

Required booster initial thrust  $\rightarrow 3.1 \times 10^6 \text{ lbf} = 13.8 \text{ MN}$   
 Initial chamber pressure  $= 700 \text{ PSI} = 4.8 \text{ MPa}$   
 mixture ratio  $= 2.0$

$$C_F = 1.75$$

$$C^* = 5912 \text{ ft/sec} = 1801.9776 \text{ m/s}$$

$$\dot{m}_{\text{propellant}} = \frac{P_c A^*}{\sqrt{T_0}} \left[ \gamma / R \left( \frac{2}{\gamma+1} \right)^{(\gamma+1)/(\gamma-1)} \right]^{1/2}$$

assuming the combustion efficiency to be 0.9  
 the  $C^*_{\text{real}} = C^* \cdot 0.9 = 1801.9776 \times (0.9)$   
 $= 1621.779 \text{ m/s}$

$$C_F (\dot{m}_{\text{prop}}) (C^*) = F$$

$$\Rightarrow 1.75 (\dot{m}_{\text{prop}}) (1621.779) = 13.8 \times 10^6$$

$$\dot{m}_{\text{prop}} = 48.6472 \text{ kg/s}$$

In the chamber we can assume the mixture to be following ideal gas law.

for calculating thrust per unit

$$\text{MR} = 2$$



$$\therefore \text{MR}_{\text{stoichiometric}} = 16/1$$

$$\therefore \text{EAR} = \frac{2}{16} = 1/8$$



$$P_t = P_c \left( 1 + \frac{(r-1)}{2} ER \right)^{2/r-1}$$

assuming  $r=1.4$

$$P_t = 4.8 \text{ MPa} \left[ 1 + \frac{0.4}{16} \right]$$

$$= 5.23 \text{ MPa}$$

$$\therefore A_t = \frac{13.8 \times 10^6 \text{ N}}{5.23 \times 10^6 \text{ Pa}} = 2.63 \text{ m}^2$$

$$\frac{\pi d^2}{4} = 2.63$$

$$d = 1.83 \text{ m}$$

$$\bullet \bullet \dot{Q}_{\text{Ox}} + \dot{Q}_F = \dot{Q}_{\text{prop}}$$

$$\dot{Q}_{\text{Ox}} / \dot{Q}_F = 2$$

$$\Rightarrow \dot{Q}_{\text{Ox}} + \frac{\dot{Q}_{\text{Ox}}}{2} = \dot{Q}_{\text{prop}}$$

$$\Rightarrow \frac{3 \dot{Q}_{\text{Ox}}}{2} = \dot{Q}_{\text{prop}}$$

$$\Rightarrow \dot{Q}_{\text{Ox}} = \frac{2(48.64)}{3} = 32.42 \text{ kg/s}$$