

- 212.** A 1250 kg car on a curve with a 35.0 m radius has a centripetal force from friction and gravity of 8.07×10^3 N. What is the car's tangential speed?
- 213.** Two wrestlers, 2.50×10^{-2} m apart, exert a 2.77×10^{-3} N gravitational force on each other. One has a mass of 157 kg. What is the other's mass?
- 214.** A 1.81×10^5 kg blue whale is 1.5 m from a 2.04×10^4 kg whale shark. What is the gravitational force between them?
- 215.** Triton's orbit around Neptune has a radius of 3.56×10^5 km. Neptune's mass is 1.03×10^{26} kg. What is Triton's period?
- 216.** Find the tangential speed in problem 215.
- 217.** A moon orbits a 1.0×10^{26} kg planet in 365 days. What is the radius of the moon's orbit?
- 218.** What force is required to produce a $1.4 \text{ N}\cdot\text{m}$ torque when applied to a door at a 60.0° angle and 0.40 m from the hinge?
- 219.** What is the maximum torque that the force in problem 218 can exert?
- 220.** A worker hanging 65.0° from the vane of a windmill exerts an $8.25 \times 10^3 \text{ N}\cdot\text{m}$ torque. If the worker weighs 587 N, what is the vane's length?

Chapter 8 Fluid Mechanics

- 221.** A cube of volume 1.00 m^3 floats in gasoline, which has a density of 675 kg/m^3 . How large a buoyant force acts on the cube?
- 222.** A cube 10.0 cm on each side has a density of $2.053 \times 10^4 \text{ kg/m}^3$. Its apparent weight in fresh water is 192 N. Find the buoyant force.
- 223.** A 1.47×10^6 kg steel hull has a base that is $2.50 \times 10^3 \text{ m}^2$ in area. If it is placed in sea water ($\rho = 1.025 \times 10^3 \text{ kg/m}^3$), how deep does the hull sink?
- 224.** What size force will open a door of area 1.54 m^2 if the net pressure on the door is 1.013×10^3 Pa?
- 225.** Gas at a pressure of 1.50×10^6 Pa exerts a force of 1.22×10^4 N on the upper surface of a piston. What is the piston's upper surface area?
- 226.** In a barometer, the mercury column's weight equals the force from air pressure on the mercury's surface. Mercury's density is $13.6 \times 10^3 \text{ kg/m}^3$. What is the air's pressure if the column is 760 mm high?
- 227.** A cube of osmium with a volume of 166 cm^3 is placed in fresh water. The cube's apparent weight is 35.0 N. What is the density of osmium?

- 228.** A block of ebony with a volume of $2.5 \times 10^{-3} \text{ m}^3$ is placed in fresh water. If the apparent weight of the block is 7.4 N, what is the density of ebony?
- 229.** One piston of a hydraulic lift holds 1.40×10^3 kg. The other holds an ice block ($\rho = 917 \text{ kg/m}^3$) that is 0.076 m thick. Find the first piston's area.
- 230.** A hydraulic-lift piston raises a 4.45×10^4 N weight by 448 m. How large is the force on the other piston if it is pushed 8.00 m downward?
- 231.** A platinum flute with a density of 21.5 g/cm^3 is submerged in fresh water. If its apparent weight is 40.2 N, what is the flute's mass?

Chapter 9 Heat

- 232.** Surface temperature on Mercury ranges from 463 K during the day to 93 K at night. Express this temperature range in degrees Celsius.
- 233.** Solve problem 233 for degrees Fahrenheit.
- 234.** The temperature in Fort Assiniboine, Montana, went from -5°F to $+37^\circ\text{F}$ on January 19, 1892. Calculate this change in temperature in kelvins.
- 235.** An acorn falls 9.5 m, absorbing 0.85 of its initial potential energy. If 1200 J/kg will raise the acorn's temperature 1.0°C , what is its temperature increase?
- 236.** A bicyclist on level ground brakes from 13.4 m/s to 0 m/s. What is the cyclist's and bicycle's mass if the increase in internal energy is 5836 J?
- 237.** A 61.4 kg roller skater on level ground brakes from 20.5 m/s to 0 m/s. What is the total change in the internal energy of the system?
- 238.** A 0.225 kg tin can ($c_p = 2.2 \times 10^3 \text{ J/kg}\cdot^\circ\text{C}$) is cooled in water, to which it transfers 3.9×10^4 J of energy. By how much does the can's temperature change?
- 239.** What mass of bismuth ($c_p = 121 \text{ J/kg}\cdot^\circ\text{C}$) increases temperature by 5.0°C when 25 J are added by heat?
- 240.** Placing a 0.250 kg pot in 1.00 kg of water raises the water's temperature 1.00°C . The pot's temperature drops 17.5°C . Find the pot's specific heat capacity.
- 241.** Lavas at Kilauea in Hawaii have temperatures of 2192°F . Express this quantity in degrees Celsius.
- 242.** The present temperature of the background radiation in the universe is 2.7 K. What is this temperature in degrees Celsius?