

SAMPLE PROBLEM A

Work Done on or by a Gas

PROBLEM

An engine cylinder has a cross-sectional area of 0.010 m^2 . How much work can be done by a gas in the cylinder if the gas exerts a constant pressure of $7.5 \times 10^5 \text{ Pa}$ on the piston and moves the piston a distance of 0.040 m ?

SOLUTION

Given: $A = 0.010 \text{ m}^2$ $d = 0.040 \text{ m}$
 $P = 7.5 \times 10^5 \text{ Pa} = 7.5 \times 10^5 \text{ N/m}^2$

Unknown: $W = ?$

Use the equation for the work done on or by a gas.

$$W = P\Delta V = PA d$$

$$W = (7.5 \times 10^5 \text{ N/m}^2)(0.010 \text{ m}^2)(0.040 \text{ m})$$

$$W = 3.0 \times 10^2 \text{ J}$$



Because W is positive, we can conclude that the work is done by the gas rather than on the gas.

PRACTICE A

Work Done on or by a Gas

1. Gas in a container is at a pressure of $1.6 \times 10^5 \text{ Pa}$ and a volume of 4.0 m^3 . What is the work done by the gas if
 - a. it expands at constant pressure to twice its initial volume?
 - b. it is compressed at constant pressure to one-quarter of its initial volume?
2. A gas is enclosed in a container fitted with a piston. The applied pressure is maintained at 599.5 kPa as the piston moves inward, which changes the volume of the gas from $5.317 \times 10^{-4} \text{ m}^3$ to $2.523 \times 10^{-4} \text{ m}^3$. How much work is done? Is the work done *on* or *by* the gas? Explain your answer.
3. A balloon is inflated with helium at a constant pressure that is $4.3 \times 10^5 \text{ Pa}$ in excess of atmospheric pressure. If the balloon inflates from a volume of $1.8 \times 10^{-4} \text{ m}^3$ to $9.5 \times 10^{-4} \text{ m}^3$, how much work is done on the surrounding air by the helium-filled balloon during this expansion?
4. Steam moves into the cylinder of a steam engine at a constant pressure and does 0.84 J of work on a piston. The diameter of the piston is 1.6 cm , and the piston travels 2.1 cm . What is the pressure of the steam?