1 L0 M-1 I+1 Θ0 N0 J0

**prefLabel:** TheoreticalSpecificCapacity

**Relations:**

- is\_a [SpecificChargeCapacity](#)

## ElectrochemicallyActiveSurfaceArea

**IRI:** [http://emmo.info/BattINFO#EMMO\\_bad1b6f4\\_1b26\\_40e2\\_b552\\_6d53873e3973](http://emmo.info/BattINFO#EMMO_bad1b6f4_1b26_40e2_b552_6d53873e3973)

**elucidation:** The area of the electrode material that is accessible to the electrolyte that is used for charge transfer and/or storage.

**physicalDimension:** T0 L+2 M0 I0 Θ0 N0 J0

**prefLabel:** ElectrochemicallyActiveSurfaceArea

**Relations:**

- is\_a [ElectrochemicalQuantity](#)

## ActiveElectrochemicalMaterialLoading

**IRI:** [http://emmo.info/BattINFO#EMMO\\_c955c089\\_6ee1\\_41a2\\_95fc\\_d534c5cfd3d5](http://emmo.info/BattINFO#EMMO_c955c089_6ee1_41a2_95fc_d534c5cfd3d5)

**elucidation:** Weight of active material in an electrode per unit electrode area.

**physicalDimension:** T0 L-2 M+1 I0 Θ0 N0 J0

**prefLabel:** ActiveElectrochemicalMaterialLoading

**Relations:**

- is\_a [ElectrochemicalQuantity](#)

## BatteryQuantity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_230809da\\_bc18\\_42ec\\_ac94\\_4ca6a86292d1](http://emmo.info/BattINFO#EMMO_230809da_bc18_42ec_ac94_4ca6a86292d1)

**elucidation:** Physical quantities defined within the domain of batteries.

**prefLabel:** BatteryQuantity

**Relations:**

- is\_a [ElectrochemicalQuantity](#)

## SpecificChargeCapacity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_1e3dc60d\\_dd6b\\_47d6\\_8161\\_70004fc5ee30](http://emmo.info/BattINFO#EMMO_1e3dc60d_dd6b_47d6_8161_70004fc5ee30)

**elucidation:** Electric charge per unit mass.

**physicalDimension:** T+1 L0 M-1 I+1 Θ0 N0 J0

**prefLabel:** SpecificChargeCapacity

**Relations:**

- is\_a [ElectrochemicalQuantity](#)
- is\_a [ISQDerivedQuantity](#)

## TheoreticalChargeCapacity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_2b09f961\\_3374\\_42e4\\_8836\\_bffc6bf522fa](http://emmo.info/BattINFO#EMMO_2b09f961_3374_42e4_8836_bffc6bf522fa)

**elucidation:** Theoretical amount of charge a cell can store. Minimum of the theoretical capacity of the positive electrode and negative electrode.

**physicalDimension:** T+1 L0 M0 I+1 Θ0 N0 J0

**prefLabel:** TheoreticalChargeCapacity

**Relations:**

- is\_a **ChargeCapacity**

## SpecificEnergy

**IRI:** [http://emmo.info/BattINFO#EMMO\\_ea0c7651\\_b58b\\_4caf\\_ae02\\_fb6a4dfe6a5d](http://emmo.info/BattINFO#EMMO_ea0c7651_b58b_4caf_ae02_fb6a4dfe6a5d)

**elucidation:** Energy per unit mass.

**physicalDimension:** T-2 L+2 M0 I0 Θ0 N0 J0

**prefLabel:** SpecificEnergy

**Relations:**

- is\_a **ElectrochemicalQuantity**
- is\_a **ISQDerivedQuantity**

## TheoteticalSpecificEnergy

**IRI:** [http://emmo.info/BattINFO#EMMO\\_1c13c786\\_35ae\\_4768\\_88fe\\_795813d465cd](http://emmo.info/BattINFO#EMMO_1c13c786_35ae_4768_88fe_795813d465cd)

**elucidation:** TheoreticalEnergy per unit mass of the cell.

**physicalDimension:** T-2 L+2 M0 I0 Θ0 N0 J0

**prefLabel:** TheoteticalSpecificEnergy

**Relations:**

- is\_a **SpecificEnergy**

## TheoreticalStoredEnergy

**IRI:** [http://emmo.info/BattINFO#EMMO\\_9ea6a862\\_131f\\_4154\\_be47\\_e7417f2fb924](http://emmo.info/BattINFO#EMMO_9ea6a862_131f_4154_be47_e7417f2fb924)

**elucidation:** Theoretical amount of energy that can be stored in a battery cell. Minimum of the theoretical energy of the positive electrode and negative electrode. Product of the Theoretical Capacity and the Theoretical Open-Circuit Voltage.

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** TheoreticalStoredEnergy

**Relations:**

- is\_a **StoredEnergy**

## Electrochemical Transport Quantity branch

### ElectronicConductivity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_6a28741c\\_ef47\\_4a11\\_ba3d\\_166aef581e86](http://emmo.info/BattINFO#EMMO_6a28741c_ef47_4a11_ba3d_166aef581e86)

**physicalDimension:** T+3 L-3 M-1 I+2 Θ0 N0 J0

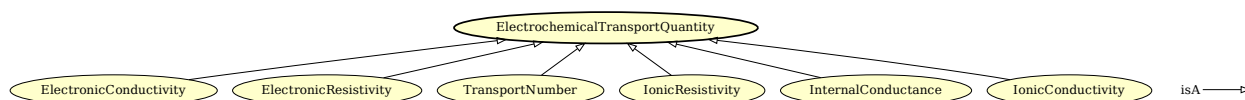


Figure 3.8: Electrochemical Transport Quantity branch.

**prefLabel:** ElectronicConductivity

**Relations:**

- is\_a [ElectricConductivity](#)
- is\_a [ElectrochemicalTransportQuantity](#)

## ElectronicResistivity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_bbcafb37\\_ceec\\_436b\\_bb45\\_080a2bc656aa](http://emmo.info/BattINFO#EMMO_bbcafb37_ceec_436b_bb45_080a2bc656aa)

**elucidation:** Inverse of ElectronicConductivity

**physicalDimension:** T-3 L+3 M+1 I-2 Θ0 N0 J0

**prefLabel:** ElectronicResistivity

**Relations:**

- is\_a [ElectrochemicalTransportQuantity](#)
- is\_a [ElectricResistivity](#)

## TransportNumber

**IRI:** [http://emmo.info/BattINFO#EMMO\\_5c0ad135\\_89ea\\_44da\\_8df7\\_f108f8ee1d75](http://emmo.info/BattINFO#EMMO_5c0ad135_89ea_44da_8df7_f108f8ee1d75)

**elucidation:** Of ions B, the current density due to ions B divided by the sum of current densities of all the ions in the electrolyte.

