

Biochemical Engineering Lab (BT 311) Quiz, Nov 14, 2023

Duration: 120 min

Total Marks: 20

Part A (10 Marks)

Answer ALL questions and each question carries EQUAL marks

- ✓1. Rochelle salt in di-nitro salicylic (DNS) acid method acts as a \_\_\_\_\_
- ✓2. The immobilization of enzyme in calcium alginate beads is mediated by \_\_\_\_\_ technique
3. Enumerate FOUR advantages of immobilizing enzymes for industrial applications.
- ✓4. *Escherichia coli* is capable of growing at a rate such that the culture doubles in biomass every 20 minutes. If 100 ml of nutrient medium is inoculated with about 1  $\mu$ g of organisms and subsequently incubated for 24 h, what quantity of organisms might you expect to be produced?
- ✓5. An absorbance value of 1 unit (at 600 nm) corresponds to \_\_\_\_\_ g/l of biomass concentration based on a thumb's rule.
- ✓6. A certain yeast cell is said to have a doubling time of 90 minutes. What is its specific growth rate in  $\text{hr}^{-1}$ ?
- ✓7. A 1 ml inoculum of same yeast cells (as mentioned in question 6) at an OD of 5 (600 nm) is used to inoculate 1 L of fresh media. How long will it take for the new culture to reach an OD of 1?
- ✓8. A typical example for structured growth model is \_\_\_\_\_
- ✓9. Define the significance of the parameter ' $K_s$ ' in Monod's model.
- ✓10. In Monod's model, specific growth rate profile principally varies with change in substrate concentration in a \_\_\_\_\_ fashion.

Part B (10 Marks)

Answer ALL questions and each question carries EQUAL marks

- ✓11. A reaction mixture consisting of crude Cellulase enzyme with a  $K_m$  of 2.5mM acts on 0.35 M CMC substrate with no product formation. After 60 seconds, the reaction mixture was subjected for product quantification and 45  $\mu\text{M}$  of glucose was observed. Find the  $V_{\max}$  and concentration of product formation after 135 seconds.
12. The absorbance of a DNS solution containing 0.500  $\text{mg/mL}$  glucose was reported as 0.3500 at 440nm.
  - a) Calculate the specific absorptivity, including units, on the assumption that a 1.00 cm cuvette was used.
  - ✓b) What will be the absorbance if the solution is diluted to twice its original volume.

~~CO~~  
 $CO = CO$   
17



- ✓ 13. A 20-l stirred fermenter containing a *Bacillus thuringiensis* culture at 30°C is used for production of microbial insecticide,  $k_{La}$  is determined using the dynamic method. Air flow is shut off for a few minutes and the dissolved-oxygen level drops; the air supply is then re-connected. The following results are obtained.

Time (sec)	Dissolved oxygen tension
5	50
15	66

→ saturation (%)

When steady state is established, the dissolved-oxygen tension is 78 % air saturation.

- (a) Estimate  $k_{La}$ .  
 (b) An error is made determining the steady-state oxygen level which, instead of 78%, is taken as 70%. What is the percentage error in  $k_{La}$  resulting from this error in steady state oxygen level?
14. A simple batch fermentation of an aerobic bacterial strain grown on glucose limited mineral salt medium was performed in a lab scale bioreactor. The experimental results are tabulated as shown below:

Time (h)	Cell Dry	Glucose
0	0.2	9.23
2	0.211	9.21
4	0.305	9.07
8	0.98	8.03
10	1.77	6.8
12	3.2	4.6
14	5.6	0.92
16	6.15	0.077
18	6.2	0

weight (gm/L)

→ concentration (gm/L)

- (a) Calculate the maximum specific growth rate  
 (b) Calculate the saturation constant  
 (c) Calculate the cell mass doubling time

- ✓ 15. An experiment was performed on an exponential phase microbial culture, where the oxygen supply was disconnected and the dissolved oxygen (DO) concentration was allowed to fall to 43.5% saturation. At this point, aeration was resumed and the increase in DO concentration was monitored with respect to time. From the following data of the reoxygenation stage, calculate the quasi-steady state DO concentration and the gas-liquid mass transfer coefficient ( $k_{La}$ ) for the reactor.

Time (s)	% DO Saturation
10	43.5
20	53.5
30	60
40	67.5
50	70.5
60	72
70	73
100	73.5
130	73.5

$$\mu = \frac{\mu_{max} S}{K_s + S}$$