b. OH
$$\xrightarrow{85\% \text{ H}_3\text{PO}_4}$$
 $+ \text{ H}_2\text{O}$

c. O
$$CH_3-C-OH + CH_3-OH \longrightarrow O$$
 $CH_3-C-O-CH_3 + H_2C$

38. Which of the following reactions is a condensation reaction?

$$CH_3C \equiv CH + HBr \xrightarrow{ether} CH_3 - C = CH_2$$
b.
$$Br$$

b.
$$+ Br_2 \xrightarrow{CCl_4} Br$$

c.
$$CH_3-CH_2-OH + CH_3-CH_2-OH \xrightarrow{H_2SO_4} CH_3-CH_2-O-CH_2-CH_3 + H_2O$$

39. Which of the following reactions is an elimination reaction?

tion reaction?
a.
$$CH_2$$
= CH - CH_2 - CH_3 + Cl_2 —

 Cl - CH_2 - CH - CH_2 - CH_3
 Cl

b. OH
$$CH_3$$
 $\xrightarrow{-CH}$ CH_3 $\xrightarrow{-H_3O^+}$ CH_3 $-CH$ $=$ CH_2 $+$ H_2O

c.
$$CH_3CH_3 + Cl_2 \xrightarrow{light \text{ or} \atop \text{heat}} CH_3CH_2Cl + HCl$$

MIXED REVIEW

40. Classify each of the following reactions as an elimination reaction or a condensation reaction:

a. Br
$$CH_3-C=CH_2 + NaNH_2 \longrightarrow CH_3C=CH + NaBr + NH_3$$

b. OH
$$CH_3-CH_2-CH-CH_3 \xrightarrow{85\% \ H_3PO_4} CH_3-CH=CH-CH_3 \ + \ H_2O$$

c.
$$CH_{3}-CH_{2}-OH + CH_{3}-C-OH \longrightarrow O$$

$$CH_{3}-CH_{2}-O-C-CH_{3} + H_{2}O$$
d.
$$O O$$

$$CH_{3}-CH_{2}-O-C-CH_{3} + H_{2}O$$

$$CH_{3}-C-CH_{3} \longrightarrow O$$

$$CH_{3}-C-CH_{3} \longrightarrow O$$

$$CH_{3}-C-CH_{3} \longrightarrow O$$

$$CH_{3}-C-CH_{3} \longrightarrow O$$

41. Classify each of the following reactions as a substitution reaction or an addition reaction:

a.
$$\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-OH} + \text{HCl} \xrightarrow{\frac{\text{ZnCl}_2}{\text{heat}}}$$
 $\text{CH}_3\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-CH}_2\text{-Cl} + \text{H}_2\text{O}$

b.
$$CH_2=CH_2 + HBr \xrightarrow{\text{ether}} CH_3-CH_2-Br$$

c.
$$CH \equiv CH + HCl \xrightarrow{HgCl_2} CH_2 = CH - Cl$$

$$\begin{array}{c|c} \text{d.} & & \text{CH}_3 \\ & + & \text{HBr} & \xrightarrow{\text{ether}} & & \\ \end{array}$$

42. Identify each of the following reactions as an addition, substitution, elimination, or condensation reaction:

a. O OH
$$CH_3-C-OH + CH_3-CH-CH_3 \longrightarrow$$
 O CH_3 $CH_3-C-O-CH-CH_3 + H_2O$ b. $CH_2=CH-CH_3 + Cl_2 \longrightarrow$

$$\begin{array}{c} : \operatorname{CH}_2 = \operatorname{CH} - \operatorname{CH}_3 + \operatorname{Cl}_2 \longrightarrow \\ & \operatorname{Cl} - \operatorname{CH}_2 - \operatorname{CH} - \operatorname{CH}_3 \\ & \operatorname{Cl} \end{array}$$

c.
$$+ Cl_2 \xrightarrow{\text{light or heat}} Cl + HCl$$

d.
$$CH_3$$
 OH CH_3 H_3O^+ , THF $+$ H_2O