**iupacEntry:** <https://goldbook.iupac.org/terms/view/T06489>

**physicalDimension:** T0 L0 M0 I0 Θ0 N0 J0

**prefLabel:** TransportNumber

**Relations:**

- is\_a [ElectrochemicalTransportQuantity](#)

## IonicResistivity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_c90a4ca0\\_493f\\_4880\\_a838\\_3a2c4b808a03](http://emmo.info/BattINFO#EMMO_c90a4ca0_493f_4880_a838_3a2c4b808a03)

**elucidation:** Inverse of IonicConductivity

**physicalDimension:** T-3 L+3 M+1 I-2 Θ0 N0 J0

**prefLabel:** IonicResistivity

**Relations:**

- is\_a [ElectricResistivity](#)
- is\_a [ElectrochemicalTransportQuantity](#)

## ElectrochemicalTransportQuantity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_4a450a27\\_b84a\\_4c70\\_a3a9\\_15ec30e2f30b](http://emmo.info/BattINFO#EMMO_4a450a27_b84a_4c70_a3a9_15ec30e2f30b)

**elucidation:** An ElectrochemicalQuantity related to the transport of mass and/or charge.

**prefLabel:** ElectrochemicalTransportQuantity

**Relations:**

- is\_a [ElectrochemicalQuantity](#)

## InternalConductance

**IRI:** [http://emmo.info/BattINFO#EMMO\\_0c9655c6\\_6b0b\\_4819\\_a219\\_f286ad196fa9](http://emmo.info/BattINFO#EMMO_0c9655c6_6b0b_4819_a219_f286ad196fa9)

**physicalDimension:** T+3 L-2 M-1 I+2  $\Theta$ 0 N0 J0

**prefLabel:** InternalConductance

**Relations:**

- is\_a [ElectricConductance](#)
- is\_a [ElectrochemicalTransportQuantity](#)

## IonicConductivity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_64e6ed6a\\_8d17\\_40ba\\_937f\\_f385a54a86c3](http://emmo.info/BattINFO#EMMO_64e6ed6a_8d17_40ba_937f_f385a54a86c3)

**physicalDimension:** T+3 L-3 M-1 I+2  $\Theta$ 0 N0 J0

**prefLabel:** IonicConductivity

**Relations:**

- is\_a [ElectrochemicalTransportQuantity](#)
- is\_a [ElectricConductivity](#)

## Electrochemical Kinetic Quantity branch

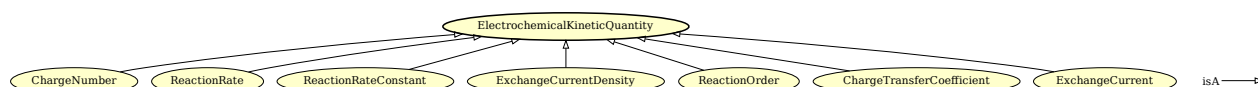


Figure 3.9: Electrochemical Kinetic Quantity branch.

## ChargeNumber

**IRI:** [http://emmo.info/BattINFO#EMMO\\_abfadc99\\_6e43\\_4d37\\_9b04\\_7fc5b0f327ae](http://emmo.info/BattINFO#EMMO_abfadc99_6e43_4d37_9b04_7fc5b0f327ae)

**elucidation:** Number of electrons transferred in a charge transfer reaction between an electrode and a single entity (ion, radical-ion, or molecule) of an electroactive substance, whose identity must be specified.

-Pingarrón et al.: Terminology of electrochemical methods of analysis, DOI: 10.1515/pac-2018-0109

**iupacEntry:** <https://goldbook.iupac.org/terms/view/C00995>

**physicalDimension:** T0 L0 M0 I0  $\Theta$ 0 N0 J0

**prefLabel:** ChargeNumber

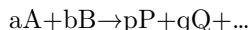
**Relations:**

- is\_a [ElectrochemicalKineticQuantity](#)

## ReactionRate

**IRI:** [http://emmo.info/BattINFO#EMMO\\_47b7d606\\_7030\\_4674\\_9828\\_cf83fb4a2995](http://emmo.info/BattINFO#EMMO_47b7d606_7030_4674_9828_cf83fb4a2995)

**elucidation:** For the general chemical reaction:



occurring under constant-volume conditions, without an appreciable build-up of reaction intermediates, the rate of reaction  $\nu$  is defined as:

$$\nu = -1/a \, d[A]/dt = -1/b \, d[B]/dt = 1/p \, d[P]/dt = 1/q \, d[Q]/dt$$

where symbols placed inside square brackets denote amount (or amount of substance) concentrations (conventionally expressed in units of mol dm<sup>-3</sup>). The symbols R and r are also commonly used in place of  $\nu$ .

**iupacEntry:** <https://goldbook.iupac.org/terms/view/R05156>

**physicalDimension:** T-1 L0 M0 I0 Θ0 N+1 J0

**prefLabel:** ReactionRate

**wikipediaEntry:** [https://en.wikipedia.org/wiki/Reaction\\_rate](https://en.wikipedia.org/wiki/Reaction_rate)

**Relations:**

- is\_a [ElectrochemicalKineticQuantity](#)

## ReactionRateConstant

**IRI:** [http://emmo.info/BattINFO#EMMO\\_dbd808a7\\_8a8f\\_43be\\_9870\\_02cc35bd1646](http://emmo.info/BattINFO#EMMO_dbd808a7_8a8f_43be_9870_02cc35bd1646)

**iupacEntry:** <https://goldbook.iupac.org/terms/view/O04322>

**prefLabel:** ReactionRateConstant

**Relations:**

- is\_a [ElectrochemicalKineticQuantity](#)

## ElectrochemicalKineticQuantity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_21745019\\_2830\\_4395\\_bca7\\_15ddfd266673](http://emmo.info/BattINFO#EMMO_21745019_2830_4395_bca7_15ddfd266673)

**elucidation:** An ElectrochemicalQuantity that relates to the kinetics of a reaction.

**prefLabel:** ElectrochemicalKineticQuantity

**Relations:**

- is\_a [ElectrochemicalQuantity](#)

## ExchangeCurrentDensity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_e9fd9ef9\\_adfe\\_46cb\\_b2f9\\_4558468a25e7](http://emmo.info/BattINFO#EMMO_e9fd9ef9_adfe_46cb_b2f9_4558468a25e7)

**elucidation:** Defined by  $j_0 = i_0/A$ , where  $i_0$  is the exchange current of the electrode reaction and  $A$  is usually taken as the geometric area of the electrode.

**iupacEntry:** <https://goldbook.iupac.org/terms/view/M03777>

**physicalDimension:** T0 L-2 M0 I+1 Θ0 N0 J0

**prefLabel:** ExchangeCurrentDensity

**wikipediaEntry:** [https://en.wikipedia.org/wiki/Exchange\\_current\\_density](https://en.wikipedia.org/wiki/Exchange_current_density)

**Relations:**

- is\_a [ElectrochemicalKineticQuantity](#)



## ReactionOrder

**IRI:** [http://emmo.info/BattINFO#EMMO\\_29a57599\\_aa0d\\_458f\\_b23e\\_666a2da55883](http://emmo.info/BattINFO#EMMO_29a57599_aa0d_458f_b23e_666a2da55883)

**elucidation:** If the macroscopic (observed, empirical or phenomenological) rate of reaction ( $v$ ) for any reaction can be expressed by an empirical differential rate equation (or rate law) which contains a factor of the form  $k [A]^\alpha [B]^\beta \dots$  (expressing in full the dependence of the rate of reaction on the concentrations  $[A]$ ,  $[B]$  ...) where  $\alpha$ ,  $\beta$  are constant exponents (independent of concentration and time) and  $k$  is independent of  $[A]$  and  $[B]$  etc. (rate constant, rate coefficient), then the reaction is said to be of order  $\alpha$  with respect to A, of order  $\beta$  with respect to B, ... , and of (total or overall) order  $n = \alpha + \beta + \dots$ . The exponents  $\alpha$ ,  $\beta$ , ... can be positive or negative integral or rational nonintegral numbers.

