Abbreviated Table of Isotopes and Atomic Masses

Z	Element	Symbol	Average atomic mass (u)	Mass number (* indicates radioactive) A	Atomic mass (u)	Percent abundance	Half-life (if radioactive) T _{1/2}
0	(Neutron)	n		1*	1.008 665		10.4 m
1	Hydrogen	Н	1.0079	1	1.007 825	99.985	
	Deuterium	D		2	2.014 102	0.015	
	Tritium	Т		3*	3.016 049		12.33 y
2	Helium	He	4.002 60	3	3.016 029	0.000 14	
				4	4.002 602	99.999 86	
				6*	6.018 886		0.81 s
3	Lithium	Li	6.941	6	6.015 121	7.5	
				7	7.016 003	92.5	
4	Beryllium	Be	9.0122	7*	7.016 928		53.3 d
				8*	8.005 305		$6.7 \times 10^{-17} \text{ s}$
				9	9.012 174	100	1.5106
				10*	10.013 584		$1.5 \times 10^6 \mathrm{y}$
5	Boron	В	10.81	10	10.012 936	19.9	
				11	11.009 305	80.1	
6	Carbon	С	12.011	10*	10.016 854		19.3 s
				11*	11.011 433	00.0	20.4 m
				12 13	12.000 000 13.003 355	98.9 1.10	
				13 14*	14. 003 242	1.10	5715 y
	NT*	N	14.0067				
/	Nitrogen	N	14.0067	13* 14	13.005 738 14.003 074	99.63	996 m
				15	15.000 108	0.37	
				16*	16.006 100	0.57	7.13 s
8	Oxygen	0	15.9994	15*	15.003 065		122 s
Ü	On gen	Ü	15.7771	16	15.994 915	99.761	122 0
				17	16.999 132	0.039	
				18	17.999 160	0.200	
				19*	19.003 577		26.9 s
9	Fluorine	F	18.998 40	18*	18.000 937		109.8 m
				19	18.998 404	100	
				20*	19.999 982		11.0 s
10	Neon	Ne	20.180	19*	19.001 880		17.2 s
				20	19.992 435	90.48	
				21	20.993 841	0.27	
				22	21.991 383	9.25	
11	Sodium	Na	22.989 87	22*	21.994 434		2.61 y
				23	22.989 767	100	14.06 h
				24*	23.990 961		14.96 h
12	Magnesium	Mg	24.305	23*	22.994 124	=0.00	11.3 s
				24	23.985 042	78.99	
				25 26	24.985 838	10.00	
	.1	, ,	24.00		25.982 594	11.01	= 1 105
13	Aluminum	Al	26.981 54	26*	25.986 892	100	$7.4 \times 10^5 \text{ y}$
				27	26.981 534	100	