

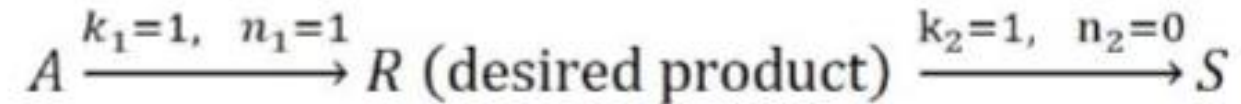
BT209

Bioreaction Engineering

05/04/2023

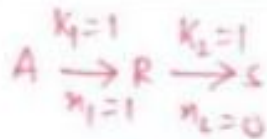
Problem 1

Consider the following reactions ($C_{A0} = 10$, $C_{R0} = 0$, $C_{S0} = 0$) in a mixed flow reactor (CSTR):



- (a) Find the operating condition (τ_m) which maximize the C_R in a mixed flow reactor?
- (b) At what space time C_R will be non-detectable?

solution



CSTR $C_A: \frac{\tau_m}{C_{A0}} = \frac{C_{A0} - C_A}{C_{A0} \times K_1 C_A} \Rightarrow \boxed{\frac{C_A}{C_{A0}} = \frac{1}{1 + K_1 \tau_m}}$

$$\begin{aligned} -r_A &= -\frac{dC_A}{dt} = K_1 C_A \\ r_R &= \frac{dC_R}{dt} = K_1 C_A - K_2 \end{aligned}$$

$$C_R: \frac{\tau_m}{C_{R0}} = \frac{-C_{R0} + C_R}{C_{R0} (K_1 C_A - K_2)}$$

$$\Rightarrow \tau_m K_1 C_A - \tau_m K_2 = C_R$$

$$\Rightarrow \boxed{\frac{C_R}{C_{A0}} = \frac{K_1 \tau_m}{1 + K_1 \tau_m} - \frac{K_2 \tau_m}{C_{A0}}} \text{ for } \tau_m \leq \frac{C_{A0}}{K_2} - \frac{1}{K_1}$$

$$\frac{d(C_R/C_{A0})}{d\tau_m} = 0 \text{ for } C_R^{\text{max}}$$

$$\Rightarrow \frac{(1 + K_1 \tau_m) K_1 - K_1 \tau_m \cdot K_1}{(1 + K_1 \tau_m)^2} - \frac{K_2}{C_{A0}} = 0$$

$$\Rightarrow \frac{K_1}{(1 + K_1 \tau_m)^2} - \frac{K_2}{C_{A0}} = 0, \Rightarrow 1 + K_1 \tau_m = \sqrt{\frac{C_{A0} K_1}{K_2}}$$

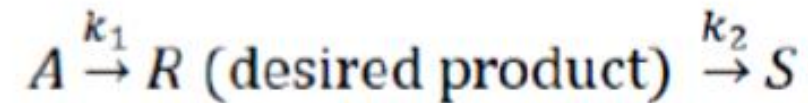
$$\Rightarrow \boxed{\tau_{m, \text{opt}} = 2.162} \text{ for } C_R^{\text{max}}$$

⑥ After first C_R become non-detectable, at $\tau_m = \frac{10}{1} - \frac{1}{1} = 9$ C_R become non-detectable

$$\tau_{m, \text{opt}} = \sqrt{\frac{C_{A0}}{K_1 K_2}} - \frac{1}{K_1} = \sqrt{10} - 1 = 2.16$$

Problem 2

Consider the following elementary reactions ($C_{R0} = 0, C_{S0} = 0$):



Derive the maximum C_R in CSTR as

$$C_{R,max} = \frac{C_{A0}}{\left(1 + \sqrt{\frac{k_2}{k_1}}\right)^2}$$

solution

Example: Irreversible first order



product distribution in CSTR / PFR ??

CSTR $C_A, C_R, C_S = ??$

Material balance of A

$$F_{A0} = F_A + (-r_A)V$$

$$vC_{A0} = vC_A + K_1 C_A V$$

$$\Rightarrow \frac{C_A}{C_{A0}} = \frac{1}{1 + K_1 \tau_m}$$

Material balance of R

$$vC_{R0} = vC_R + (-r_R)V$$

$$0 = vC_R + (-K_1 C_A + K_2 C_R)V$$

$$vC_R + K_2 C_R V = K_1 V \frac{C_{A0}}{1 + K_1 \tau_m} \Rightarrow C_R (v + K_2 V) = \frac{K_1 V C_{A0}}{1 + K_1 \tau_m}$$

$$\frac{C_R}{C_{A0}} = \frac{K_1 V}{(1 + K_1 \tau_m)(v + K_2 V)} = \frac{K_1 \tau_m}{(1 + K_1 \tau_m)(1 + K_2 \tau_m)} = \frac{K_1 \tau_m}{(1 + K_1 \tau_m)(1 + K_2 \tau_m)}$$