**iupacEntry:** <https://goldbook.iupac.org/terms/view/O04322>

**physicalDimension:** T0 L0 M0 I0 Θ0 N0 J0

**prefLabel:** ReactionOrder

**Relations:**

- is\_a [ElectrochemicalKineticQuantity](#)

## ChargeTransferCoefficient

**IRI:** [http://emmo.info/BattINFO#EMMO\\_a4dfa5c1\\_55a9\\_4285\\_b71d\\_90cf6613ca31](http://emmo.info/BattINFO#EMMO_a4dfa5c1_55a9_4285_b71d_90cf6613ca31)

**elucidation:** The fraction of the electrostatic potential energy affecting the reduction rate in an electrode reaction, with the remaining fraction affecting the corresponding oxidation rate.

- Guidelli et al.: Transfer coefficient: An assessment, DOI: 10.1515/pac-2014-5026

**physicalDimension:** T0 L0 M0 I0 Θ0 N0 J0

**prefLabel:** ChargeTransferCoefficient

**wikipediaEntry:** [https://en.wikipedia.org/wiki/Charge\\_transfer\\_coefficient](https://en.wikipedia.org/wiki/Charge_transfer_coefficient)

**Relations:**

- is\_a [ElectrochemicalKineticQuantity](#)

## ExchangeCurrent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_ccde24bb\\_790a\\_40ca\\_a06e\\_cea156a61031](http://emmo.info/BattINFO#EMMO_ccde24bb_790a_40ca_a06e_cea156a61031)

**elucidation:** The common value ( $i_0$ ) of the anodic and cathodic partial currents when the reaction is at equilibrium:

$$i_0 = i_a = -i_c$$

For an electrode at equilibrium at which only one reaction is significant  $i = 0$ . When more than one reaction is significant at a given electrode, subscripts to  $i_0$  may be used to distinguish exchange currents.  $i$  is not usually zero when only one of these reactions is at equilibrium.

**iupacEntry:** <https://goldbook.iupac.org/terms/view/E02238>

**physicalDimension:** T0 L0 M0 I+1 Θ0 N0 J0

**prefLabel:** ExchangeCurrent

**Relations:**

- is\_a [ElectrochemicalKineticQuantity](#)

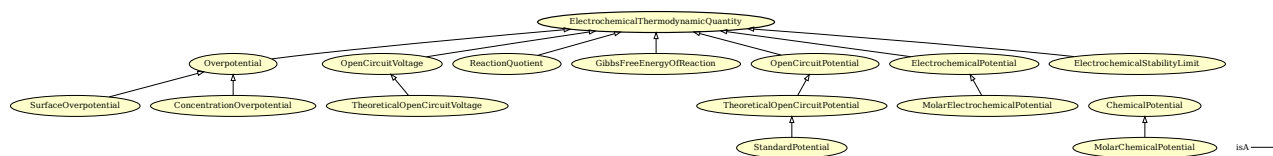


Figure 3.10: Electrochemical Thermodynamic Quantity branch.

## Electrochemical Thermodynamic Quantity branch

### SurfaceOverpotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_60741c58\\_a10d\\_4aa6\\_bb68\\_0066a6ff8e30](http://emmo.info/BattINFO#EMMO_60741c58_a10d_4aa6_bb68_0066a6ff8e30)

**elucidation:** The potential of a working electrode relative to a reference electrode of the same kinds placed in the solution adjacent to the surface of the working electrode (just outside the double layer).

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** SurfaceOverpotential

**Relations:**

- is\_a **Overpotential**

### TheoreticalOpenCircuitVoltage

**IRI:** [http://emmo.info/BattINFO#EMMO\\_34e440e0\\_b720\\_4585\\_a915\\_fbe5abb8615d](http://emmo.info/BattINFO#EMMO_34e440e0_b720_4585_a915_fbe5abb8615d)

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** TheoreticalOpenCircuitVoltage

**Relations:**

- is\_a **OpenCircuitVoltage**

### Overpotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_1cd1d777\\_e67b\\_47eb\\_81f1\\_edac35d9f2c6](http://emmo.info/BattINFO#EMMO_1cd1d777_e67b_47eb_81f1_edac35d9f2c6)

**elucidation:** Deviation of the potential of an electrode from its equilibrium value required to cause a given current to flow through the electrode.

**iupacEntry:** <https://goldbook.iupac.org/terms/view/O04358>

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** Overpotential

**Relations:**

- is\_a **ElectrochemicalThermodynamicQuantity**

### OpenCircuitVoltage

**IRI:** [http://emmo.info/BattINFO#EMMO\\_0c0c623c\\_43b8\\_426d\\_a536\\_168108e2353a](http://emmo.info/BattINFO#EMMO_0c0c623c_43b8_426d_a536_168108e2353a)

**elucidation:** Measured difference between two electrodes without external current flow.

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** OpenCircuitVoltage

**Relations:**

- is\_a [ElectricPotential](#)
- is\_a [ElectrochemicalThermodynamicQuantity](#)

## ReactionQuotient

**IRI:** [http://emmo.info/BattINFO#EMMO\\_740d5817\\_3fa7\\_464a\\_90c3\\_55552e51a3df](http://emmo.info/BattINFO#EMMO_740d5817_3fa7_464a_90c3_55552e51a3df)

**physicalDimension:** T0 L0 M0 I0 Θ0 N0 J0

**prefLabel:** ReactionQuotient

**wikipediaEntry:** [https://en.wikipedia.org/wiki/Reaction\\_quotient](https://en.wikipedia.org/wiki/Reaction_quotient)

**Relations:**

- is\_a [ElectrochemicalThermodynamicQuantity](#)

## TheoreticalOpenCircuitPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_d91940f0\\_c8b6\\_4505\\_9b68\\_6bf6cfc5c544](http://emmo.info/BattINFO#EMMO_d91940f0_c8b6_4505_9b68_6bf6cfc5c544)

**elucidation:** Theoretical electrode potential considering a given electrochemical charge-transfer reaction.

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** TheoreticalOpenCircuitPotential

**Relations:**

- is\_a [OpenCircuitPotential](#)

## GibbsFreeEnergyOfReaction

**IRI:** [http://emmo.info/BattINFO#EMMO\\_d62ff300\\_26ac\\_4b00\\_bfcd\\_04a68aff5dc3](http://emmo.info/BattINFO#EMMO_d62ff300_26ac_4b00_bfcd_04a68aff5dc3)

**elucidation:** Change in the Gibbs free energy between the products and reactants in a reaction.

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** GibbsFreeEnergyOfReaction

**Relations:**

- is\_a [ElectrochemicalThermodynamicQuantity](#)

## OpenCircuitPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_9c657fdc\\_b9d3\\_4964\\_907c\\_f9a6e8c5f52b](http://emmo.info/BattINFO#EMMO_9c657fdc_b9d3_4964_907c_f9a6e8c5f52b)

