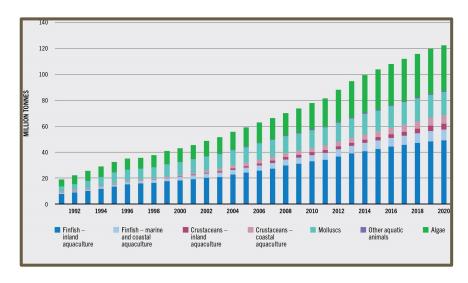


The Effects of CaCO<sub>3</sub> on *Hemigrapsus*oregonensis
Bailey Snodgrass, John Plinka, Sean Berry

## Introduction

- Global harvest in 2018: 6.9 million tonnes of crustaceans (\$36.1 b USD)
   (NOAA, 2021)
- Between 2018 and 2020, aquaculture live weight globally jumped from 115.9 million tons to 122.6 million tons (FAO, 2022)



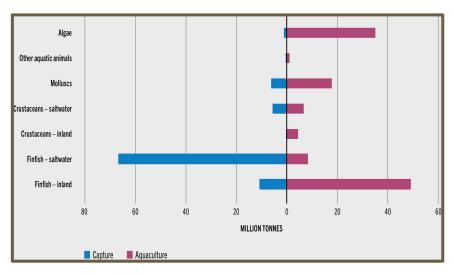


Fig 1). World Aquaculture Production 1991-2020 (Source FAO)

Fig 2). Aquaculture Composition by Functional Species Group 2020 (Source FAO)

# **Background**

- Mineral accumulation of heavy metals have been observed in RAS (Recirculating Aquaculture Systems), and have been shown to have effects on other aquaculture relevant species such as the Nile tilapia (*Oreochromis niloticus*) (Martins et al. 2010)
- Alkalinity increases can result from increased CaCO<sub>3</sub> levels, and alkaline conditions can have adverse effects on internal physiological functions
- Higher [Ca<sup>2+</sup>] levels were found to facilitate molt in *S. paramamosain* (Zhang *et al.*, 2024), and build up within the exoskeleton and the gills was observed

## **Research Question**

Research Question: How will dissolved calcium carbonate affect Hairy Shore Crab (*Hemigrapsus oregonesis*)?

Hypothesis 1 (Calcium Carbonate): Excess dissolved calcium carbonate in the water will bind to both their shells and to their gills internally, limiting their ability to respire

Null Hypothesis 1 (Calcium Carbonate): Excess calcium carbonate will have no effect on the crabs

Hypothesis 2 (Temperature): Increased temperature stress will further exacerbate the the effects of the calcium carbonate and lead to higher mortality rates

Null Hypothesis 2 (Temperature): Temperature will have no effect on the physiological impacts of the calcium carbonate

### **Materials**

- Hairy shore crabs (*H. oregonensis*) n=15;
   control population
- Three 2-L tubs: n=5/ea.
  - Heat controlling device
  - Air stone
  - CaCO<sub>3</sub> lime powder
  - Instant Ocean mix
- Saltwater calcium concentration test kit
- Disposable syringe
- Cayman chemical L-Lactate assay kit



Fig. 3) The hairy shore crab, *Hemigrapsus oregonensis* 

## **Methods**

- CaCO<sub>3</sub> lime weighed out to 1.000g and dissolved into 2-L tubs
  - $\circ$  1 mg CaCO<sub>3</sub> = 0.4 mg Ca<sup>2+</sup>
  - Raised [CaCO<sub>3</sub>] from ~400 mg/L to 600 mg/L
- Placed into one of two tanks: 13° C or 27° C
  - Air stone maintained dissolved O<sub>2</sub>
- Samples and tests conducted at weekly intervals (day 7 and 14)
  - One hemolymph extraction per tub when possible
- Hemolymph processed with L-Lactate assay kit and results recorded

## **Results - Calcium accumulation**

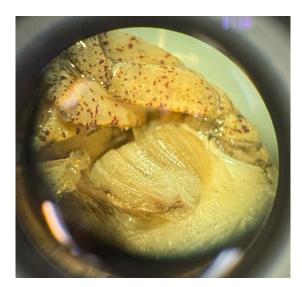


Fig. 4) Typical gill of *H.* oregonensis under baseline regime

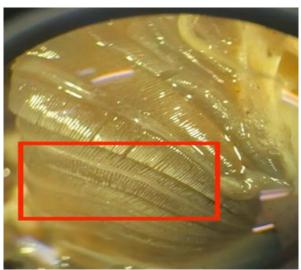


Fig. 5) Accumulation on gills of *H. Oregonensis* in H-Ca<sup>+</sup>



Fig. 6) Accumulation on carapace of *H. Oregonensis* in C-Ca

## **Results - Mortalities**

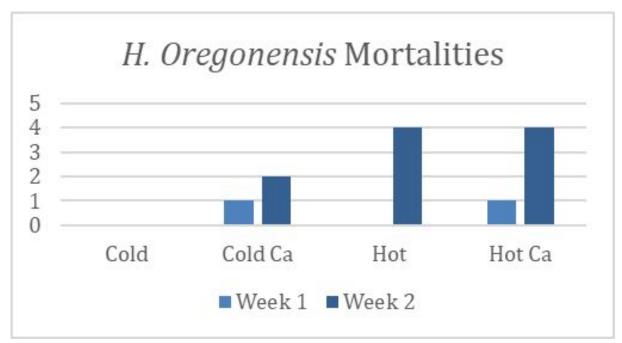


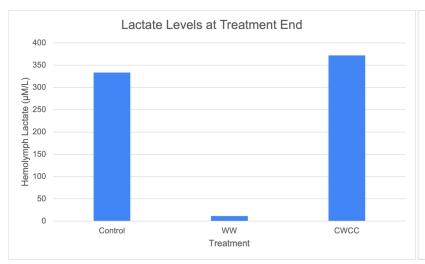
Fig. 7) Mortalities of *H. oregonensis* under each experimental regime

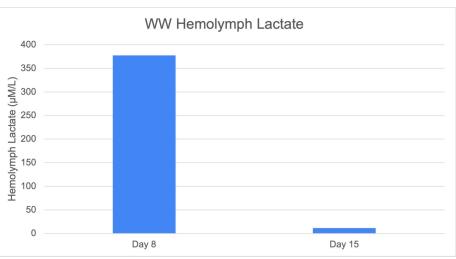
#### What does it mean?

- Foundational study
- Evidence of particulate build up in gills
- Temperature increases may accelerate build up
- No connection between lactate and treatment
- Methodology hindered physiological assessment

### What's next?

- Resolve temperature impacts on particulate accumulation
- Repeat experiment with other popular ph buffers
- Investigate unexpected lactate assay results





#### Who cares?

Aquaculture Managers!

- Most valuable stock for least expensive conditions
- Industry expansion
- More humans = more potential human error

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