

**Table 6 Rules for Rounding in Calculations**

What to do	When to do it	Examples
round down	• whenever the digit following the last significant figure is a 0, 1, 2, 3, or 4	30.24 becomes 30.2
	• if the last significant figure is an even number and the next digit is a 5, with no other nonzero digits	32.25 becomes 32.2 32.65000 becomes 32.6
round up	• whenever the digit following the last significant figure is a 6, 7, 8, or 9	22.49 becomes 22.5
	• if the digit following the last significant figure is a 5 followed by a nonzero digit	54.7511 becomes 54.8
	• if the last significant figure is an odd number and the next digit is a 5, with no other nonzero digits	54.75 becomes 54.8 79.3500 becomes 79.4

## SECTION REVIEW

- Which SI units would you use for the following measurements?
  - the length of a swimming pool
  - the mass of the water in the pool
  - the time it takes a swimmer to swim a lap
- Express the following measurements as indicated.
  - 6.20 mg in kilograms
  - $3 \times 10^{-9}$  s in milliseconds
  - 88.0 km in meters
- Perform these calculations, following the rules for significant figures.
  - $26 \times 0.02584 = ?$
  - $15.3 \div 1.1 = ?$
  - $782.45 - 3.5328 = ?$
  - $63.258 + 734.2 = ?$
- Critical Thinking** The following students measure the density of a piece of lead three times. The density of lead is actually  $11.34 \text{ g/cm}^3$ . Considering all of the results, which person's results were accurate? Which were precise? Were any both accurate and precise?
  - Rachel:  $11.32 \text{ g/cm}^3$ ,  $11.35 \text{ g/cm}^3$ ,  $11.33 \text{ g/cm}^3$
  - Daniel:  $11.43 \text{ g/cm}^3$ ,  $11.44 \text{ g/cm}^3$ ,  $11.42 \text{ g/cm}^3$
  - Leah:  $11.55 \text{ g/cm}^3$ ,  $11.34 \text{ g/cm}^3$ ,  $11.04 \text{ g/cm}^3$