**elucidation:** Measured electric potential of an electrode without external current flow.

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** OpenCircuitPotential

**Relations:**

- is\_a [ElectricPotential](#)
- is\_a [ElectrochemicalThermodynamicQuantity](#)

## ElectrochemicalPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_1422cde1\\_929e\\_46b6\\_b0dc\\_1010eebc5dfd](http://emmo.info/BattINFO#EMMO_1422cde1_929e_46b6_b0dc_1010eebc5dfd)

**iupacEntry:** <https://goldbook.iupac.org/terms/view/E01945>

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** ElectrochemicalPotential

**Relations:**

- is\_a [ElectrochemicalThermodynamicQuantity](#)

## ElectrochemicalThermodynamicQuantity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_2d896559\\_eee3\\_447c\\_9759\\_87c854a4266a](http://emmo.info/BattINFO#EMMO_2d896559_eee3_447c_9759_87c854a4266a)

**elucidation:** A thermodynamically derived ElectrochemicalQuantity.

**prefLabel:** ElectrochemicalThermodynamicQuantity

**Relations:**

- is\_a [ElectrochemicalQuantity](#)

## ChemicalPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_17e305af\\_52a9\\_4255\\_a70f\\_700ba1088f13](http://emmo.info/BattINFO#EMMO_17e305af_52a9_4255_a70f_700ba1088f13)

**elucidation:** Energy that can be absorbed or released due to a change of the particle number of the given species

**iupacEntry:** <https://goldbook.iupac.org/terms/view/C01032>

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** ChemicalPotential

**wikipediaEntry:** [https://en.wikipedia.org/wiki/Chemical\\_potential](https://en.wikipedia.org/wiki/Chemical_potential)

**Relations:**

- is\_a [ElectrochemicalThermodynamicQuantity](#)

## MolarChemicalPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_68dc1bf8\\_9813\\_43c8\\_b428\\_6bd614c3161d](http://emmo.info/BattINFO#EMMO_68dc1bf8_9813_43c8_b428_6bd614c3161d)

**elucidation:** ChemicalPotential per mole.

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N-1 J0

**prefLabel:** MolarChemicalPotential

**Relations:**

- is\_a [ChemicalPotential](#)

## StandardPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_7fc10197\\_41d9\\_4c1e\\_a107\\_928f03eb2d36](http://emmo.info/BattINFO#EMMO_7fc10197_41d9_4c1e_a107_928f03eb2d36)

**elucidation:** Theoretical equilibrium potential under standard conditions.

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** StandardPotential

**Relations:**

- is\_a [TheoreticalOpenCircuitPotential](#)

## TheoreticalOpenCircuitVoltage

**IRI:** [http://emmo.info/BattINFO#EMMO\\_367a4916\\_d03a\\_483c\\_9f2c\\_6588370fc9d9](http://emmo.info/BattINFO#EMMO_367a4916_d03a_483c_9f2c_6588370fc9d9)

**elucidation:** Difference between the theoretical electric potentials of the positive electrode and negative electrode under no current flow.

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** TheoreticalOpenCircuitVoltage

**Relations:**

- is\_a [OpenCircuitVoltage](#)

## ConcentrationOverpotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_9ed7210c\\_c4fa\\_467b\\_822d\\_ba12f885bdf4](http://emmo.info/BattINFO#EMMO_9ed7210c_c4fa_467b_822d_ba12f885bdf4)

**elucidation:** The concentration overpotential of an electrode reaction at a given electrode current density (c.d.) is basically the difference in equilibrium potentials across the diffusion layer. More precisely, it is the potential of a reference electrode (of the same electrode reaction as the working electrode ) with the interfacial concentrations which establish themselves at c.d., relative to the potential of a similar reference electrode with the concentrations of the bulk solution. From such a measured potential difference, with c.d. flowing, one needs to subtract the ohmic potential drop prevailing between the two electrodes.

**iupacEntry:** <https://goldbook.iupac.org/terms/view/C01230>

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** ConcentrationOverpotential

**Relations:**

- is\_a [Overpotential](#)

## MolarElectrochemicalPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_7fe804b8\\_6126\\_4132\\_be8f\\_b4985d61b1f6](http://emmo.info/BattINFO#EMMO_7fe804b8_6126_4132_be8f_b4985d61b1f6)

**elucidation:** ElectrochemicalPotential per mole.

**iupacEntry:** <https://goldbook.iupac.org/terms/view/E01945>

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N-1 J0

**prefLabel:** MolarElectrochemicalPotential

**Relations:**

- is\_a [ElectrochemicalPotential](#)

## ElectrochemicalStabilityLimit

**IRI:** [http://emmo.info/BattINFO#EMMO\\_8f4b90ef\\_fea4\\_47c9\\_99f5\\_a9b3290a505d](http://emmo.info/BattINFO#EMMO_8f4b90ef_fea4_47c9_99f5_a9b3290a505d)

**elucidation:** Electric potential at which a material undergoes an oxidation or reduction decomposition.

**example:** For water, the electrochemical stability limits are: Reduction: 0 V Oxidation: 1.23 V

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** ElectrochemicalStabilityLimit

**Relations:**

- is\_a [ElectrochemicalThermodynamicQuantity](#)

## Electrochemical Constant branch

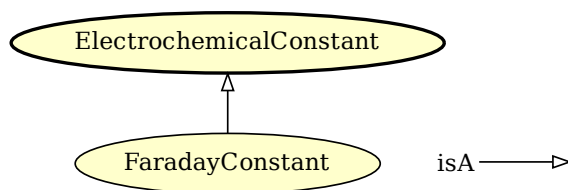


Figure 3.11: Electrochemical Constant branch.

### ElectrochemicalConstant

**IRI:** [http://emmo.info/BattINFO#EMMO\\_cdaf1d61\\_b5df\\_43a9\\_91a4\\_a5b7f719e2b4](http://emmo.info/BattINFO#EMMO_cdaf1d61_b5df_43a9_91a4_a5b7f719e2b4)

**prefLabel:** ElectrochemicalConstant

**Relations:**

- is\_a **PhysicalConstant**

### FaradayConstant

**IRI:** [http://emmo.info/BattINFO#EMMO\\_499a652b\\_5be6\\_4931\\_be7b\\_15d42e544b0b](http://emmo.info/BattINFO#EMMO_499a652b_5be6_4931_be7b_15d42e544b0b)

**definition:** Product of ElectronCharge and AvagadroConstant

**elucidation:** Fundamental physical constant representing molar elementary charge:  $F=9.648\,533\,99(24)\times 10^4$  C mol<sup>-1</sup>.