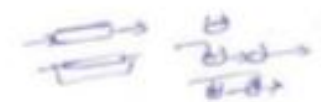
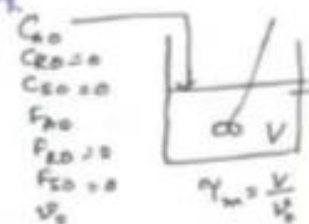
$$\Rightarrow \frac{C_R}{C_{A0}} = \frac{K_1 \tau_m}{(1 + K_1 \tau_m)(1 + K_2 \tau_m)}$$

$$C_S = C_{A0} - C_A - C_R$$

$$\frac{C_S}{C_{A0}} = 1 - \frac{1}{1 + K_1 \tau_m} - \frac{K_1 \tau_m}{(1 + K_1 \tau_m)(1 + K_2 \tau_m)}$$

$$\Rightarrow \frac{C_S}{C_{A0}} = \frac{K_1 K_2 \tau_m^2}{(1 + K_1 \tau_m)(1 + K_2 \tau_m)}$$

When you have first order reaction, you can use the same equations for CSTR and PFR.



From kinetics

Alternatively

$$\tau_m = \frac{X_R}{-r_A} = \frac{C_{A0} X_R}{C_{A0} (-r_A)}$$

$$\tau_m = \frac{X_R}{-r_A} = \frac{C_R - C_{R0}}{C_{A0} (K_1 C_A - K_2 C_R)}$$

KINETICS

$$r_A = -K_1 C_A$$

$$r_R = K_1 C_A - K_2 C_R$$

$$r_S = K_2 C_R$$

$$r_A + r_R + r_S = \frac{dC_A}{dt} + \frac{dC_R}{dt} + \frac{dC_S}{dt}$$

$$C_A + C_R + C_S = C_{A0}$$

$$\Rightarrow C_{A0} = C_{A0}$$

$C_{R, \max} = ??$ and what γ_m
 $\frac{d}{d\gamma_m}(C_R) = 0 \Rightarrow C_{R, \max}$ or $\gamma_{m, \text{opt}}$ will get

$$C_R = C_{A0} \frac{K_1 \gamma_m}{(1 + K_1 \gamma_m)(1 + K_2 \gamma_m)}$$

$$d\left(\frac{y}{x}\right) = \frac{x dy - y dx}{x^2}$$

$$\frac{dC_R}{d\gamma_m} = C_{A0} \left[\frac{(1 + K_1 \gamma_m)(1 + K_2 \gamma_m) K_1 - K_1 \gamma_m \{K_1(1 + K_2 \gamma_m) + K_2(1 + K_1 \gamma_m)\}}{(1 + K_1 \gamma_m)^2 (1 + K_2 \gamma_m)^2} \right]$$

$$\Rightarrow (1 + K_1 \gamma_m)(1 + K_2 \gamma_m) = \gamma_m K_1(1 + K_2 \gamma_m) + K_2 \gamma_m(1 + K_1 \gamma_m)$$

$$\Rightarrow 1 + K_1 \gamma_m + K_2 \gamma_m + K_1 K_2 \gamma_m^2 = K_1 \gamma_m + K_1 K_2 \gamma_m^2 + K_2 \gamma_m + K_1 K_2 \gamma_m^2$$

$$\Rightarrow 1 = K_1 K_2 \gamma_m^2$$

$$\Rightarrow \gamma_m = \frac{1}{\sqrt{K_1 K_2}}$$

$$\therefore \gamma_{m, \text{opt}} = \frac{1}{\sqrt{K_1 K_2}}$$

1st order
for a series rxn
in CSTR

$$\therefore C_{R, \max} = C_{A0} \frac{K_1 \gamma_{m, \text{opt}}}{(1 + K_1 \gamma_{m, \text{opt}})(1 + K_2 \gamma_{m, \text{opt}})}$$

$$\frac{C_{R, \max}}{C_{A0}} = \frac{1}{(1 + K_1 \gamma_{m, \text{opt}}) \left(\frac{1}{K_1 \gamma_{m, \text{opt}}} + \frac{K_2}{K_1} \right)}$$

$$= \frac{1}{\left(1 + K_1 \cdot \frac{1}{\sqrt{K_1 K_2}} \right) \left(\frac{\sqrt{K_1 K_2}}{K_1} + \frac{K_2}{K_1} \right)}$$

$$= \frac{1}{\left(1 + \sqrt{\frac{K_1}{K_2}} \right) \left(\sqrt{\frac{K_2}{K_1}} + \frac{K_2}{K_1} \right)} = \frac{1}{\sqrt{\frac{K_2}{K_1}} + 1 + \frac{K_2}{K_1} + \sqrt{\frac{K_2}{K_1}}}$$

$$= \frac{1}{\frac{K_2}{K_1} + 2\sqrt{\frac{K_2}{K_1}} + 1}$$

$$\therefore \boxed{\frac{C_{R, \max}}{C_{A0}} = \frac{1}{\left(\sqrt{\frac{K_2}{K_1}} + 1 \right)^2}}$$