

PEM ELECTROLYSER

(PROTON EXCHANGE MEMBRANE)

RSI make PEM electrolyser is a type of electrolytic device used to produce hydrogen gas (H2) by using distilled water.







WORKING PRINCIPLE:

- When Water reacts at the anode to form oxygen and positively charged hydrogen ions (protons).
- ▼ The electrons flow through an external circuit and the hydrogen ions selectively move across the PEM to the cathode.
- At the cathode, hydrogen ions combine with electrons from the external circuit to form hydrogen gas.

Anode Reaction: $2H_2O \rightarrow O_2 + 4H^+ + 4e^-$

Cathode Reaction: 4H⁺ + 4e⁻ → 2H₂



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PEM ELECTROLYSER SPECIFICATIONS	
Membrane	PEM based(Proton exchange membrane)
Material	Stainless steel
Electrolyte	Distilled water
Gasket	PTFE
Flow field	Titanium mesh
Working area	3 x 3 cm & 5 x 5 cm
Ports	H ₂ O Inlet O ₂ Outlet H ₂ Outlet
Current collector	SS 316
Safety Insulated	YES



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MANUFACTURING AND R&D HQ No.63/1, Club House Road, Maruthi Nagar, Hosur- 635109,Tamil Nadu ,India Mail id :sales@rsindia.net, Ph: 8148274261

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sales@rsindia.net

Ph: +91 81482 74261