**iupacEntry:** <https://goldbook.iupac.org/terms/view/F02325>

**physicalDimension:** T+1 L0 M0 I+1 Θ0 N-1 J0

**prefLabel:** FaradayConstant

**Relations:**

- is\_a **ElectrochemicalConstant**

## Additional physical quantities

### IonicResistivity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_c90a4ca0\\_493f\\_4880\\_a838\\_3a2c4b808a03](http://emmo.info/BattINFO#EMMO_c90a4ca0_493f_4880_a838_3a2c4b808a03)

**elucidation:** Inverse of IonicConductivity

**physicalDimension:** T-3 L+3 M+1 I-2 Θ0 N0 J0

**prefLabel:** IonicResistivity

**Relations:**

- is\_a **ElectricResistivity**
- is\_a **ElectrochemicalTransportQuantity**

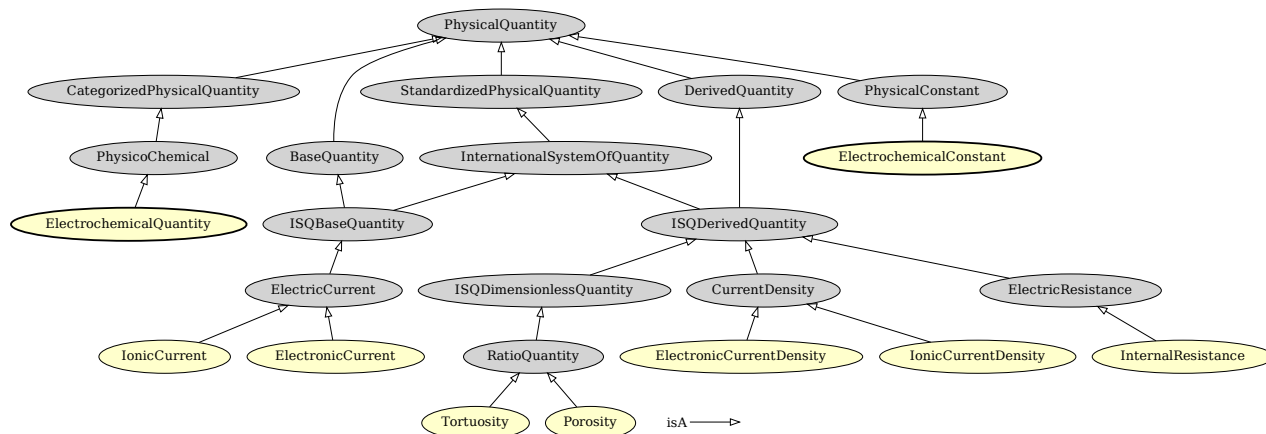


Figure 3.12: Additional physical quantities defined in BattINFO. Parent classes belonging to EMMO are shown in gray.

## TheoreticalOpenCircuitVoltage

**IRI:** [http://emmo.info/BattINFO#EMMO\\_34e440e0\\_b720\\_4585\\_a915\\_fbe5abb8615d](http://emmo.info/BattINFO#EMMO_34e440e0_b720_4585_a915_fbe5abb8615d)

**physicalDimension:** T-3 L+2 M+1 I-1  $\Theta$ 0 N0 J0

**prefLabel:** TheoreticalOpenCircuitVoltage

**Relations:**

- is\_a **OpenCircuitVoltage**

## ElectronicConductivity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_6a28741c\\_ef47\\_4a11\\_ba3d\\_166aef581e86](http://emmo.info/BattINFO#EMMO_6a28741c_ef47_4a11_ba3d_166aef581e86)

**physicalDimension:** T+3 L-3 M-1 I+2  $\Theta$ 0 N0 J0

**prefLabel:** ElectronicConductivity

**Relations:**

- is\_a **ElectricConductivity**
- is\_a **ElectrochemicalTransportQuantity**

## ElectronicResistivity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_bbcafb37\\_ceec\\_436b\\_bb45\\_080a2bc656aa](http://emmo.info/BattINFO#EMMO_bbcafb37_ceec_436b_bb45_080a2bc656aa)

**elucidation:** Inverse of ElectronicConductivity

**physicalDimension:** T-3 L+3 M+1 I-2  $\Theta$ 0 N0 J0

**prefLabel:** ElectronicResistivity

**Relations:**

- is\_a **ElectrochemicalTransportQuantity**
- is\_a **ElectricResistivity**

## OpenCircuitPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_9c657fdc\\_b9d3\\_4964\\_907c\\_f9a6e8c5f52b](http://emmo.info/BattINFO#EMMO_9c657fdc_b9d3_4964_907c_f9a6e8c5f52b)

**elucidation:** Measured electric potential of an electrode without external current flow.

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** OpenCircuitPotential

**Relations:**

- is\_a [ElectricPotential](#)
- is\_a [ElectrochemicalThermodynamicQuantity](#)

## Tortuosity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_caa0969a\\_1e27\\_4950\\_8af6\\_5b72fd20e504](http://emmo.info/BattINFO#EMMO_caa0969a_1e27_4950_8af6_5b72fd20e504)

**elucidation:** A measure of deviation from a straight line. It is the ratio of the actual distance traveled divided by the straight line distance.

**physicalDimension:** T0 L0 M0 I0 Θ0 N0 J0

**prefLabel:** Tortuosity

**Relations:**

- is\_a [RatioQuantity](#)

## Porosity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_3a38e30d\\_4c97\\_49d4\\_b0f4\\_661c9779e039](http://emmo.info/BattINFO#EMMO_3a38e30d_4c97_49d4_b0f4_661c9779e039)

**elucidation:** Porosity or void fraction is a measure of the void (i.e. “empty”) spaces in a material, and is a fraction of the volume of voids over the total volume, between 0 and 1, or as a percentage between 0% and 100%.

**physicalDimension:** T0 L0 M0 I0 Θ0 N0 J0

**prefLabel:** Porosity

**Relations:**

- is\_a [RatioQuantity](#)

## TheoreticalStoredEnergy

**IRI:** [http://emmo.info/BattINFO#EMMO\\_9ea6a862\\_131f\\_4154\\_be47\\_e7417f2fb924](http://emmo.info/BattINFO#EMMO_9ea6a862_131f_4154_be47_e7417f2fb924)

**elucidation:** Theoretical amount of energy that can be stored in a battery cell. Minimum of the theoretical energy of the positive electrode and negative electrode. Product of the Theoretical Capacity and the Theoretical Open-Circuit Voltage.

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** TheoreticalStoredEnergy

**Relations:**

- is\_a [StoredEnergy](#)

## TheoreticalOpenCircuitVoltage

**IRI:** [http://emmo.info/BattINFO#EMMO\\_367a4916\\_d03a\\_483c\\_9f2c\\_6588370fc9d9](http://emmo.info/BattINFO#EMMO_367a4916_d03a_483c_9f2c_6588370fc9d9)

**elucidation:** Difference between the theoretical electric potentials of the positive electrode and negative electrode under no current flow.

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** TheoreticalOpenCircuitVoltage

**Relations:**



- is\_a **OpenCircuitVoltage**

## IonicCurrent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_569a62a5\\_3b7e\\_4099\\_8a4c\\_f76e229a0347](http://emmo.info/BattINFO#EMMO_569a62a5_3b7e_4099_8a4c_f76e229a0347)

**elucidation:** A flow of electric charge, in which ions are the charge carrier.

**physicalDimension:** T0 L0 M0 I+1 Θ0 N0 J0

**prefLabel:** IonicCurrent

**Relations:**

- is\_a **ElectricCurrent**

## SpecificChargeCapacity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_1e3dc60d\\_dd6b\\_47d6\\_8161\\_70004fc5ee30](http://emmo.info/BattINFO#EMMO_1e3dc60d_dd6b_47d6_8161_70004fc5ee30)

**elucidation:** Electric charge per unit mass.

**physicalDimension:** T+1 L0 M-1 I+1 Θ0 N0 J0

**prefLabel:** SpecificChargeCapacity

**Relations:**

- is\_a **ElectrochemicalQuantity**
- is\_a **ISQDerivedQuantity**

## TheoreticalChargeCapacity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_2b09f961\\_3374\\_42e4\\_8836\\_bffc6bf522fa](http://emmo.info/BattINFO#EMMO_2b09f961_3374_42e4_8836_bffc6bf522fa)

**elucidation:** Theoretical amount of charge a cell can store. Minimum of the theoretical capacity of the positive electrode and negative electrode.

