



**IIT Guwahati**

**Lecture 19**

**Course BT 631**

# **Protein Structure, function and Crystallography**

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# Enzyme

The term enzyme was coined by F.W. Kuhne in 1878.



F.W. Kuhne

The name *enzyme* (In Greek, *en* = in; *zyme* = yeast) means “in yeast”.

This was referred to denote the production of ethyl alcohol and  $\text{CO}_2$  that involved “zymase” present in yeast.

This reaction is most popularly known as *alcoholic fermentation by yeast*.

Enzymes were then designated as Biological catalysts.

# Enzyme

**Catalyst:** A substance that increases the rate of a chemical reaction without itself undergoing any permanent chemical change.

**Substrate:** A substance or material on which an enzyme acts.

## Enzymes (Biological Catalysts)

### 1. Shared properties with chemical catalysts

- a. Enzymes are neither consumed nor produced during the course of reaction.
- b. Enzyme do not cause reaction to take place, but they greatly enhance the rate of reaction that would proceed much slower in their absence.
- c. They alter the rate but not the equilibrium constant of reaction that they catalyze.

# Enzyme

## 2. Differences between enzyme and chemical catalysts

- i. Enzymes are proteins or RNA.
- ii. Enzymes are highly specific and produce only the expected products from the given reactants or substrate (i.e. there are no side reactions).
- iii. Enzymes may show a high specificity towards one substrate or exhibit a broad specificity, using more than one substrate.
- iv. Enzymes usually function within a moderate pH and temperature range.

# Enzyme

Enzymes execute two basic functions in a biological system.

They are i) Catalytic and ii) Regulation.

## Catalytic efficiency

- Most of the enzyme-catalyzed reaction are highly efficient, proceeding from  $10^3$  to  $10^8$  times faster than un-catalyzed reactions.
- Typically, each enzyme molecule is capable of transforming 100 to 1000 substrate molecule into product each second.
- The number of molecules of substrate converted to product per enzyme molecule per second is called the **Turnover number**.

## Regulation

Enzyme activity can be regulated – that is, the enzyme can be activated or inhibited so that the rate of product formation responds to the needs of the cell.

# Enzyme Classification

## Classification of enzymes

New enzymes are classified by the Nomenclature Committee of **IUBMB**.  
(**International Union of Biochemistry and Molecular Biology**)

Enzyme nomenclature classifies the catalytic activities, not the protein structure.

The **Enzyme Commission Number (EC Number)** is a numerical classification scheme for enzymes, based on the chemical reactions they catalyze.

The chemical reaction catalyzed is a specific property that distinguishes one enzyme from the other.

EC numbers specify the enzyme-catalyzed reactions.

# Enzyme System of Classification

All Enzymes are placed in **6 classes with numerous sub-classes and sub-sub-classes**, of which only the most important are given below in the Table. *The sub-class and sub-sub-class normally define the nature of the reaction more precisely*, or specify a particular set of donor or acceptors.

Class	Reaction type	Style of systemic name (example)	Principle sub-classes
1. Oxido-reductase	Oxidation/ Reduction	Donor: acceptor oxidoreductase (EC1.1.1.1, alcohol: NAD <sup>+</sup> Oxido-reductase) <b>EC1.1.1.27 Lactate dehydrogenase</b>	1.1 acting on CH-OH group of donors. 1.2 acting on aldehyde or oxo group of donors 1.3 Acting on CH-CH group of donors 1.4 Acting on CH-NH <sub>2</sub> group of donor <b>Reduction of pyruvate to lactate with formation of NAD<sup>+</sup></b>
2. Transferase	Group Transfer Reaction	Donor: acceptor group transferase (EC 2.1.3.2, carbamoyl phosphate: <b>L- aspartate carbamoyltransferase</b>	<b>2.1 Transferring one-carbon groups</b> 2.3 Acyl transferases 2.4 Glycosyltransferases 2.6.1 Transaminases 2.7 Transferring Phosphate groups <b>2.7.1 -OH group as acceptor</b>
3. Hydrolase	Hydrolysis reaction	Substrate Hydrolysis (E.C. 3.1.1.7, acetylcholine acetylhydrolase)	3.1 Esterases acting on ester bonds. 3.2 Glycosidases 3.4 Peptidases

EC 2....-

**2 is class for Transferases.**

EC 2.7.-.-

**7 is sub-class for Transferring phosphorous-containing groups.**

EC 2.7.1.-

**1 is sub-sub-class for Phosphotransferases with an alcohol (OH) group as acceptor.**

EC 2.7.1.11

**11 is the number assigned to 6-Phosphofructokinase.**

# Enzyme System of Classification

Class	Reaction type	Style of systemic name (example)	Principle sub-classes
4. Lyases	Elimination Reaction	Substrate Group- Lyase EC 4.3.1.5 <b>Phenyl-alanine ammonia lyase</b>	4.1 Carbon-carbon lyases 4.2 Carbon-oxygen lyases 4.3 Carbon-Nitrogen lyases <b>Splits Phenylalanine into Ammonia and trans-cinnamate.</b>
5. Isomerase	Isomerization Reactions	Substrate Reactionase (E.C. 5.3.1.1, D-glyceraldehyde-3-Phosphate <b>ketol-isomerase</b> )	5.1 Racemases and epimerases 5.3 <b>Intra-molecular oxidoreductases</b> 5.4 Intra-molecular transferases (mutases)
6. Ligases	Bond Synthesis Coupled to hydrolysis (e.g. ATP)	X-Y ligase (Product Forming )  EC 6.5.1.1 <b>T4 DNA Ligase</b>	6.1 Forming carbon-oxygen bonds 6.2 Forming carbon-sulfur bonds 6.3 Forming carbon-nitrogen bonds <b>6.5 Phosphodiester bond formation and conversion of ATP into AMP</b>