

INTRO TO CH1 : PHYSICAL QUANTITIES & UNITS

Physical quantity:

Any quantity that can be measured and has units.

Is further divided into :

- 1. Base Quantities: those that can't be simplified further
- 2. Derived Quantities: can be expressed using one or more base quantities.

The seven (7) base quantities: quantity (unit)

1. length (m)
2. time (s)
3. temperature (K)
4. mass (kg)
5. amount of substance (mol)
6. current (A)
7. light intensity (cd) → "candela"

Important derived quantities: quantity (base units)

- | | |
|-------------------------------------|---|
| - Area (m^2) | - Power ($\text{kgm}^2\text{s}^{-3}$) |
| - Density (kgm^{-3}) | - Energy ($\text{kgm}^2\text{s}^{-2}$) |
| - Volume (m^3) | - Charge (As) |
| - Speed (ms^{-1}) | - Voltage ($\text{kgm}^2\text{s}^{-3}\text{A}^{-1}$) |
| - Acceleration (ms^{-2}) | - Resistance ($\text{kgm}^2\text{s}^{-3}\text{A}^{-2}$) |
| - Force (kgms^{-2}) | - Specific Heat Capacity ($\text{m}^2\text{s}^{-2}\text{K}^{-1}$) |

- Pressure ($\text{kgm}^{-1}\text{s}^{-2}$)

- Specific Latent Heat (m^2s^{-2})

- Work Done ($\text{kgm}^2\text{s}^{-2}$)

Example Question:

Given that $F = \frac{Q^2}{4\pi\epsilon r^2}$ and $F = \frac{\gamma I^2}{2\pi d}$
where....

F = force, Q = charge, I = current, r = radius, d = distance,
find the base units of $\frac{1}{\epsilon\gamma}$.

$$\frac{Q^2}{4\pi\epsilon r^2} = \frac{\gamma I^2}{2\pi d}$$

* Remove 4π and 2π since they're dimensionless quantities.

$$\frac{Q^2}{\epsilon r^2} = \frac{\gamma I^2}{d}$$

$$\frac{Q^2 d^2}{I^2 r^2} = \frac{\gamma I^2 r^2}{I^2 r^2}$$

$$\frac{I^2 r^2}{Q^2 d} = \frac{1}{\gamma}$$

* Substitute units in place of quantities

$$\frac{(\text{A})^2 (\text{m})^2}{(\text{As})^2 \text{m}} = \frac{1}{\gamma}$$

$$\frac{\cancel{\text{A}^2} \text{m}^2}{\cancel{\text{A}^2} \cancel{\text{s}^2} \cancel{\text{m}}} = \frac{1}{\gamma}$$

\therefore , base units of $(\gamma)^{-1} = \text{ms}^{-2}$

$$\frac{\text{m}}{\text{s}^2} = \frac{1}{\gamma}$$

Example Question:

In the given equation, find the base units of x

$$x = kr^3(P_1 - P_2) \sqrt{\frac{M}{R \cdot T}} \quad \text{where ...}$$

r = radius, P_1 and P_2 = Pressure, M = Mass per mol, T = temperature,

R = Joules per Kelvin \cdot Mole, k = Dimensionless quantity.

$$x = kr^3(P_1 - P_2) \sqrt{\frac{M}{R \cdot T}}$$

$$x = m^3($$