**physicalDimension:** T+1 L0 M0 I+1 Θ0 N0 J0

**prefLabel:** TheoreticalChargeCapacity

**Relations:**

- is\_a **ChargeCapacity**

## SpecificEnergy

**IRI:** [http://emmo.info/BattINFO#EMMO\\_ea0c7651\\_b58b\\_4caf\\_ae02\\_fb6a4dfe6a5d](http://emmo.info/BattINFO#EMMO_ea0c7651_b58b_4caf_ae02_fb6a4dfe6a5d)

**elucidation:** Energy per unit mass.

**physicalDimension:** T-2 L+2 M0 I0 Θ0 N0 J0

**prefLabel:** SpecificEnergy

**Relations:**

- is\_a **ElectrochemicalQuantity**
- is\_a **ISQDerivedQuantity**

## ChargeCapacity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_791c1915\\_a791\\_4450\\_acd8\\_7f94764743b5](http://emmo.info/BattINFO#EMMO_791c1915_a791_4450_acd8_7f94764743b5)

**elucidation:** Amount of electric charge that can be stored.

**physicalDimension:** T+1 L0 M0 I+1  $\Theta$ 0 N0 J0

**prefLabel:** ChargeCapacity

**Relations:**

- is\_a [ElectricCharge](#)
- is\_a [ElectrochemicalQuantity](#)

## InternalConductance

**IRI:** [http://emmo.info/BattINFO#EMMO\\_0c9655c6\\_6b0b\\_4819\\_a219\\_f286ad196fa9](http://emmo.info/BattINFO#EMMO_0c9655c6_6b0b_4819_a219_f286ad196fa9)

**physicalDimension:** T+3 L-2 M-1 I+2  $\Theta$ 0 N0 J0

**prefLabel:** InternalConductance

**Relations:**

- is\_a [ElectricConductance](#)
- is\_a [ElectrochemicalTransportQuantity](#)

## OpenCircuitVoltage

**IRI:** [http://emmo.info/BattINFO#EMMO\\_0c0c623c\\_43b8\\_426d\\_a536\\_168108e2353a](http://emmo.info/BattINFO#EMMO_0c0c623c_43b8_426d_a536_168108e2353a)

**elucidation:** Measured difference between two electrodes without external current flow.

**physicalDimension:** T-3 L+2 M+1 I-1  $\Theta$ 0 N0 J0

**prefLabel:** OpenCircuitVoltage

**Relations:**

- is\_a [ElectricPotential](#)
- is\_a [ElectrochemicalThermodynamicQuantity](#)

## TheoreticalOpenCircuitPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_d91940f0\\_c8b6\\_4505\\_9b68\\_6bf6cfc5c544](http://emmo.info/BattINFO#EMMO_d91940f0_c8b6_4505_9b68_6bf6cfc5c544)

**elucidation:** Theoretical electrode potential considering a given electrochemical charge-transfer reaction.

**physicalDimension:** T-3 L+2 M+1 I-1  $\Theta$ 0 N0 J0

**prefLabel:** TheoreticalOpenCircuitPotential

**Relations:**

- is\_a [OpenCircuitPotential](#)

## InternalResistance

**IRI:** [http://emmo.info/BattINFO#EMMO\\_9bf40017\\_3f58\\_4030\\_ada7\\_cb37a3dfda2d](http://emmo.info/BattINFO#EMMO_9bf40017_3f58_4030_ada7_cb37a3dfda2d)

**elucidation:** Impetance associated with a power source.

**physicalDimension:** T-3 L+2 M+1 I-2  $\Theta$ 0 N0 J0

**prefLabel:** InternalResistance

**Relations:**

- is\_a [ElectricResistance](#)

## IonicCurrentDensity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_683e4991\\_38f3\\_42e1\\_84de\\_5ee25942d2e8](http://emmo.info/BattINFO#EMMO_683e4991_38f3_42e1_84de_5ee25942d2e8)

**elucidation:** Current density in which the charge carriers are ions.

**physicalDimension:** T0 L-2 M0 I+1 Θ0 N0 J0

**prefLabel:** IonicCurrentDensity

**Relations:**

- is\_a [CurrentDensity](#)

## StoredEnergy

**IRI:** [http://emmo.info/BattINFO#EMMO\\_4f1ed4ee\\_06ba\\_44a4\\_8ece\\_1ee56bf12afe](http://emmo.info/BattINFO#EMMO_4f1ed4ee_06ba_44a4_8ece_1ee56bf12afe)

**elucidation:** Amount of energy stored in a physical object.

**physicalDimension:** T-2 L+2 M+1 I0 Θ0 N0 J0

**prefLabel:** StoredEnergy

**Relations:**

- is\_a [ElectrochemicalQuantity](#)
- is\_a [InternalEnergy](#)

## StandardPotential

**IRI:** [http://emmo.info/BattINFO#EMMO\\_7fc10197\\_41d9\\_4c1e\\_a107\\_928f03eb2d36](http://emmo.info/BattINFO#EMMO_7fc10197_41d9_4c1e_a107_928f03eb2d36)

**elucidation:** Theoretical equilibrium potential under standard conditions.

**physicalDimension:** T-3 L+2 M+1 I-1 Θ0 N0 J0

**prefLabel:** StandardPotential

**Relations:**

- is\_a [TheoreticalOpenCircuitPotential](#)

## TheoreticalSpecificCapacity

**IRI:** [http://emmo.info/BattINFO#EMMO\\_8632dee1\\_0adf\\_4a47\\_8400\\_820b48b86732](http://emmo.info/BattINFO#EMMO_8632dee1_0adf_4a47_8400_820b48b86732)

**elucidation:** TheoreticalCapacity divided by the mass of the cell.

**physicalDimension:** T+1 L0 M-1 I+1 Θ0 N0 J0

**prefLabel:** TheoreticalSpecificCapacity

**Relations:**

- is\_a [SpecificChargeCapacity](#)

## ElectronicCurrent

**IRI:** [http://emmo.info/BattINFO#EMMO\\_e73063fe\\_30a4\\_4ed5\\_b9f6\\_11979f807a42](http://emmo.info/BattINFO#EMMO_e73063fe_30a4_4ed5_b9f6_11979f807a42)

**elucidation:** A flow of electric charge, in which electrons are the charge carrier

**physicalDimension:** T0 L0 M0 I+1 Θ0 N0 J0

**prefLabel:** ElectronicCurrent

**Relations:**

- is\_a **ElectricCurrent**

**ElectronicCurrentDensity**

**IRI:** [http://emmo.info/BattINFO#EMMO\\_bfc8c075\\_246e\\_4633\\_ba8e\\_906a9f5f2e3a](http://emmo.info/BattINFO#EMMO_bfc8c075_246e_4633_ba8e_906a9f5f2e3a)

**elucidation:** Current density in which the charge carriers are electrons.

**physicalDimension:** T0 L-2 M0 I+1 Θ0 N0 J0

**prefLabel:** ElectronicCurrentDensity

**Relations:**

- is\_a **CurrentDensity**

**TheoteticalSpecificEnergy**

**IRI:** [http://emmo.info/BattINFO#EMMO\\_1c13c786\\_35ae\\_4768\\_88fe\\_795813d465cd](http://emmo.info/BattINFO#EMMO_1c13c786_35ae_4768_88fe_795813d465cd)

**elucidation:** TheoreticalEnergy per unit mass of the cell.

**physicalDimension:** T-2 L+2 M0 I0 Θ0 N0 J0

**prefLabel:** TheoteticalSpecificEnergy

**Relations:**

- is\_a **SpecificEnergy**

**IonicConductivity**

**IRI:** [http://emmo.info/BattINFO#EMMO\\_64e6ed6a\\_8d17\\_40ba\\_937f\\_f385a54a86c3](http://emmo.info/BattINFO#EMMO_64e6ed6a_8d17_40ba_937f_f385a54a86c3)

**physicalDimension:** T+3 L-3 M-1 I+2 Θ0 N0 J0

**prefLabel:** IonicConductivity

**Relations:**

- is\_a **ElectrochemicalTransportQuantity**
- is\_a **ElectricConductivity**

**Material Relation branch**

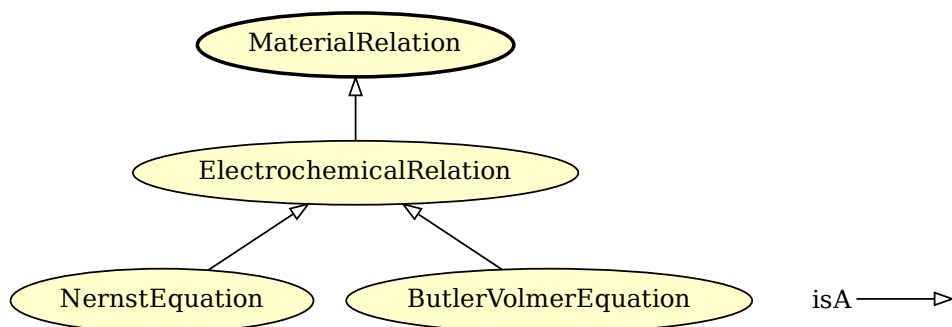


Figure 3.13: Material Relation branch.

## NernstEquation

**IRI:** [http://emmo.info/BattINFO#EMMO\\_fe3a6c9a\\_85b8\\_4da6\\_aa4f\\_71c8de74939e](http://emmo.info/BattINFO#EMMO_fe3a6c9a_85b8_4da6_aa4f_71c8de74939e)

**elucidation:** An equation that describes the equilibrium potential of an electrode at which a given electrochemical charge-transfer reaction occurs, considering the activity of the reacting species and the temperature of the system.

**prefLabel:** NernstEquation

**wikipediaEntry:** [https://en.wikipedia.org/wiki/Nernst\\_equation](https://en.wikipedia.org/wiki/Nernst_equation)

### Relations:

- is\_a **ElectrochemicalRelation**
- hasSpatialDirectPart some **ChargeNumber**
- hasSpatialDirectPart some **FaradayConstant**
- hasSpatialDirectPart some **ThermodynamicTemperature**
- hasSpatialDirectPart some **TheoreticalOpenCircuitPotential**
- hasSpatialDirectPart some **MolarGasConstant**
- hasSpatialDirectPart some **ReactionQuotient**
- hasSpatialDirectPart some **StandardPotential**

## MaterialRelation

**IRI:** [http://emmo.info/emmo#EMMO\\_e5438930\\_04e7\\_4d42\\_ade5\\_3700d4a52ab7](http://emmo.info/emmo#EMMO_e5438930_04e7_4d42_ade5_3700d4a52ab7)

**elucidation:** An ‘equation’ that stands for a physical assumption specific to a material, and provides an expression for a ‘physics\_quantity’ (the dependent variable) as function of other variables, physics\_quantity or data (independent variables).

**example:** The Lennard-Jones potential.

A force field.

An Hamiltonian.

**prefLabel:** MaterialRelation

### Relations:

- is\_a **Equation**
- hasSpatialDirectPart some **PhysicalQuantity**

## ButlerVolmerEquation

**IRI:** [http://emmo.info/BattINFO#EMMO\\_d48ea516\\_5cac\\_4f86\\_bc88\\_21b6276c0938](http://emmo.info/BattINFO#EMMO_d48ea516_5cac_4f86_bc88_21b6276c0938)

**elucidation:** The standard phenomenological model for electrode kinetics, describing the relation between the electrode current from an electrochemical charge-transfer reaction and the surface overpotential of the electrode.

**prefLabel:** ButlerVolmerEquation

### Relations:

- is\_a **ElectrochemicalRelation**
- hasSpatialDirectPart some **MolarGasConstant**
- hasSpatialDirectPart some **ElectricCurrent**
- hasSpatialDirectPart some **ExchangeCurrent**
- hasSpatialDirectPart some **SurfaceOverpotential**
- hasSpatialDirectPart some **FaradayConstant**
- hasSpatialDirectPart some **ChargeNumber**
- hasSpatialDirectPart some **ThermodynamicTemperature**

## ElectrochemicalRelation

**IRI:** [http://emmo.info/BattINFO#EMMO\\_3d805c2a\\_4801\\_440e\\_9e4d\\_0fa5585c76ae](http://emmo.info/BattINFO#EMMO_3d805c2a_4801_440e_9e4d_0fa5585c76ae)

**elucidation:** A material relation in electrochemistry.

**prefLabel:** ElectrochemicalRelation

**Relations:**

- is\_a **MaterialRelation**

## Chemical Species branch

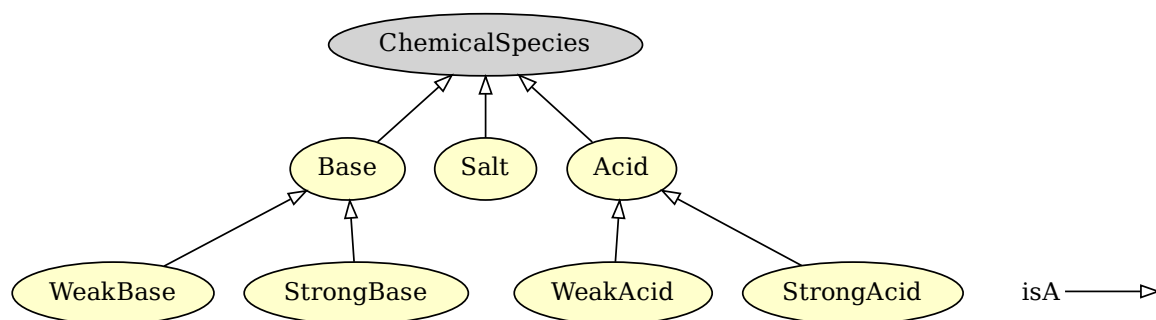


Figure 3.14: Chemical Species branch.

## Acid

**IRI:** [http://emmo.info/BattINFO#EMMO\\_c230694a\\_04ce\\_4719\\_88a4\\_ecfa85167c30](http://emmo.info/BattINFO#EMMO_c230694a_04ce_4719_88a4_ecfa85167c30)

**definition:** “A molecular entity or chemical species capable of donating a hydron (proton) (see Brønsted acid) or capable of forming a covalent bond with an electron pair (see Lewis acid).” - IUPAC Gold Book

**iupacEntry:** <https://goldbook.iupac.org/terms/view/A00071>

**prefLabel:** Acid

**Relations:**

- is\_a **ChemicalSpecies**

## WeakBase

**IRI:** [http://emmo.info/BattINFO#EMMO\\_ce548161\\_c987\\_4beb\\_9091\\_adcf80027310](http://emmo.info/BattINFO#EMMO_ce548161_c987_4beb_9091_adcf80027310)

**definition:** “A base that partially dissociates in water.”

**prefLabel:** WeakBase

**Relations:**

- is\_a **Base**

## StrongBase

**IRI:** [http://emmo.info/BattINFO#EMMO\\_a1bbb273\\_bc05\\_4e80\\_8817\\_82479178bb41](http://emmo.info/BattINFO#EMMO_a1bbb273_bc05_4e80_8817_82479178bb41)

**definition:** “A base that completely dissociates in water.”

**prefLabel:** StrongBase

**Relations:**

- is\_a **Base**

## Base

**IRI:** [http://emmo.info/BattINFO#EMMO\\_af499b32\\_68a7\\_4b8c\\_972e\\_4ebdba8b314e](http://emmo.info/BattINFO#EMMO_af499b32_68a7_4b8c_972e_4ebdba8b314e)

**definition:** “A chemical species or molecular entity having an available pair of electrons capable of forming a covalent bond with a hydron (proton) (see Brønsted base) or with the vacant orbital of some other species (see Lewis base).” - IUPAC Gold Book

**iupacEntry:** <https://goldbook.iupac.org/terms/view/B00601>

**prefLabel:** Base

**Relations:**

- is\_a **ChemicalSpecies**

## Salt

**IRI:** [http://emmo.info/BattINFO#EMMO\\_b6a52fdb\\_ba40\\_4caf\\_a8d9\\_523a467eb799](http://emmo.info/BattINFO#EMMO_b6a52fdb_ba40_4caf_a8d9_523a467eb799)

**definition:** “A chemical compound consisting of an assembly of cations and anions.” IUPAC Gold Book

**iupacEntry:** <https://goldbook.iupac.org/terms/view/S05447>

**prefLabel:** Salt

**Relations:**

- is\_a **ChemicalSpecies**

## WeakAcid

**IRI:** [http://emmo.info/BattINFO#EMMO\\_e3ec1307\\_09d7\\_4b61\\_97e3\\_a69ec87fb408](http://emmo.info/BattINFO#EMMO_e3ec1307_09d7_4b61_97e3_a69ec87fb408)

**definition:** “An acid that partially dissociates in water.”

**prefLabel:** WeakAcid

**Relations:**

- is\_a **Acid**

## StrongAcid

**IRI:** [http://emmo.info/BattINFO#EMMO\\_c9e0fb9b\\_c11e\\_48ab\\_9245\\_04b45e15dcfb](http://emmo.info/BattINFO#EMMO_c9e0fb9b_c11e_48ab_9245_04b45e15dcfb)

**definition:** An acid that completely dissociates in water.

**prefLabel:** StrongAcid

**Relations:**

- is\_a **Acid**

## Real world objects

### ElectrodePore

**IRI:** [http://emmo.info/BattINFO#EMMO\\_4f3a2ba3-7abc-4150-ba98-3973d865690f](http://emmo.info/BattINFO#EMMO_4f3a2ba3-7abc-4150-ba98-3973d865690f)

**elucidation:** A pore that exists within an electrode host domain.

**prefLabel:** ElectrodePore

**Relations:**

- is\_a **Pore**
- hasContactWith some **PorousElectrode**

### ElectrochemicalDevice

**IRI:** [http://emmo.info/BattINFO#EMMO\\_0acd0fc2\\_1048\\_4604\\_8e90\\_bf4e84bd87df](http://emmo.info/BattINFO#EMMO_0acd0fc2_1048_4604_8e90_bf4e84bd87df)

**elucidation:** A device whose primary function is facilitating the conversion between chemical and electrical energy.

**prefLabel:** ElectrochemicalDevice

**Relations:**

- is\_a **Device**
- hasPart some **ElectrochemicalComponent**

## Physical dimensions

### ChargePerMassDimension

**IRI:** [http://emmo.info/BattINFO#EMMO\\_7bfcbe2d\\_eac6\\_4953\\_86d6\\_6f075334cf29](http://emmo.info/BattINFO#EMMO_7bfcbe2d_eac6_4953_86d6_6f075334cf29)

**prefLabel:** ChargePerMassDimension

**Relations:**

- is\_a **PhysicalDimension**
- equivalent\_to hasSymbolData value “T+1 L0 M-1 I+1 Θ0 N0 J0”



## Chapter 4

# Appendix

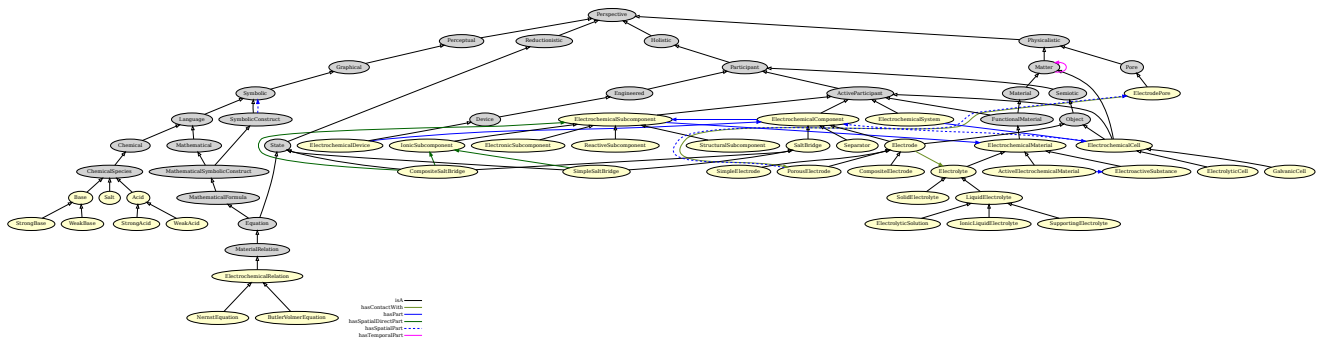


Figure 4.1: All classes defined with the BattINFO namespace, except physical quantities. In addition parent classes belonging to EMMO are shown in gray.

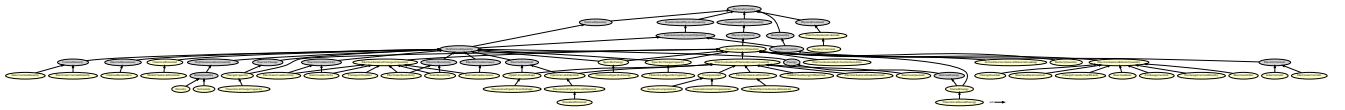


Figure 4.2: All physical quantities defined with the BattINFO namespace. In addition parent classes belonging to EMMO are shown in gray.